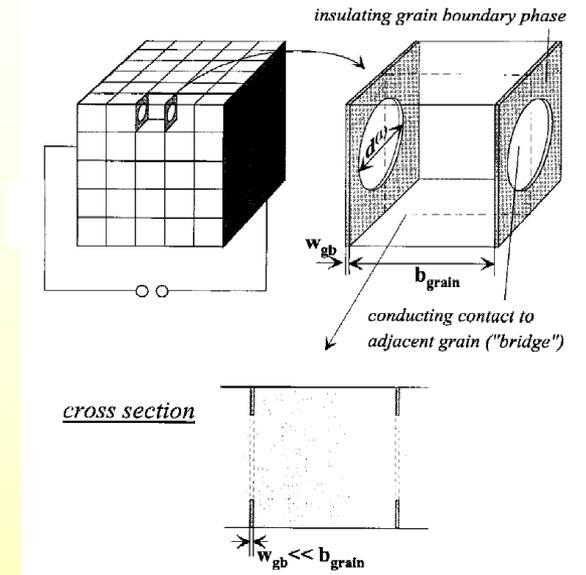
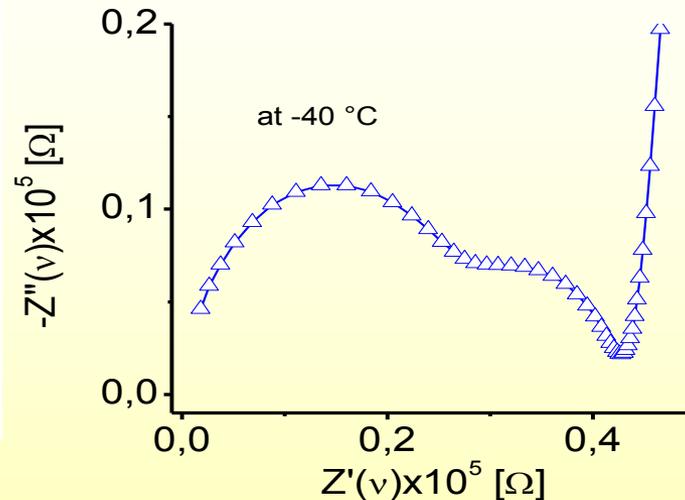
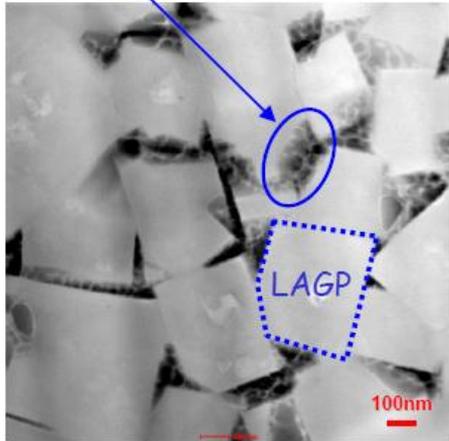


# Ion Transport across Grain Boundaries in Fast Lithium Ion Conducting Glass Ceramics

Amorphous or low crystallinity phase



**Bernhard Roling, Michael Gellert**

Department of Chemistry, University of Marburg

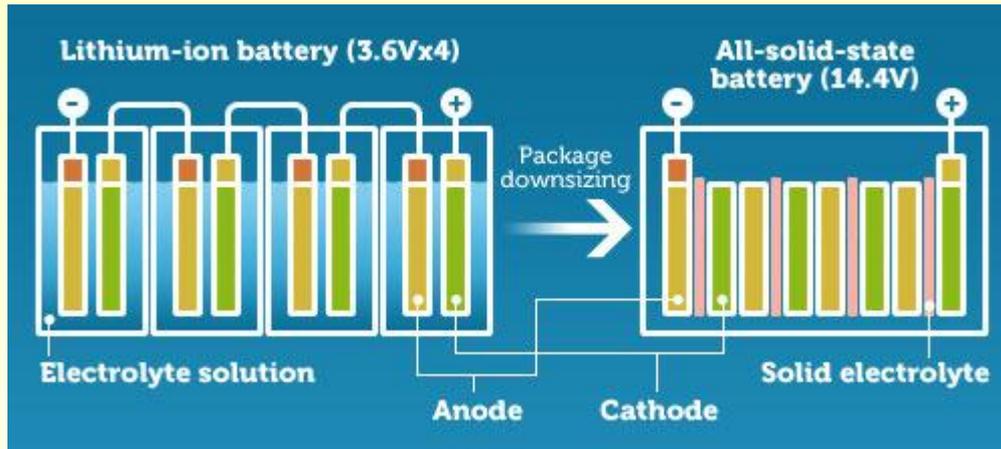
**Katharina I. Gries, Kerstin Volz**

Department of Physics, University of Marburg

**Fabio Rosciano, Chihiro Yada**

Advanced Technology Division, Toyota Motor Europe, Belgium

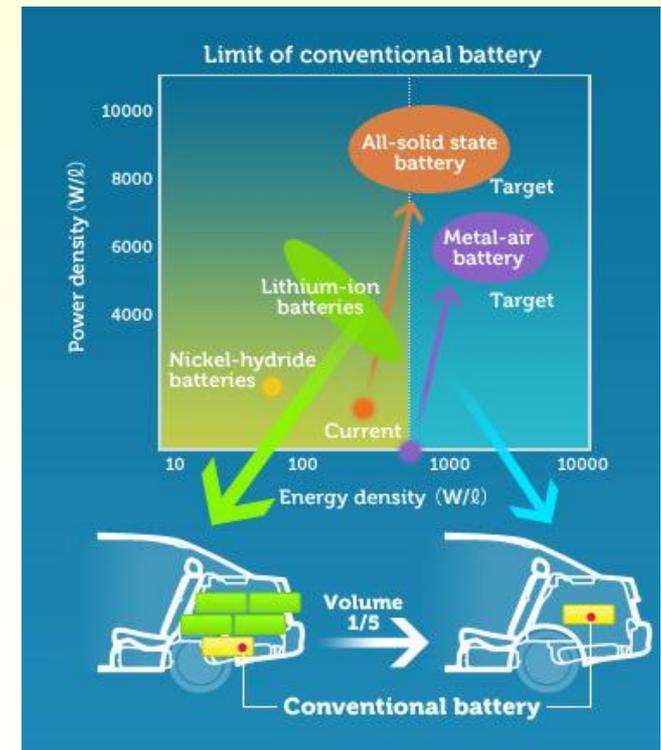
# Solid-State Batteries



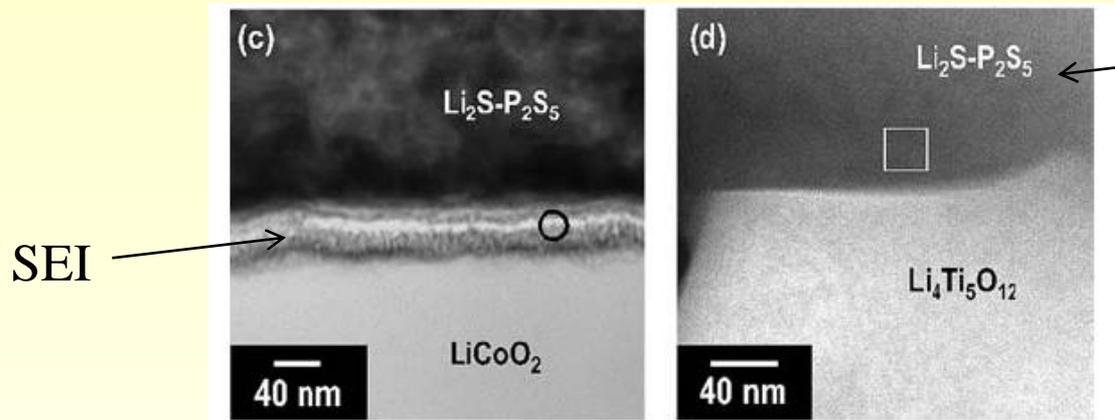
No separation of individual cells

→ More compact packaging

[http://www.toyota-global.com/innovation/environmental\\_technology/next\\_generation\\_secondary\\_batteries.html](http://www.toyota-global.com/innovation/environmental_technology/next_generation_secondary_batteries.html)



# Solid-State Batteries

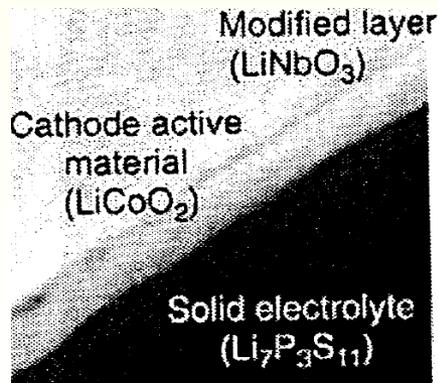


Room-temperature  
ionic conductivity

$$\sigma \approx 10^{-3} \text{ S/cm}$$

Tatsumisago and coworkers,  
J. Mater. Chem. 21 (2011) 118.

Fig. 7 HAADF-STEM images of the interface between the  $\text{LiCoO}_2$  or  $\text{Li}_4\text{Ti}_5\text{O}_{12}$  active material and the solid electrolyte in the composite electrodes pressed at room temperature ((a) and (b)), and pressed at 210 °C ((c) and (d)).



Coating of cathode with protective layer,  
e.g.  $\text{LiNbO}_3$  (thickness in the range of 10 nm)

# Solid-State Li-S-Batteries with $\text{Li}_2\text{S-P}_2\text{S}_5$ Glass as Electrolyte

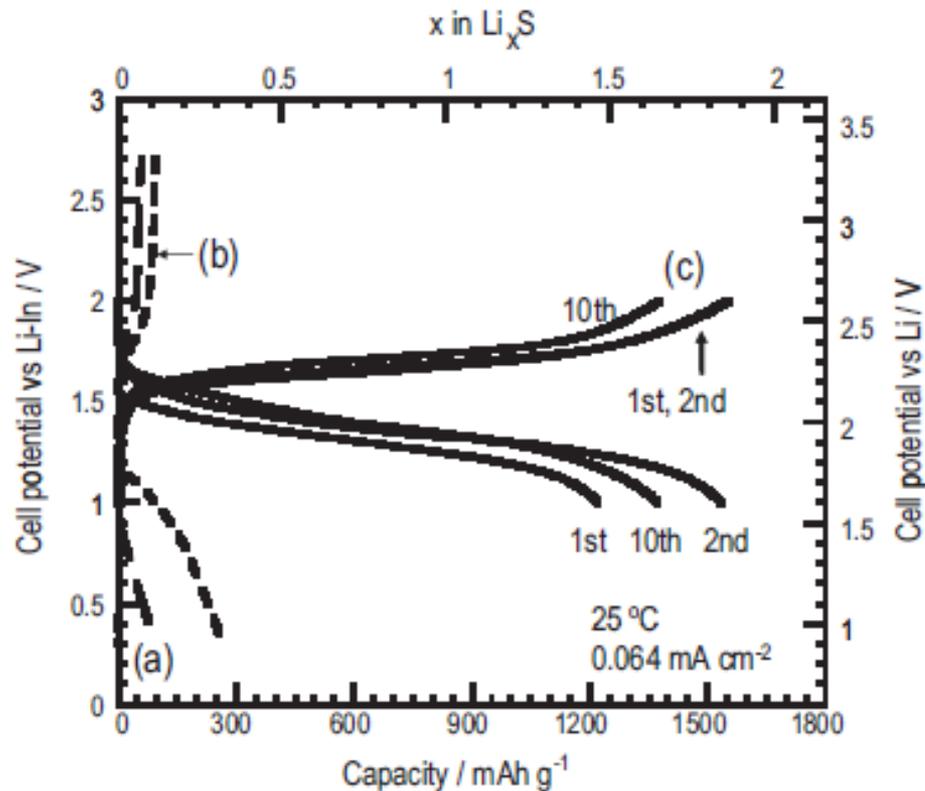
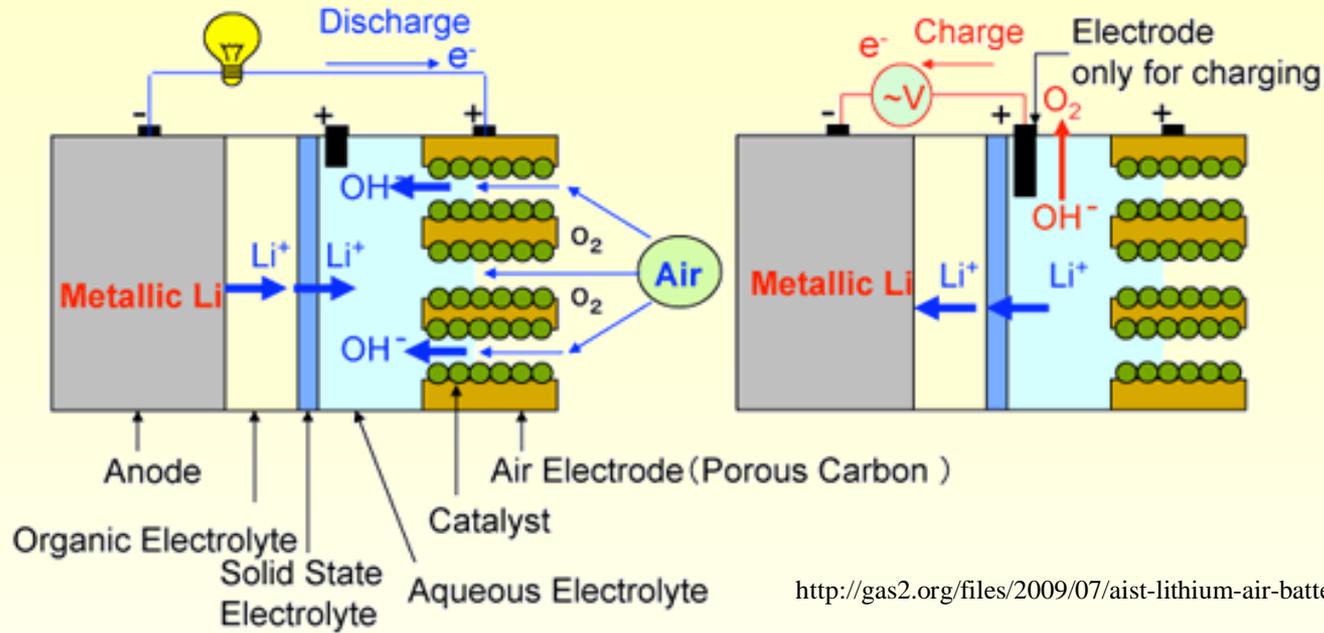


Fig. 3. Charge-discharge curves of all-solid-state cells of Li-In/80 $\text{Li}_2\text{S}$ -20 $\text{P}_2\text{S}_5$  glass-ceramic/S using (a) (S+AB+SE), (b) (S-AB+SE), and (c) S-AB-SE electrodes as the working electrode.

Tatsumisago and coworkers,  
Electrochim. Acta 56 (2011) 6055.

At low charge/discharge rates, the battery capacity is close to the theoretical capacity.

# Lithium-Air Batteries



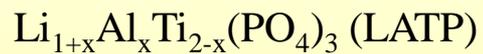
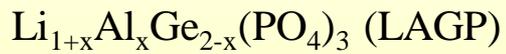
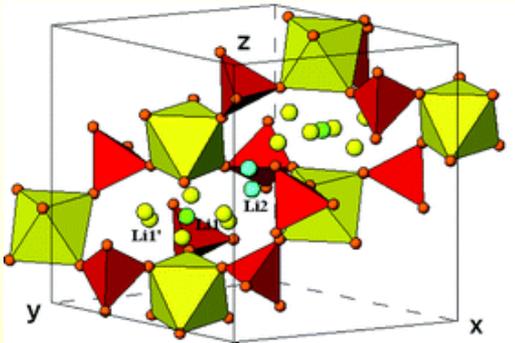
$$EMF \approx 3.1 \text{ V}$$

$$\text{theoretical energy density} \approx 3,6 \frac{\text{kWh}}{\text{kg } Li_2O_2}$$

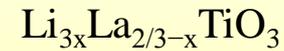
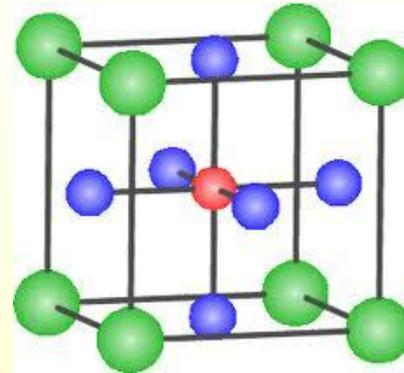
(comparable to mechanical energy from  
1 kg gasoline)

# Crystalline Fast Lithium-Ion Conductors

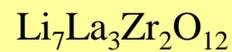
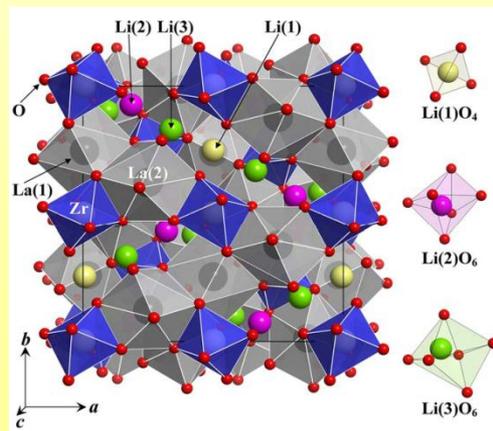
## NASICON



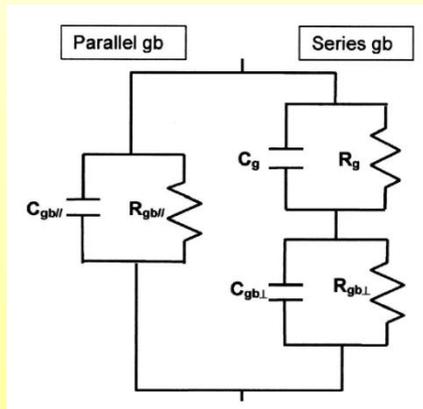
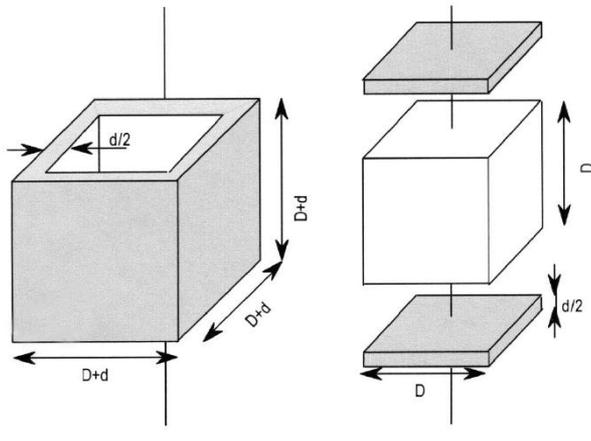
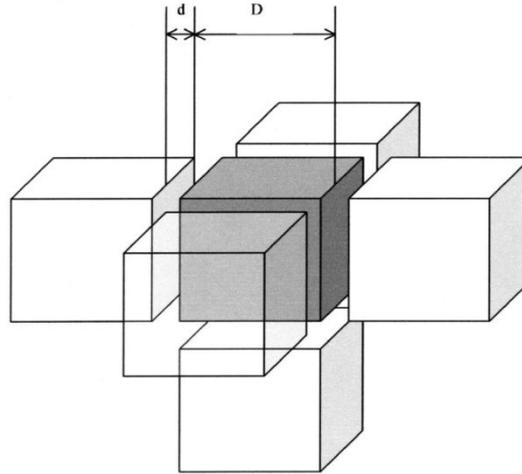
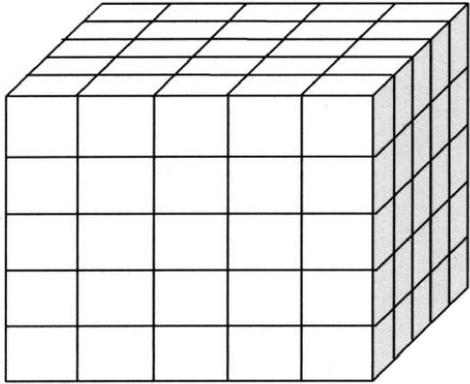
## Perovskite



## Garnet



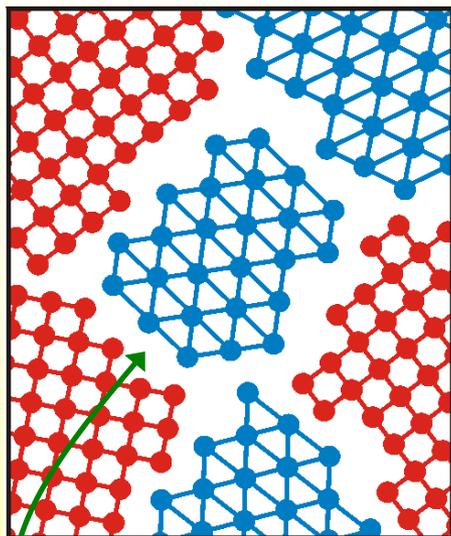
# Brick Layer Model for Grain Boundary Ion Transport



Parallel gb conduction is only then relevant, if

- $\sigma_{gb} \gg \sigma_g$  or
- $D$  is comparable to  $d$

# Parallel Grain Boundary Conduction in Lithium Ion Conductors

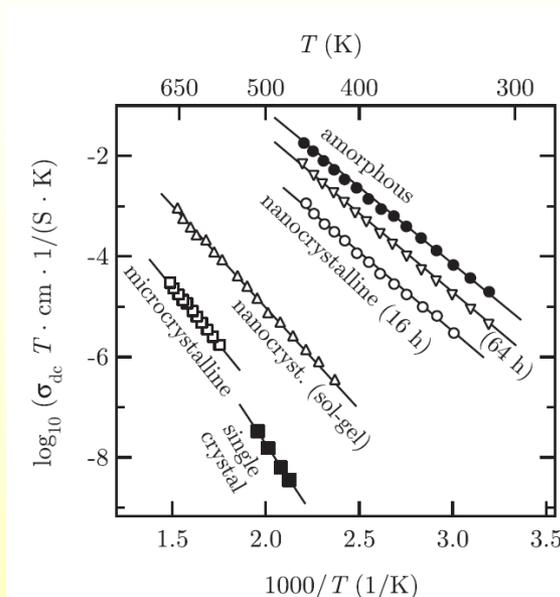


**Li<sub>2</sub>O - B<sub>2</sub>O<sub>3</sub>  
nanocomposite**

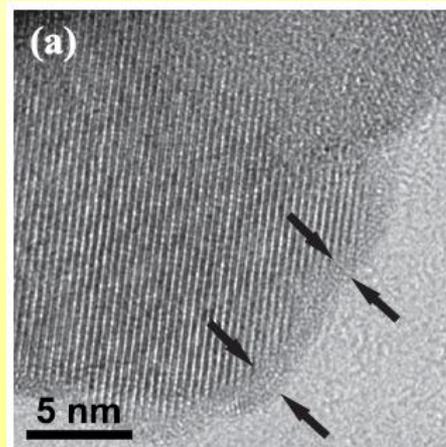
Fast Li<sup>+</sup> ion conduction  
at interfaces

P. Heitjans,  
S. Indris,  
*Phys. Cond. Mat.*  
**15** (2003) R1257.

## LiNbO<sub>3</sub>

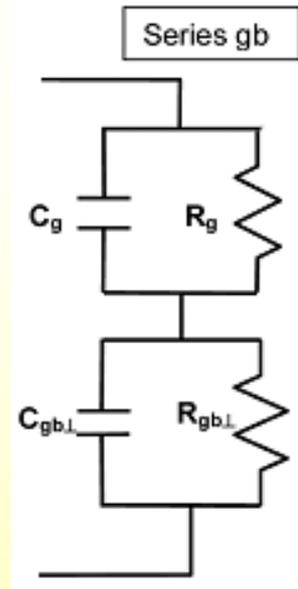
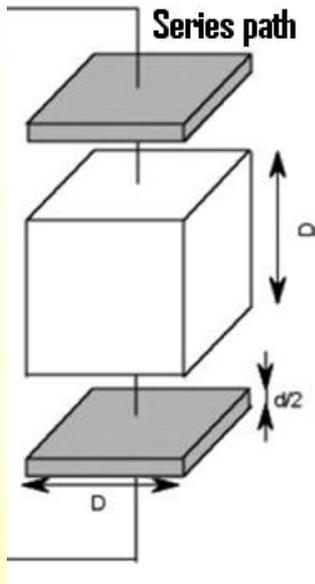


P. Heitjans,  
M. Masoud,  
A. Feldhoff,  
M. Wilkening,  
*Faraday Discuss.*  
**134** (2007) 67.



Large amount of  
*amorphous* LiNbO<sub>3</sub>  
at the boundaries  
of the nanograins

If parallel grain boundary conduction is negligible:

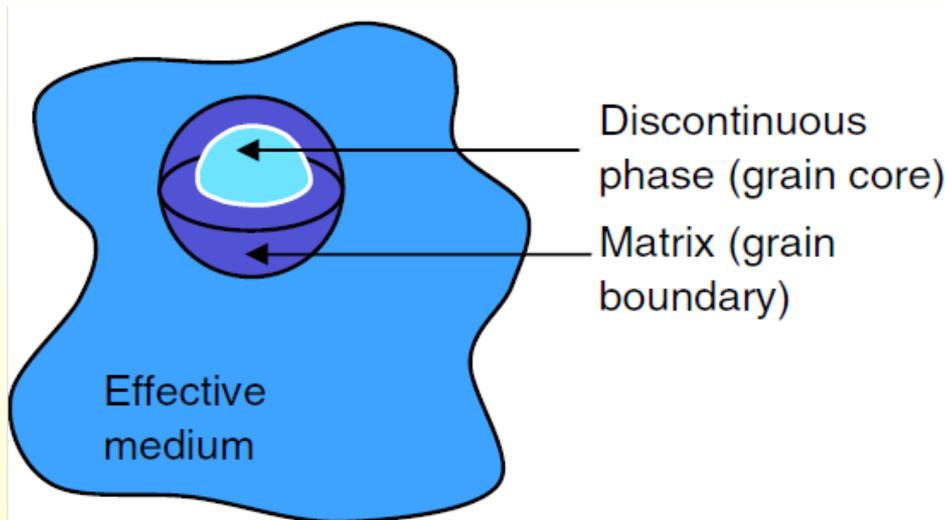


Simple results for ratios of capacitances and resistances:

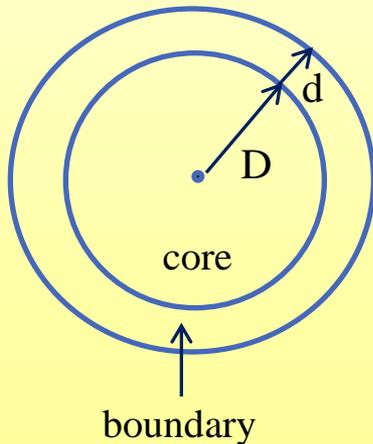
$$\frac{R_g}{R_{gb,\perp}} = \frac{\sigma_{gb}}{\sigma_g} \cdot \frac{D}{d}$$

$$\frac{C_g}{C_{gb,\perp}} = \frac{\epsilon_g}{\epsilon_{gb}} \cdot \frac{d}{D}$$

# Nano-Grain Composite Model



Kidner et al.,  
 J. Am. Ceram. Soc.  
 91 (2008) 1733.



if  $D \gg d$

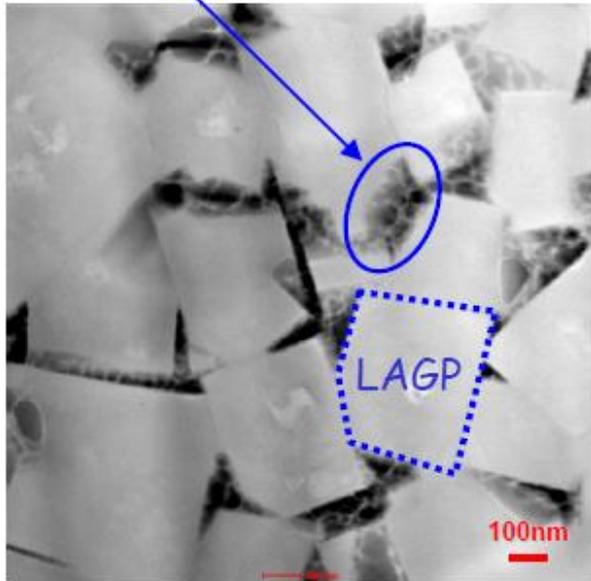
$$\frac{C_{gb}}{C_{gc}} = \frac{\epsilon_{gb}}{\epsilon_{gc}} \cdot \frac{D}{d}$$

$$\frac{R_{gb}}{R_{gc}} = \frac{\sigma_{gc}}{\sigma_{gb}} \cdot \frac{d}{D}$$

in agreement with  
 brick layer model

## TEM Images of $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)

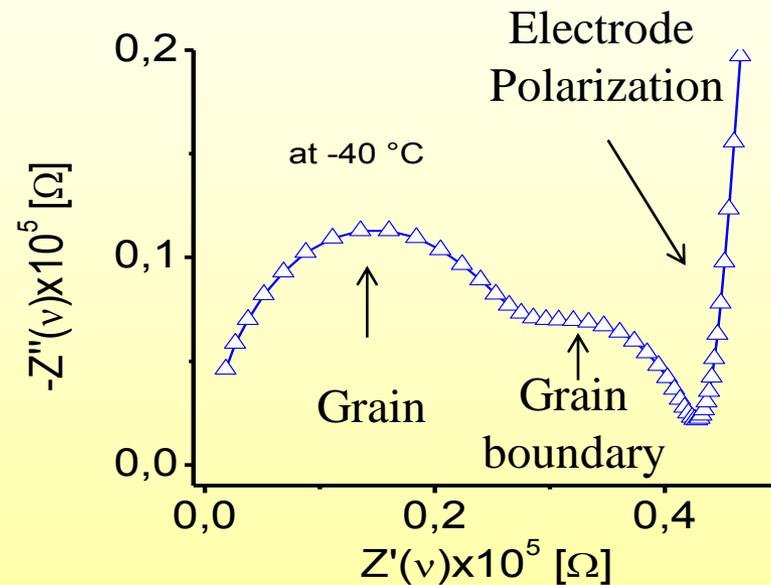
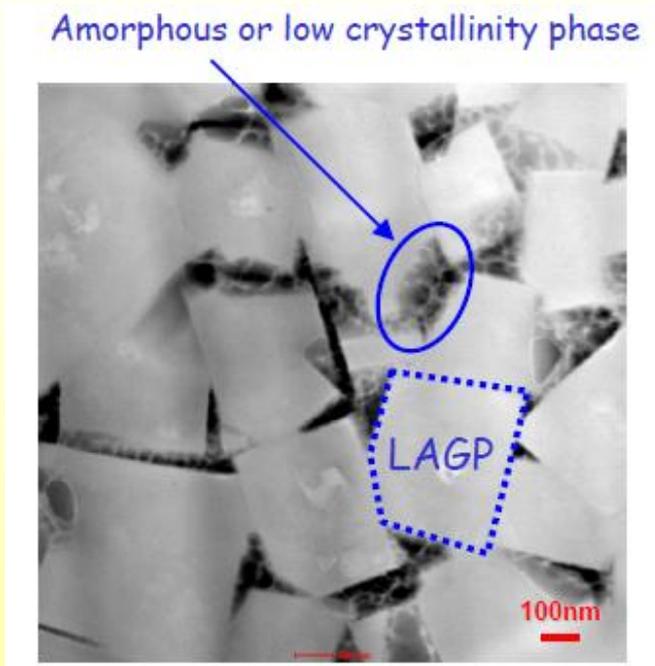
Amorphous or low crystallinity phase



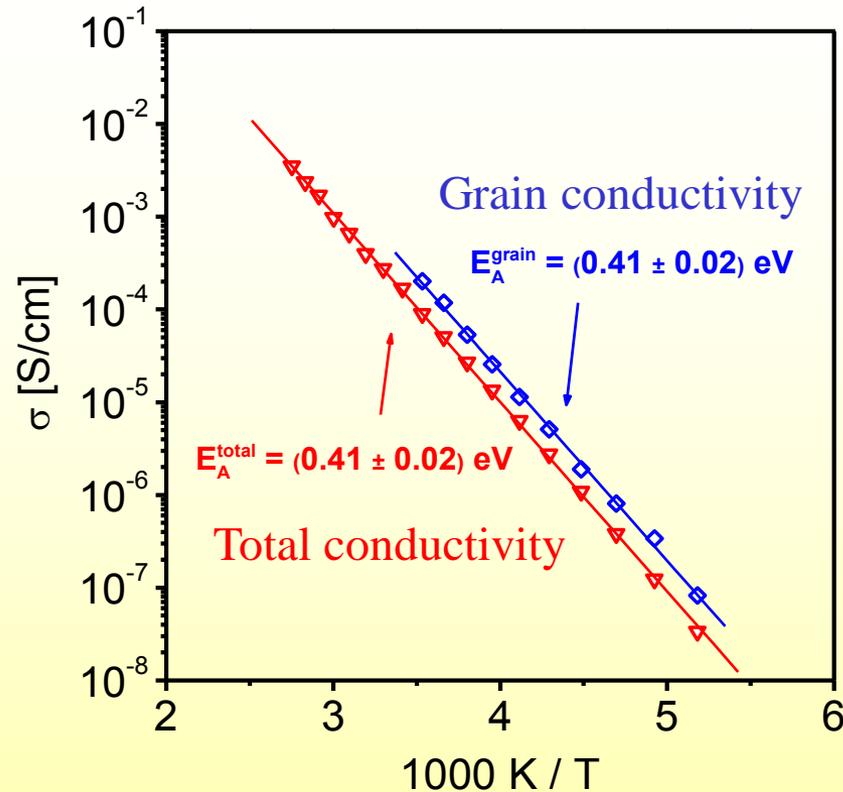
C. R. Mariappan, C. Yada,  
F. Rosciano, B. Roling,  
J. Power Sources 196 (2011) 6455.

- (i) Existence of impurity phases (e.g.  $\text{AlPO}_4$ ) and amorphous phases with low ionic conductivity
- (ii) Contact area between grains is lower than assumed in the BLM

# Impedance Spectrum of $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)



# Grain Conductivity and Total Conductivity of LAGP



Grain conductivity

$$\sigma_{\text{g}} = \frac{1}{R_{\text{g}}} \left( \frac{d}{A} \right)$$

Total conductivity

$$\sigma_{\text{t}} = \frac{1}{(R_{\text{g}} + R_{\text{gb}})} \left( \frac{d}{A} \right)$$

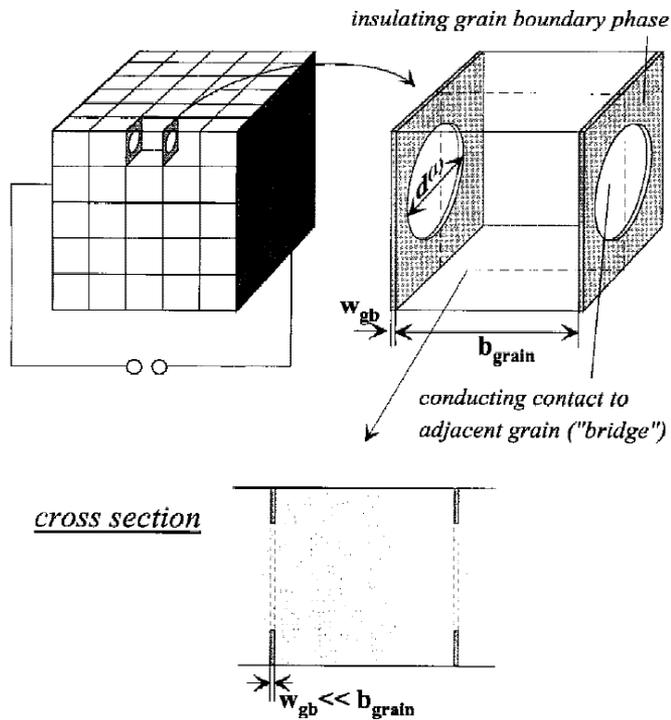
Grain and grain boundary resistance are almost identical.



Activation energies are identical.

Purely geometrical current constriction due to limited grain contact area

# Finite-Element Calculations by Fleig and Maier



J. Fleig, J. Maier, J. Am. Ceram. Soc  
82 (1999) 3485.

$$\frac{R_{gb}}{R_g} \approx \frac{1}{\sqrt{4\alpha}}$$

Fraction of contacted area

$$\alpha = A_{\text{contact}} / A_{\text{grain}}$$

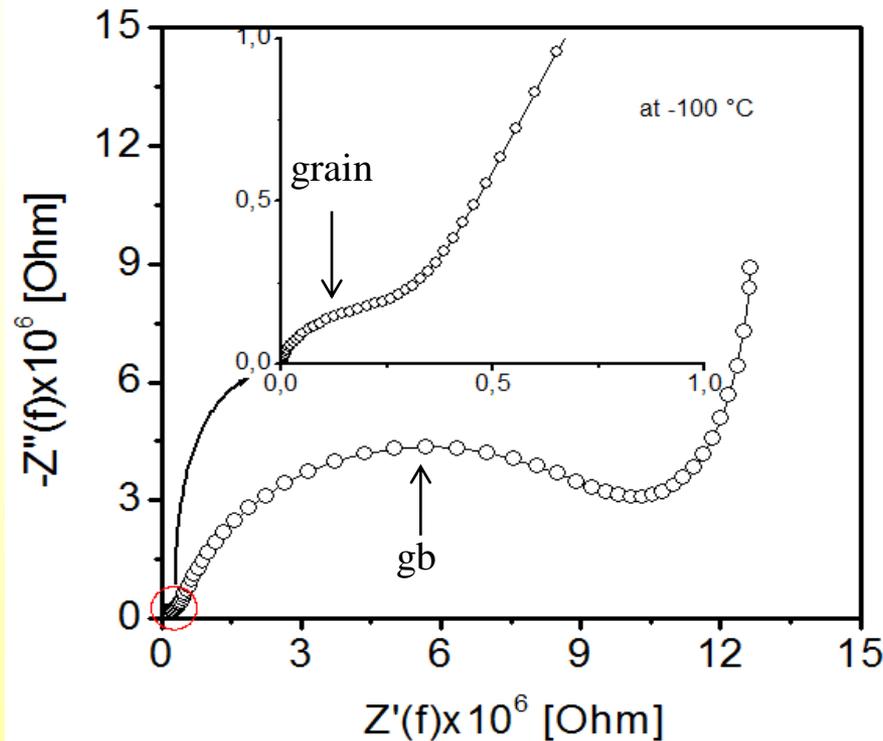
For LAGP:

$$R_{gb} \approx R_g \quad \longrightarrow \quad \alpha \approx 0.25$$

in reasonable agreement  
with TEM results

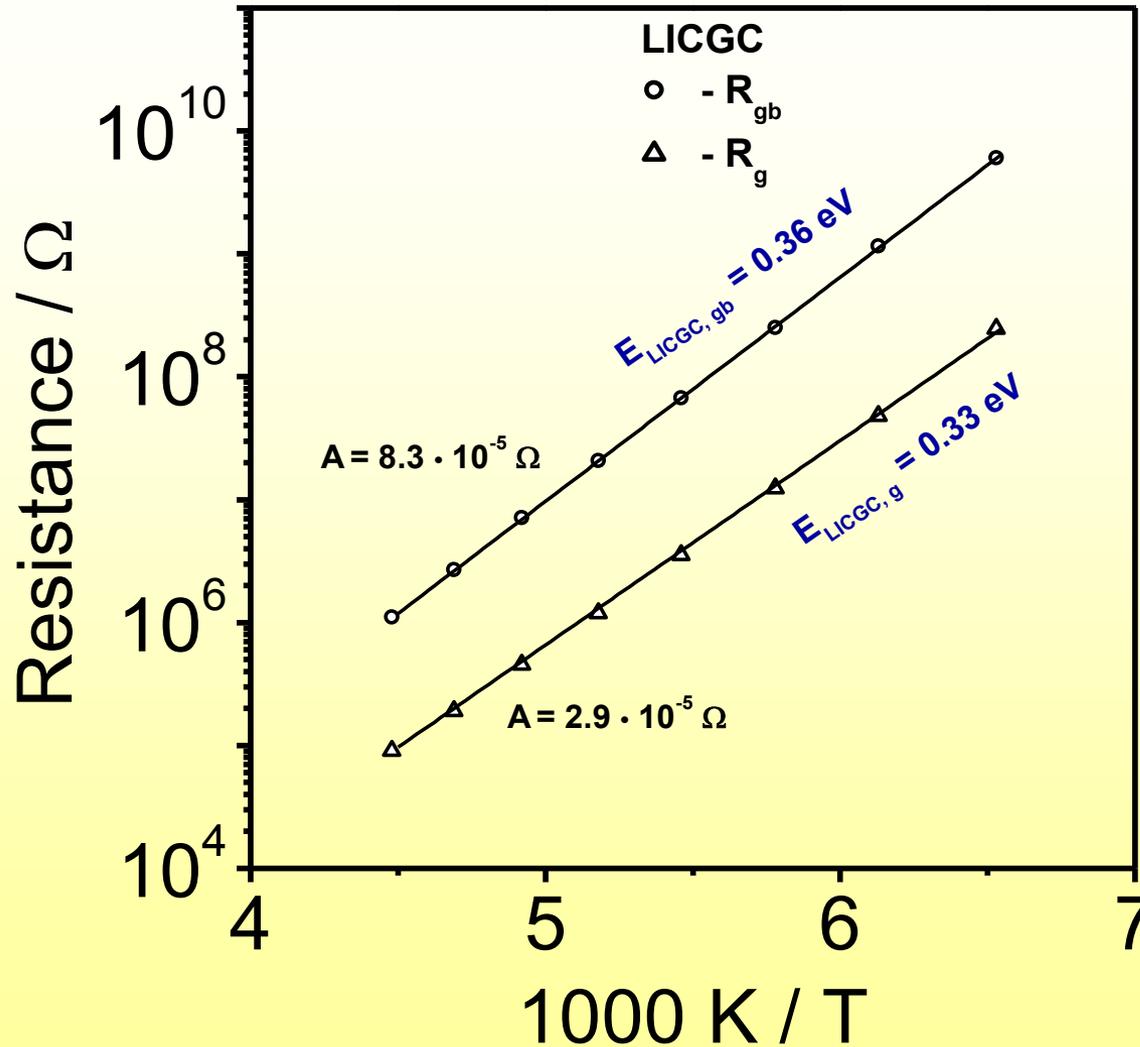
# Impedance Spectroscopy on Ohara Glass Ceramic (commercial)

$\text{Li}_{1+x}\text{Al}_x\text{Ti}_{2-x}(\text{PO}_4)_3$   
doped with various  
other oxides

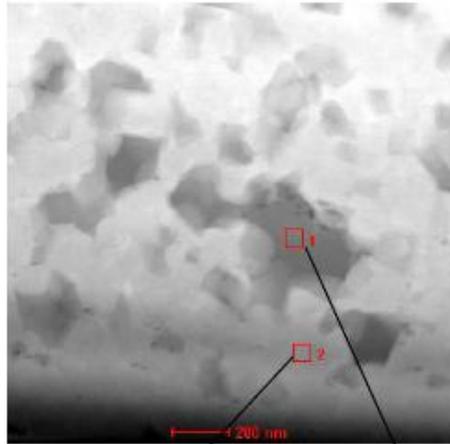


Grain boundary resistance  
is more than one order of magnitude  
higher than grain resistance

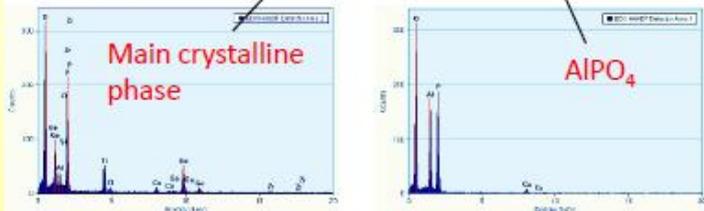
# Arrhenius Plot of Grain and Grain Boundary Resistance



# Ohara Glass Ceramic



(x 68k)



Grain conductivity at room temperature:

$$\sigma_{\text{grain}} \approx 10^{-3} \text{ S/cm}$$

$$E_{\text{A}}^{\text{grain}} = 0.33 \text{ eV}$$

Grain boundary conductivity at room temperature:

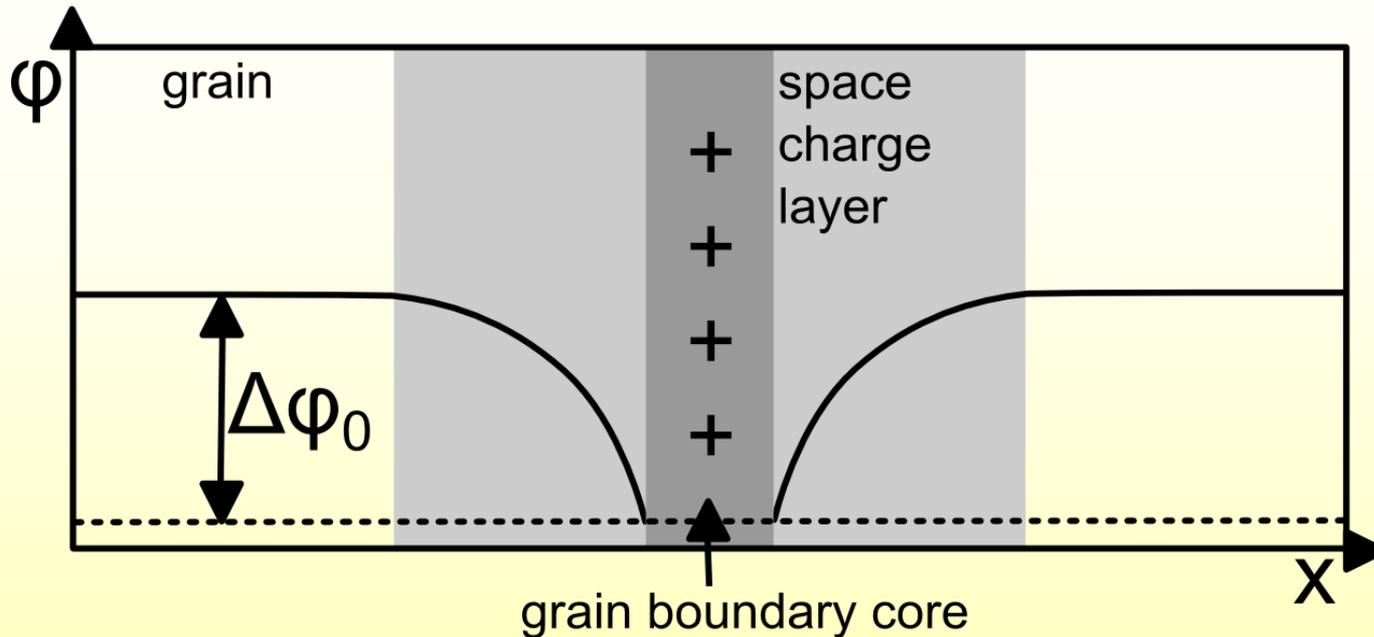
$$\sigma_{\text{grain boundary}} \approx 10^{-4} \text{ S/cm}$$

$$E_{\text{A}}^{\text{grain boundary}} = 0.36 \text{ eV}$$

**Origin of higher activation energy of grain boundary conductivity?**

**Space charge layers? Mechanical stresses?**

## Space Charge Model



- Charged gb core with oppositely charged space charge layer
- Space charge layer results in electrostatic barrier for ion transport;

**Important: Single barrier**

**Well established in the field of oxide ion conductors**

# Measurements with DC Bias on CeO<sub>2</sub> doped with 1% Y<sub>2</sub>O<sub>3</sub>



Figure 1. HRTEM of grain boundary in 1.0 mol % Y<sub>2</sub>O<sub>3</sub>-doped CeO<sub>2</sub>. The moiré rings are also visible.

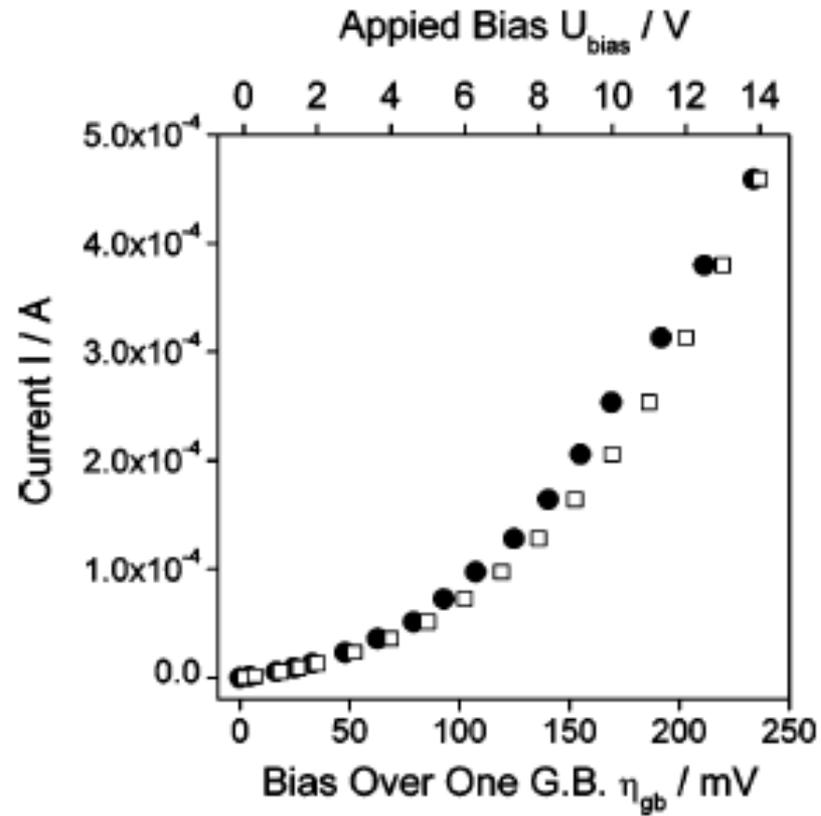


Figure 3. Current  $I$  as a function of bias over one grain boundary  $\eta_{gb}$  (filled circles). For comparison the current  $I$ -applied bias  $U_{bias}$  relation is also plotted (open squares).

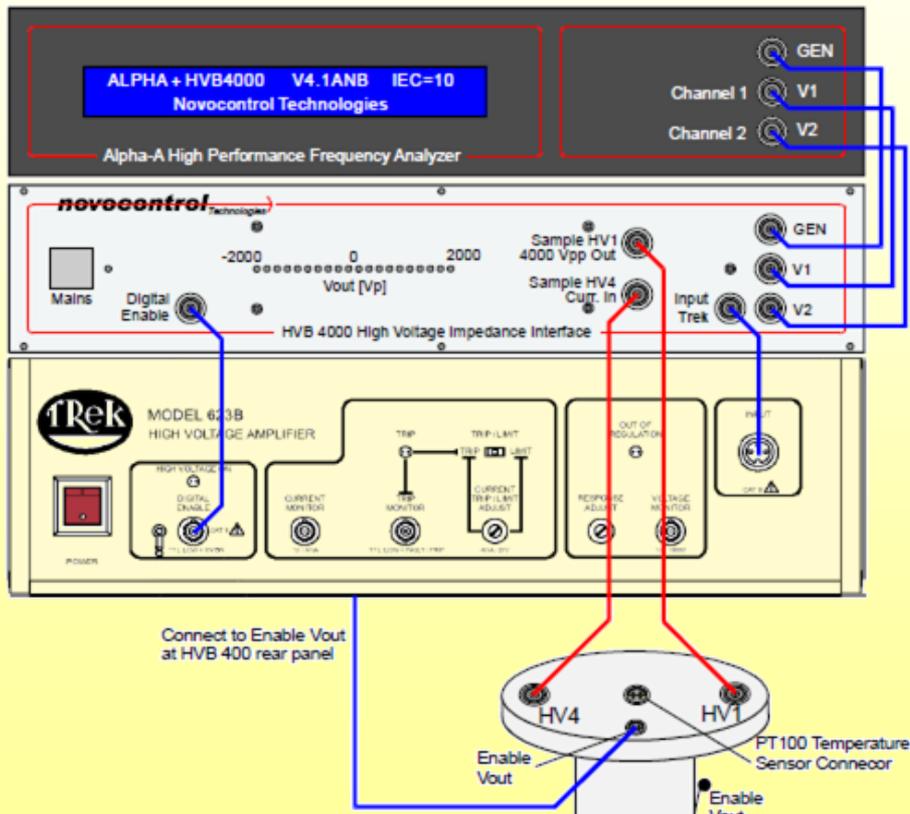
Guo, Waser et al.,  
 Electrochem. Solid State Lett.  
 8 (2005) J1 and 8 (2005) E67.

**Height of  
 space charge barrier:  
 about 0.4 eV**

# High-Voltage Measurement System

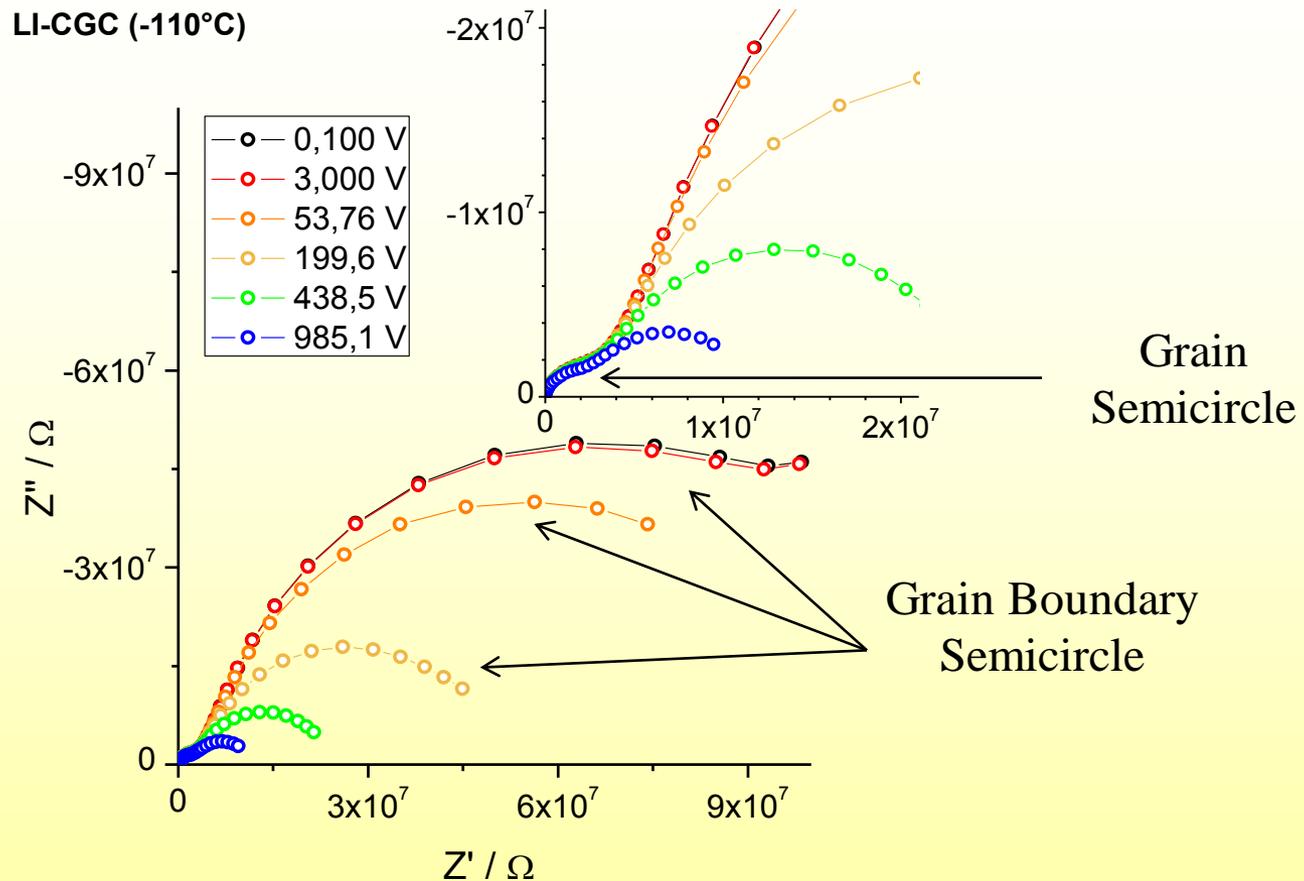
Novocontrol Alpha-AK High Performance Frequency Analyser, equipped with:

- High-Voltage Amplifier Trek model 623B
- Novocontrol HVB4000 High-Voltage Impedance Interface



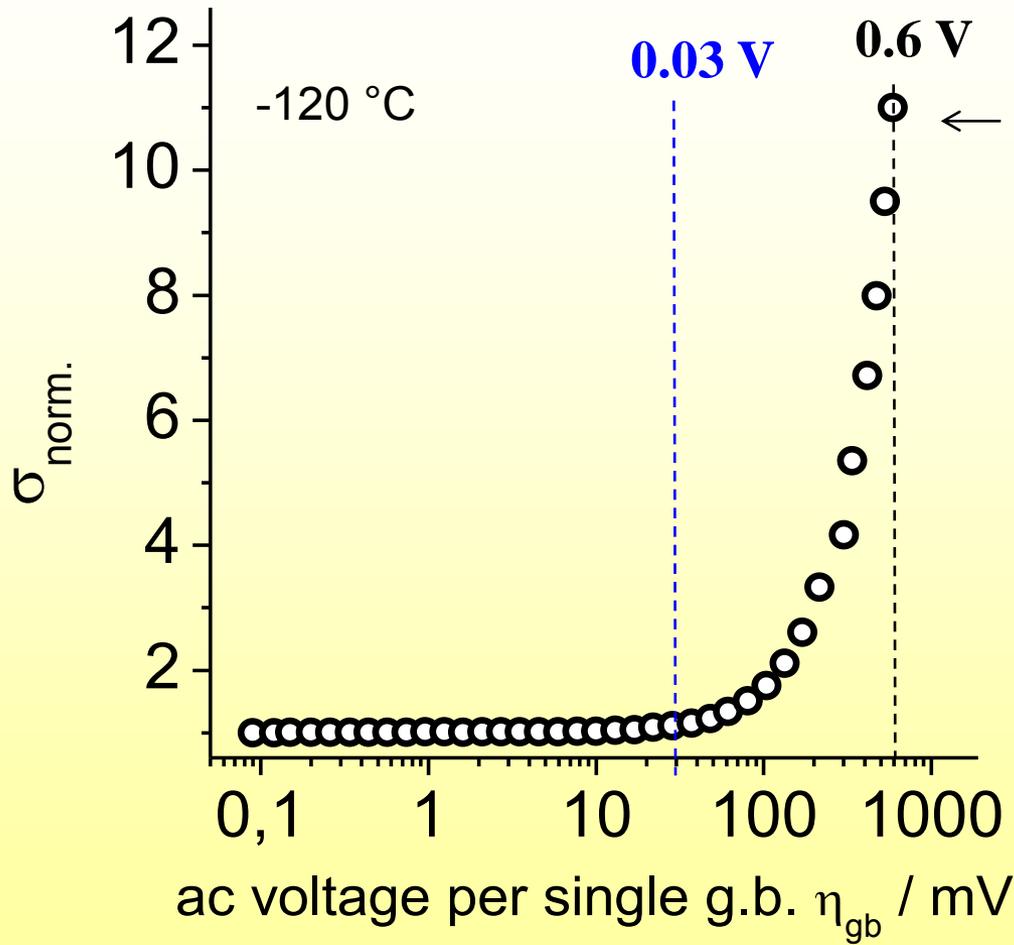
- Frequency range: 3  $\mu$ Hz - 10 kHz
- Maximum amplitude of ac voltage: 2 kV
- Current resolution: 5 fA

# Nonlinear Impedance Spectroscopy with High AC Voltages



- **Almost 1 V ac voltage per single grain boundary can be applied without any irreversible changes of the grain boundary properties**
- **Grain boundary resistance decreases with increasing voltage**

# Grain Boundary Conductivity vs. AC Voltage per Single Grain Boundary



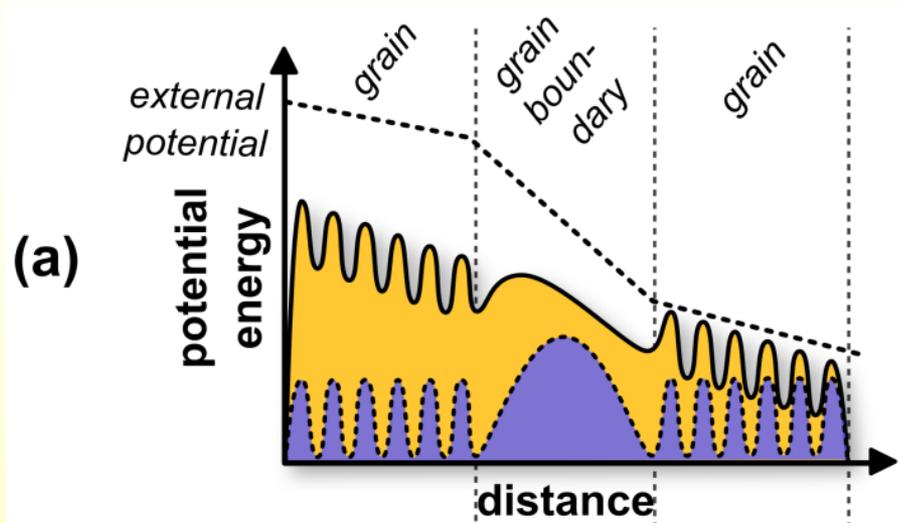
← Grain boundary resistance is still slightly higher than grain resistance

If a **single higher barrier** existed at each grain boundary

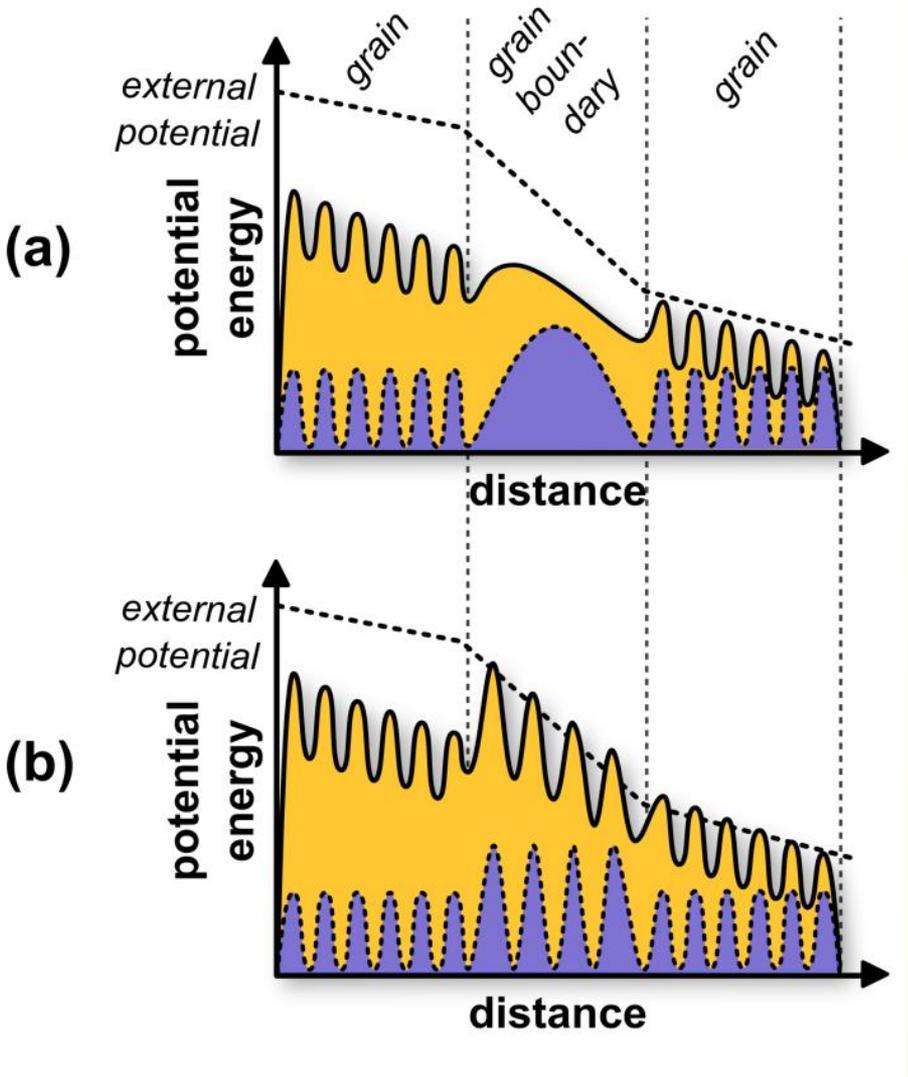


An external voltage of about 30 mV should be sufficient for decreasing the grain boundary resistance to the level of the grain resistance

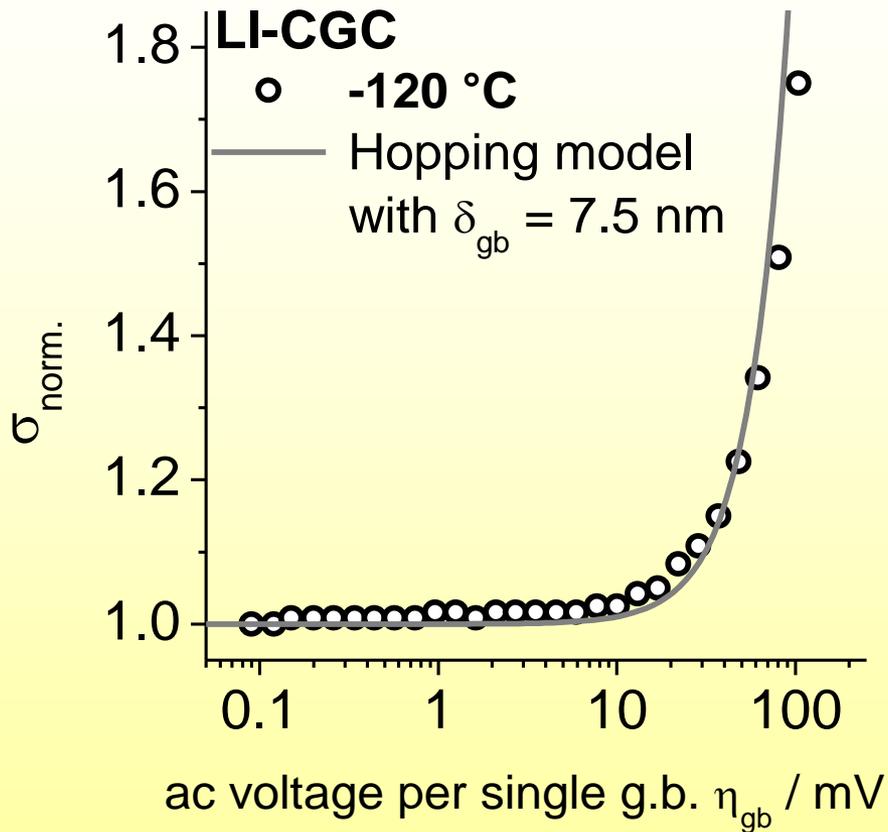
# Influence of High Voltages on Ion Transport Across Grain Boundary



# Influence of High Voltages on Ion Transport Across Grain Boundary



# Estimation of Grain Boundary Thickness

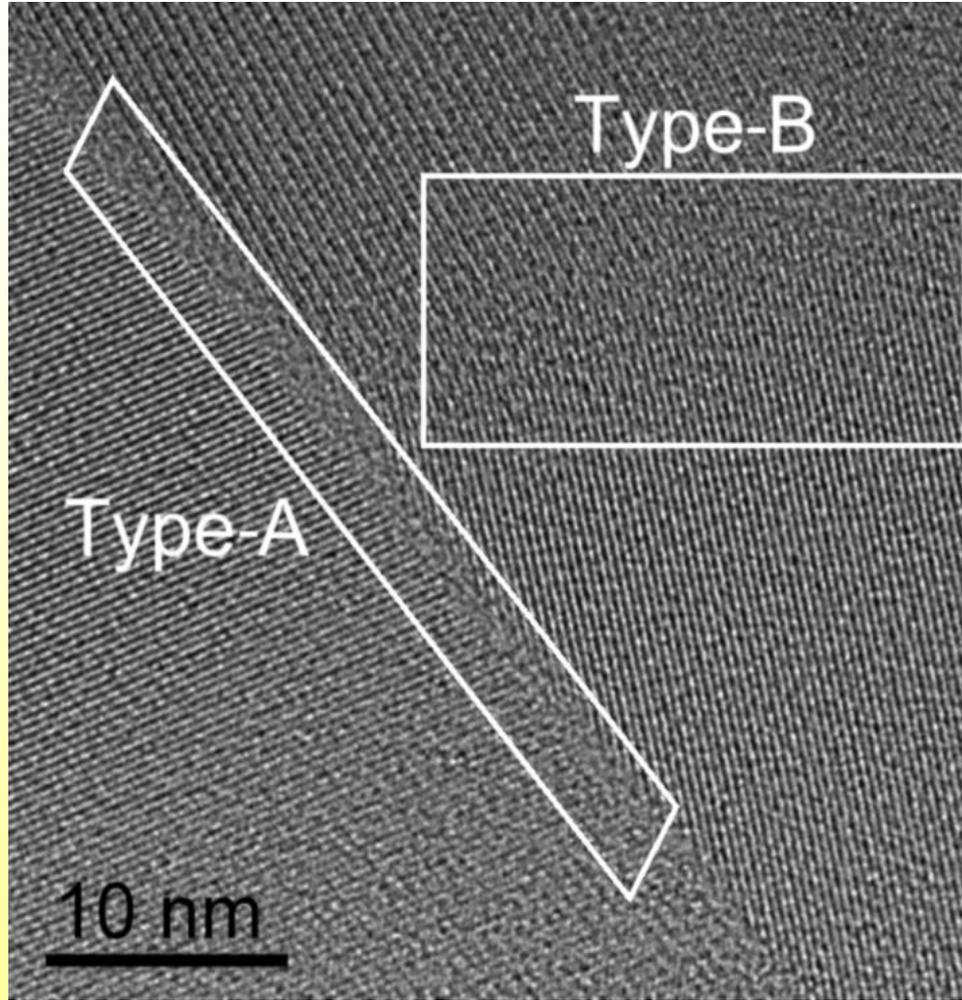


**Assumption:**  
Field dependence of  
gb conductivity follows  
universal behaviour  
of thin ion-conducting  
layers.



Thickness of  
grain boundary:  
about 7.5 nm

## HR-TEM Images of Grain Boundaries



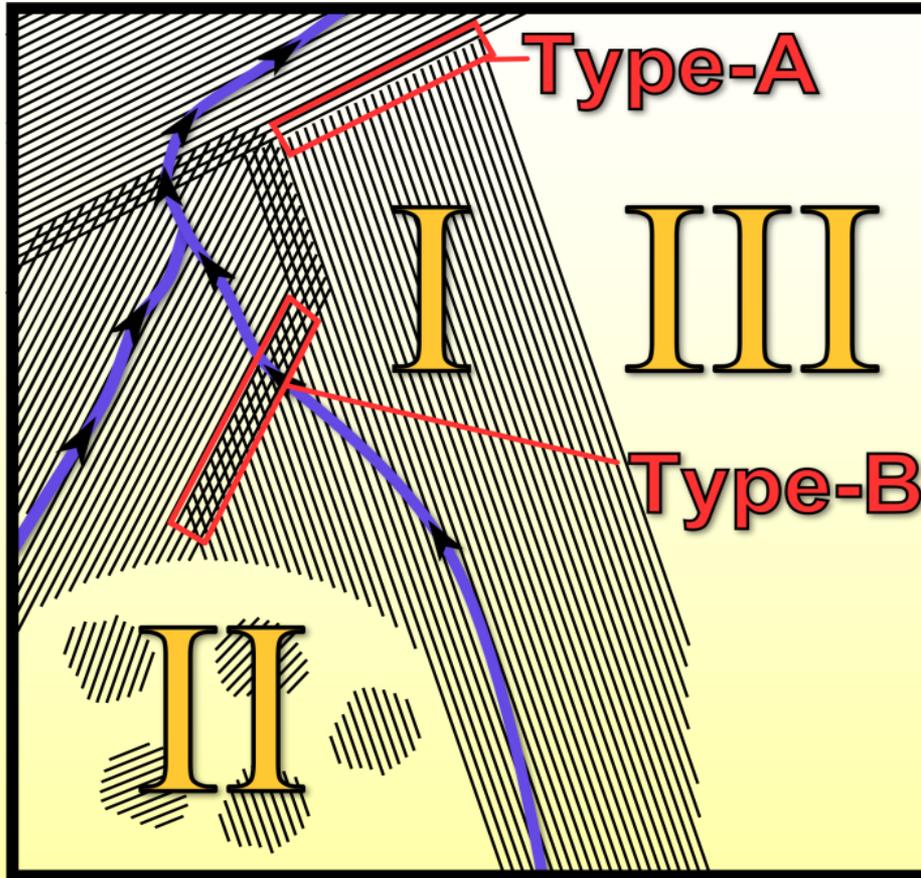
### **Type-B Grain Boundary:**

- Layer between grains with similar lattice orientation:
- High degree of crystallinity
- Mechanical stresses may lead to slightly higher activation energy
- Thickness: about 5-10 nm

### **Type-A Grain Boundary:**

Amorphous layer between grains with strongly dissimilar lattice orientation  
→ Highly resistive

# Model for Grain Boundary Transport



Blue lines: Ion transport pathways

# Conclusions

## $\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$ (LAGP)

- Grain and grain boundary resistance exhibit the same activation energy.  
→ **Purely geometrical current constriction; fraction of contacted area: about 25%**

## Ohara Glass Ceramic ( $\text{Li}_{1+x}\text{Al}_x\text{Ti}_{2-x}(\text{PO}_4)_3$ doped with other oxides)

- Grain boundary resistance exhibits a slightly higher activation energy than the grain resistance.
- Nonlinear impedance spectra provide strong indication that the grain boundary resistance is **not** caused by a **single (space charge) barrier**, but by **several serial barriers**
- Fit of nonlinear impedance data suggests that the **thickness of the grain boundaries is in the range 5-10 nm.**
- HR-TEM images reveal type-A (amorphous) and **type-B** (high degree of crystallinity) **grain boundaries.**
- **Thickness of type-B** grain boundaries is, in fact, in the range 5-10 nm.  
**Mechanical stresses** may lead to slightly higher activation energy.

Many thanks

for

your attention!