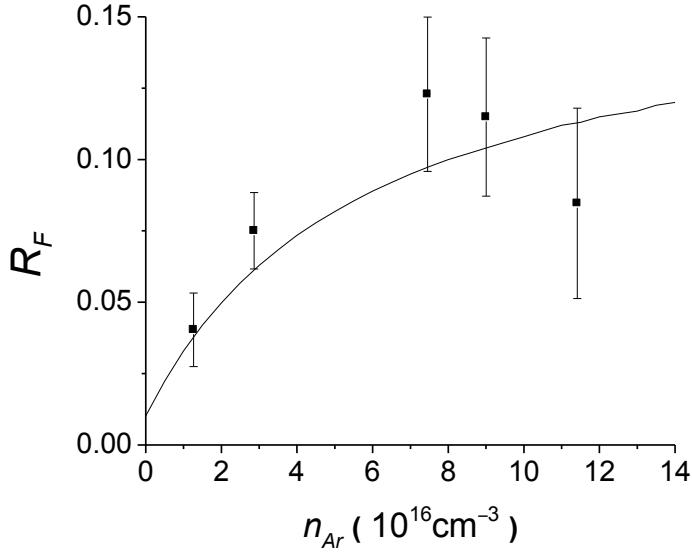
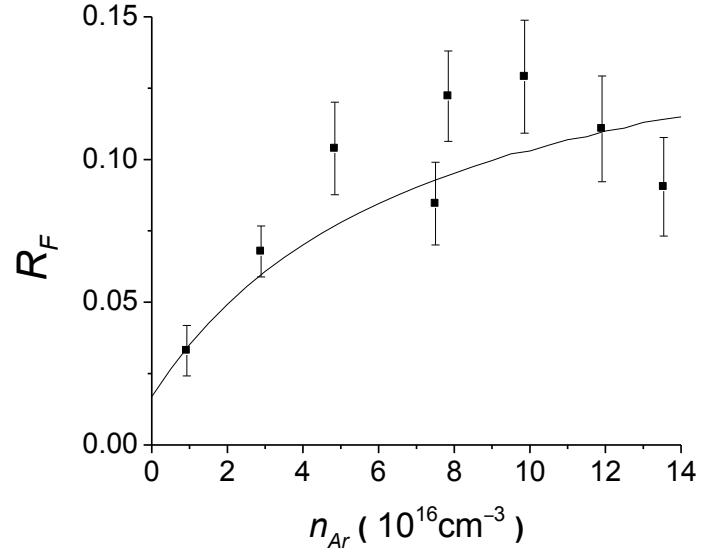


Supplementary Material Fig. 35: (a) Plot of rate coefficients $k_{Ar}^{\Delta J}$ for rotationally inelastic collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon atoms, comparing the results of fits obtained with various fixed values of k_{Cs}^Q/Γ or with k_{Cs}^Q/Γ allowed to vary. (b) Plot of rate coefficients for rotationally inelastic collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with cesium atoms, in units of the cesium quenching rate coefficient k_{Cs}^Q , comparing the results of fits obtained with various fixed values of k_{Cs}^Q/Γ or with k_{Cs}^Q/Γ allowed to vary. k_{Cs}^Q/Γ values are in units of cm^3 . For cases listed as “vary k_{Cs}^Q/Γ within limits” the range of allowed values was $1 \times 10^{-17} \text{ cm}^3 < k_{Cs}^Q/\Gamma < 1 \times 10^{-15} \text{ cm}^3$. The value of Γ was taken to be $2.82 \times 10^7 \text{ s}^{-1}$.

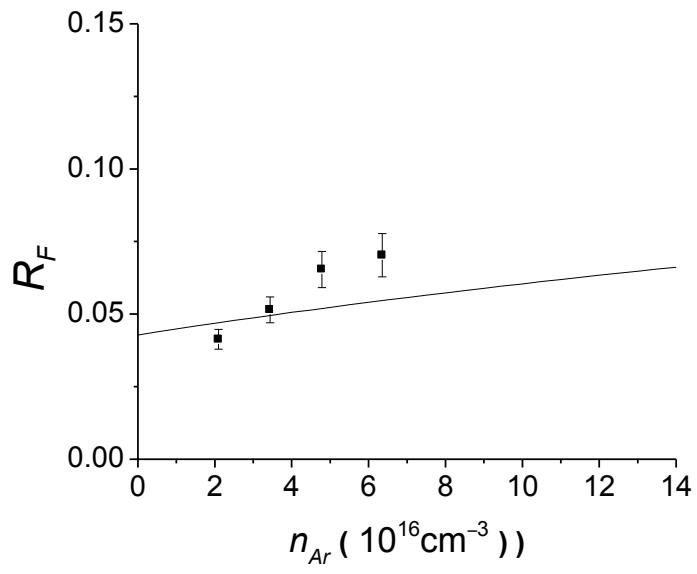
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = +1$



(a) $n_{Cs} = 9.4 \times 10^{14} \text{ cm}^{-3}$



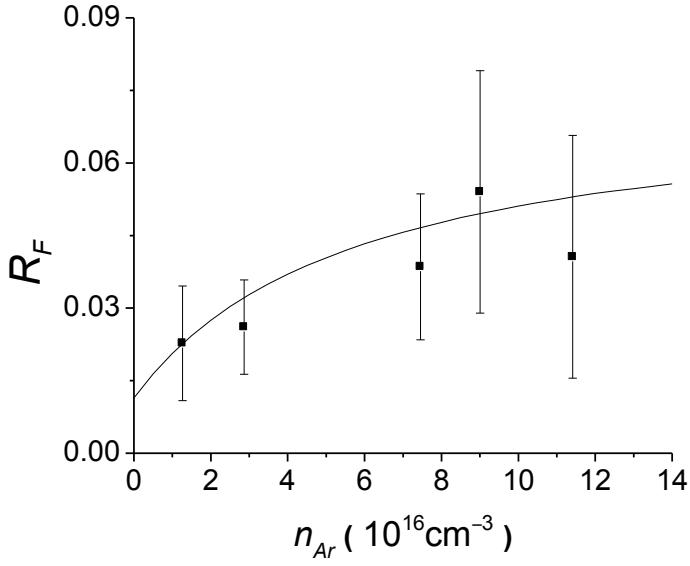
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



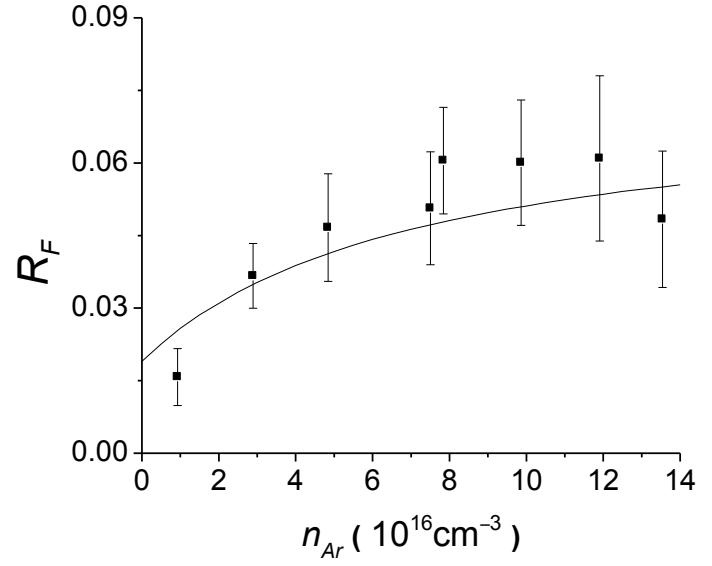
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 36: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = +1$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

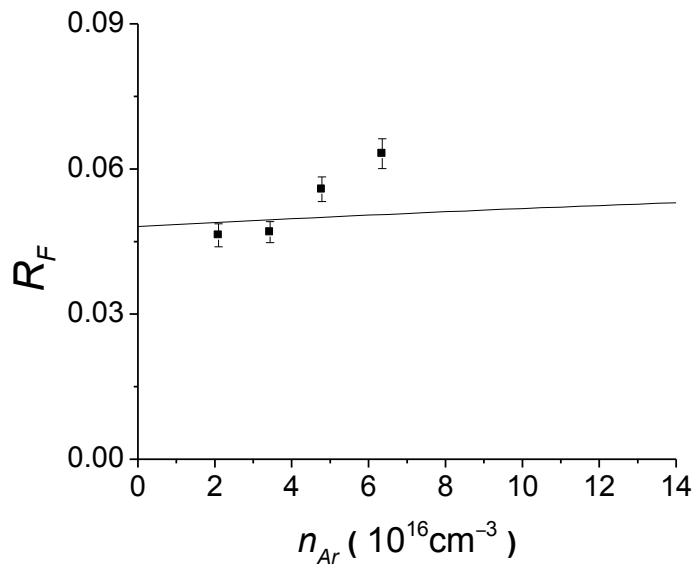
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = +2$



(a) $n_{Cs} = 9.4 \times 10^{14} \text{ cm}^{-3}$



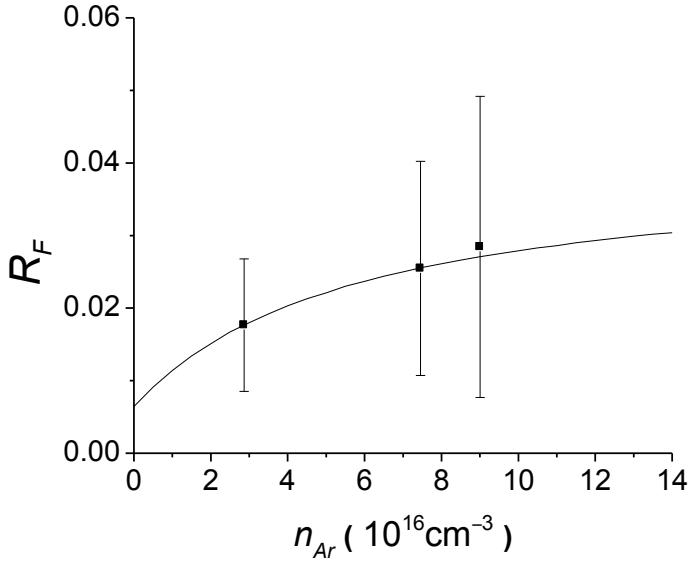
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



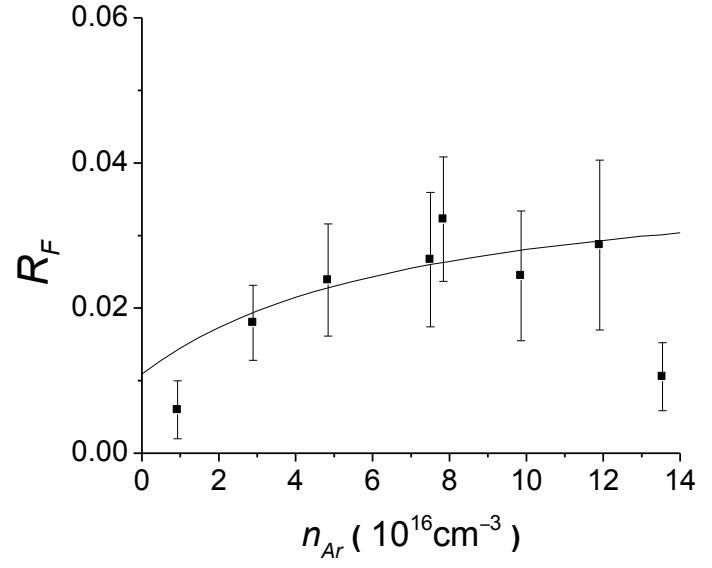
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 37: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = +2$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

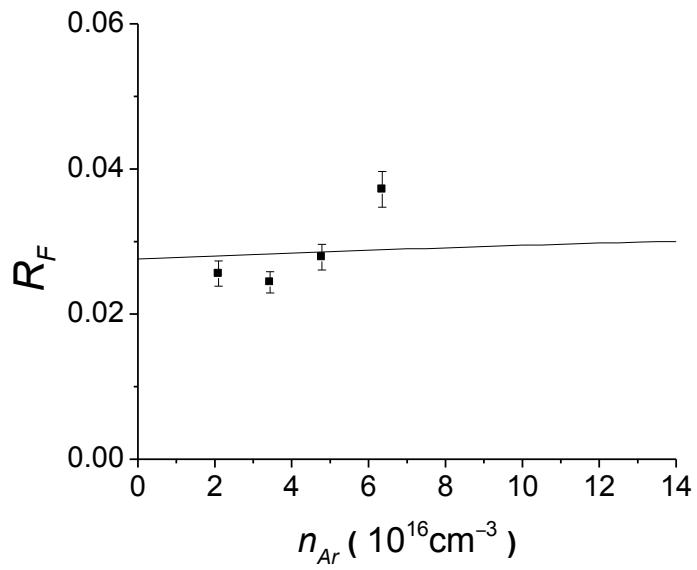
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = +3$



(a) $n_{Cs} = 9.1 \times 10^{14} \text{ cm}^{-3}$



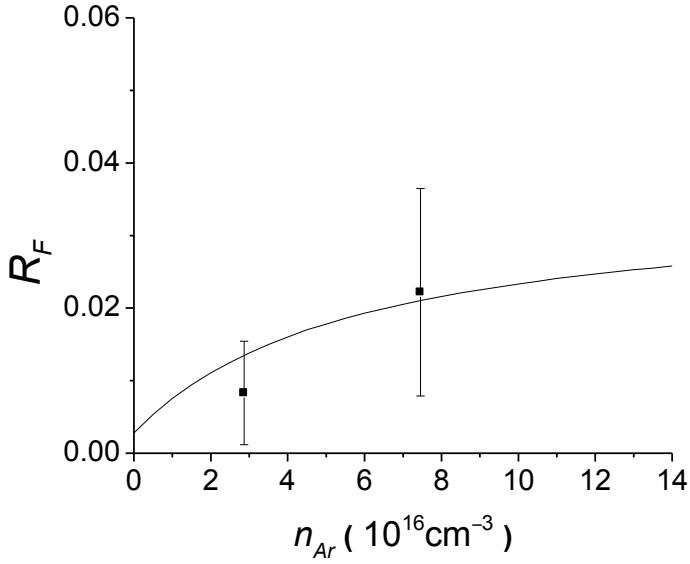
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



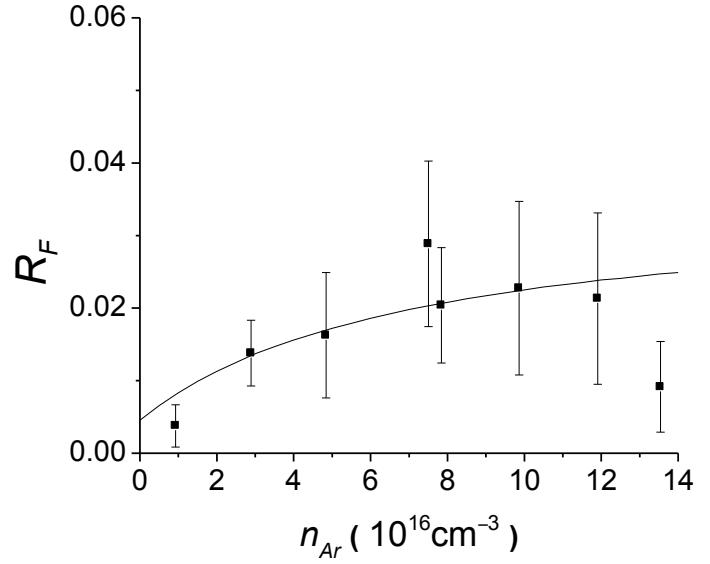
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 38: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = +3$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

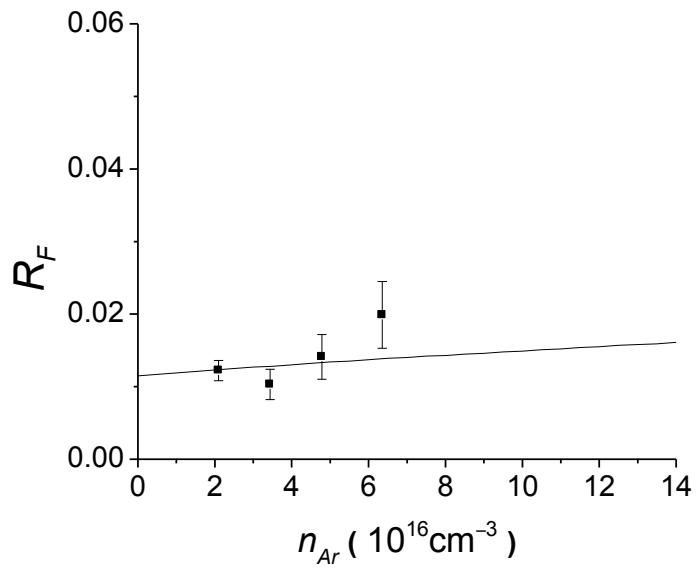
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = +4$



(a) $n_{Cs} = 9.6 \times 10^{14} \text{ cm}^{-3}$



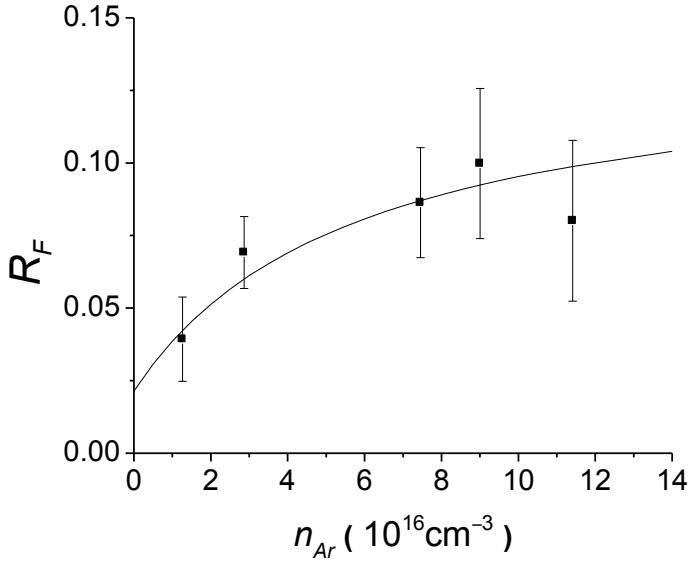
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



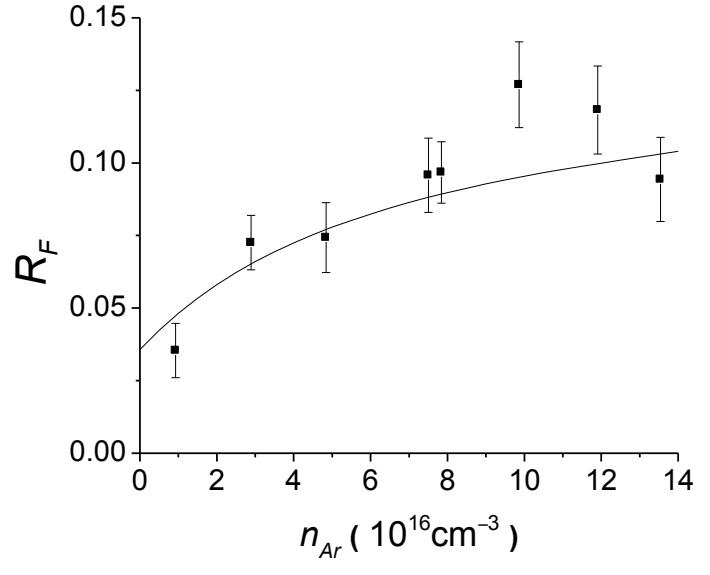
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 39: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = +4$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

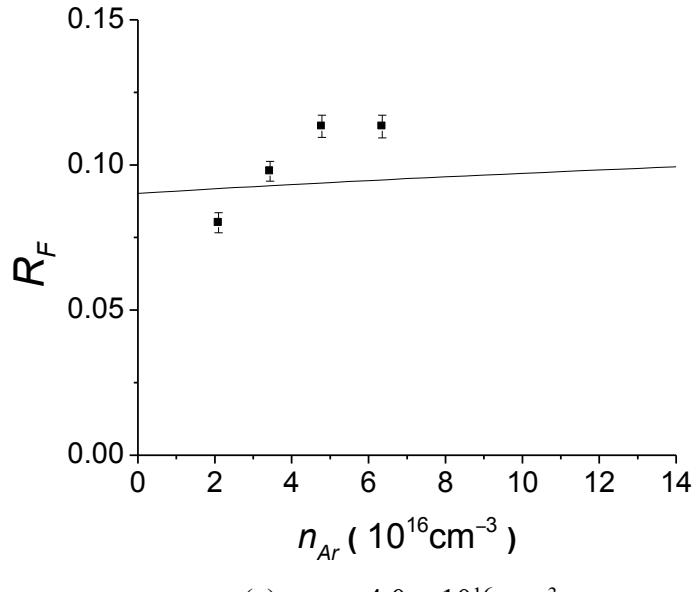
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = -1$



(a) $n_{Cs} = 9.4 \times 10^{14} \text{ cm}^{-3}$



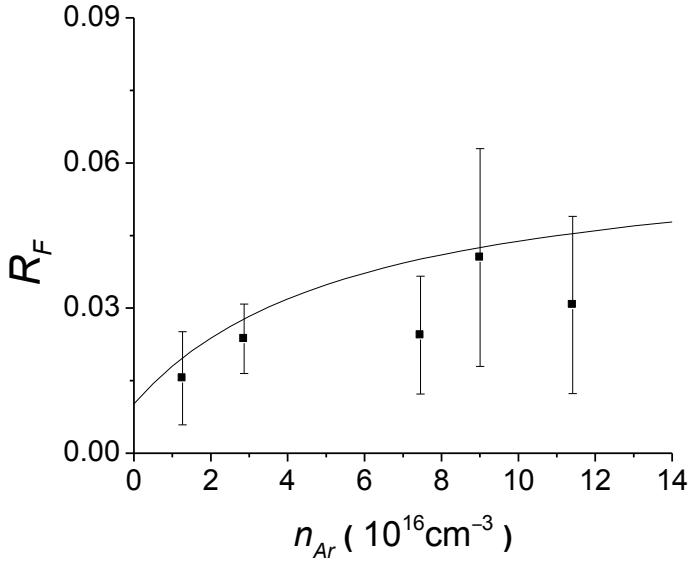
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



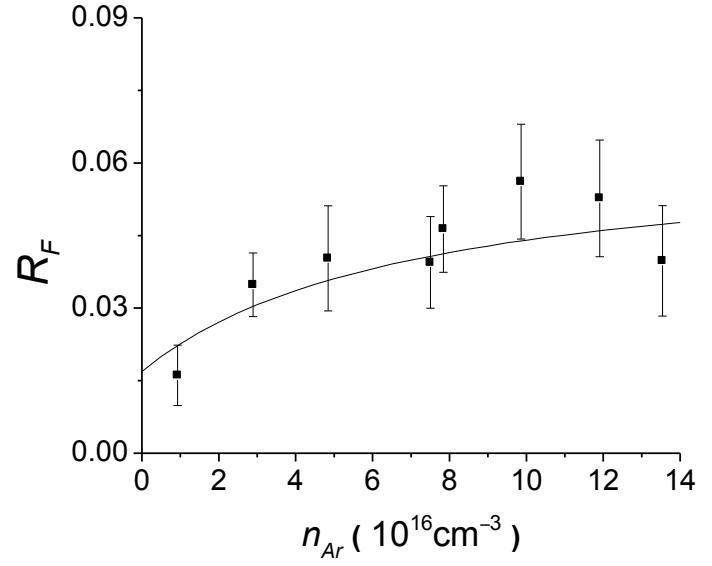
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 40: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = -1$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

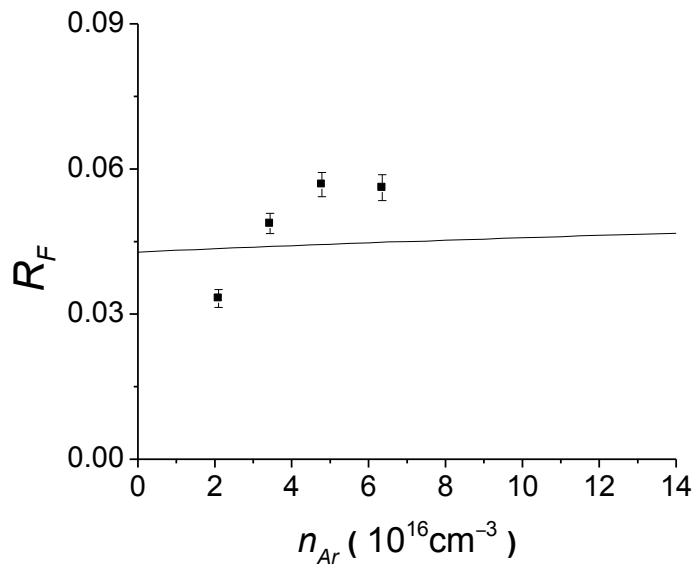
R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = -2$



(a) $n_{Cs} = 9.4 \times 10^{14} \text{ cm}^{-3}$



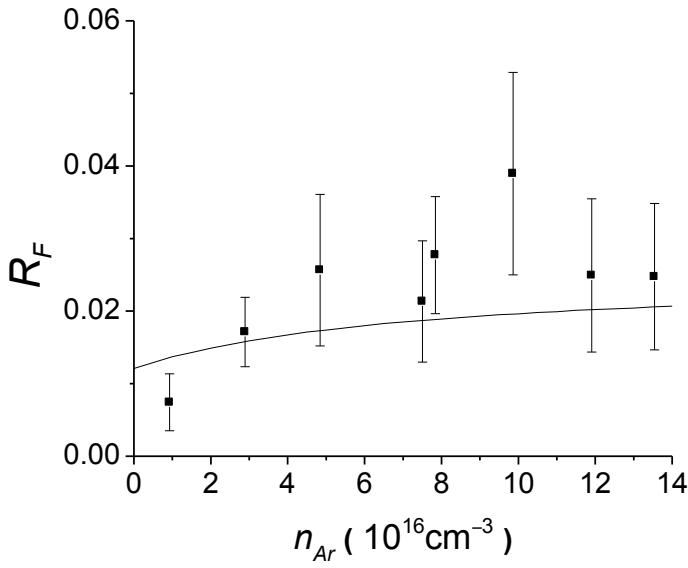
(b) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



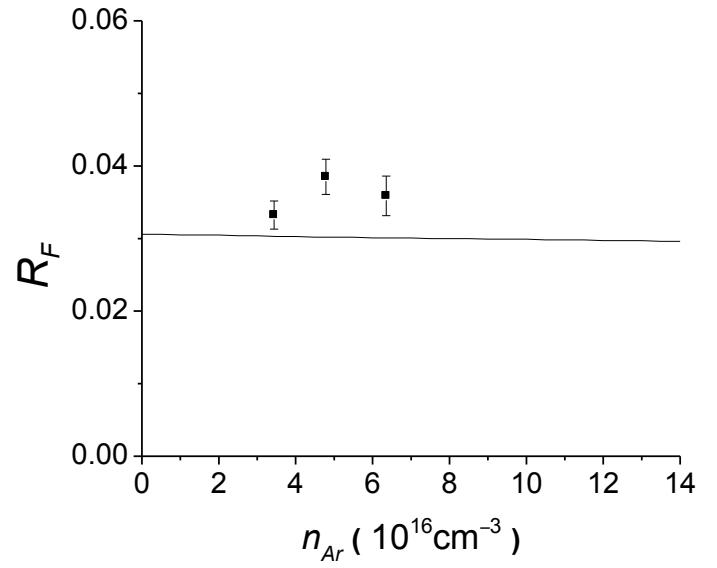
(c) $n_{Cs} = 4.0 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 41: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = -2$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = -3$



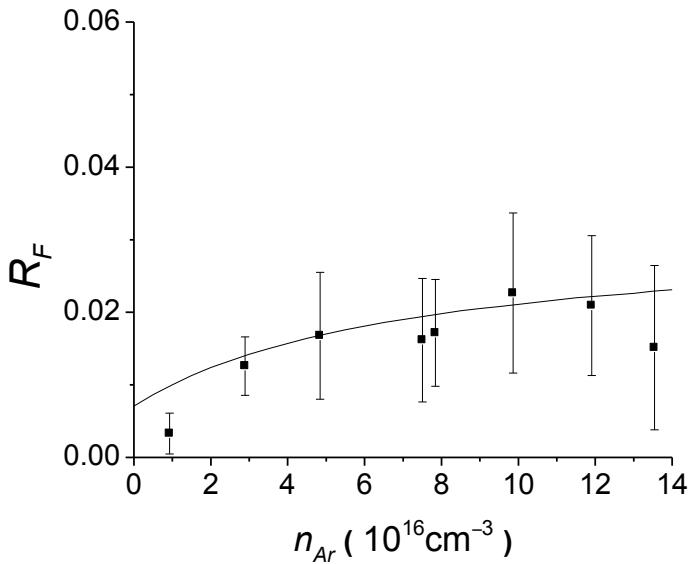
(a) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$



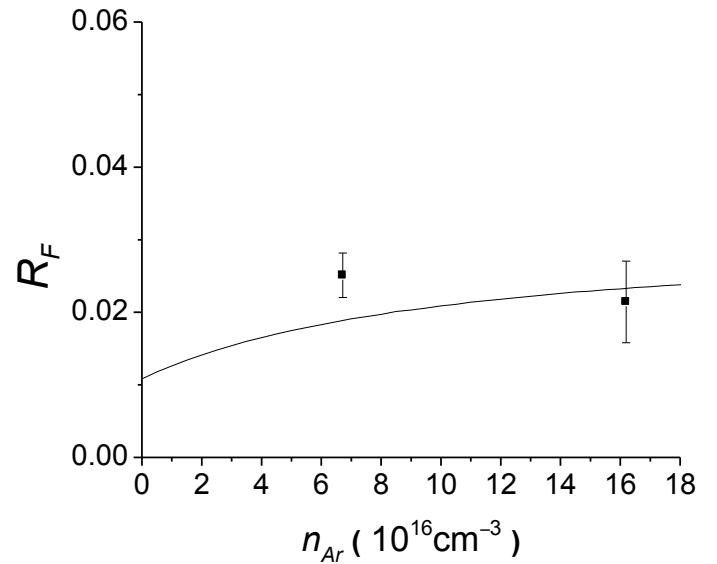
(b) $n_{Cs} = 4.2 \times 10^{16} \text{ cm}^{-3}$

Supplementary Material Fig. 42: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = -3$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .

R_F vs. n_{Ar} for NaCs $2(A)^1\Sigma^+(v=14, J=32)$, $\Delta J = -4$

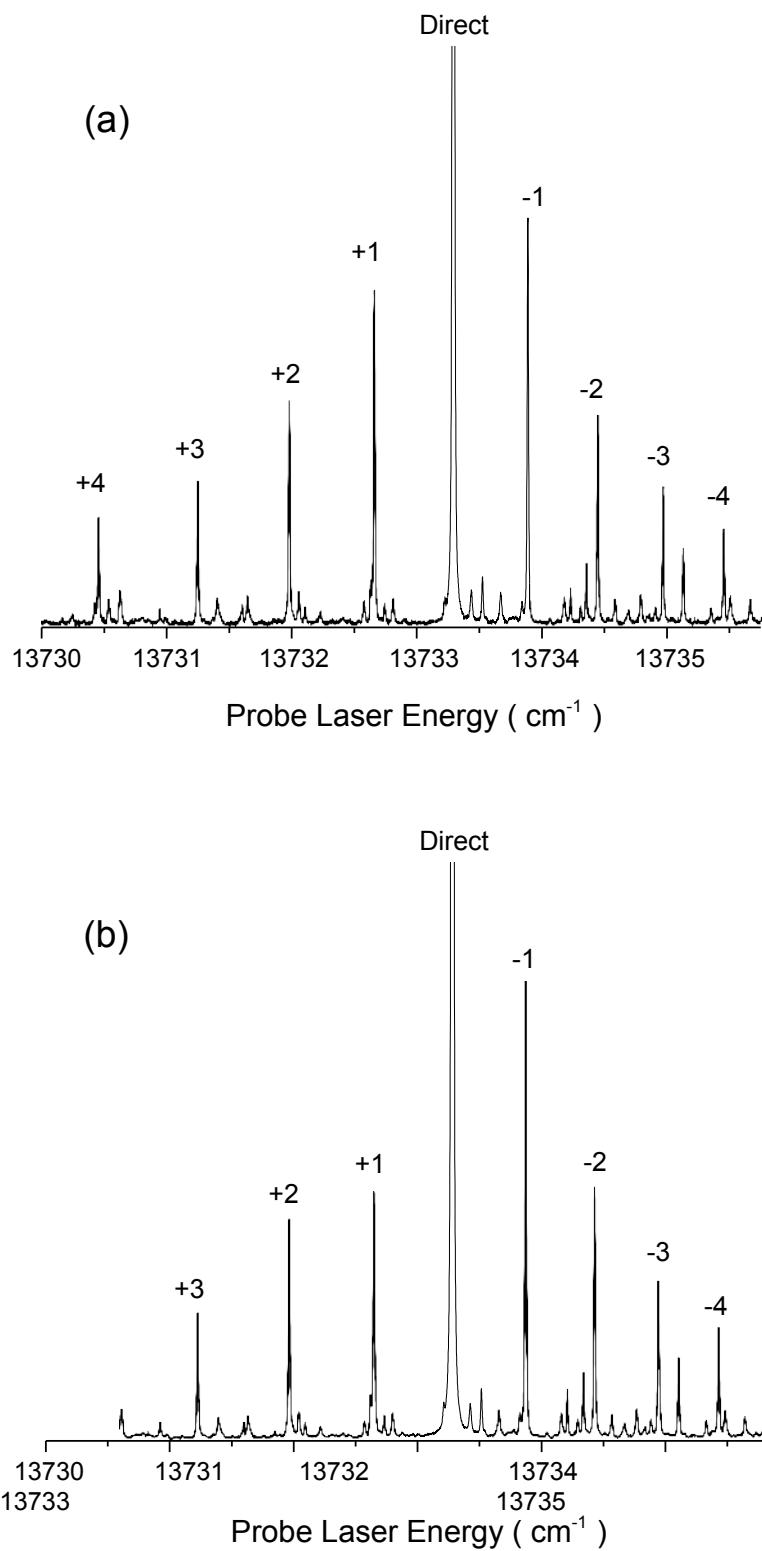


(a) $n_{Cs} = 1.9 \times 10^{15} \text{ cm}^{-3}$

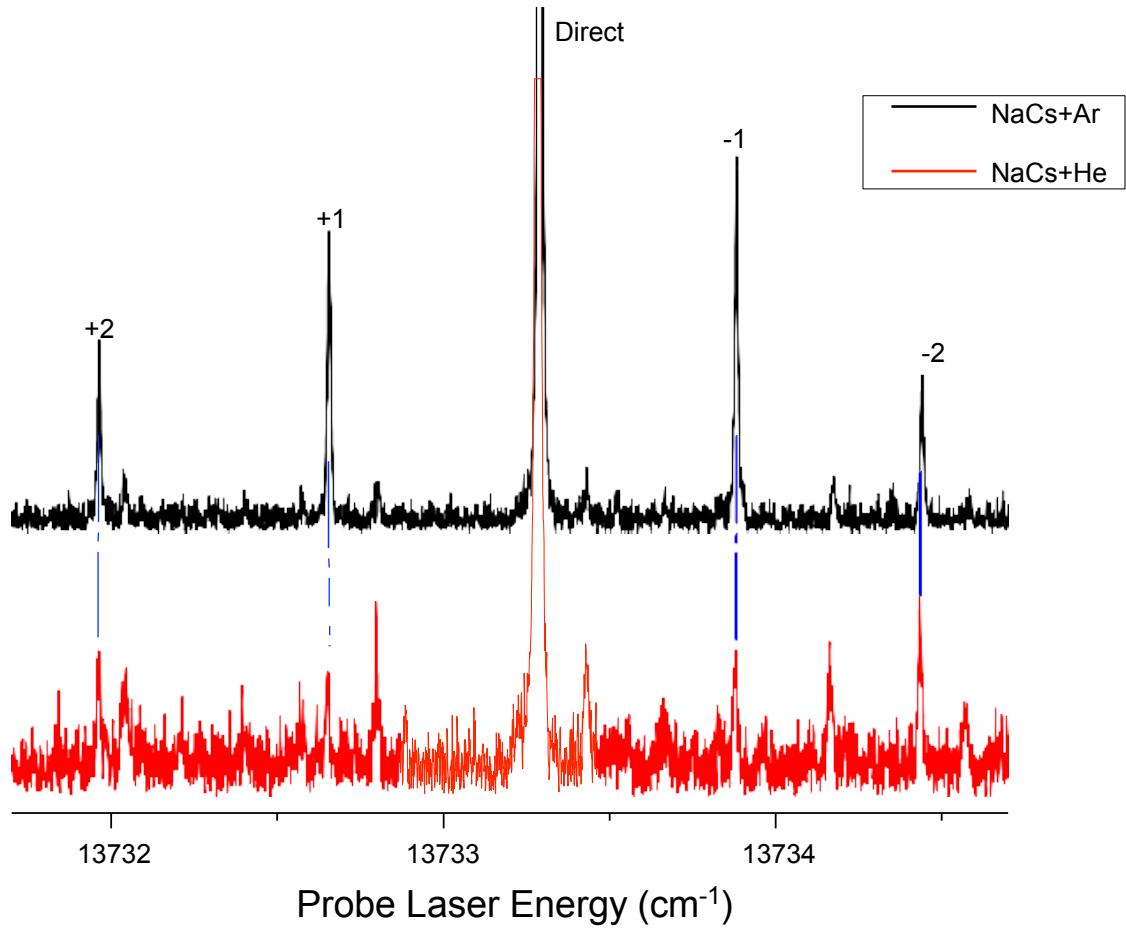


(b) $n_{Cs} = 4.2 \times 10^{15} \text{ cm}^{-3}$

Supplementary Material Fig. 43: Plots of fluorescence ratio data (R_F) versus argon density for $\Delta J = -4$ collisions of NaCs $2(A)^1\Sigma^+(v=14, J=32)$ molecules with argon and cesium perturbers. Each panel represents a fixed cesium density n_{Cs} .



Supplementary Materials Fig. 44: Plot of NaCs excitation spectra for two different cesium densities and similar argon densities ($n_{Ar} \sim 6.0 \times 10^{16} \text{ cm}^{-3}$) comparing the difference in relative peak intensities for $\Delta J = +1$ and $\Delta J = -1$ collisional lines. a) $n_{Cs} = 1.04 \times 10^{16} \text{ cm}^{-3}$, $I_{col}^{\Delta J=+1}/I_{col}^{\Delta J=-1} = 0.78$. b) $n_{Cs} = 4.37 \times 10^{16} \text{ cm}^{-3}$, $I_{col}^{\Delta J=+1}/I_{col}^{\Delta J=-1} = 0.62$.



Supplementary Materials Fig. 45: Spectra showing $\Delta J = \pm 1, \pm 2$ NaCs $2(A)^1\Sigma^+(v = 14, J = 32)$ rotationally inelastic collisions with argon and helium buffer gases. The top trace shows a spectrum obtained using argon as the buffer gas, where a decrease in the intensity of the collisional peaks with increasing $|\Delta J|$ can be seen. The bottom trace shows the same NaCs collisional transitions, except with helium as the buffer gas. The number of $\Delta J = \pm 2$ collisions, relative to the number of $\Delta J = \pm 1$ collisions, appears to be greater for helium than for argon perturbers.