

**Unique opportunities for new insight in the
outer surfaces and interfaces by
High Sensitivity Low Energy Ion Scattering
(HS-LEIS)**

Hide Brongersma

ION-TOF GmbH / Eindhoven University of Technology /
Imperial College (London)

H.H.Brongersma@tue.nl

March 23, 2011
Lehigh University

ION-TOF GmbH
Heisenbergstr. 15
D-48149 Münster / Germany
www.iontof.com

➤ Introduction

➤ Principles and features of LEIS

- 1st atom, In – depth (0 – 10 nm), quantitative

➤ Applications

- Organics (surface modific., anti-wetting, SAMs)
- Catalysts (mixed oxides, coke, NP's, oxid. states)
- Ceramics (SOFC, membranes)
- ALD growth (high-k, inter-diffusion)
- and many more

Why High-Sensitivity LEIS ?

IONTOF

CONTROL at the ATOMIC LEVEL



requires

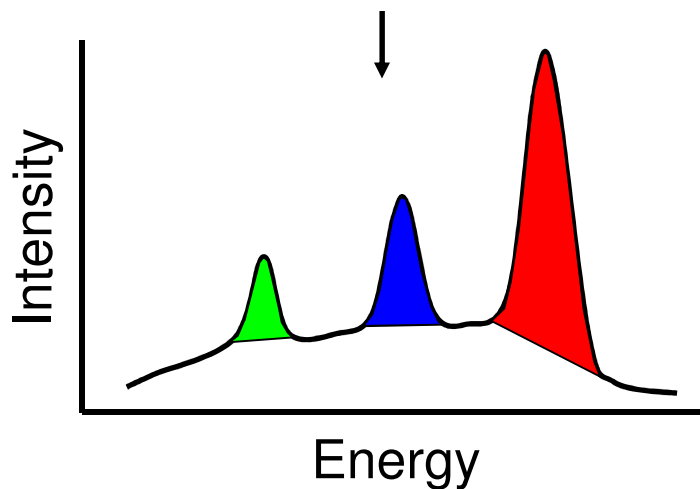
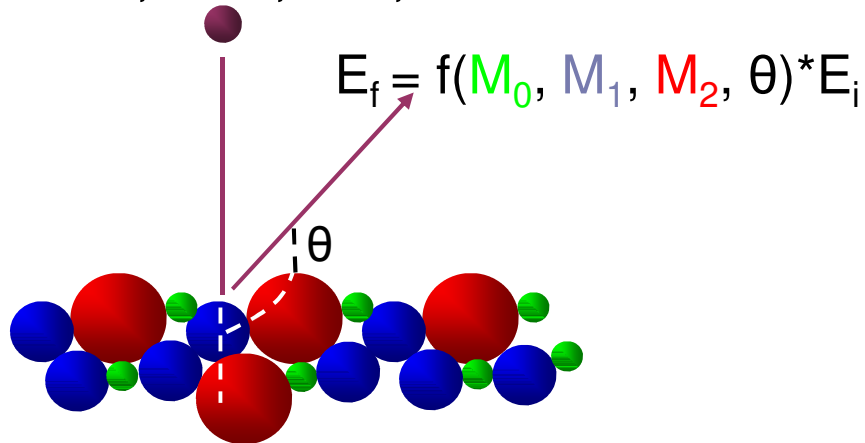
QUANTITATIVE ANALYSIS at the ATOMIC LEVEL

Low Energy Ion Scattering

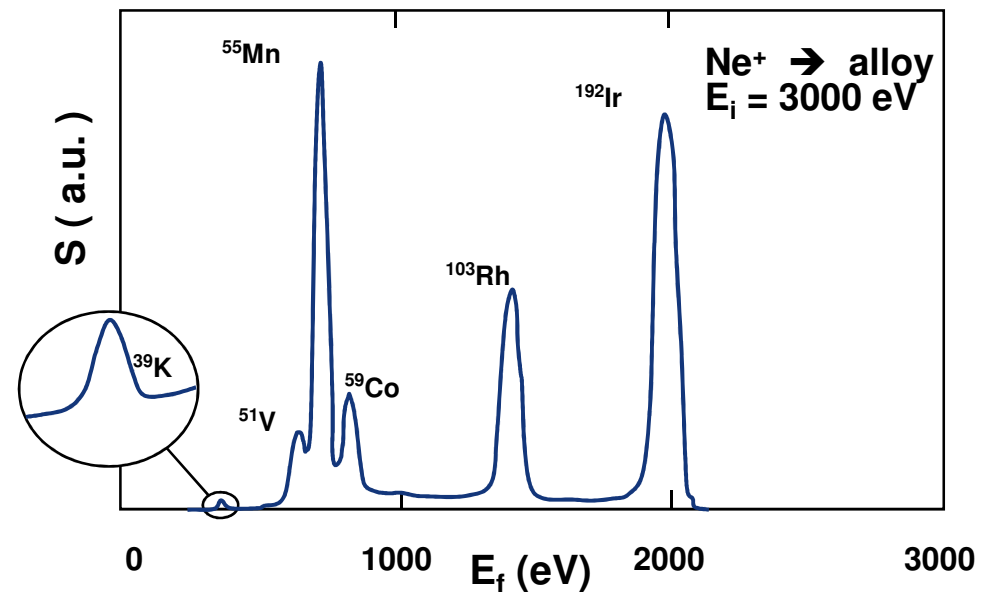
IONTOF

Analytical Capabilities

$^3\text{He}^+$, $^4\text{He}^+$, Ne^+ , Ar^+



- Atomic composition of the outermost atomic layer
- Energy 1 – 8 keV
- Lateral resolution 0.01 – 1 mm
- Static in-depth (0 – 10 nm)
- Quantitative !!
- No matrix effects

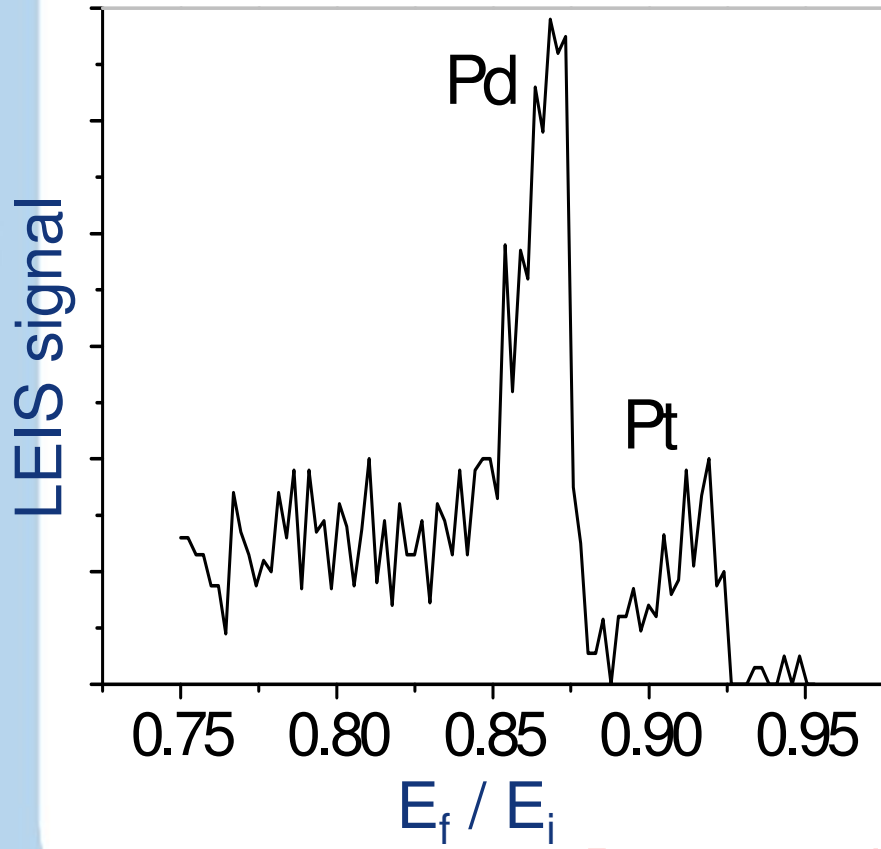


Conventional LEIS vs HS - LEIS

Pd / Pt-C (1000 m²/g)

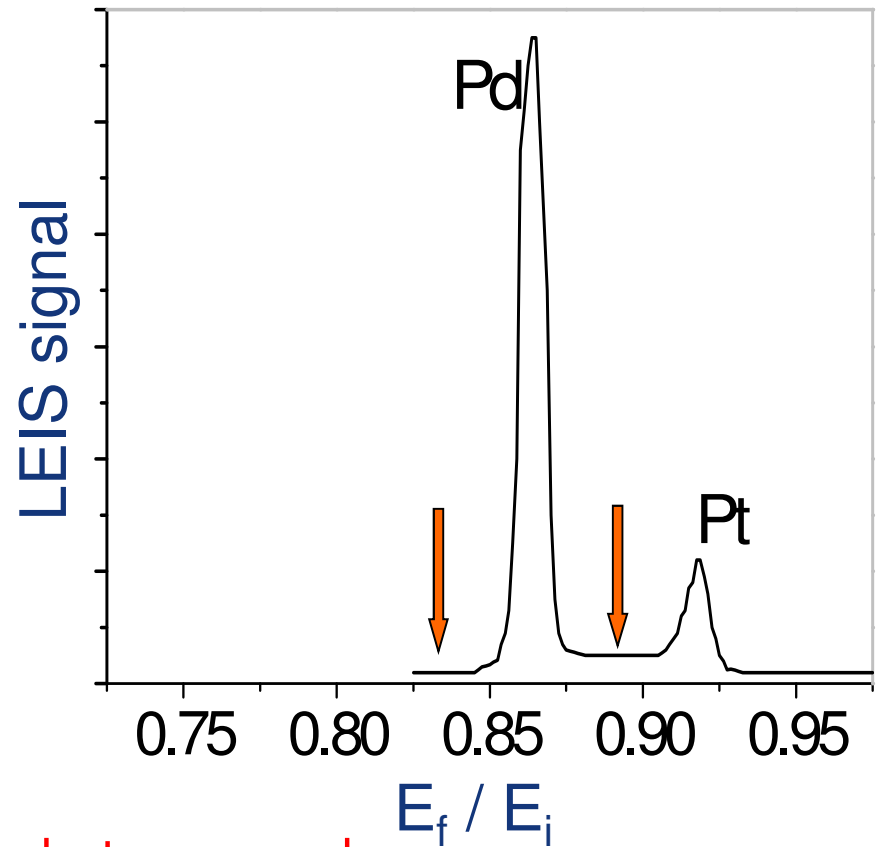
Conventional

⁴He 10,000 nC



High-Sensitivity LEIS

⁴He 5.4 nC



Promoters visible, but removed

Qtac : Unique new Analyser



High – Sensitivity LEIS

Energy image:

- parallel energy detection
- only low dose needed



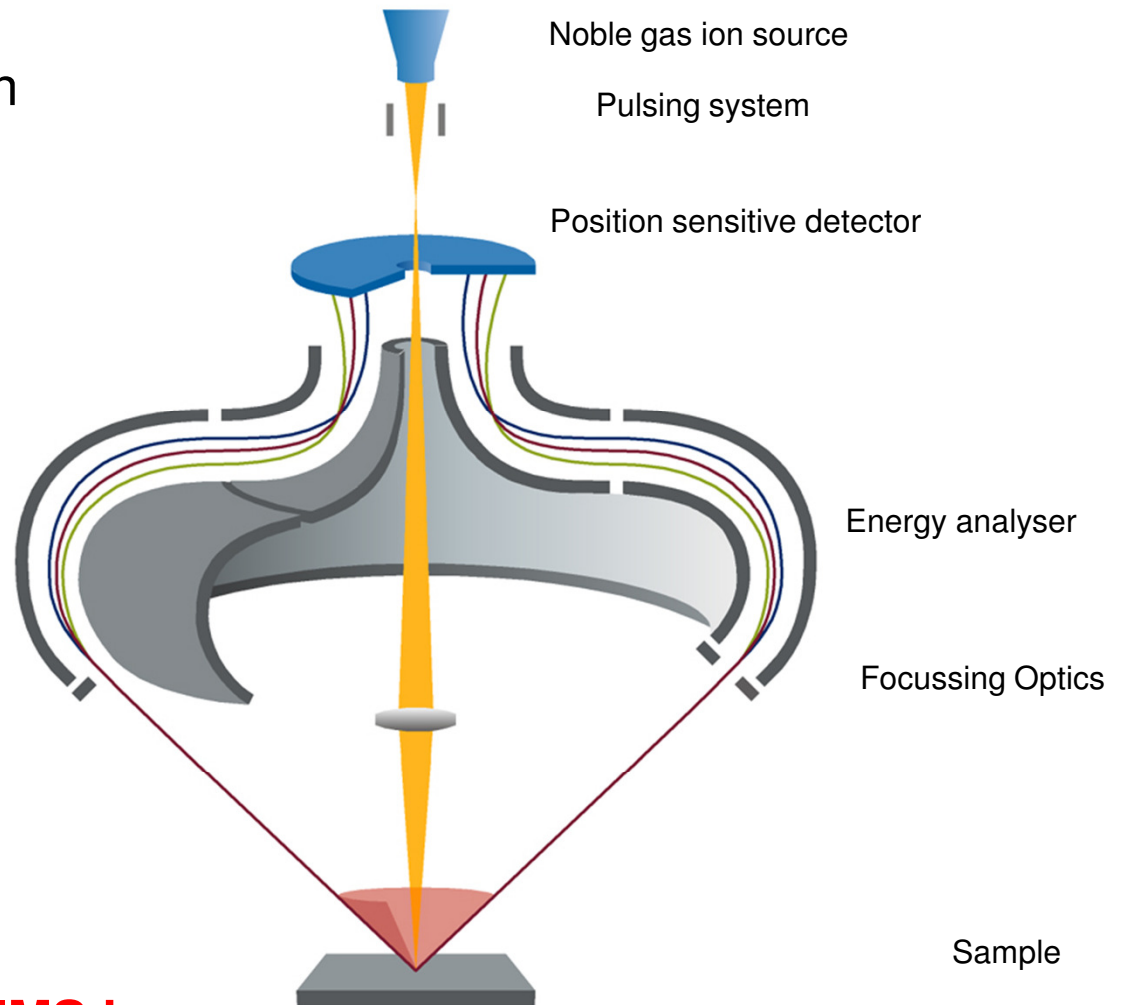
STATIC LEIS

“ Analysis *before* Damage ”

(Molecular Dynamics simulation)

“ Hit same place only once ”

ToF filtering eliminates SIMS ions

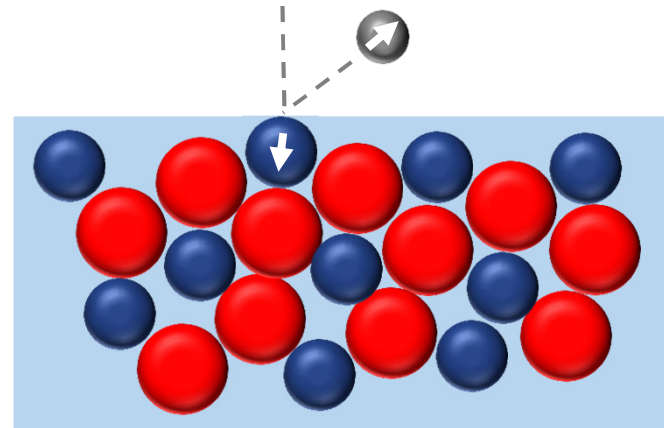
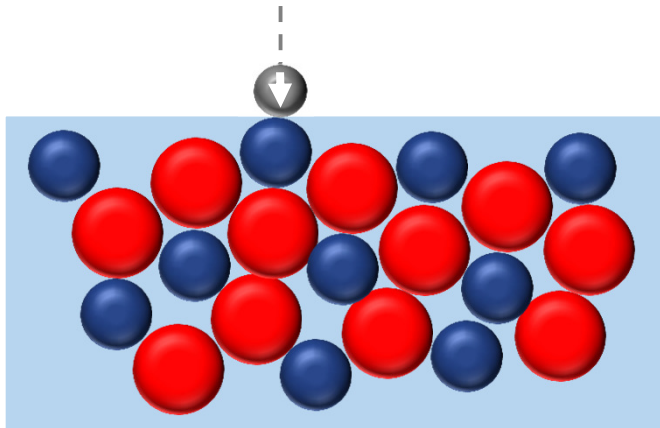


- Analysis *before* damage (static)
 - 1st Monolayer
 - Quantitative (peak \longrightarrow concentration / coverage)
 - Sensitivity
 - Mass resolution
 - ToF-filtered LEIS
 - Imaging
-
- In-depth: Static (0 – 6 nm) *or* with sputtering

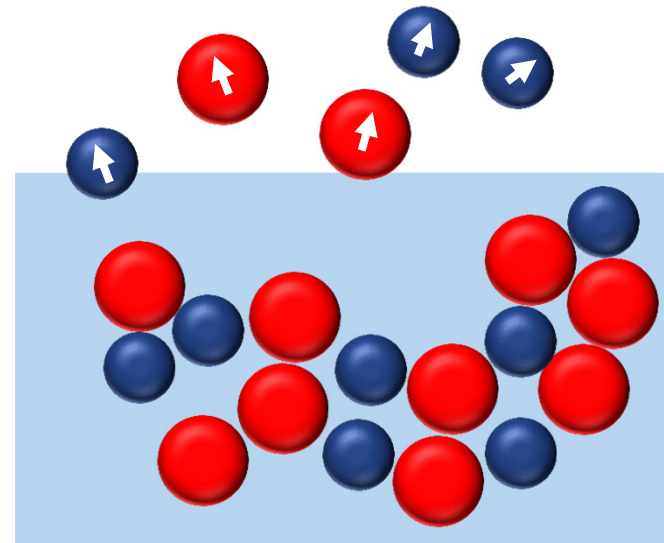
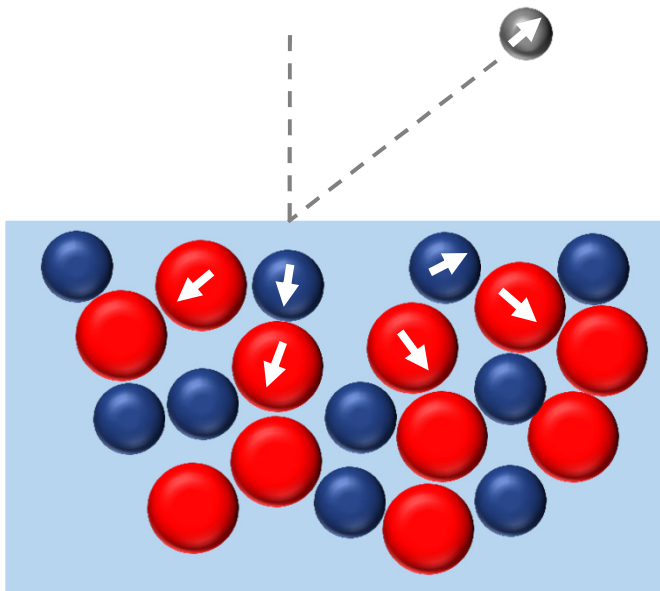
LEIS and SIMS

IONTOF

Time resolved: **LEIS analysis *before* damage !**



LEIS



SIMS

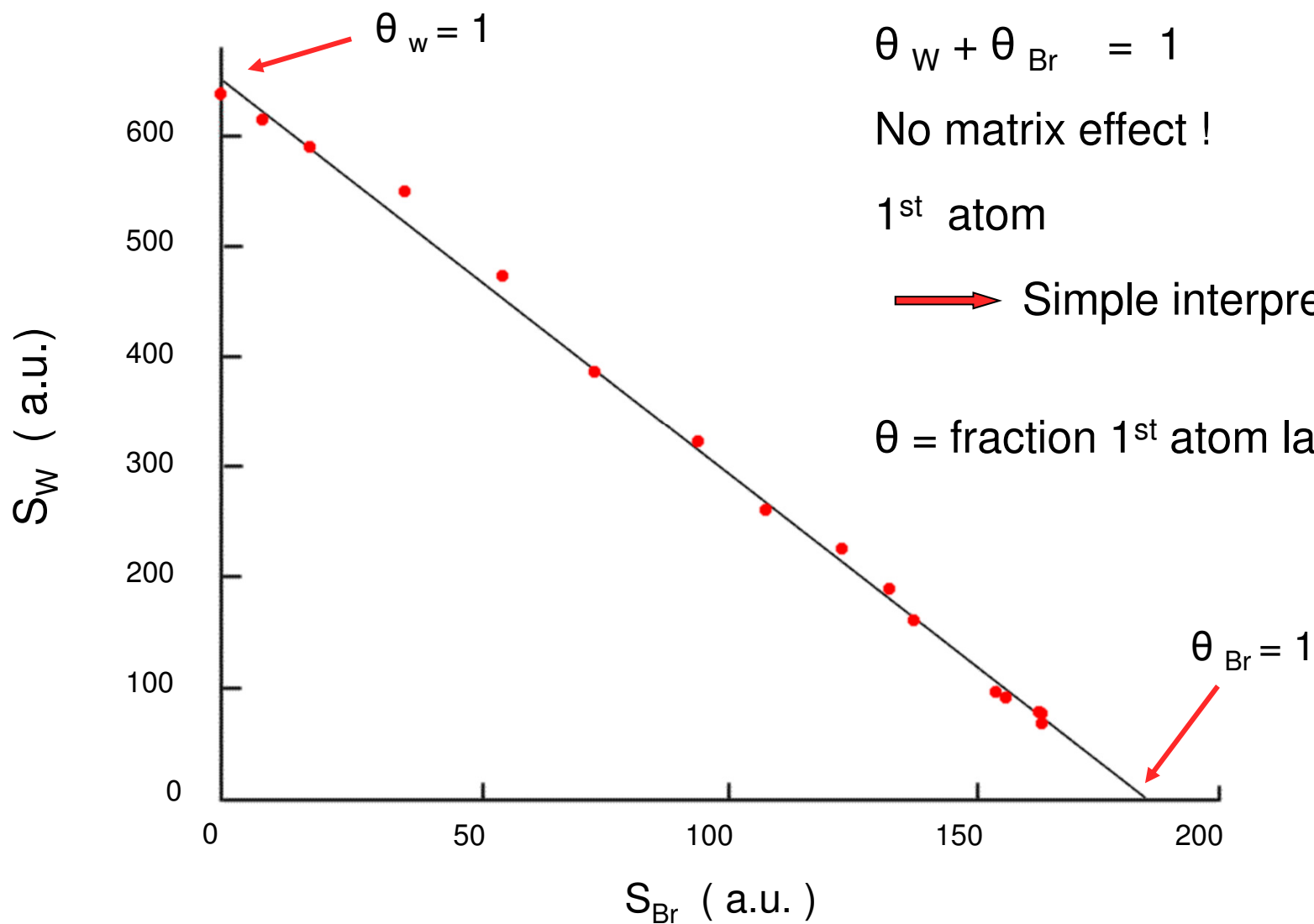
Quantification

Review:

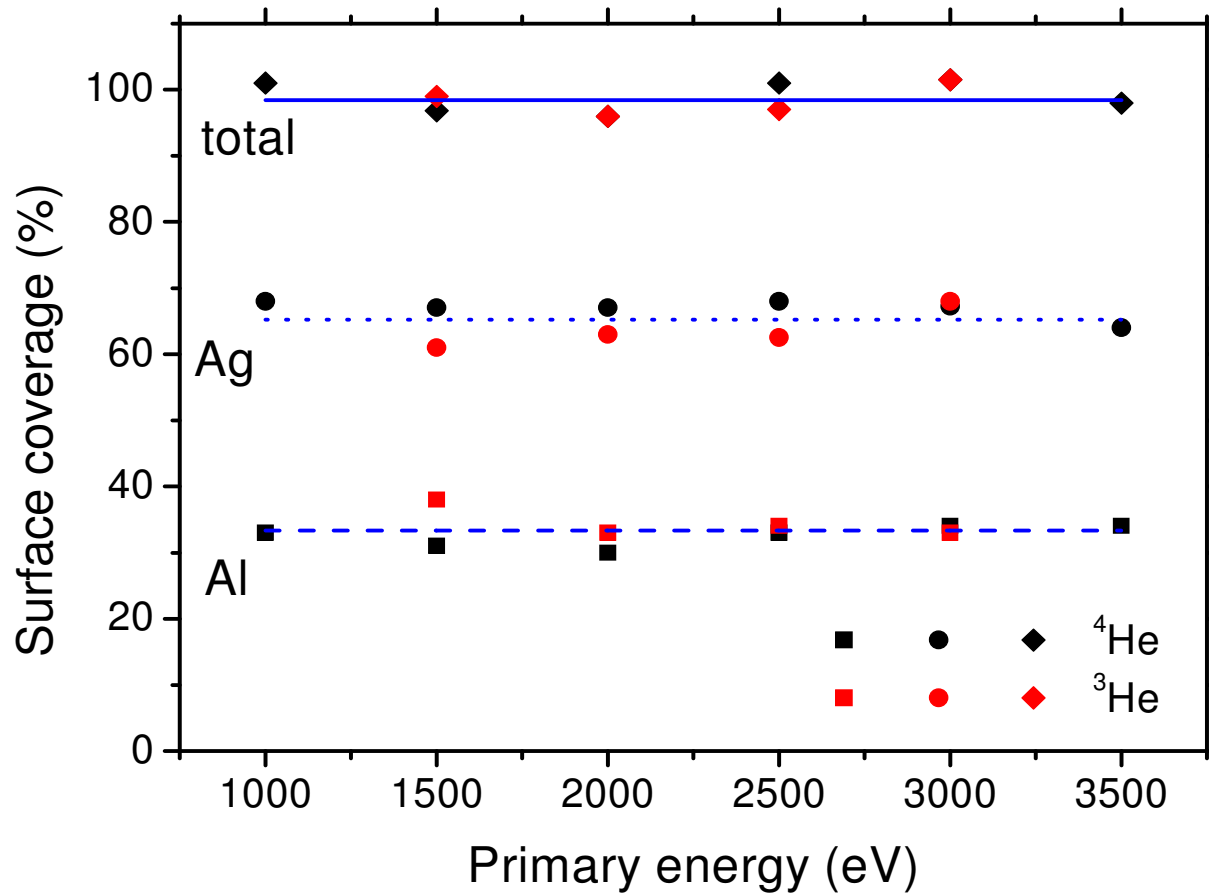
Brongersma et al., Surf. Sci. Reports, 62 (2007) 63 – 109.

Quantification

Bromine adsorption on Tungsten



Bulk Composition $\text{Ag}_{80}\text{Al}_{20}$



Surface Composition



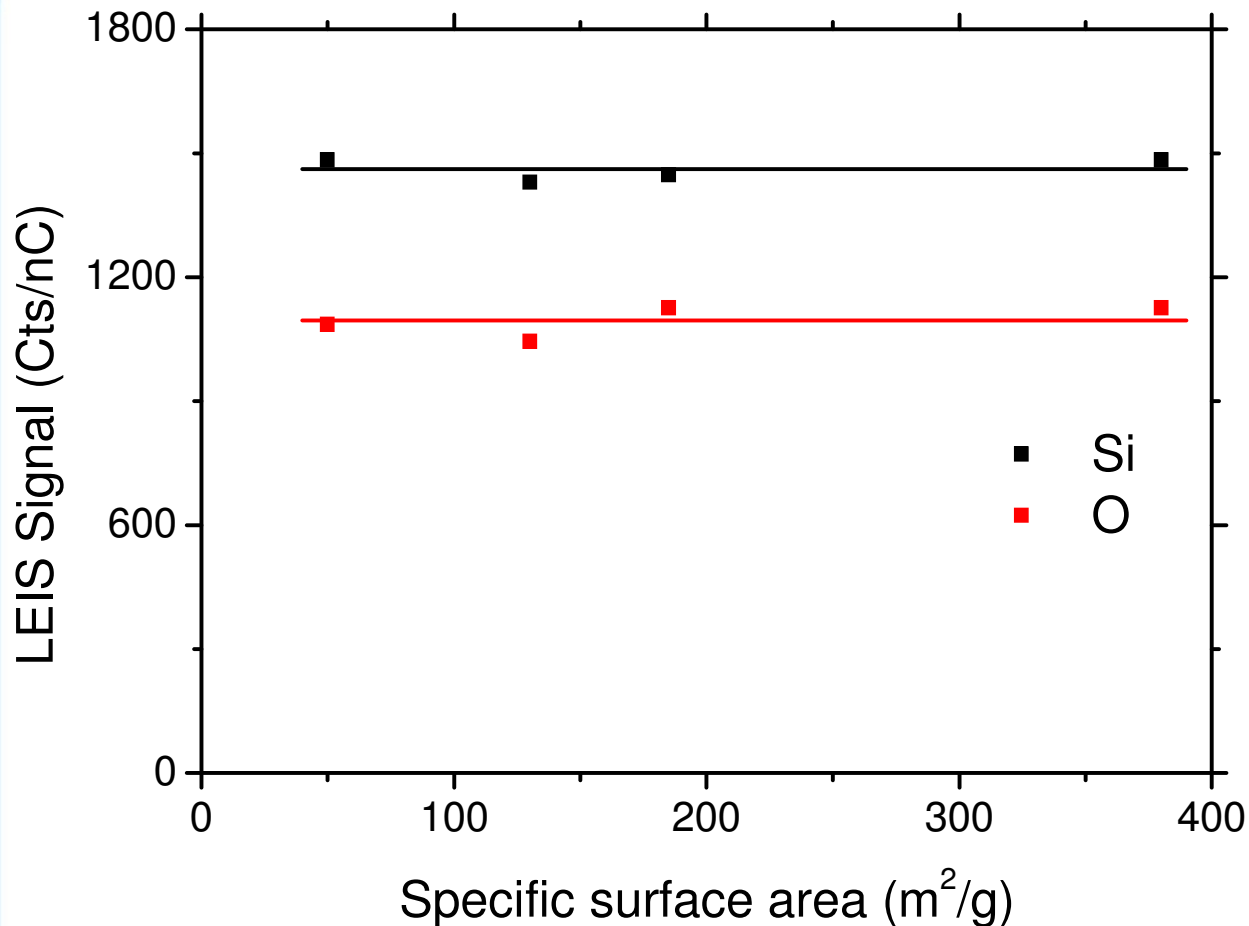
(independent of primary energy)



NO matrix effects

Rough silica: 50 – 380 m²/g

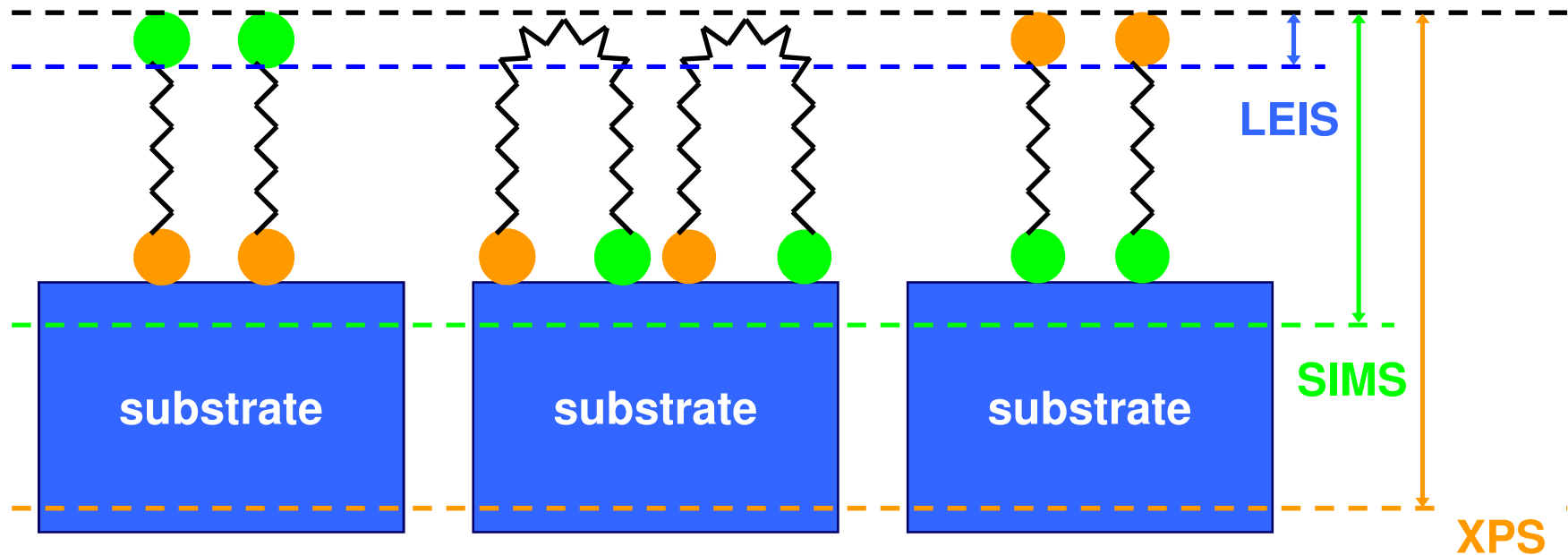
HS-LEIS: Insensitive to roughness



LEIS Signals:
rough silica about 77%
of flat silica (quartz)

Monolayer sensitivity

IONTOF



LEIS 1st atom and in-depth; quantitative, sensitive

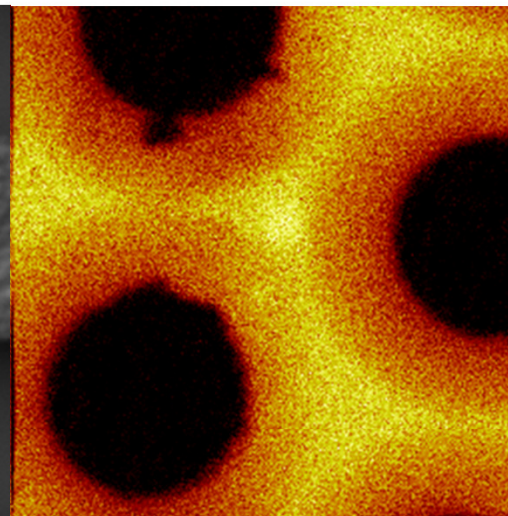
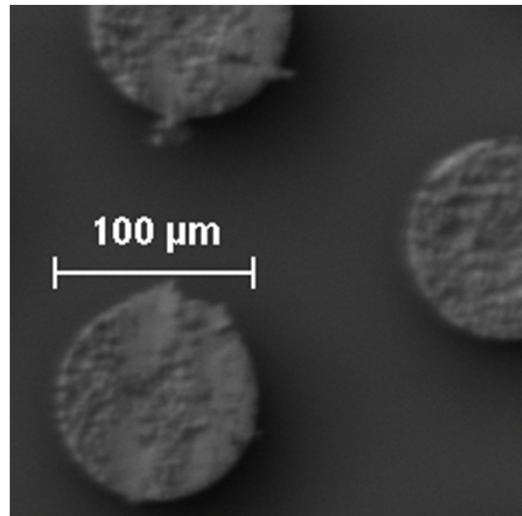
SIMS not quantitative for near-surface / interface

XPS average over 3 – 10 nm; chemical info

Elemental mapping by LEIS and SE Image

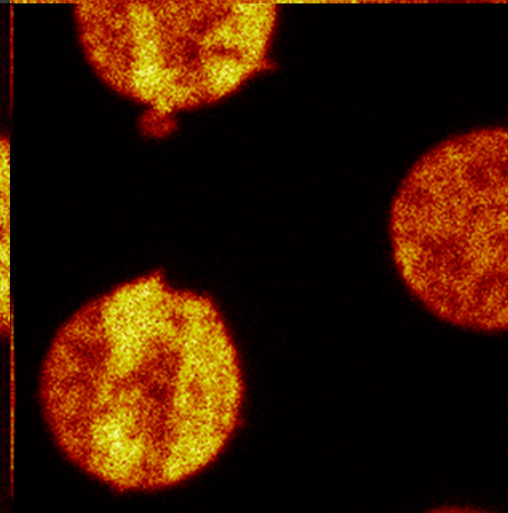
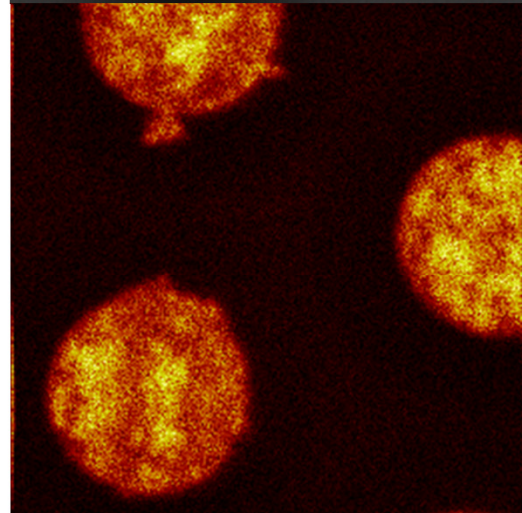
Solder bumps

SE image



Ti

Sn



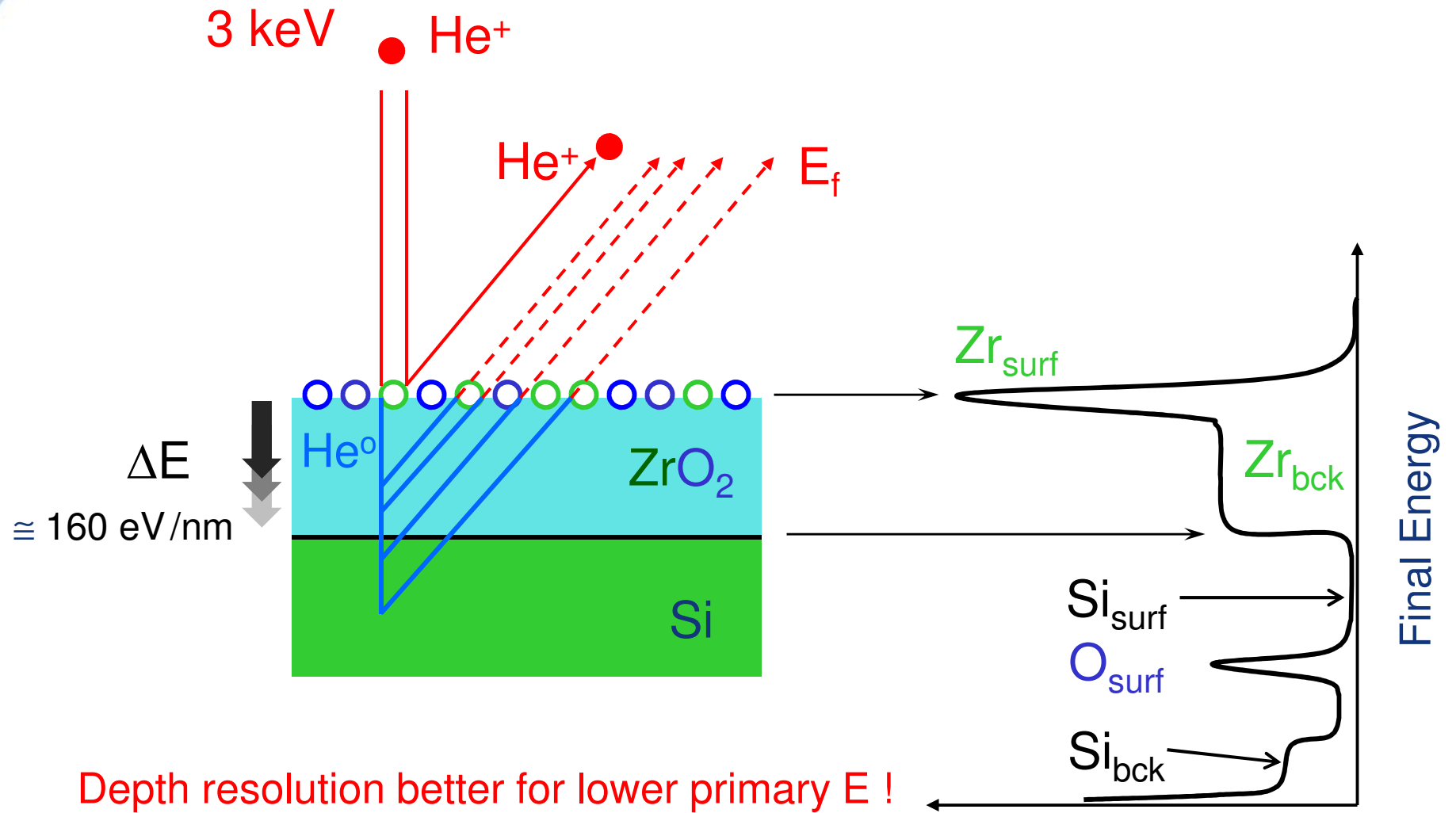
Pb

Depth info for
Ultra thin layers and interfaces

Two possibilities:

1. Static LEIS + sputter depth profiling with dual ion beam
(advantage of quantification, depth resolution LEIS)
2. Static LEIS
(analogous to RBS and MEIS, but better depth resolution)

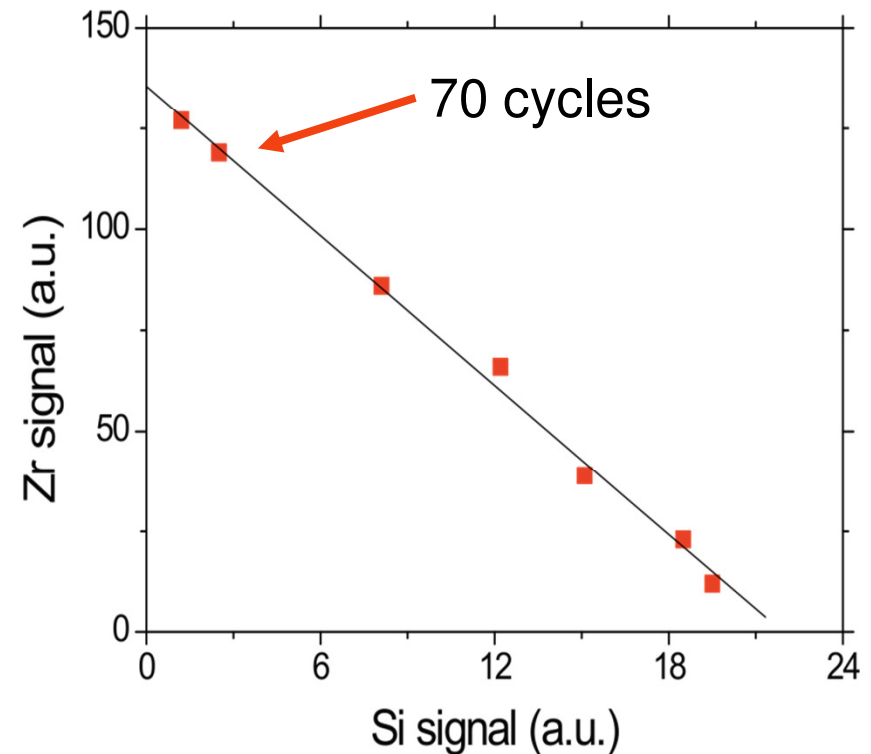
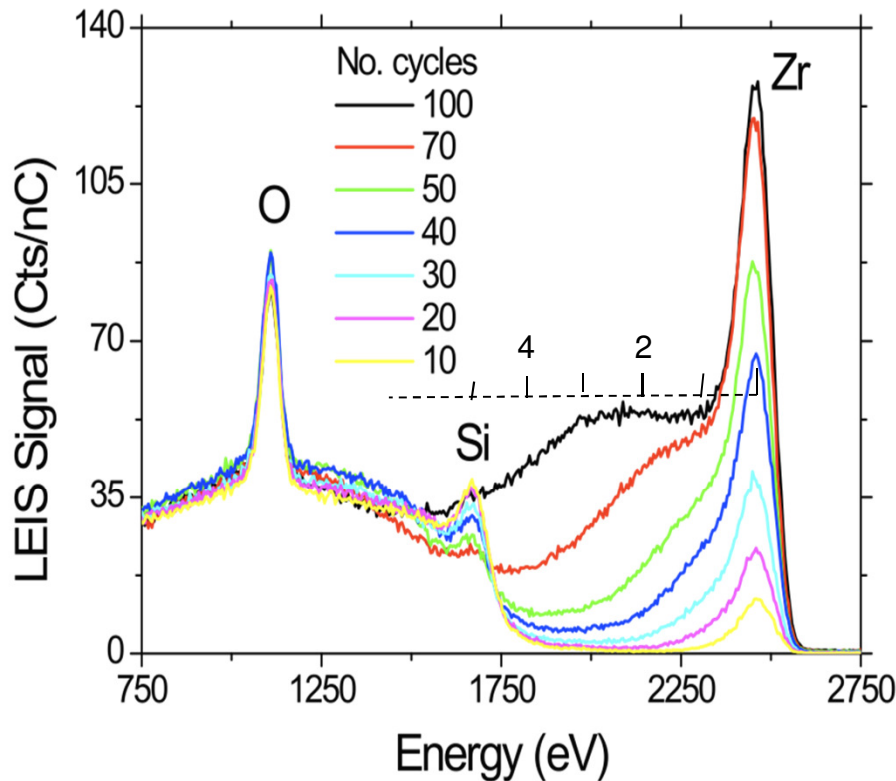
Depth info



1st atom and Static Depth Profile

ZrO₂ Atomic Layer Deposition on Silicon


- Closure / quantification pinholes (still present after 70 cycles)
- Thickness distribution ZrO₂ layer (160 eV/nm)
- No matrix effect
- Example: calibration / quantification for a 2 component system



In LEIS only backscattered *ions* are detected

Peaks: Ions backscattered from 1st atom.
(*one well-defined collision*)

Tails: Backscattering in deeper layers + reionization
(Scattering by oxygen atoms: efficient reionization)

 **Shape:** f (in-depth distribution Zr)

Intensity: f (oxygen concentration in
1st atomic layer)

LEIS Technique

IONTOF

Features of Low Energy Ion Scattering (LEIS)

LEIS Features

Ultra-high surface sensitivity,
top atomic layer analysis

Static depth profiling information (up to 10 nm)

Reliable and straight-forward ***quantification***

Detection of all elements ***> He***

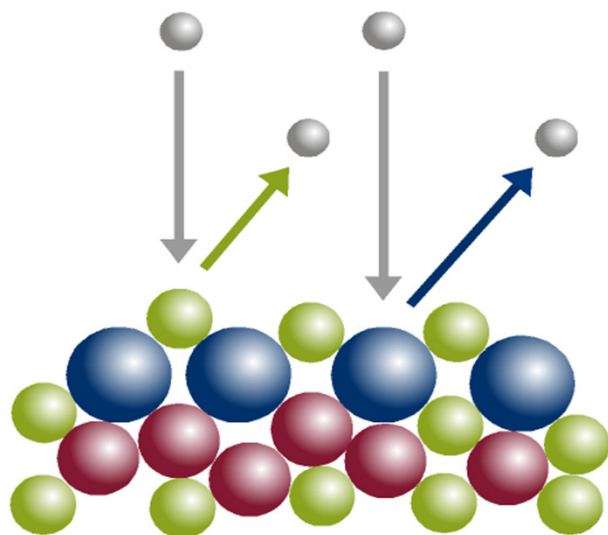
Detection limits:

Li - O ≥ 1 % of 1 ML

F - Cl 1 % - 0.05 % of 1 ML

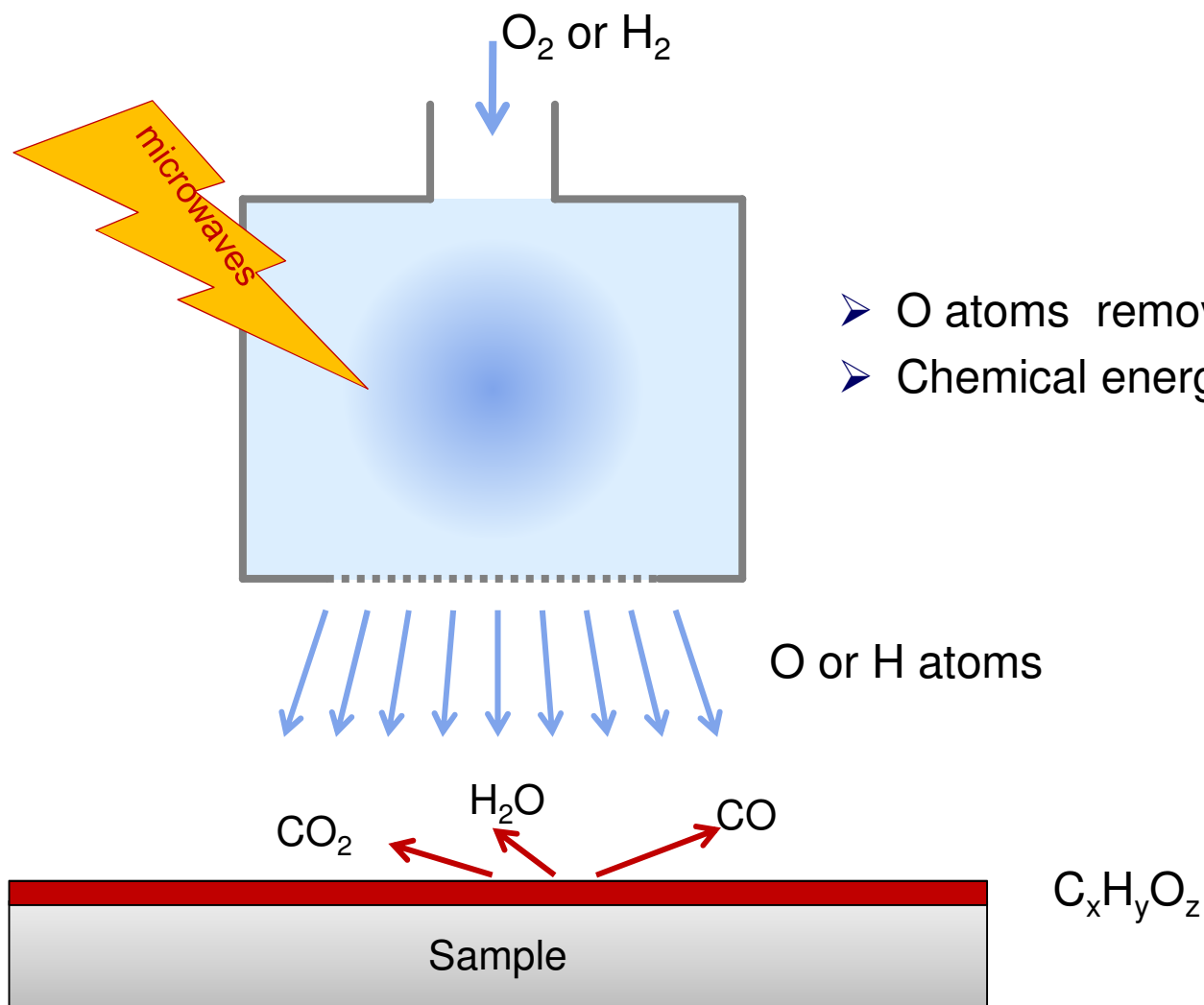
K - U 500 ppm - 10 ppm of 1 ML

He⁺, Ne⁺, Ar⁺, Kr⁺
1 - 8 keV



Sample Treatment

Atom Source for Surface Cleaning



- O atoms remove organics, coke
- Chemical energy: no sputtering

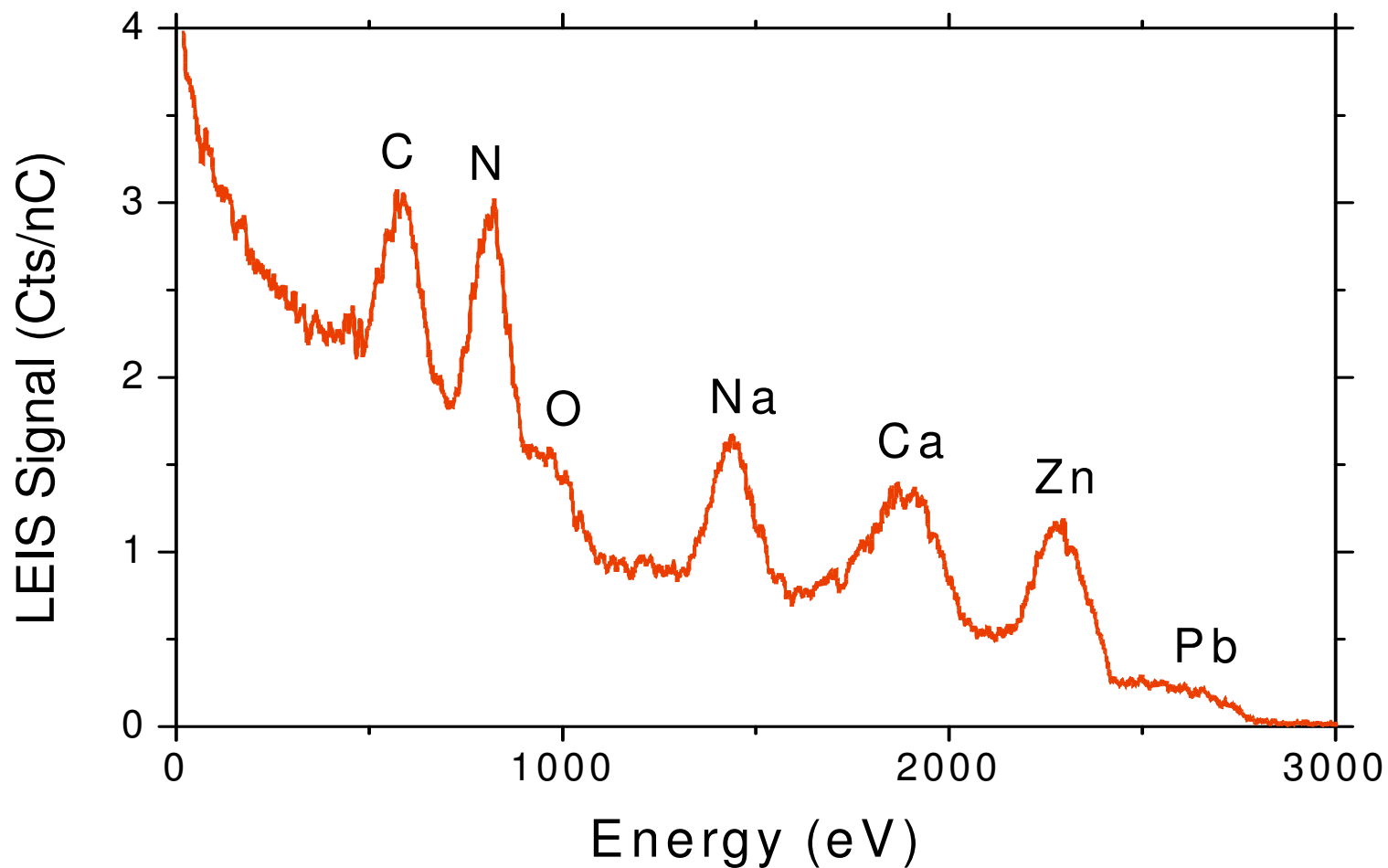
Organics

- Surface segregation
 - Dendrimers
- Antiwetting
- Surface modification
- Metal / polymer interface
- SAMs

- Inter - molecular segregation
 - Segregation impurities, additives (0.1 s - days - ..; up to 10^8 x !!)
- Intra - molecular segregation (0.1 s - days - ..)
 - Aging plasma oxidized PE
- Anti - wetting layers
- Metal diffusion in polymers
- SAM's

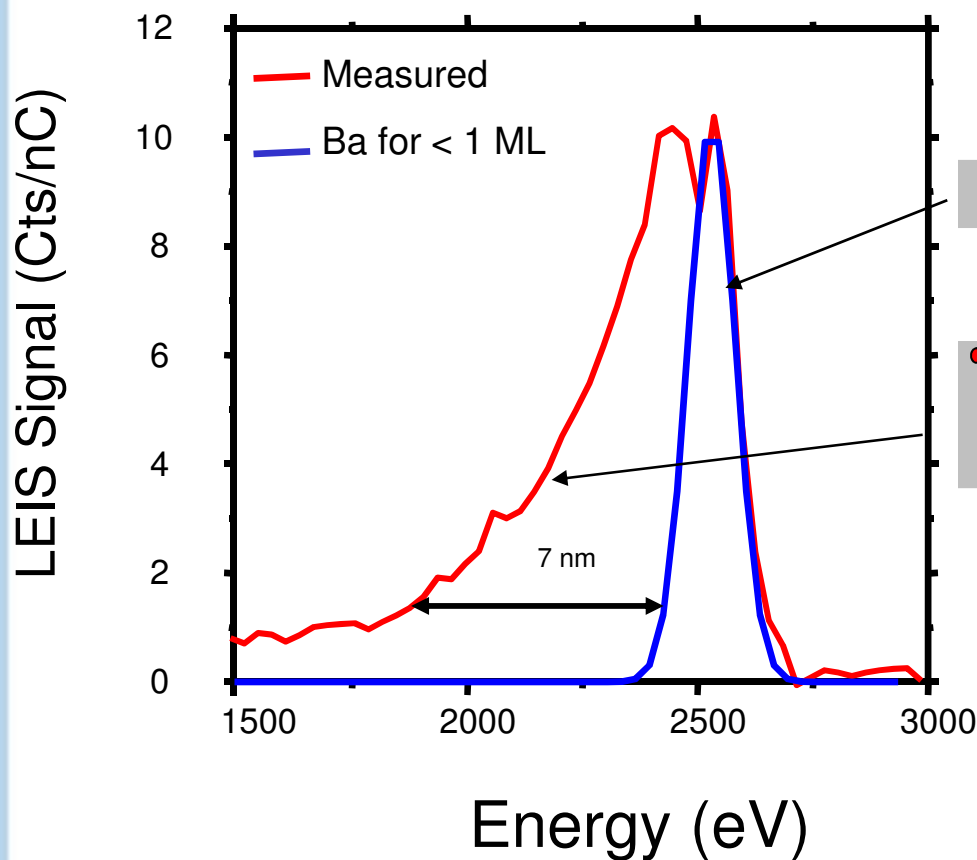
Acrylonitrile-Butadiene-Styrene (ABS)

Surface segregation of additives



Metal - polymer interface in ultra - thin layers

PLED: Ba evaporation on PPV



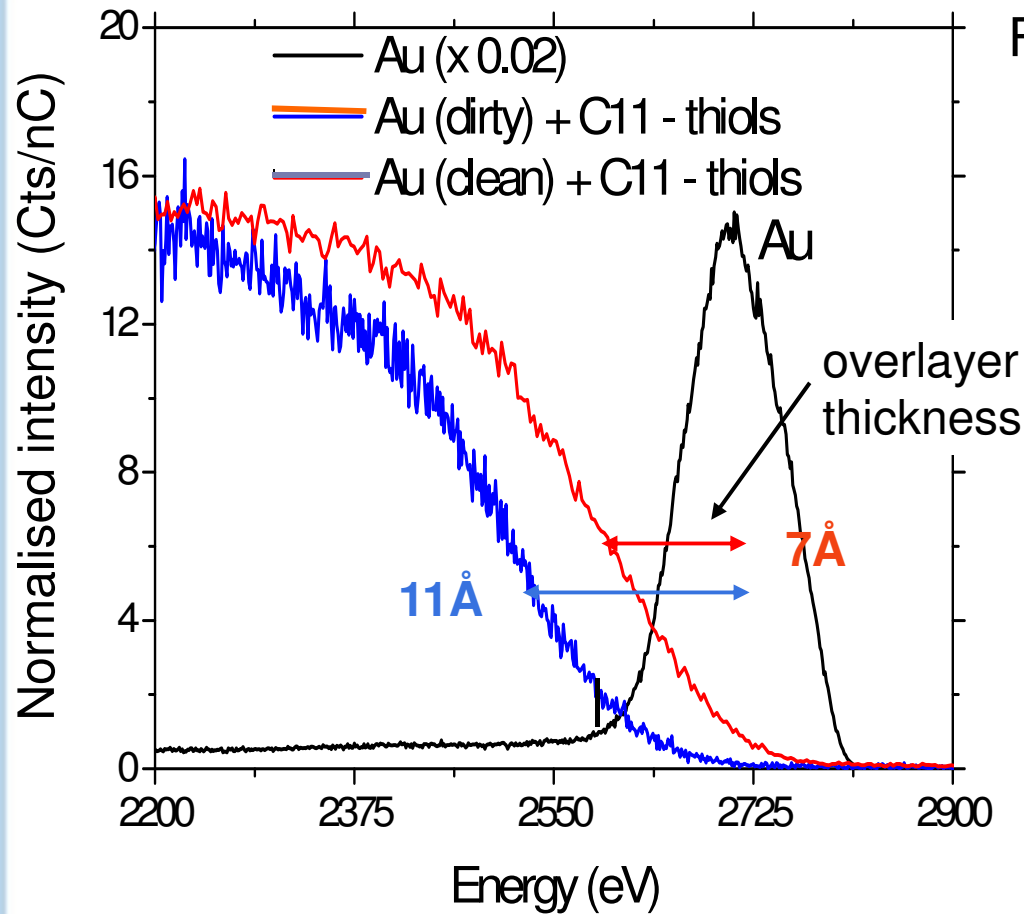
During evaporation of barium on PPV, most of the Ba diffuses into the PPV.

Compare the peakshape of a sub-monolayer of Ba (blue) with the actual peak (red)

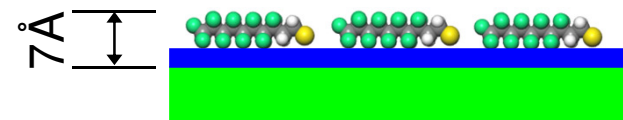
Peak shape ↔ depth distribution

PLED: higher light output for narrow depth distribution

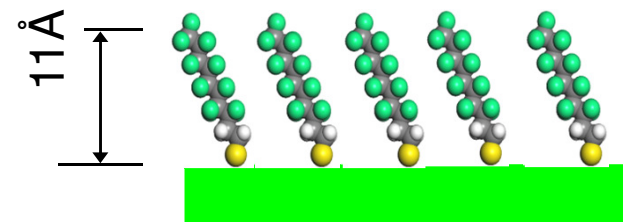
High-energy edge of SAMs on Au



Fluorinated thiol on **dirty** Au surface

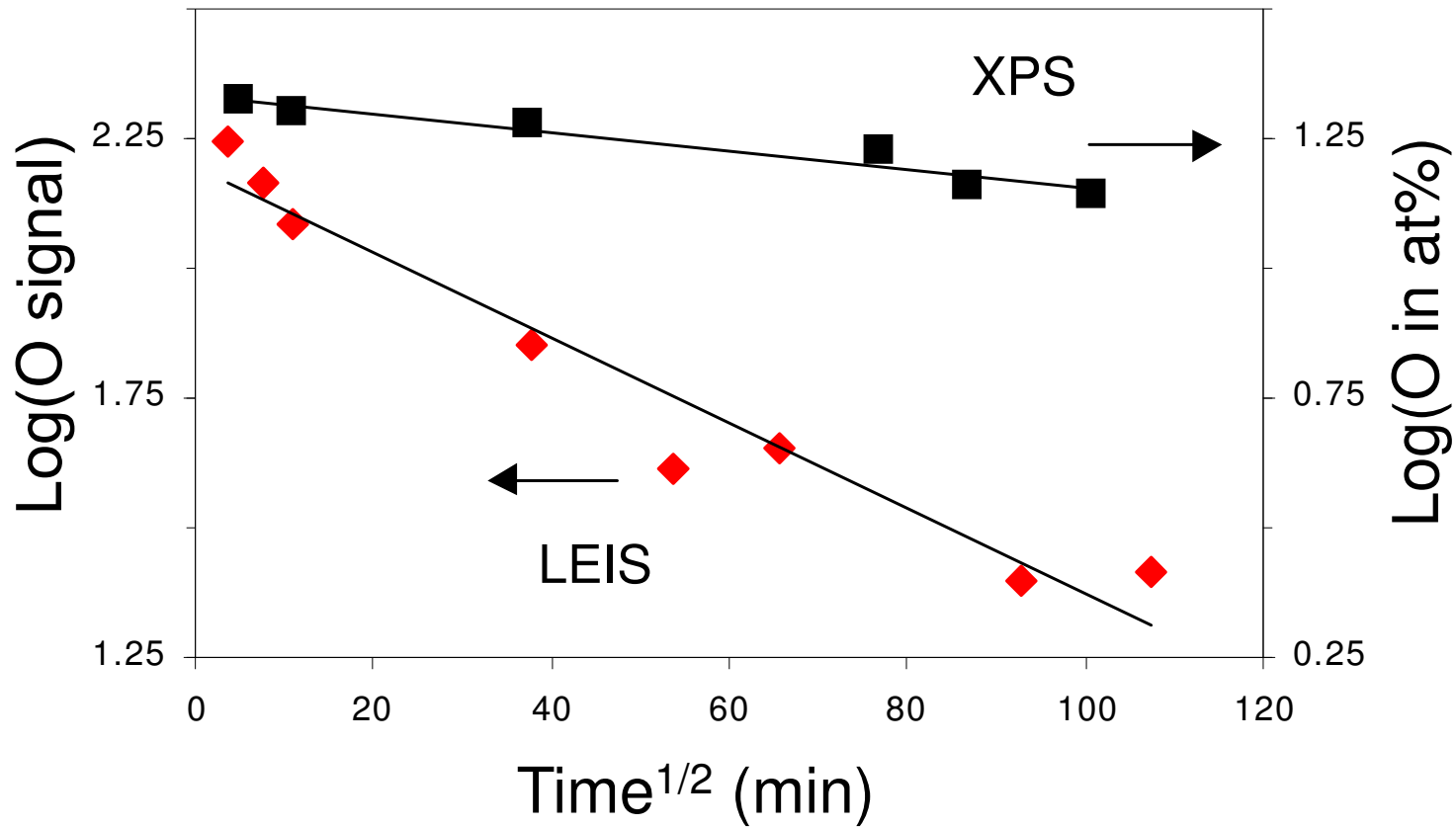


Fluorinated thiol on **clean** Au surface



Aging of plasma oxidized HDPE

- Aging (LEIS) faster than aging (XPS) !
- “Straight line” → diffusion process



Selection of examples:

- Pt/Au
 - Mixed oxides
- γ -alumina
 - Poison (Coke on TWC)
 - Poison (oxygen membranes, SOFC)
 - Use of probe molecules
 - NP's , core/shell
 - Oxidation state 1st atom

Important / unique applications for catalysis

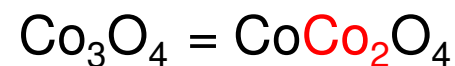
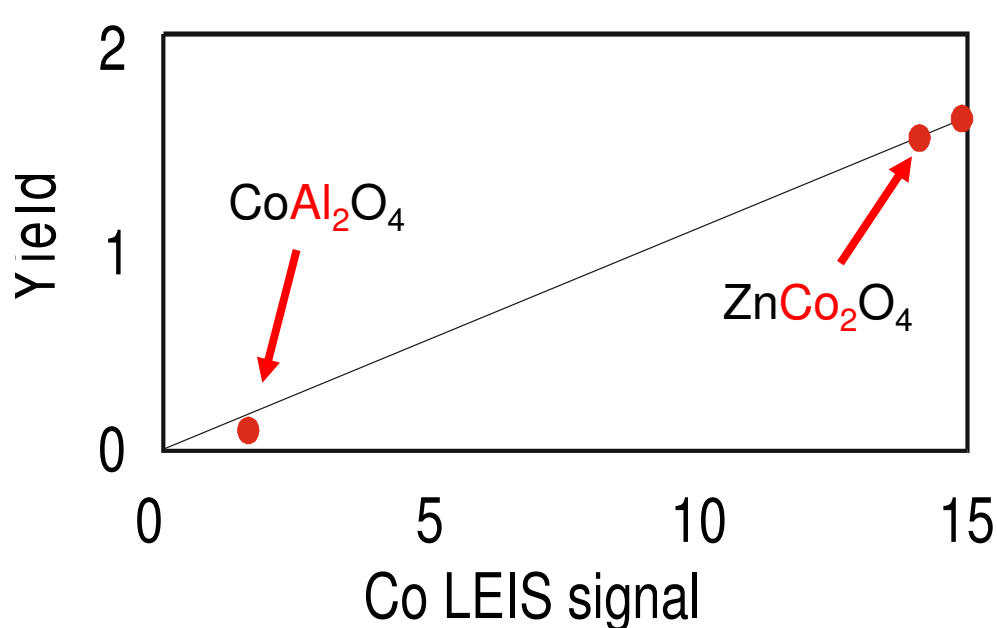
IONTOF

Mixed oxides and catalysis

The atomic composition of the 1st atom layer controls catalysis.

In a spinel (AB_2O_4) only the B-cations (octahedral site) are **catalytically active and visible** for LEIS (1st at.).

The A-cations (tetrahedral sites) are in 2nd layer (not active, no LEIS peak).



Test reaction:

only Co catalytically active

Co signals:

XPS: 1 : 2 : 3

LEIS: 0.3 : 1.9 : 2.0

LEIS



Catalysis



XPS

Importance of the outer surface

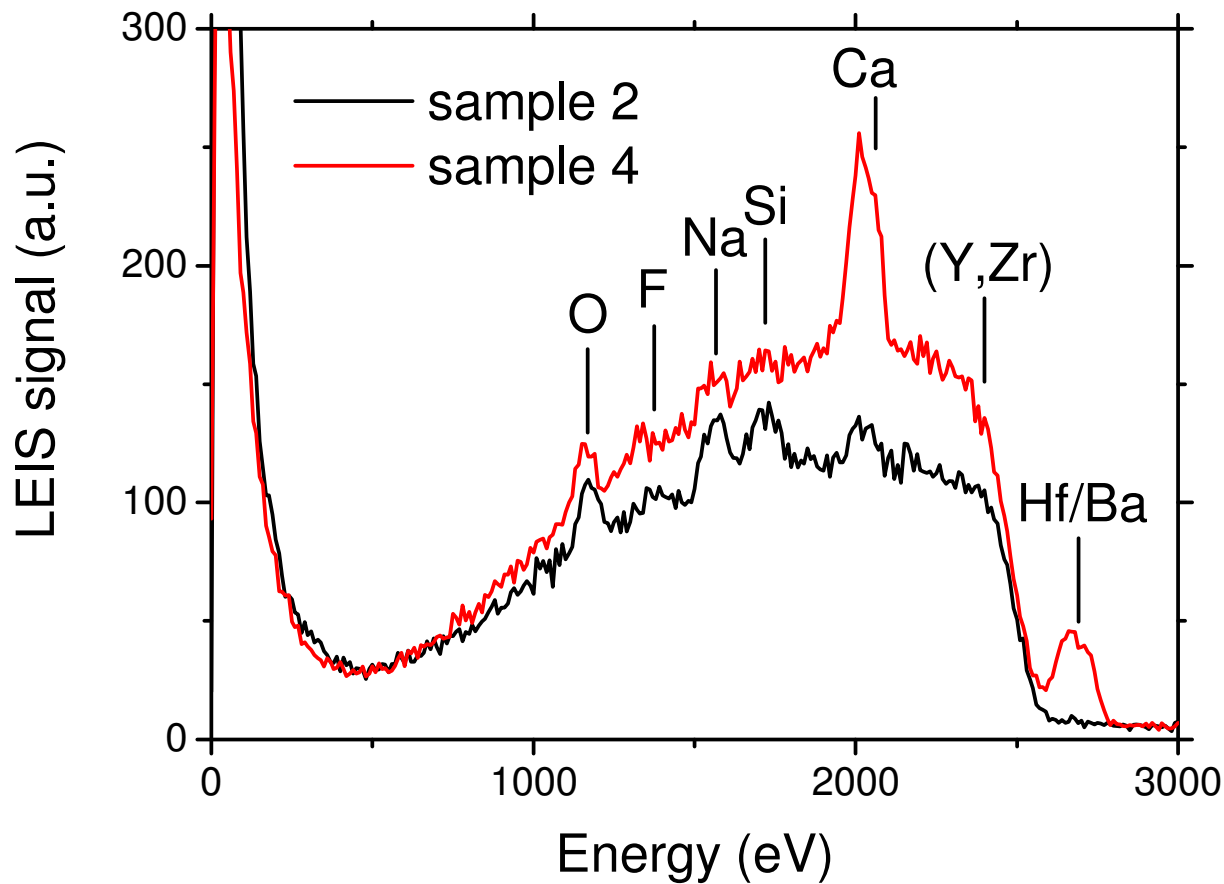
- Performance relies on oxygen transport
- Performance: “ Hampered by the surface ”
- Why ? What is the surface ??

M. de Ridder et al., J. Appl. Phys. 92 (2002) 3056 - 3064

M. de Ridder et al., Solid State Ionics 156 (2003) 255 - 262

Fuel Cells

Ytria stabilized Zirconia (YSZ) after calcination



Calcination for 5 hours at 1000°C in an oxygen flow of 1.5 bar.



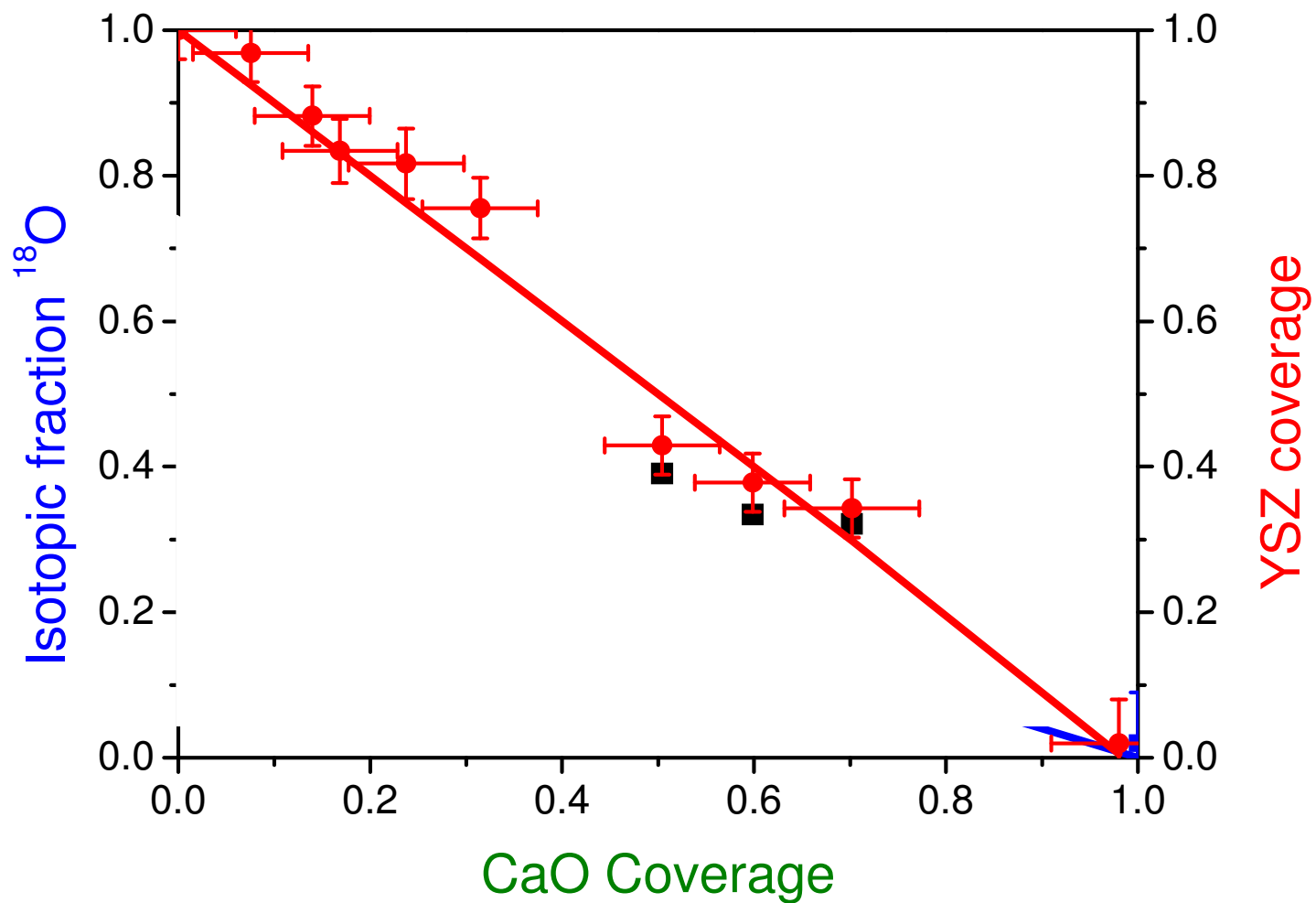
Segregation of monolayer of impurities

For $T > 700$ C: No Y, Zr in 1st atom !

XPS: Ca not visible (\leftrightarrow Zr)

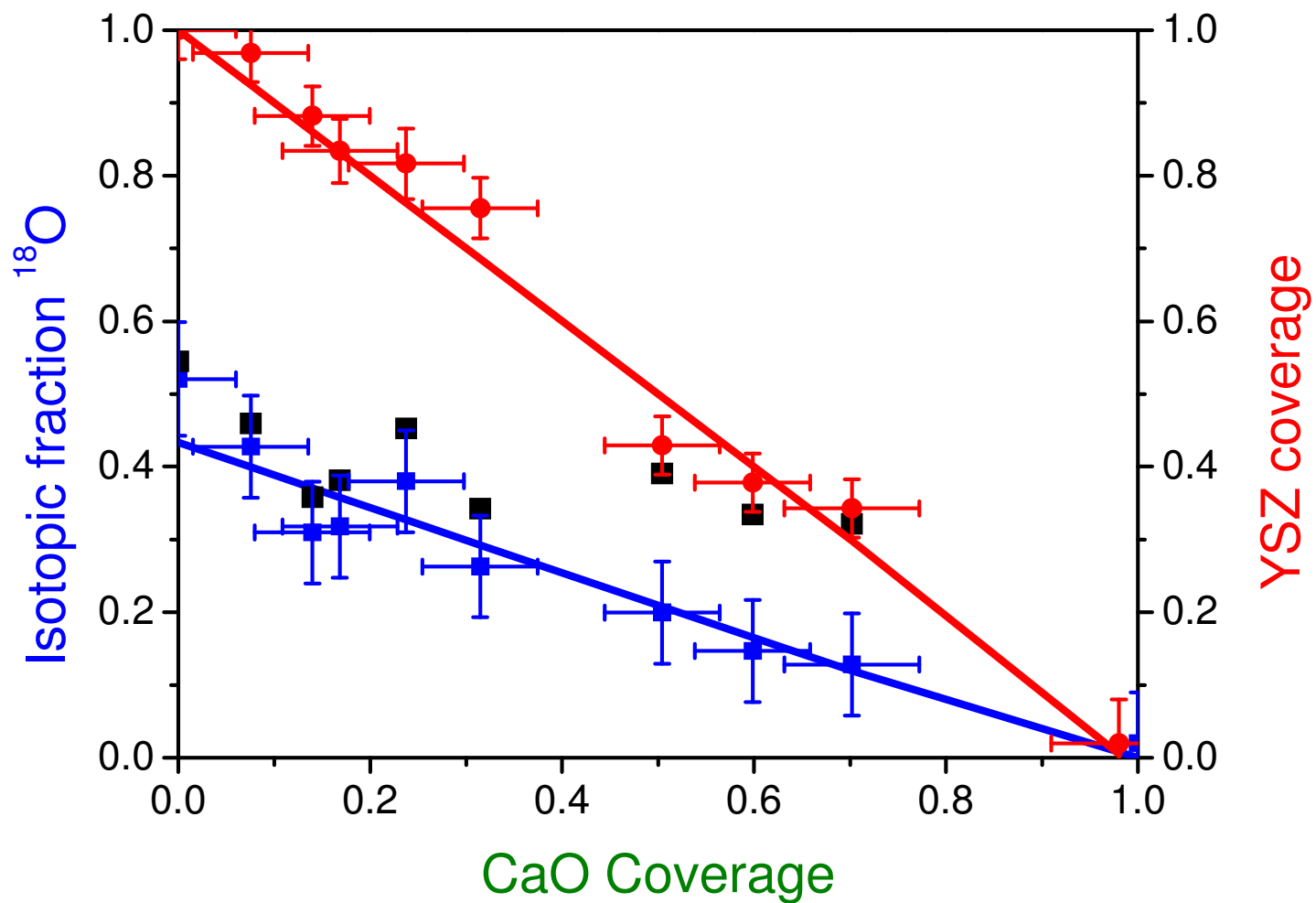
Fuel Cells

CaO coverage blocks $^{16}\text{O} - ^{18}\text{O}$ exchange



Fuel Cells

CaO coverage blocks ^{16}O – ^{18}O exchange




Coke Formation on Commercial TWC

IONTOF

Three Way Catalyst (TWC) (Pt, Rh / CeO₂ / γ - Al₂O₃)

Cold start: 50% loss of Pt signal — *sintering or coke formation ?*

Room temperature oxidation with atomic oxygen gives complete recovery of Pt signal  *loss is due to coke.*

Detection of C with “any” surface technique.

But: WHERE is the coke ??!

LEIS determines which fraction of Pt is covered by coke !

Applications:

➤ Number of Pt atoms available for catalysis.

Quality control of catalysts !

➤ Detection of nucleation site for coke (active phase, support, binder, ...)

Diameter \longleftrightarrow TON; size often related to failure

TEM:

- excellent catalyst characterisation
- detailed info, but local
- contrast required (high Z cluster on low Z support)


Chemisorption:

- requires known probe / surface interaction

HS - LEIS:

- new technique; any material; clusters: 1 atom - 10 nm

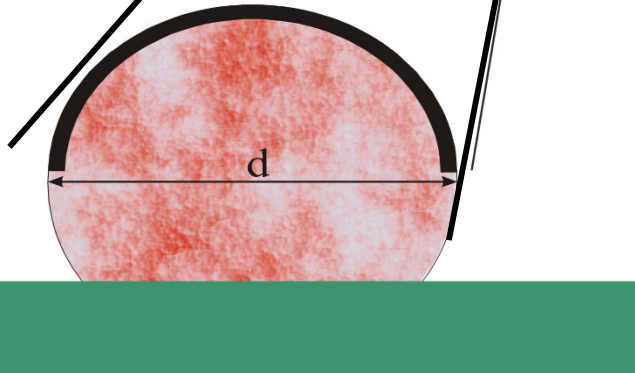
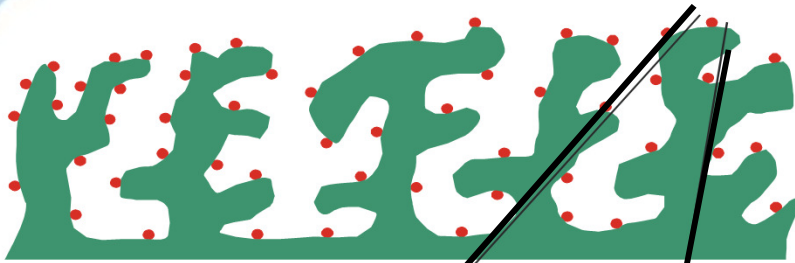
Comparison:  Richard A. P. Smith (J&M), ECASIA 2009

 T. Tanabe et al. (Toyota), Appl. Catal. A370 (2009) 108

Important / unique applications for catalysis

IONTOF

4. Nanoclusters



- Average diameter nanoclusters
- Surface segregation in alloy clusters
- Core/shell particles
(verification, closure, thickness shell)

Example: Three-Way catalyst (exhaust)

Pt clusters on CeO_2 // γ -alumina

Loading = 0.004 g Pt / γ -alumina

Cluster diameter: 1.6 nm (average)

Accurate for $d < 10$ nm

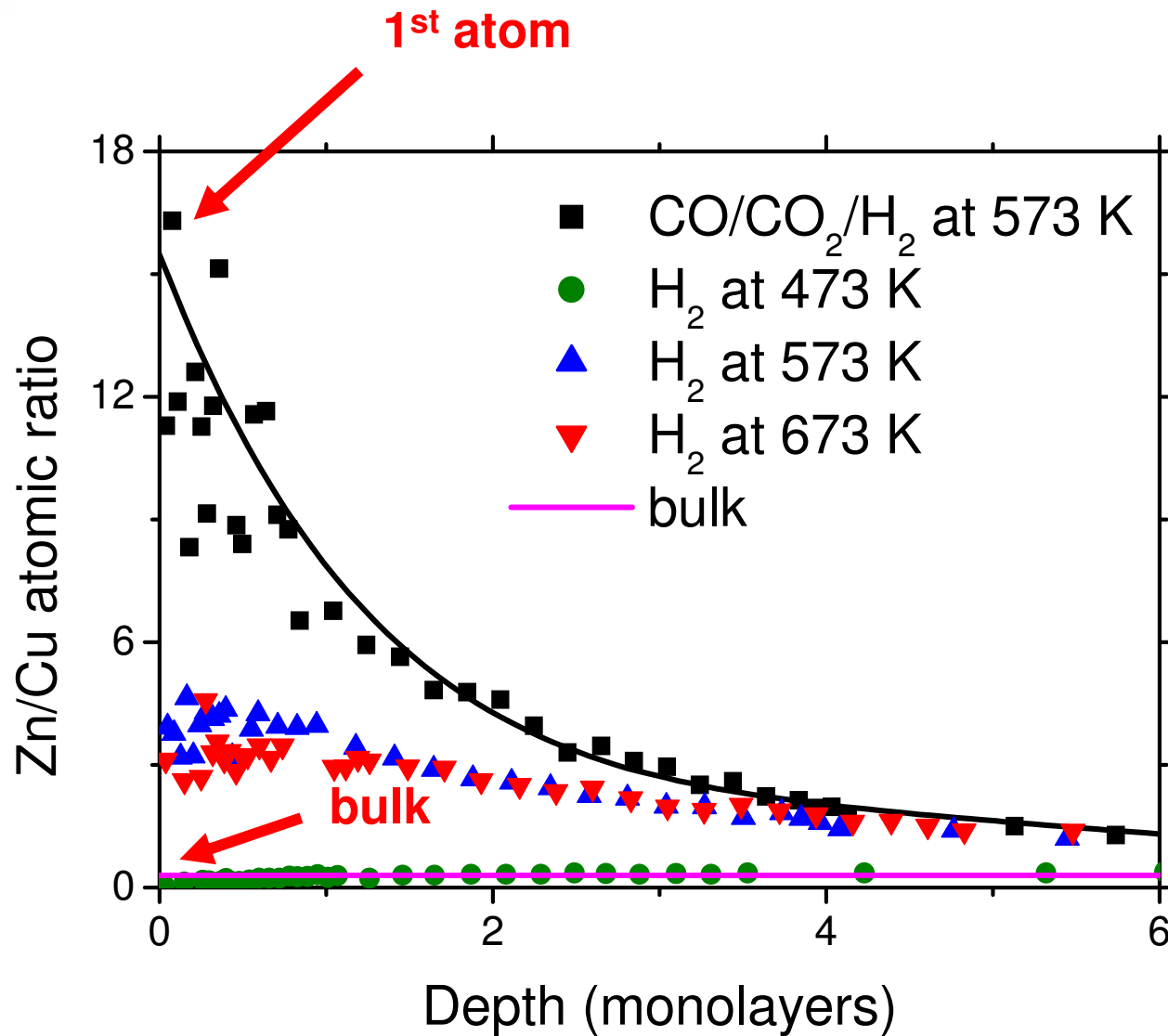
The diameter is derived from the ratio of the bulk loading (volume) to the LEIS signal (surface area)

This method is possible where TEM fails ($d \leq 2$ nm; high Z support)

Cu / ZnO / SiO₂ Catalysts

IONTOF

Synthesis of Methanol, Fatty Acids



Strong Zn(O)
segregation

Zn(O) on top of Cu is
thermodynamically
favorable

Oxidation states Cu and Zn in outer surface ?

IONTOF

LEIS + chemical titration !

XPS:

Oxidation states, **BUT** averaged over 10 – 20 atomic layers.

LEIS:

Elemental composition outer atomic layer, **BUT** no chemical info

Oxidation of metallic Cu, Zn gives shielding by oxygen.

Signal decrease: factor 5 resp. 3.7.

Chemical titration:

Information on oxidation states, **BUT** not only the outer surface (?)



???

Oxidation states Cu and Zn in outer surface ?

IONTOF

LEIS + chemical titration !

XPS:

Oxidation states, **BUT** averaged over 10 – 20 atomic layers.

LEIS:


Elemental composition outer atomic layer, **BUT** no chemical info

Oxidation of metallic Cu, Zn gives shielding by oxygen.

Signal decrease: factor 5 resp. 3.7.

Chemical titration:

Information on oxidation states, **BUT** not only the outer surface (?)

LEIS + Chemical titration:  oxidation states in the outer surface !

➤ N₂O for oxidation

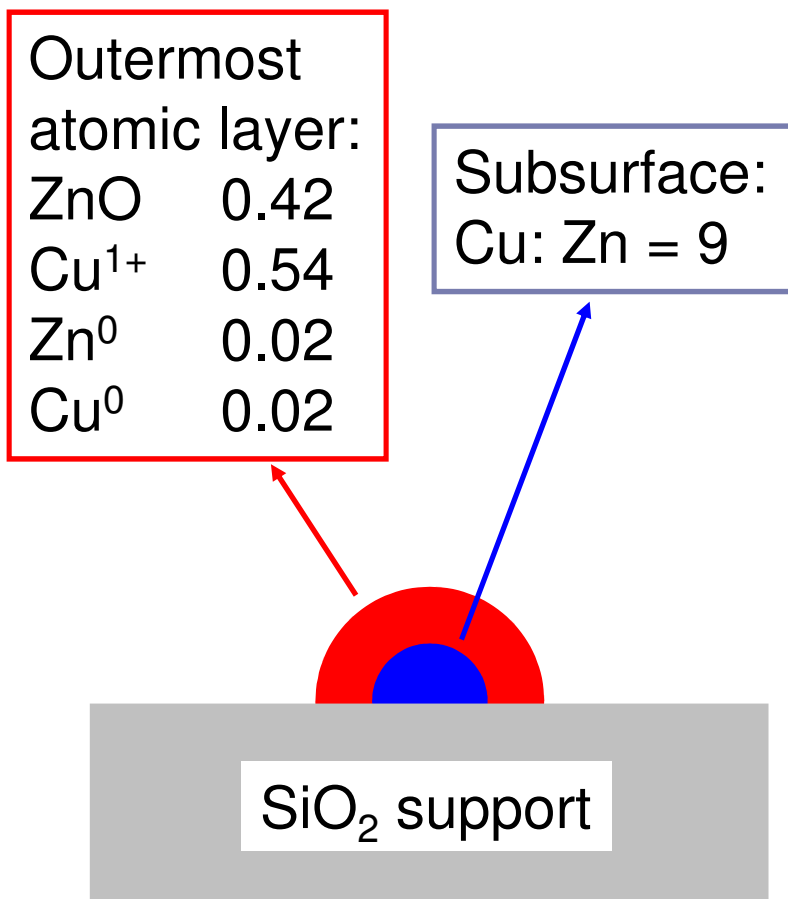
➤ LEIS for detection increase in shielding after N₂O treatment

Cu / ZnO / SiO₂ - Catalyst

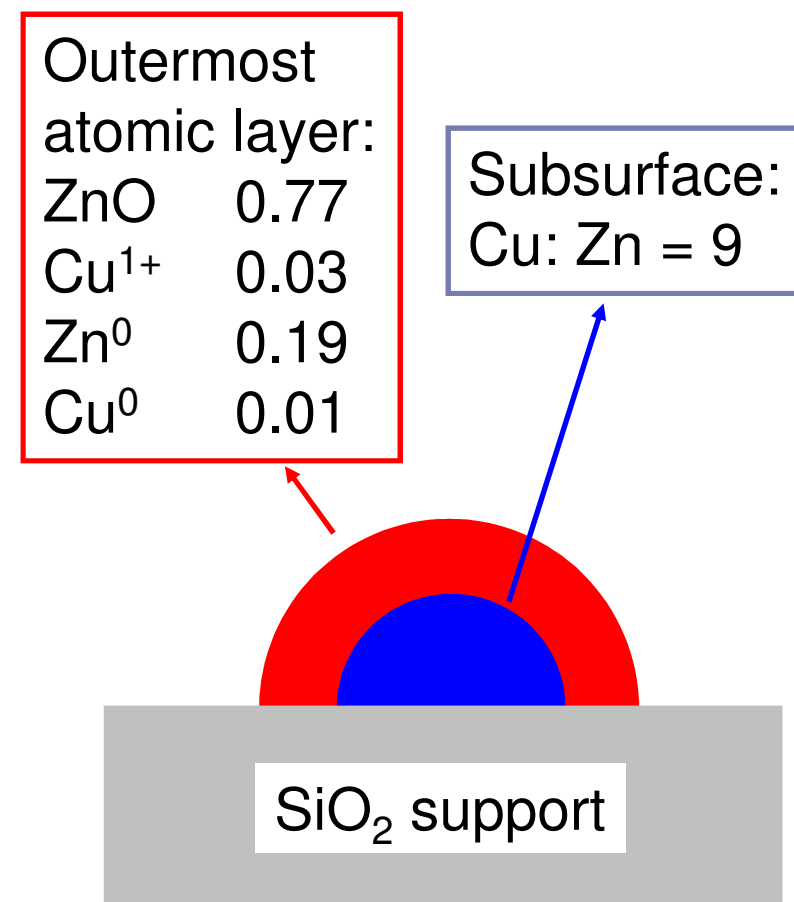
IONTOF

Determination of cluster size and oxidation states by LEIS

Cu/Zn/SiO₂ reduced at 473 K



Cu/Zn/SiO₂ reduced at 673 K



Atomic Layer Deposition (ALD)

“ Growth with Digital Accuracy “

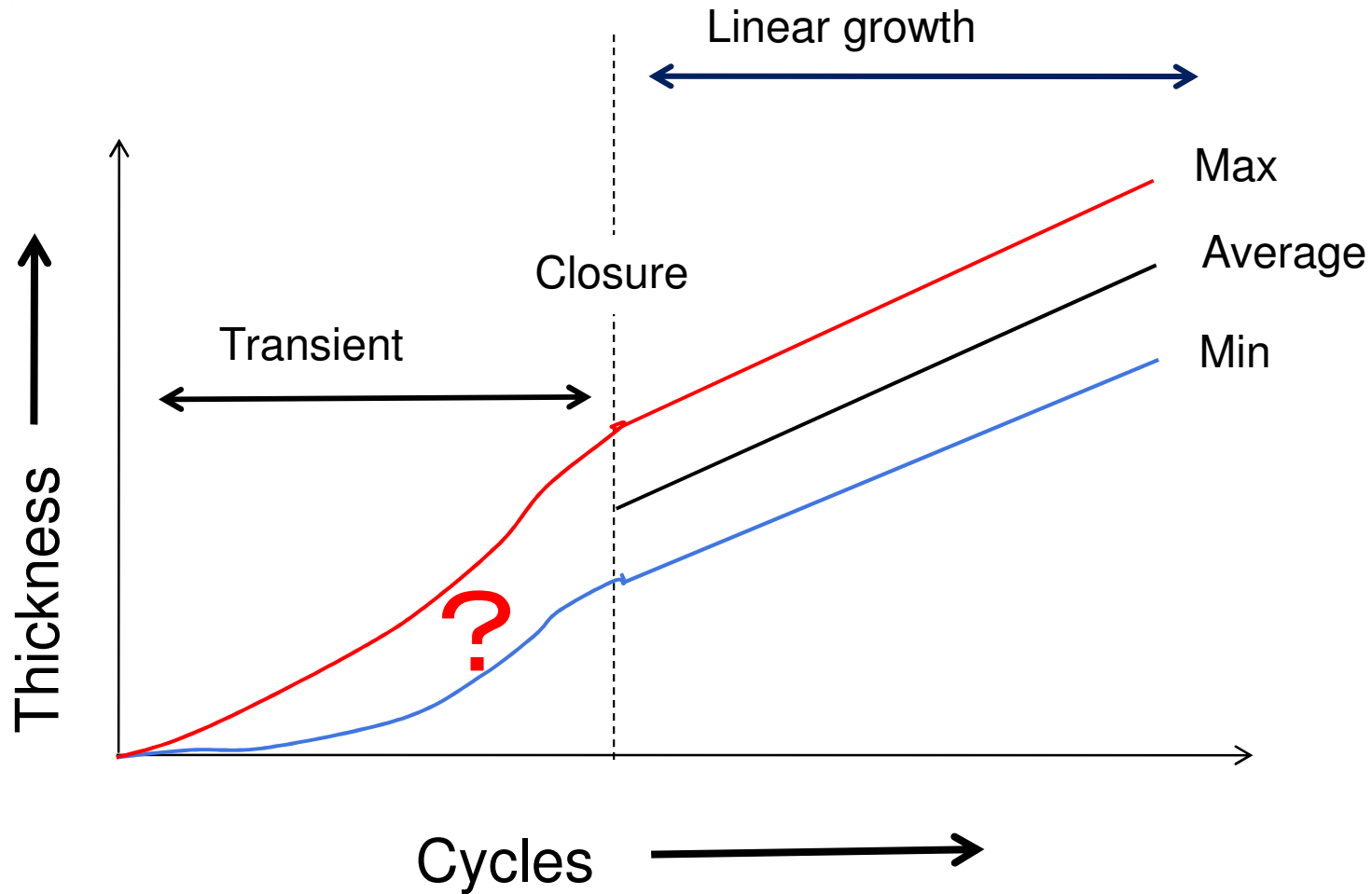
How many cycles for Closure ?

ALD cartoons: (often) show closed layer after 1 cycle
In practice: closure after a few up to > 100 cycles !

Typical examples (depending on ALD conditions !):

- 6 cycles $\text{CrOx} / \text{Al}_2\text{O}_3$
- 6-9 cycles HfO_2 / Si
- ~ 15 cycles $\text{ALN} / \text{SiO}_2$
- ~ 40 cycles $\text{Al}_2\text{O}_3 / \text{Si}$
- ~ 70 cycles $\text{Fe}_2\text{O}_3 / \text{ZrO}_2$
- ~ 150 cycles $\text{TiN} / \text{SiO}_2$

Layer thickness versus cycle



The transient regime determines closure and uniformity

Characterization of MOCVD vs. ALD HfO₂ layer closure and growth mode on Silicon: a new model for preferential deposition

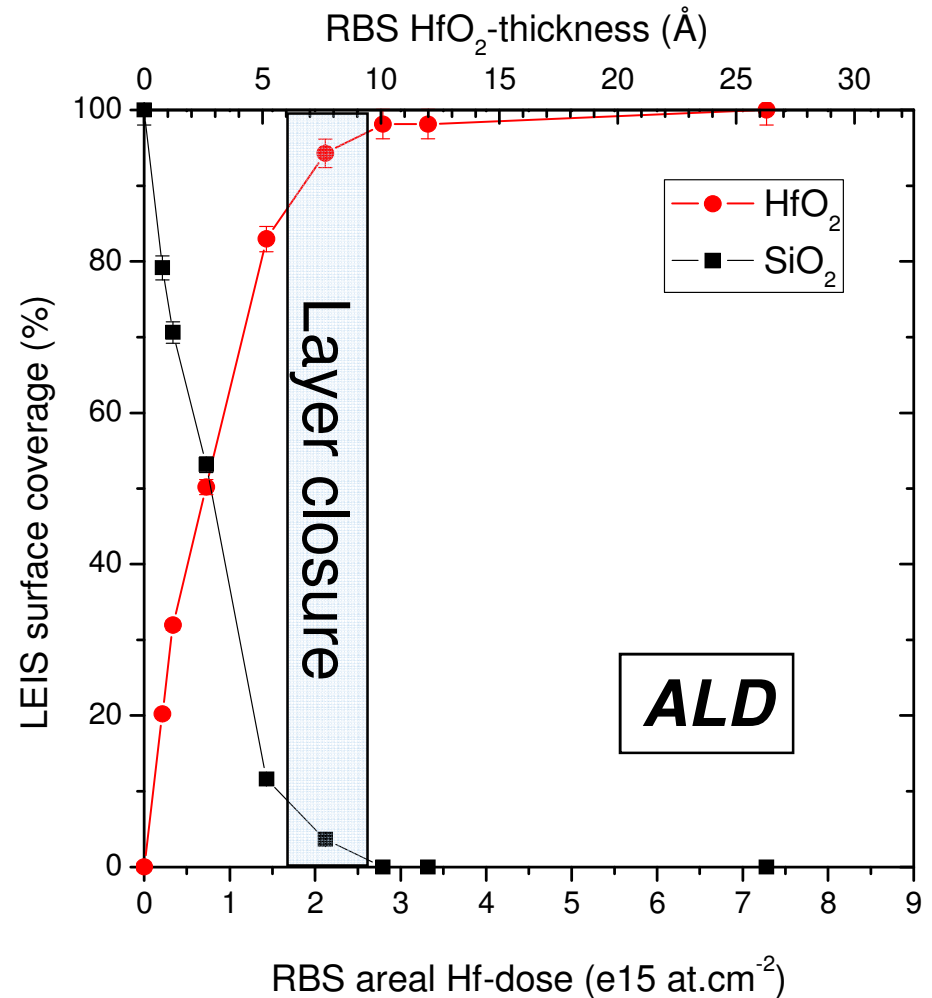
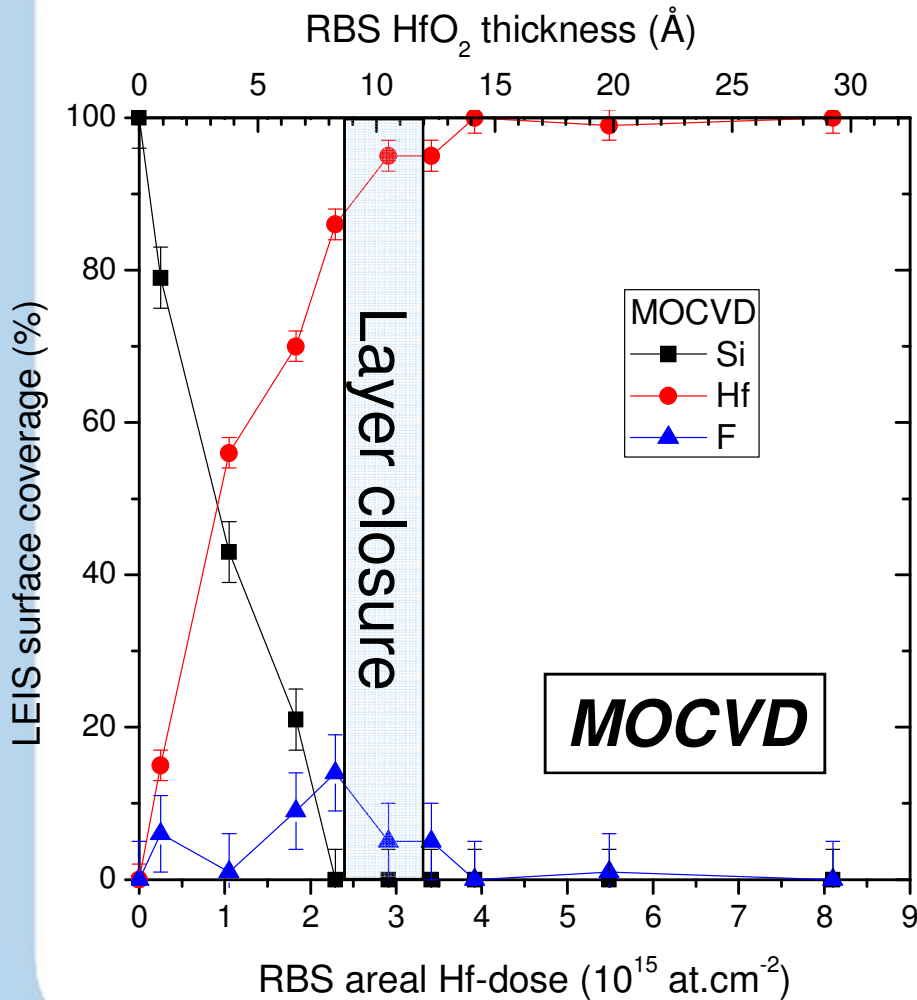
M.J.P. Hopstaken¹, M.S. Gordon¹, J. Schaeffer², H. Jagannathan¹,
T. Grehl³, H.H. Brongersma^{3,4}, M. Copel¹, M.M. Frank¹, V. Narayanan¹,
K. Choi², M. Fartmann⁴, D. Breitenstein⁴

¹IBM Research, ²GLOBALFOUNDRIES, ³ION -TOF, ⁴Tascon

HfO₂ layer closure: MOCVD vs ALD

Surface fractions (LEIS) as function of coverage

➔ Earlier layer closure for ALD-HfO₂



LEIS and Growth

- Initial growth; growth mode
- Poisoning, activation
- Closure, pinholes
- Thickness distribution

Conclusion IBM / Global Foundries / ION-TOF / Tascon study:

- Origin of the superior quality of the ALD grown layers revealed by HS-LEIS
(other analytic tools have insufficient depth resolution)

Summary: Why do you need LEIS ?? !!

- Any material, any T
- Quantitative
- 1st atom and high-resolution in-depth !!

Unique applications of High Sensitivity LEIS (NOT's)

- Segregation, Anti-wetting
- Adhesion: “ 5% vs 100% “
- Follow ultrathin growth
- Pinholes in ultrathin; Nano pinholes
- Metal / polymer in-depth diffusion
- Catalysis: poison, promoter, probe molecules, core-shell,
- Nanoclusters (diameter; outer atoms)
- Inorganics: oxidation states
- Improve cleaning strategies

Complementarity to XPS, ToF-SIMS,

Miscellaneous applications

- Microelectronics, polymers, ceramics, catalysis, sensors,

But also:

- Pinholes in coatings
- Candy wrappers
- Gold mining
- F 16 Dome
- Bone tissue, implants, stents,
- Ageing of Linoleum (“ Linowonder”)
- Anti-wetting (watches,
- Floor wax

Complementary Cutting Edge techniques

HS-LEIS

Qtac¹⁰⁰

+

HR-XPS

Scienta ESCA 300

Lehigh University



qtac¹⁰⁰

Thank you for your attention.