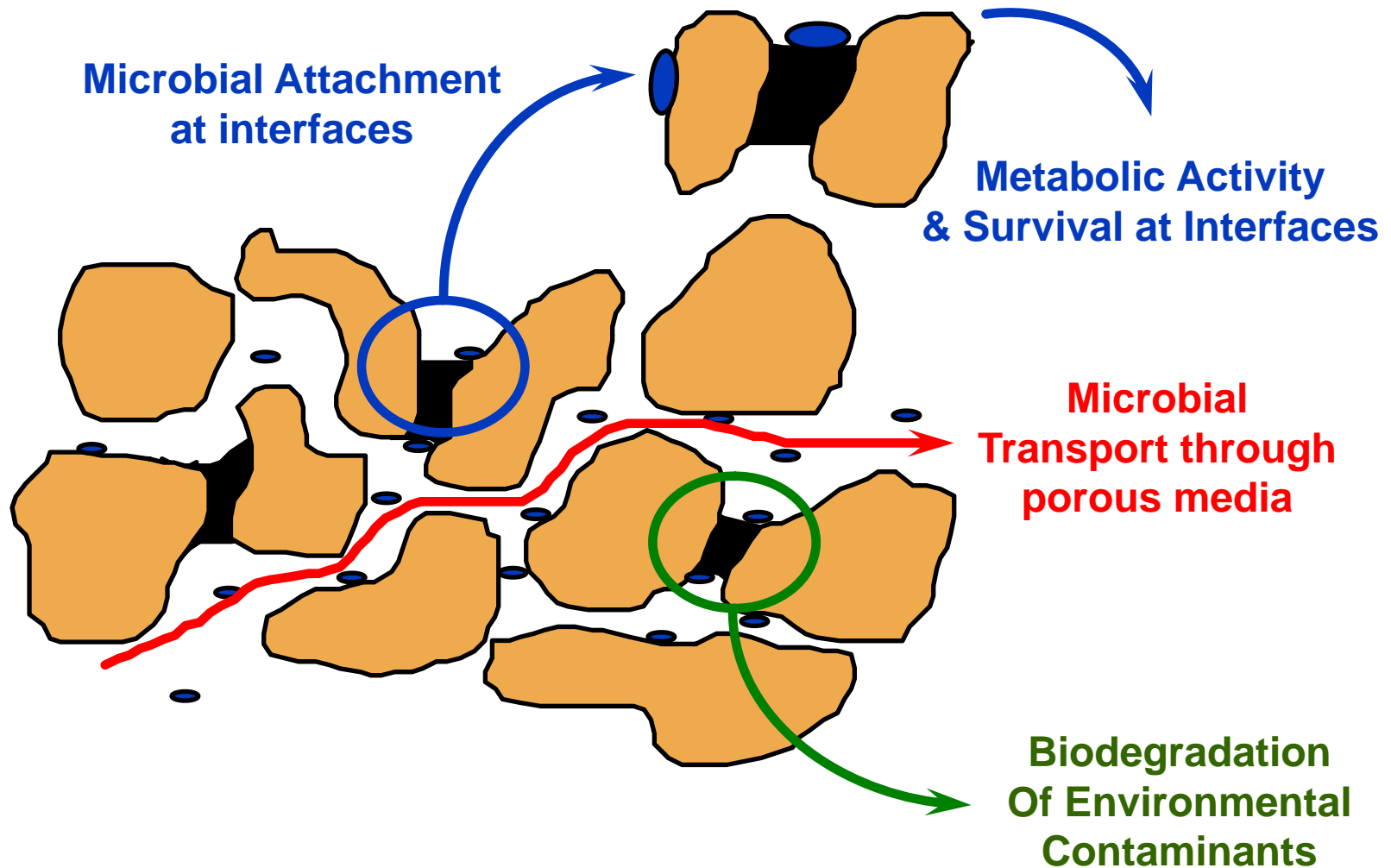

Environmental Biotechnology

Derick G. Brown, Ph.D., P.E.

**Associate Professor and
Co-Director of the Environmental Initiative
Department of Civil & Environmental Engineering**

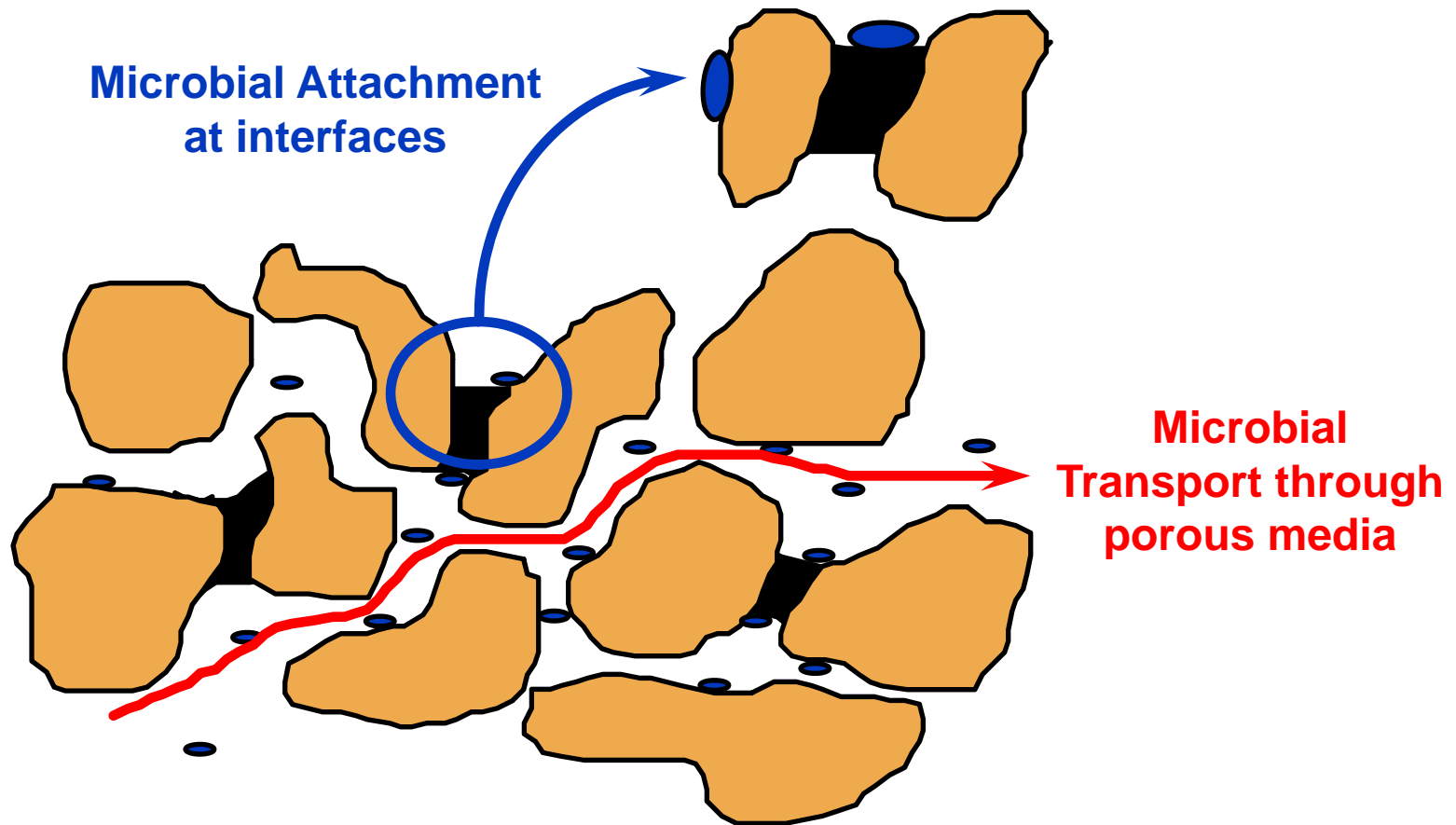
Environmental Biotechnology and Complex Systems



Theme...

You can learn a lot from
simple experiments

Microbial Attachment and Transport



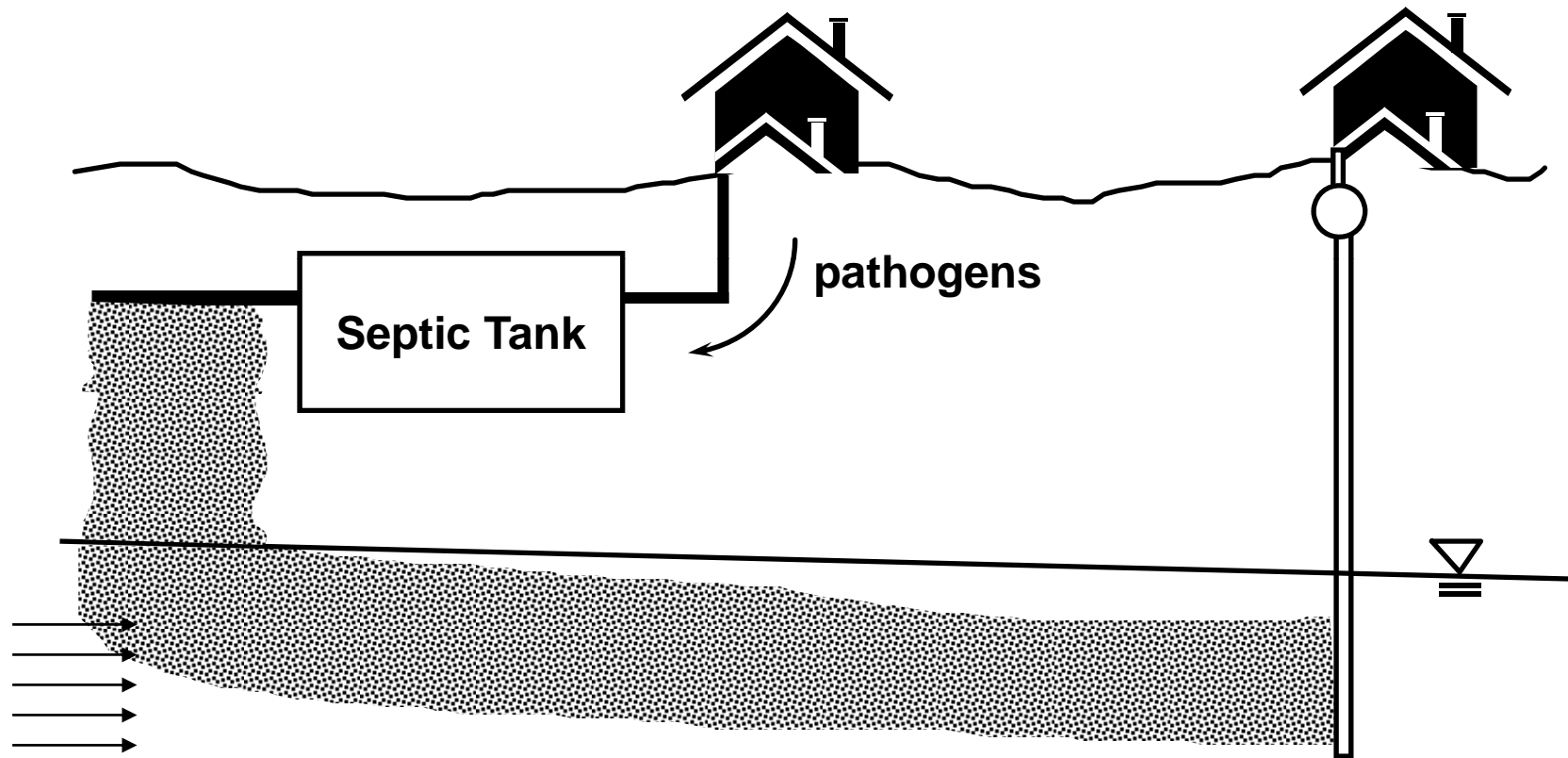
Protection of drinking water supplies

- 77.5 billion gallons of groundwater drawn daily for use in the U.S.
- Groundwater supplies ~50% of the nation's drinking water.



- Typically assumed that groundwater is free from microbial contamination.
- However, a major fraction of drinking water disease outbreaks in the U.S. are due to consumption of untreated groundwater.

The question is why do pathogens travel farther than expected?

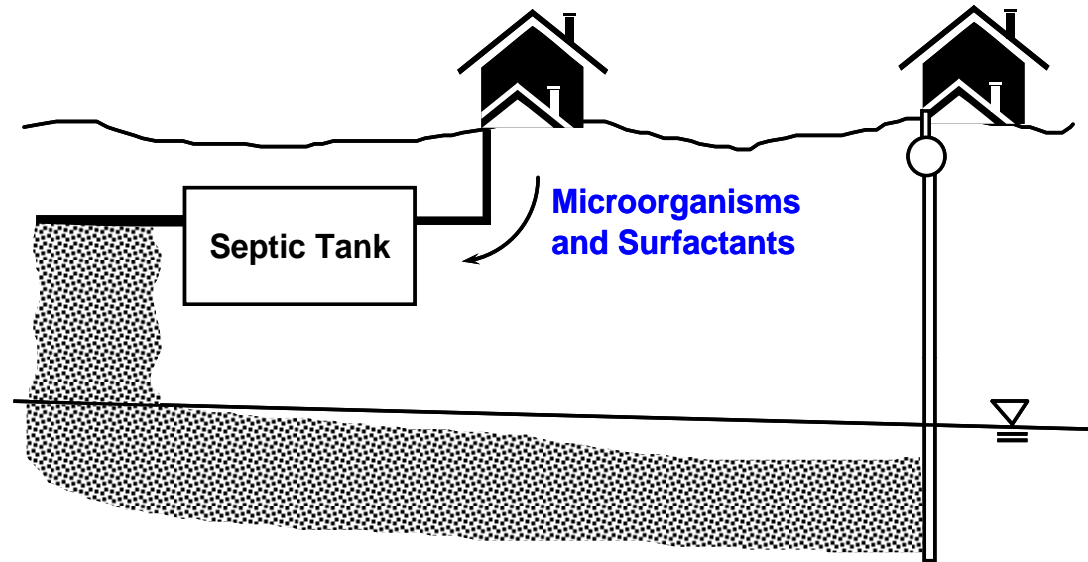


We are investigating bacterial transport through porous media using lab and meso-scale systems



Surfactants and domestic sewage

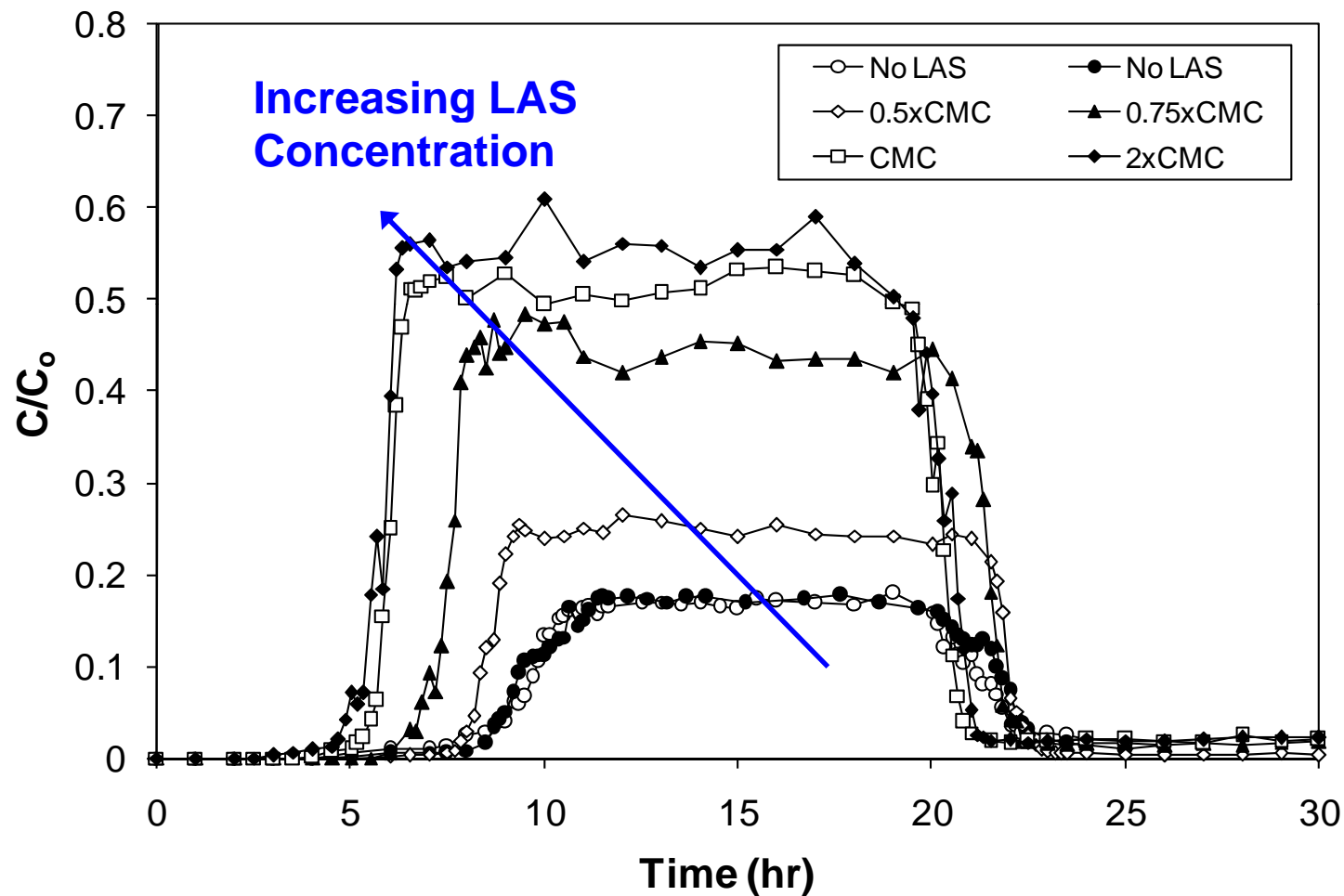
- There are $\sim 7 \times 10^6$ metric tons of surfactants used for laundry detergents in the U.S. every year.
- Surfactants are the major man-made organic constituent in domestic wastewater.
- Grey water contains 150 – 600 mg/L of LAS.
- Total volume of septic waste released into the subsurface in the U.S. is estimated to be one trillion gallons per year.



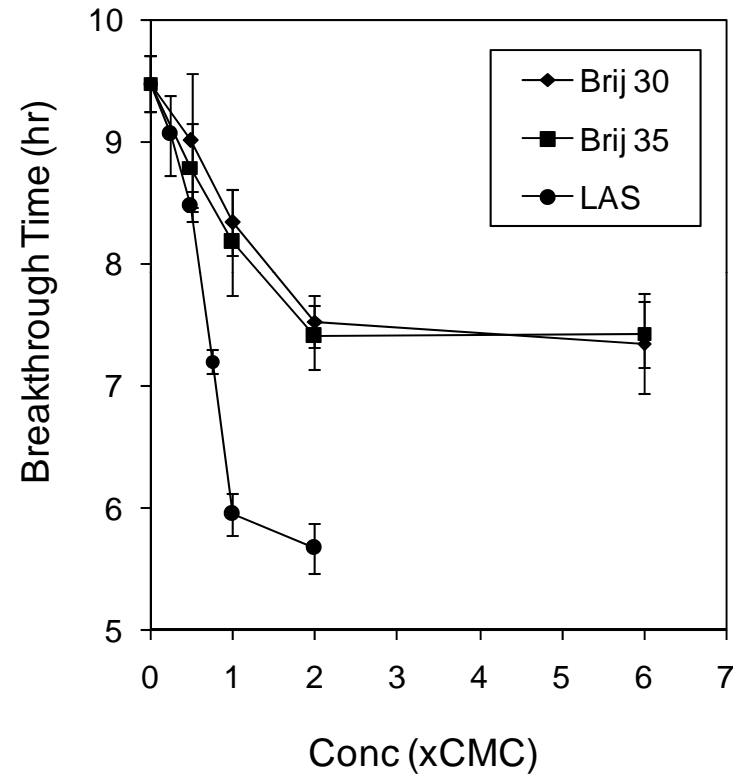
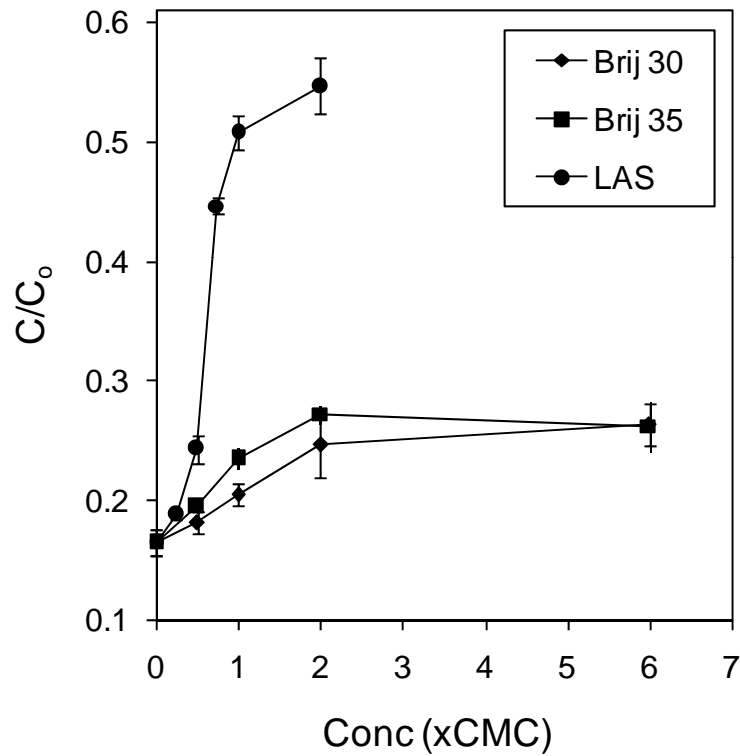
Examining effects of surfactants on bacterial transport through unsaturated porous media



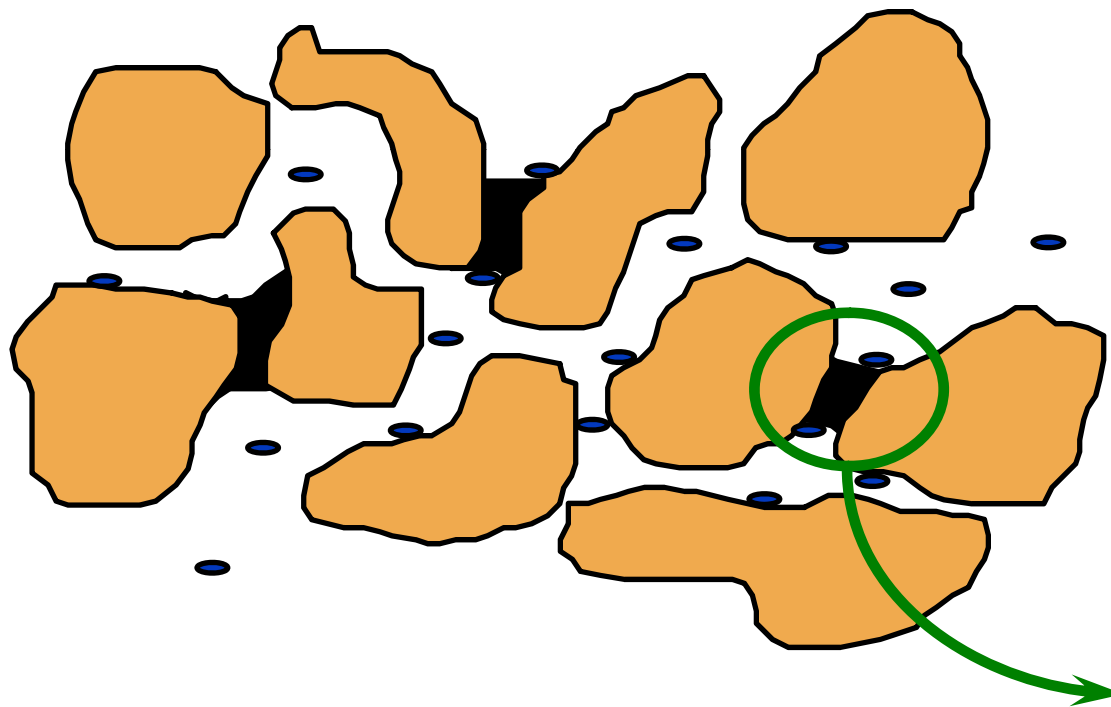
Transport of *E. coli* through unsaturated porous media in the presence of Linear Alkylbenzene Sulfonate (LAS)



Surfactant affect both increase transport rate and effluent concentration

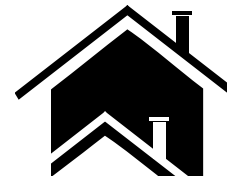
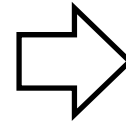
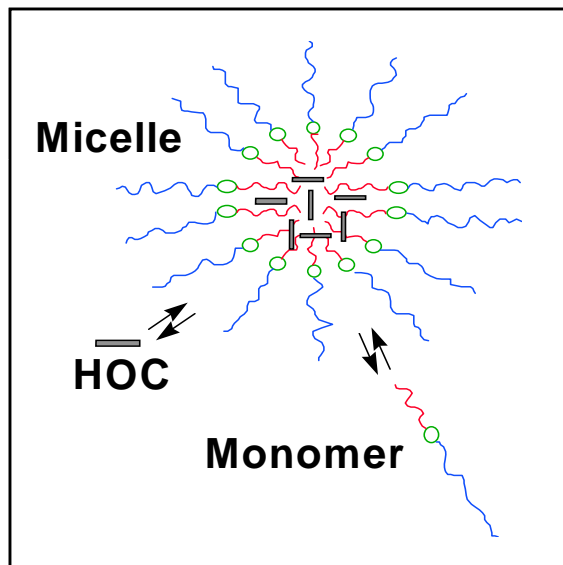
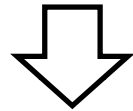
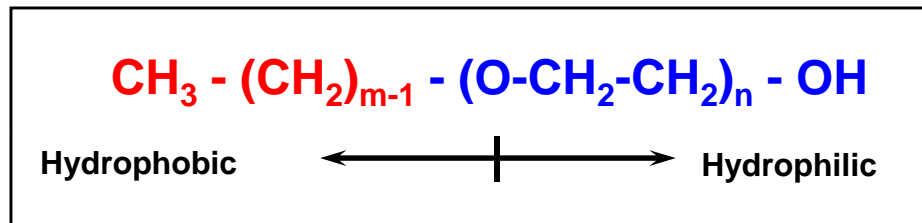


Enhanced biodegradation of environmental contaminants (microbial kinetics)

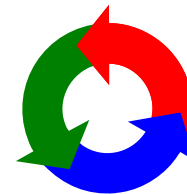
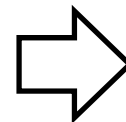


Biodegradation
Of Environmental
Contaminants

Use of surfactants to enhance biodegradation of hydrophobic substrates

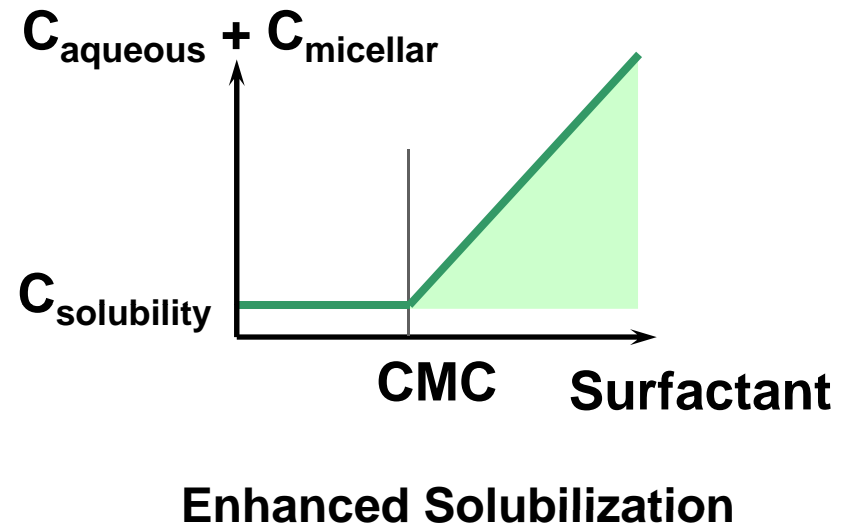
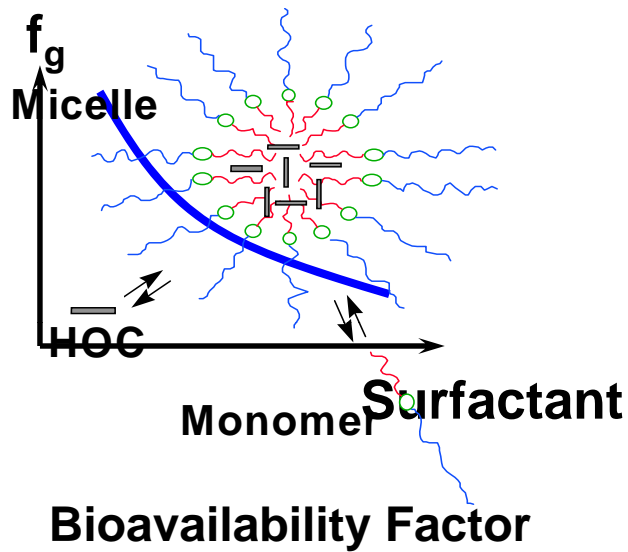


Laundry and Dish Detergents



Enhanced solubility and bioavailability

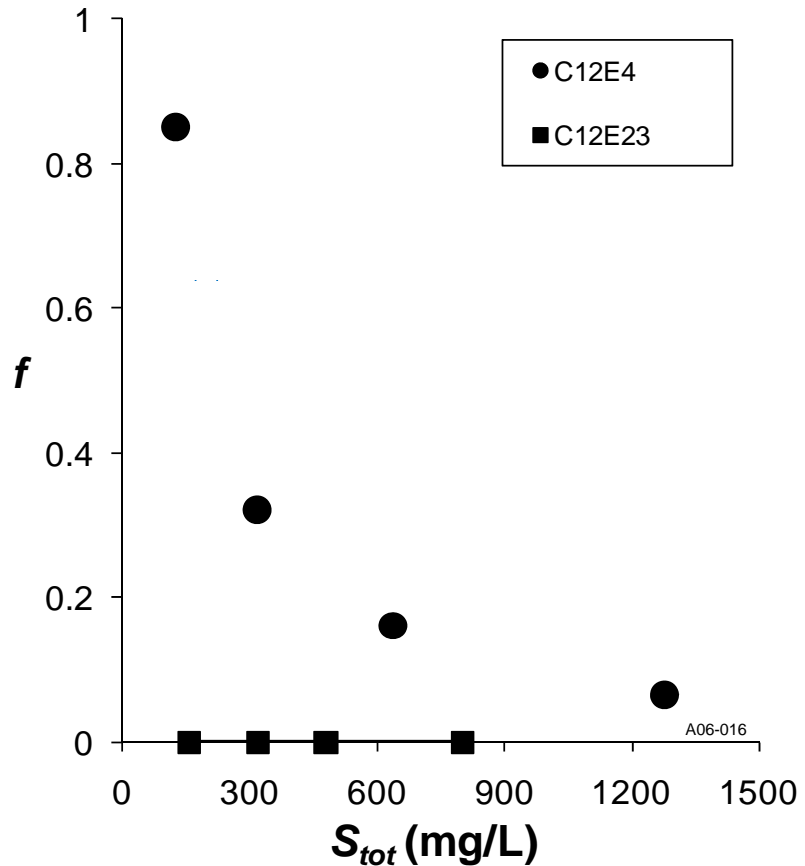
The bioavailable factor and the enhanced solubilization define the regime of enhanced bioavailability



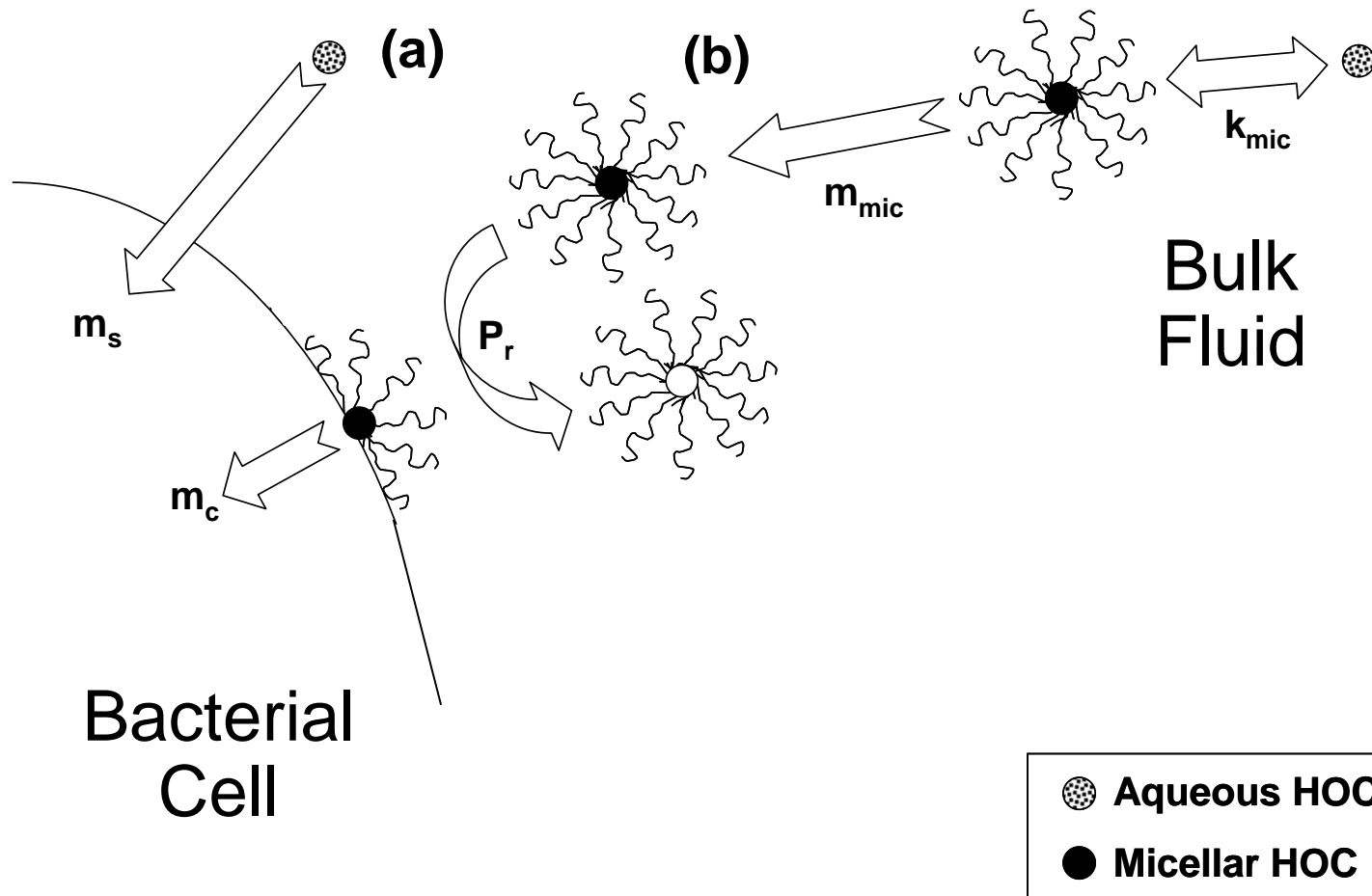
$$C_{\text{bioavailable}} = C_{\text{aqueous}} + f_g C_{\text{micellar}}$$

Bioavailability factor experimentally determined for $C_{12}E_y$ surfactants

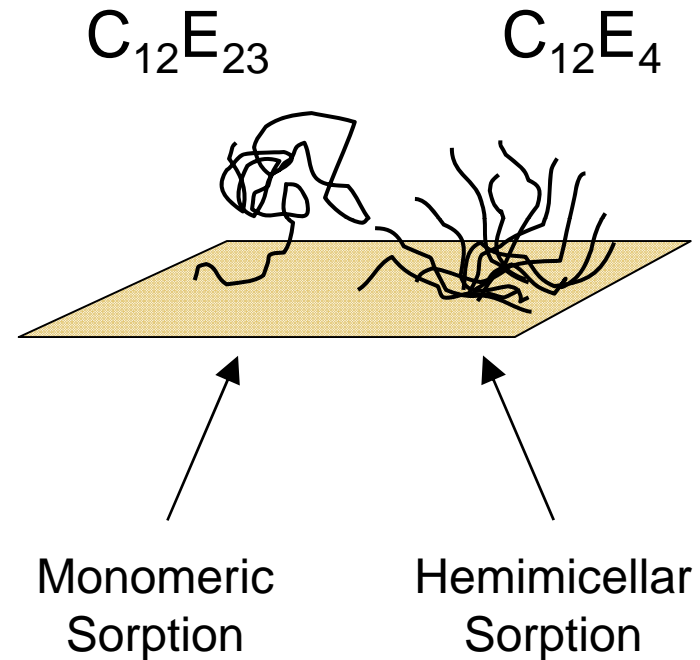
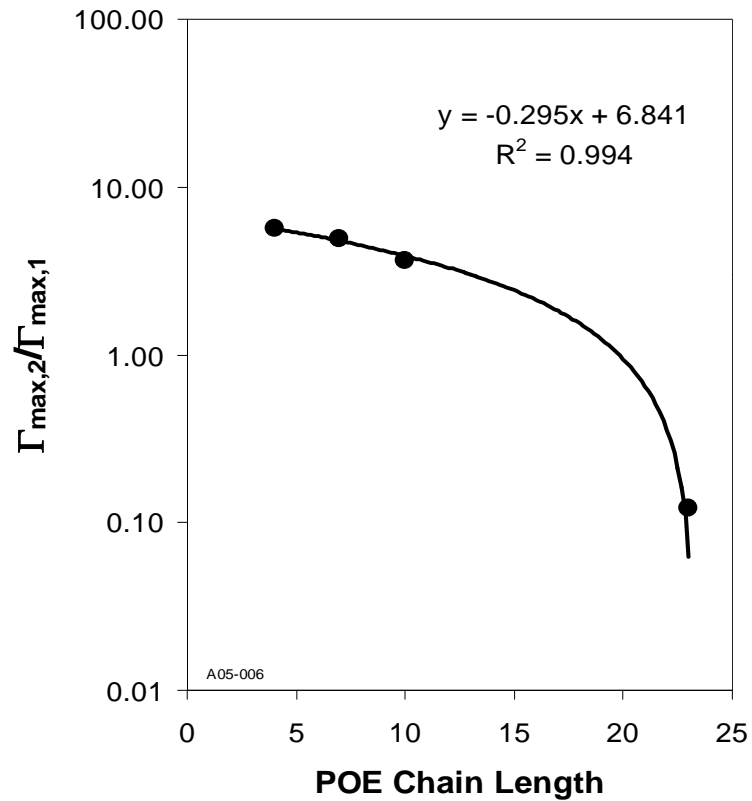
- **Guha and Jaffe** experimentally determined f for $C_{12}E_4$ and $C_{12}E_{23}$ surfactants.
- **Sriwatanapongse** experimentally determined f for $C_{12}E_7$, $C_{12}E_9$, and $C_{12}E_{10}$.



Theory on surfactant-enhanced bioavailability

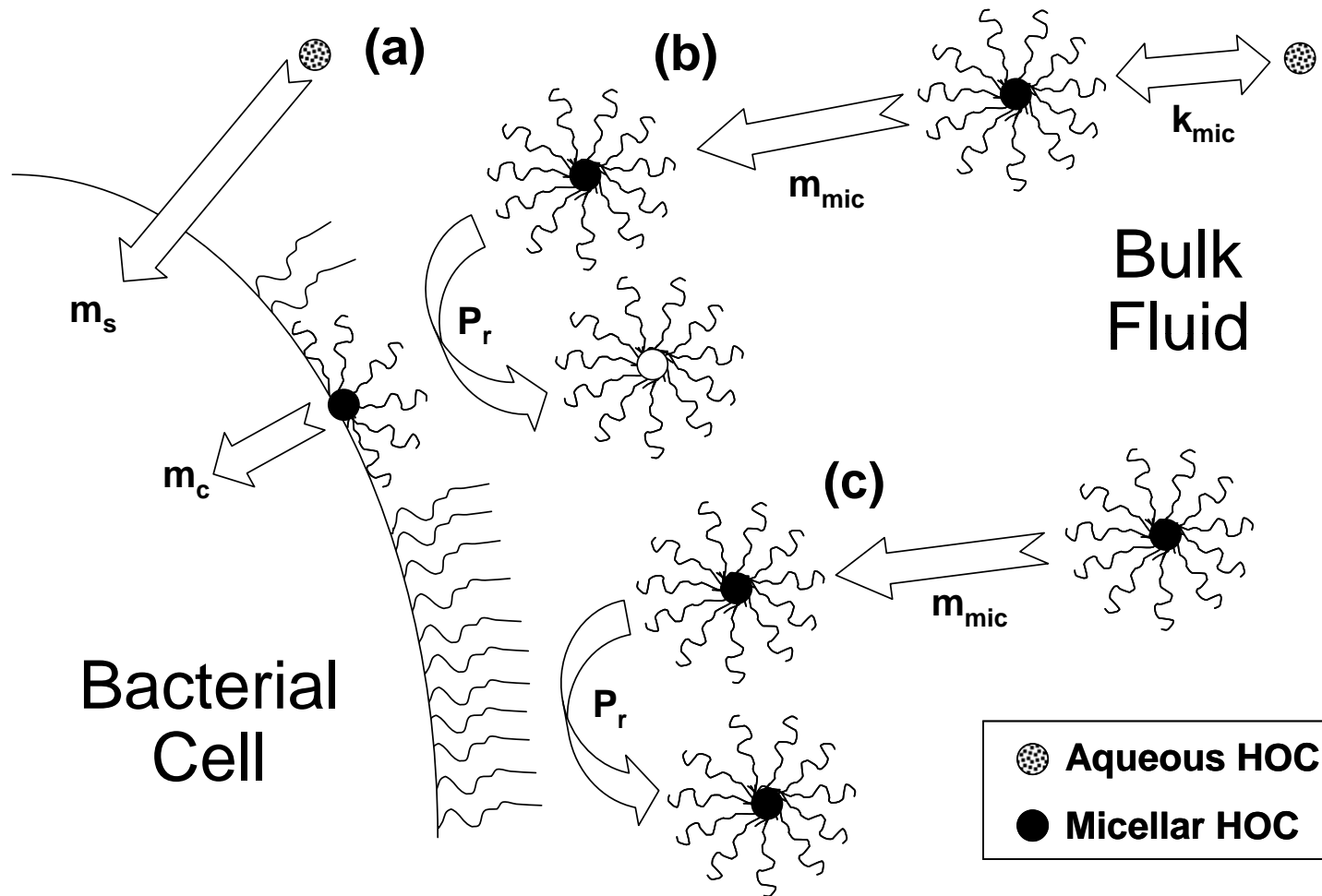


Monolayer and Lateral Interactions Vary as a Function of POE Chain Length

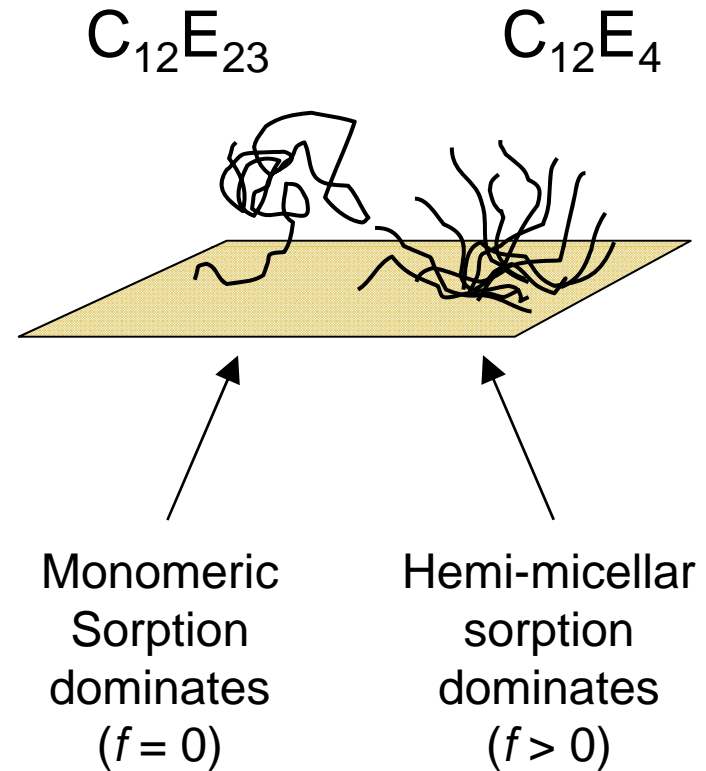
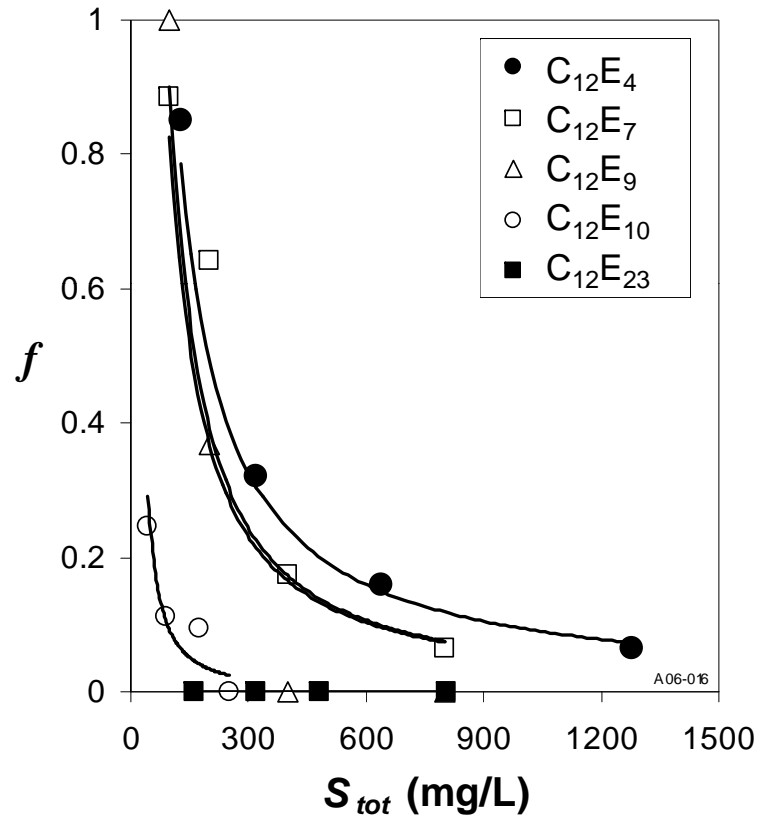


Brown and Al Nuaimi. 2005. *Langmuir* 21(24):11368-11372.

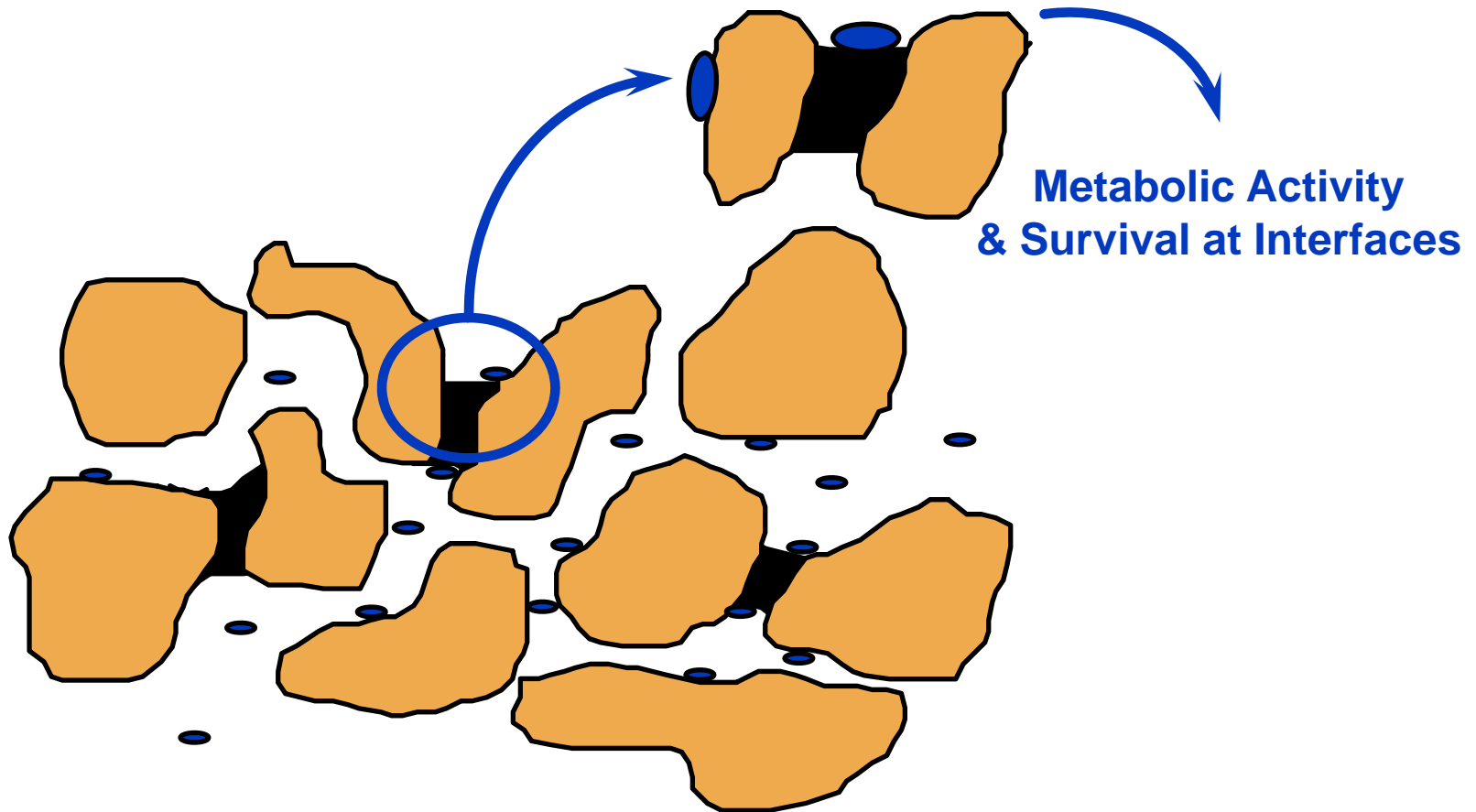
Developed theory on surfactant-enhanced bioavailability



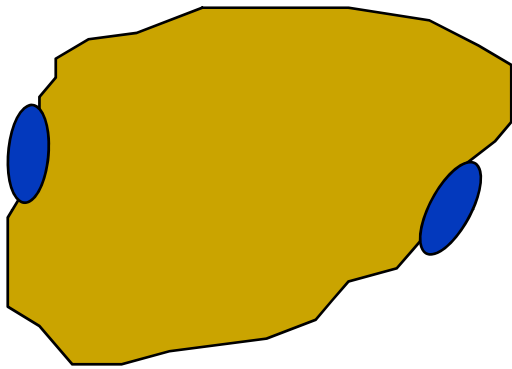
Surfactant-enhanced bioavailability theory is able to replicate multiple experimental data sets



Effects of surfaces on metabolic activity and survival



Surfaces have been observed to stimulate bacterial metabolic activity and survival

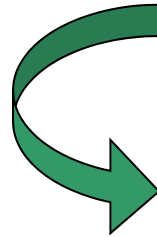


Bacterial adhesion has been shown to provide:

- Enhanced Metabolic Activity
- Enhanced Survivability

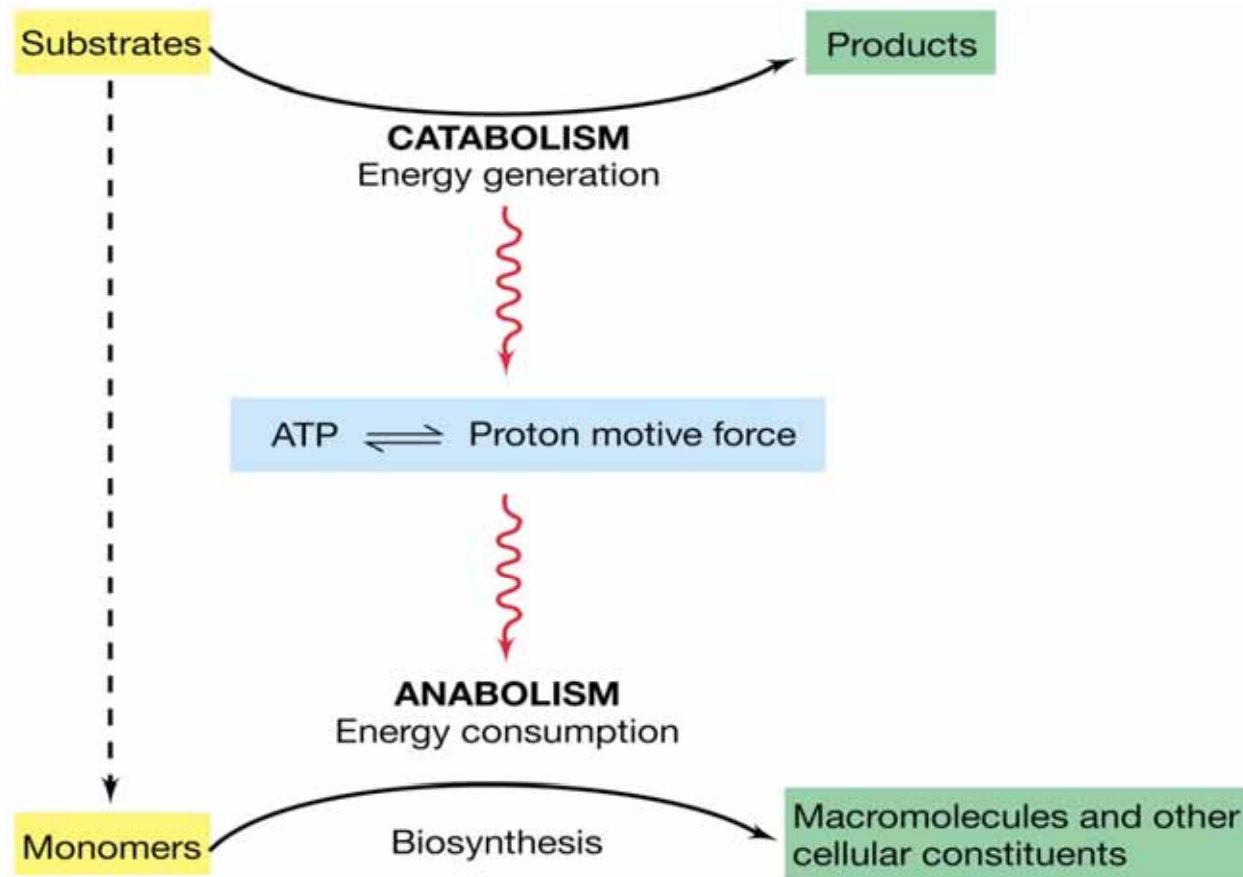


Increased ATP levels



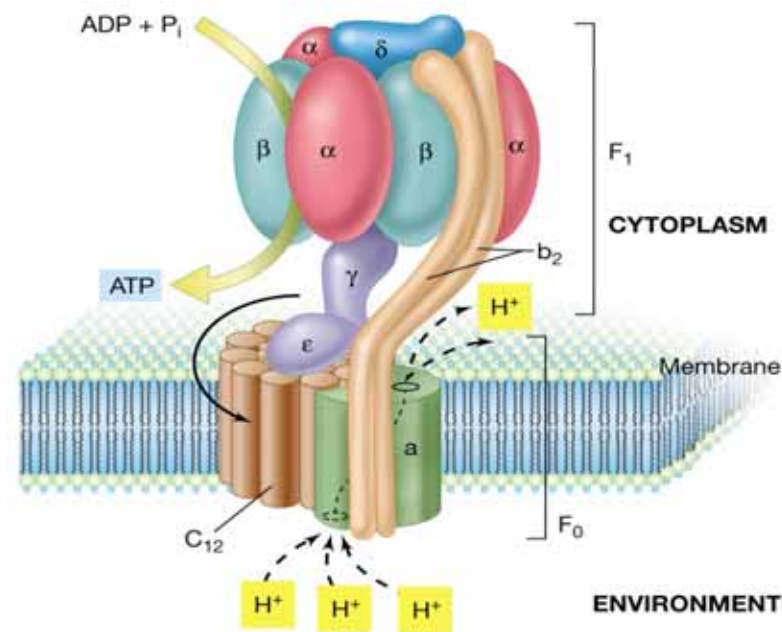
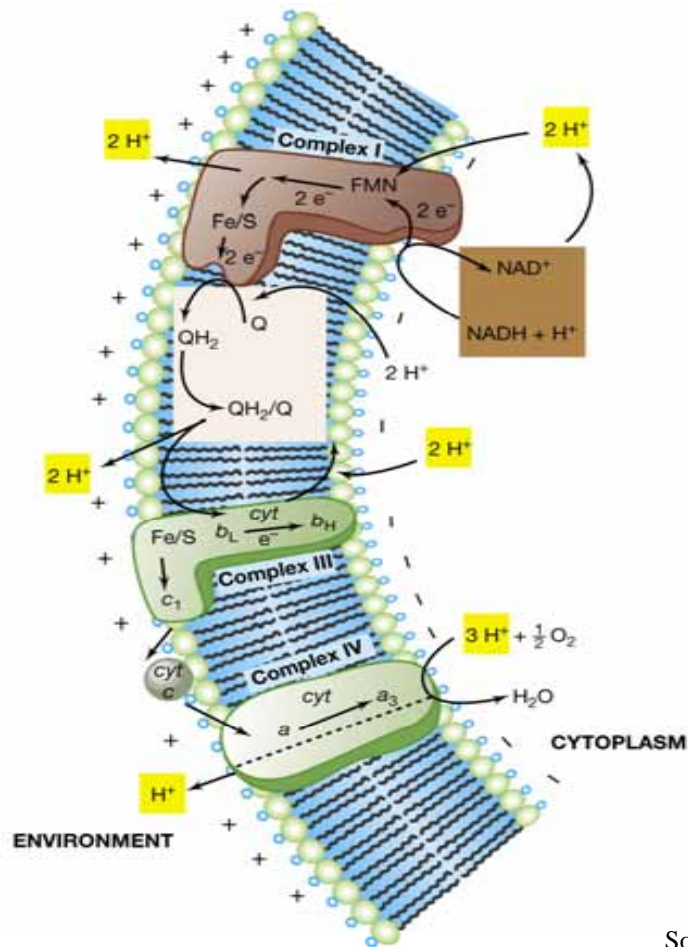
- Not due to nutrients
- Not due to substrate availability

Cellular bioenergetics links catabolism and anabolism



Source: Madigan et al. 2003. [Brock Biology of Microorganisms](#), 10th Edition. Prentice Hall.

Cellular bioenergetics uses a proton gradient across the cell membrane

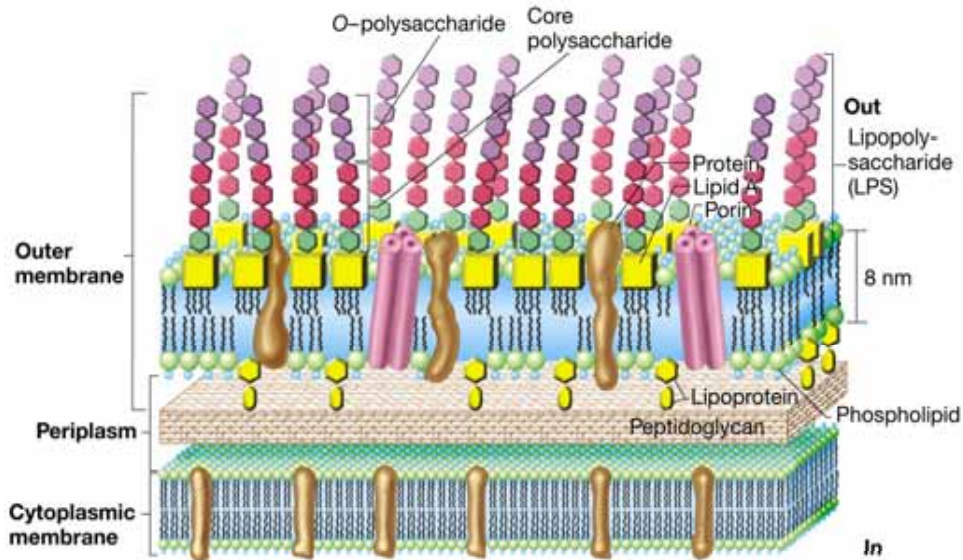


$$\Delta p = \Delta \psi - 59 \Delta pH$$

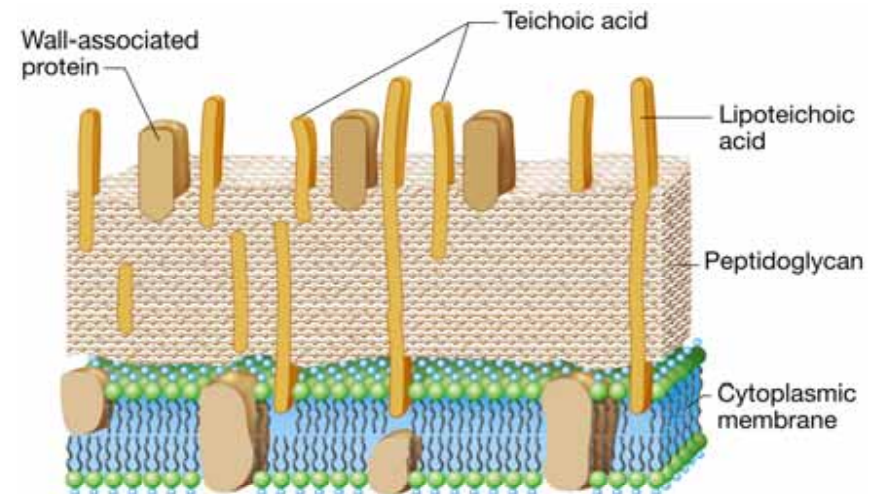
Source: Madigan et al. 2003. *Brock Biology of Microorganisms*, 10th Edition. Prentice Hall.

Cell wall structure of Gram-negative and Gram-positive bacteria

Gram-negative

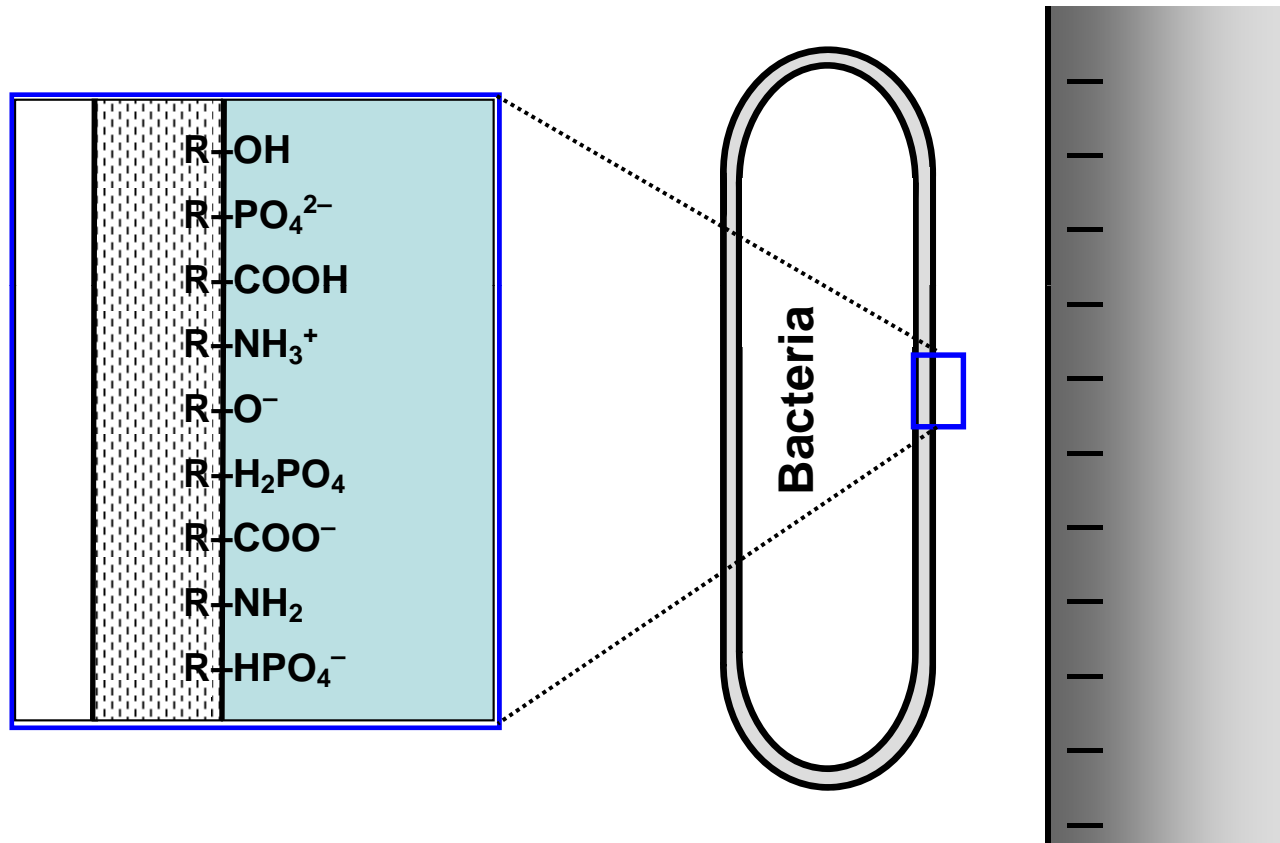


Gram-positive

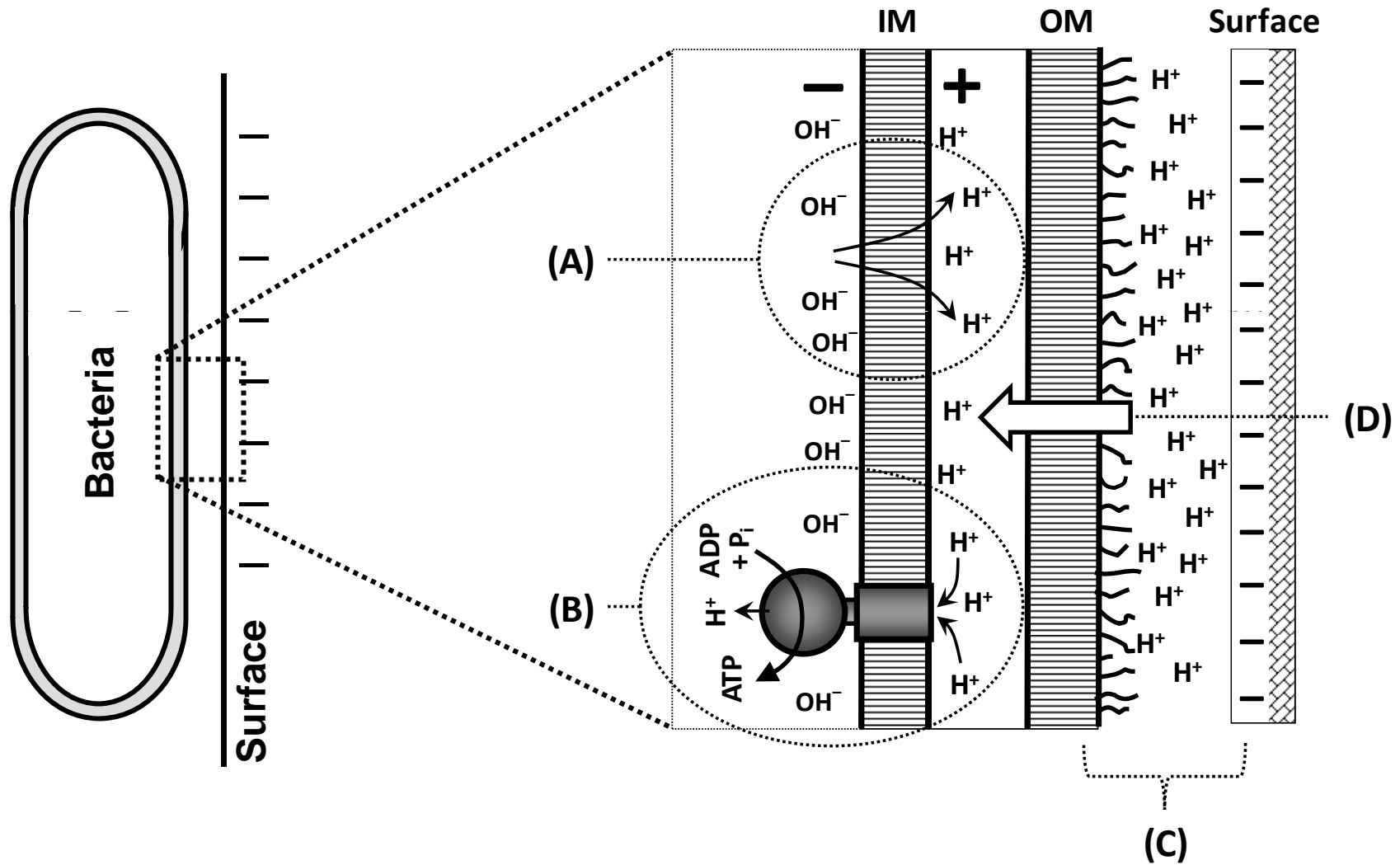


Source: Madigan et al. 2003. Brock Biology of Microorganisms, 10th Edition. Prentice Hall.

Charge-regulation effect can result in surface pH changes upon adhesion

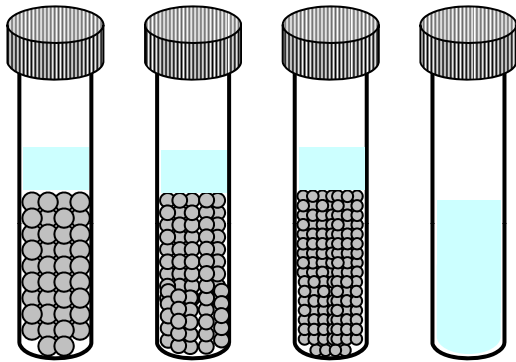


Hypothesis: Charge regulation process alters surface pH, which alters the proton motive force

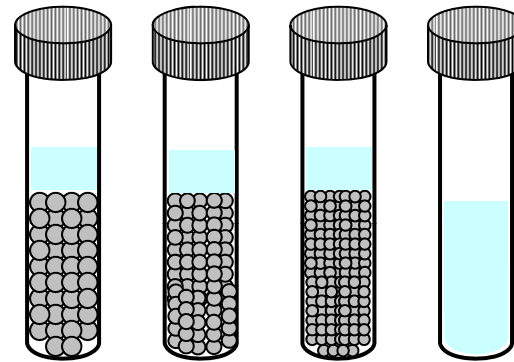


Experimental Approach

Bacteria



Bacteria + Surfactant

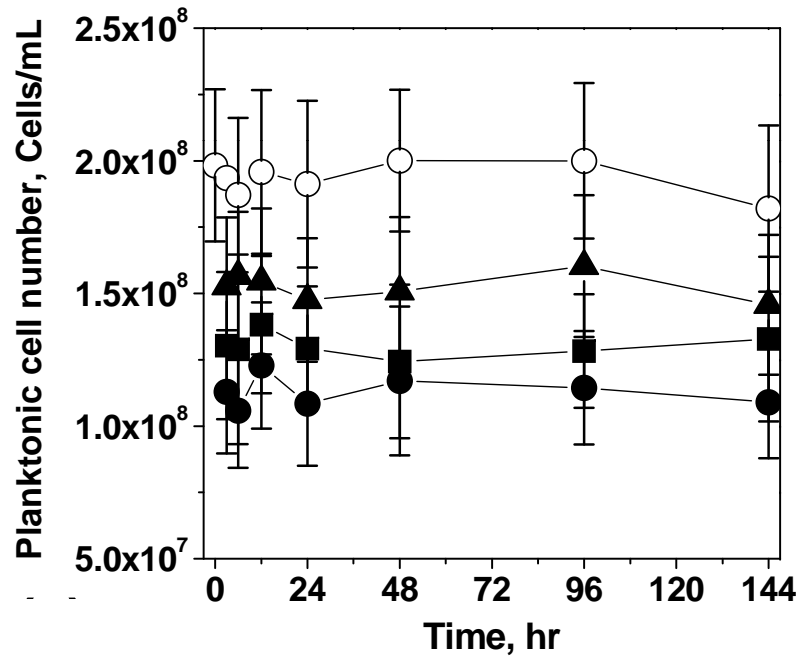


Diameter	Surface Area
2.5 mm	76.8 cm ²
1.0 mm	192 cm ²
0.5 mm	384 cm ²

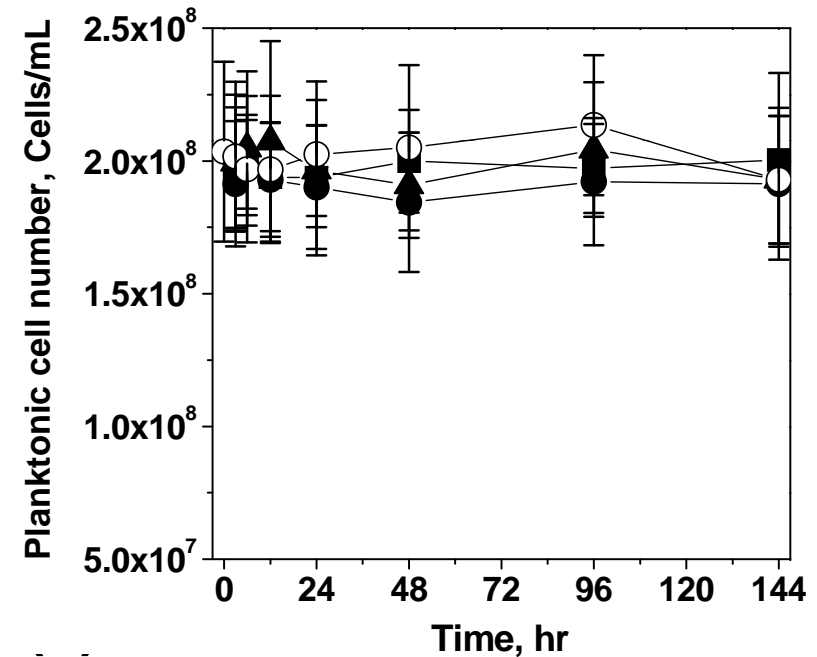
- Orbitron Shaker
- Samples taken over time
 - Adhesion
 - ATP

Adhesion as a function of surface area

E. coli in 0.01 M Phosphate buffer



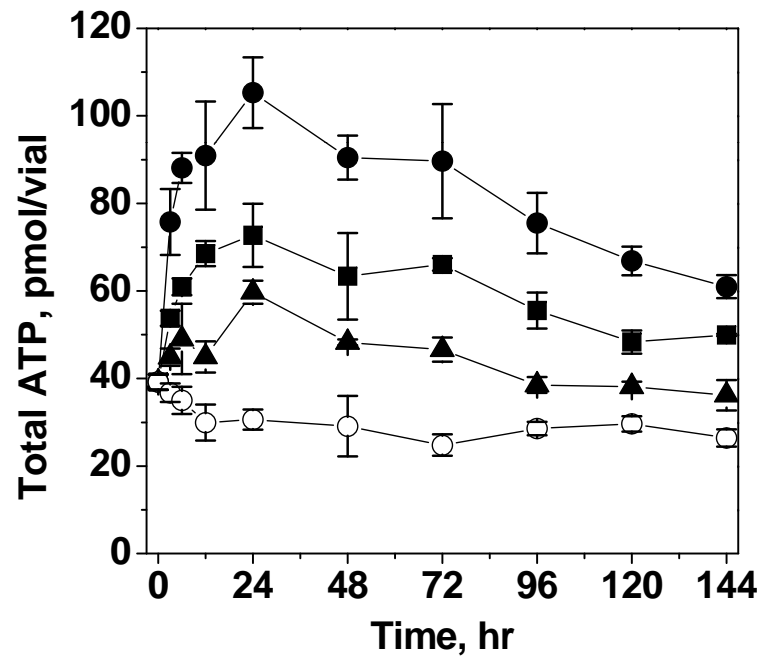
E. coli in 0.01 M Phosphate buffer + Tween 80



● 0.5 mm ■ 1.0 mm ▲ 2.0 mm ○ No beads

Total ATP as a function of adhesion

E. coli in 0.01 M Phosphate buffer + Tween 80



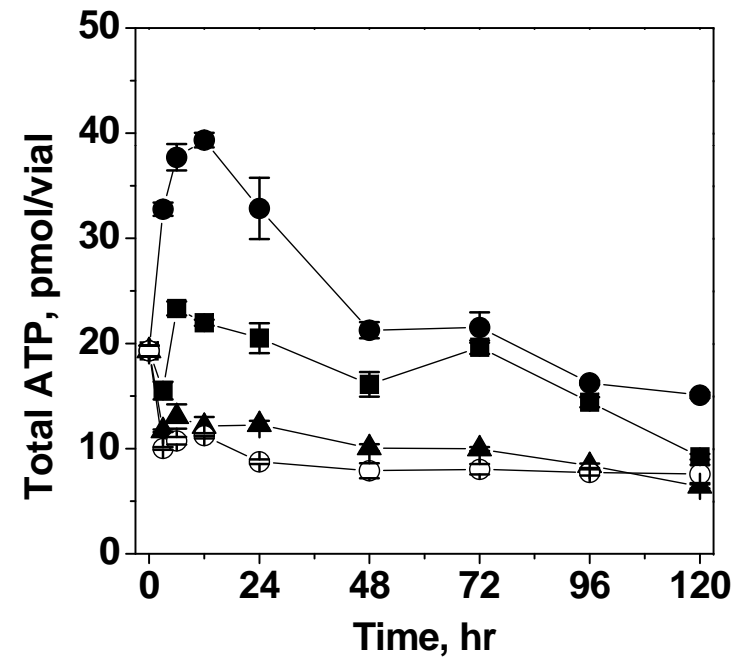
● 0.5 mm

■ 1.0 mm

▲ 2.0 mm

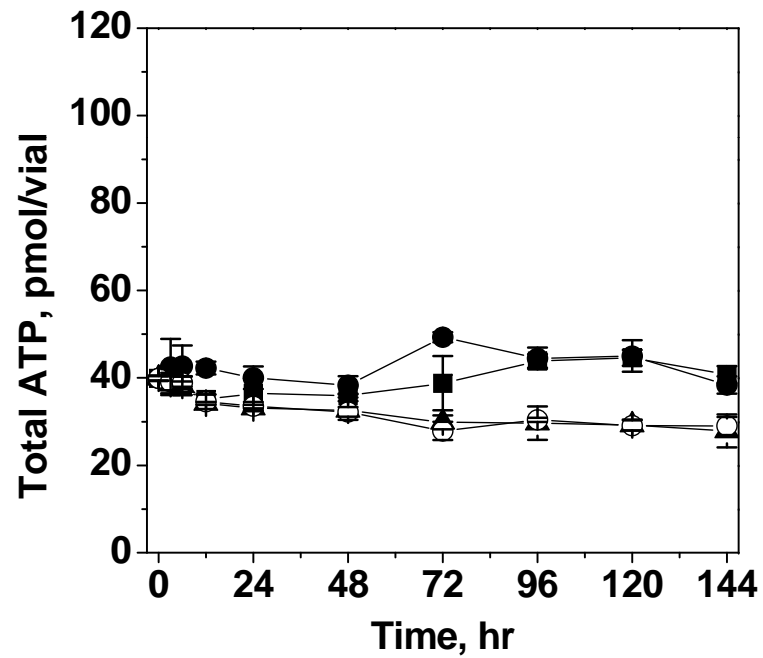
○ No beads

B. brevis in 0.01 M Phosphate buffer + Tween 80

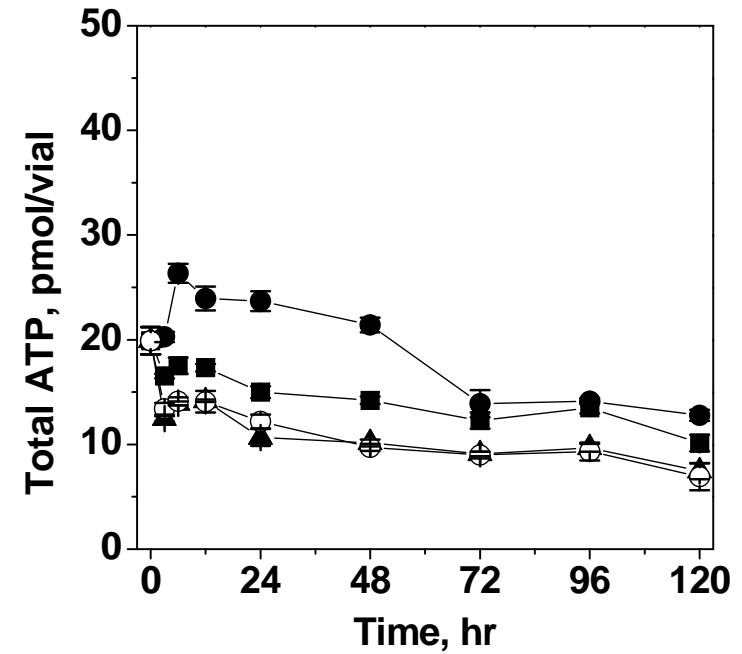


Total ATP as a function of adhesion

E. coli in 0.01 M Phosphate buffer + Tween 80



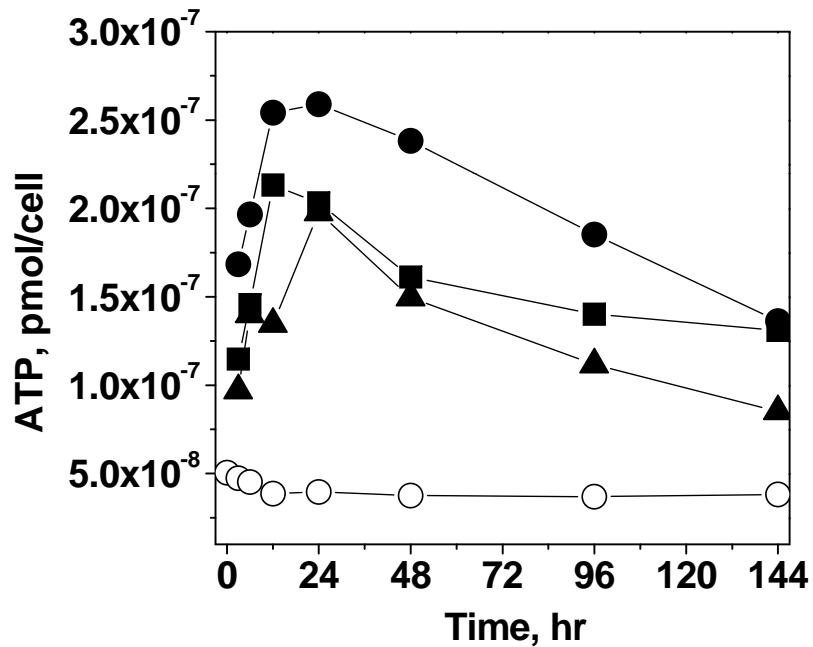
B. brevis in 0.01 M Phosphate buffer + Tween 80



● 0.5 mm ■ 1.0 mm ▲ 2.0 mm ○ No beads

ATP per cell for attached and planktonic bacteria

E. coli in 0.01 M Phosphate buffer



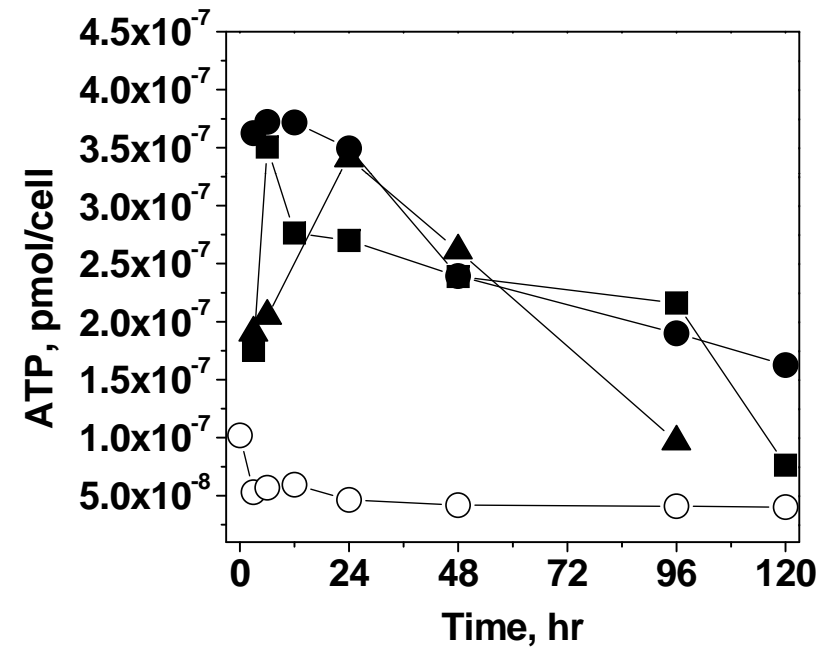
● 0.5 mm

■ 1.0 mm

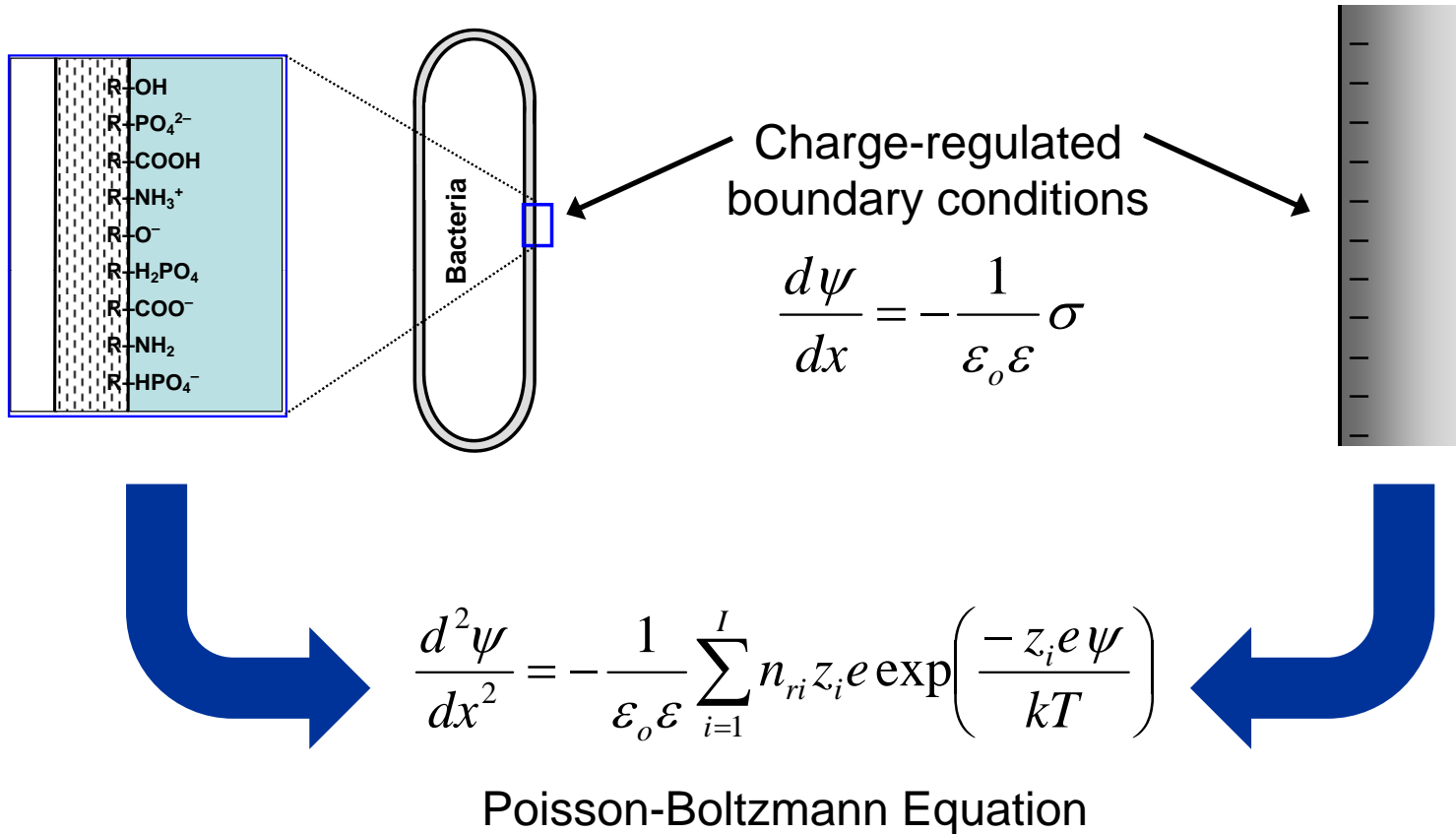
▲ 2.0 mm

○ No beads

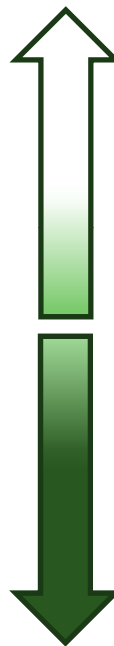
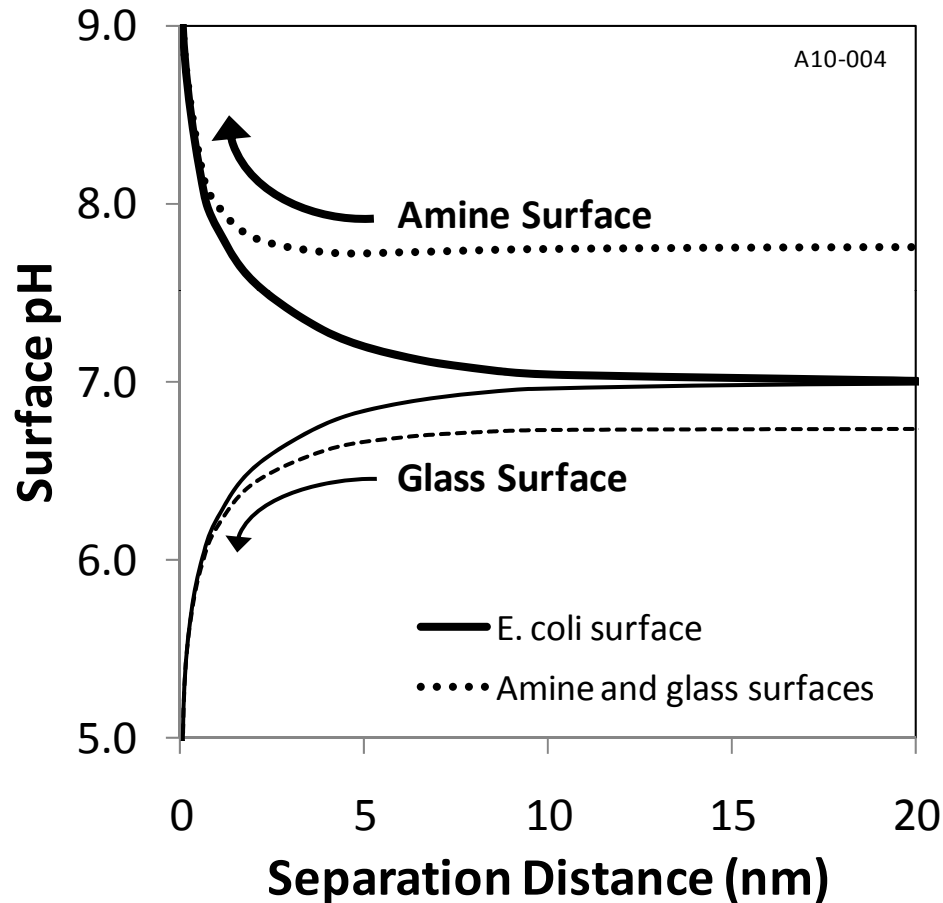
B. brevis in 0.01 M Phosphate buffer



Charge-regulation model accounts for acid/base properties of the glass and bacterial surfaces



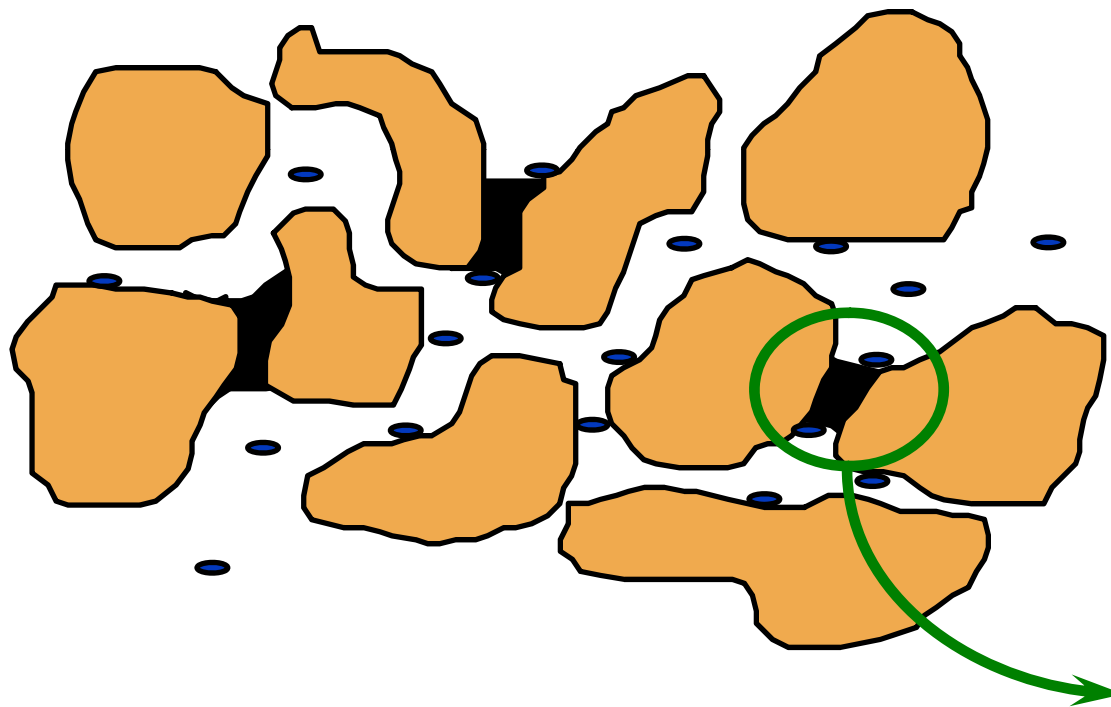
How surface charge affects surface pH and cellular ATP



$\psi \rightarrow$ less negative
ATP Decreases

$\psi \rightarrow$ more negative
ATP Increases

Impacts of microbial viruses (bacteriophages) on the design and operation of engineered systems



Biodegradation
Of Environmental
Contaminants

Wastewater treatment plants are big bioreactors!

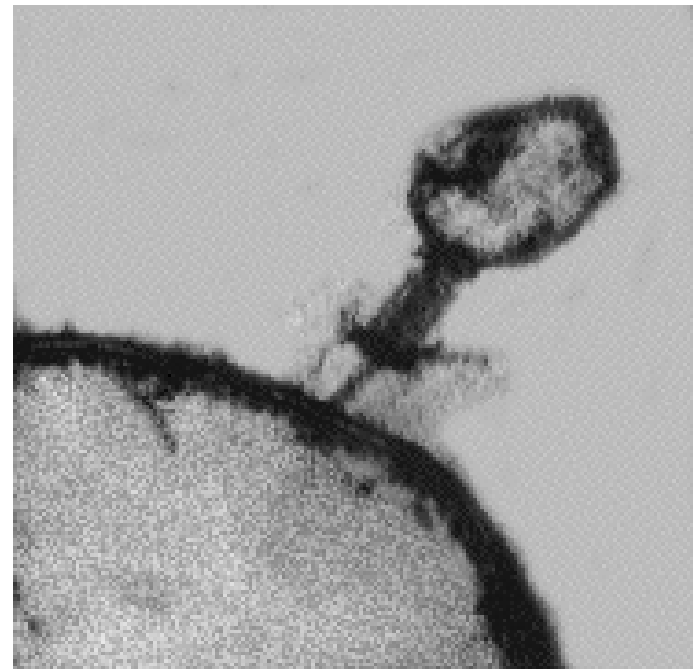
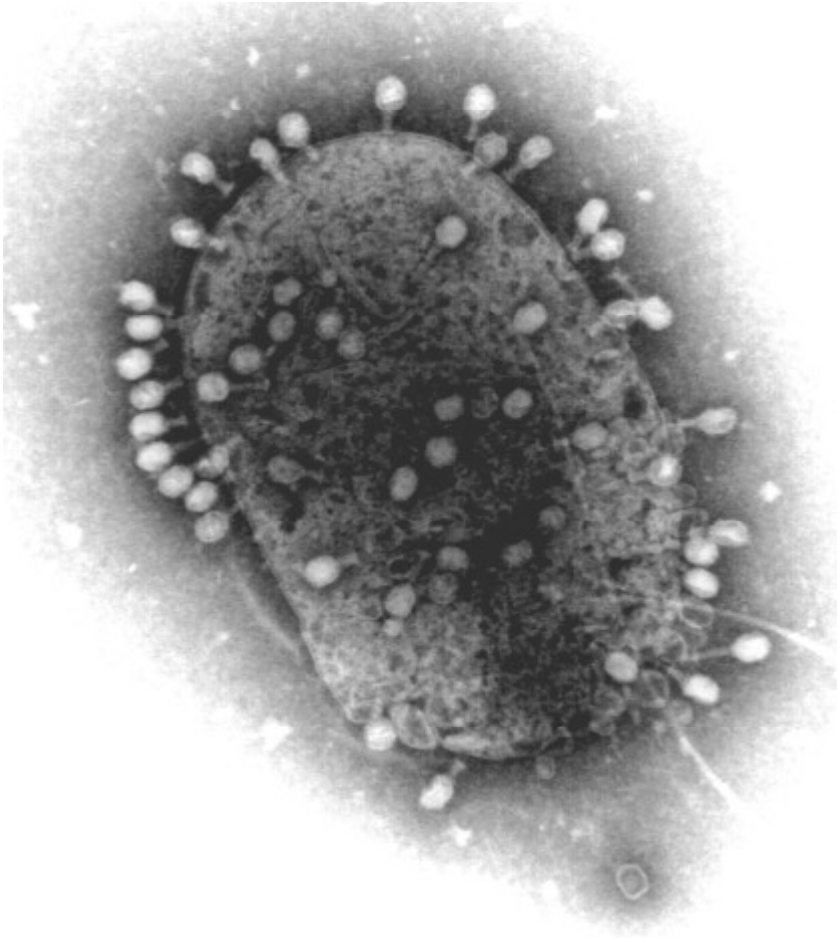


We design systems like these considering biological kinetics

Bacteria:
$$\frac{dX}{dt} = \mu_{\max} \frac{SX}{K_s + S} - k_d X$$

Growth Substrate:
$$\frac{dS}{dt} = -\frac{\mu_{\max}}{Y} \frac{SX}{K_s + S}$$

This approach does not consider the impacts of bacteriophage on the system dynamics



Copyright: CIMC

Add a few more equations and we can model the systems by considering the impact of bacteriophages

Susceptible Bacteria:
$$\frac{dX}{dt} = \mu_{\max}^X \frac{SX}{K_S^X + S} - k_d^X X - \delta P X (1 - e^{-\lambda})$$

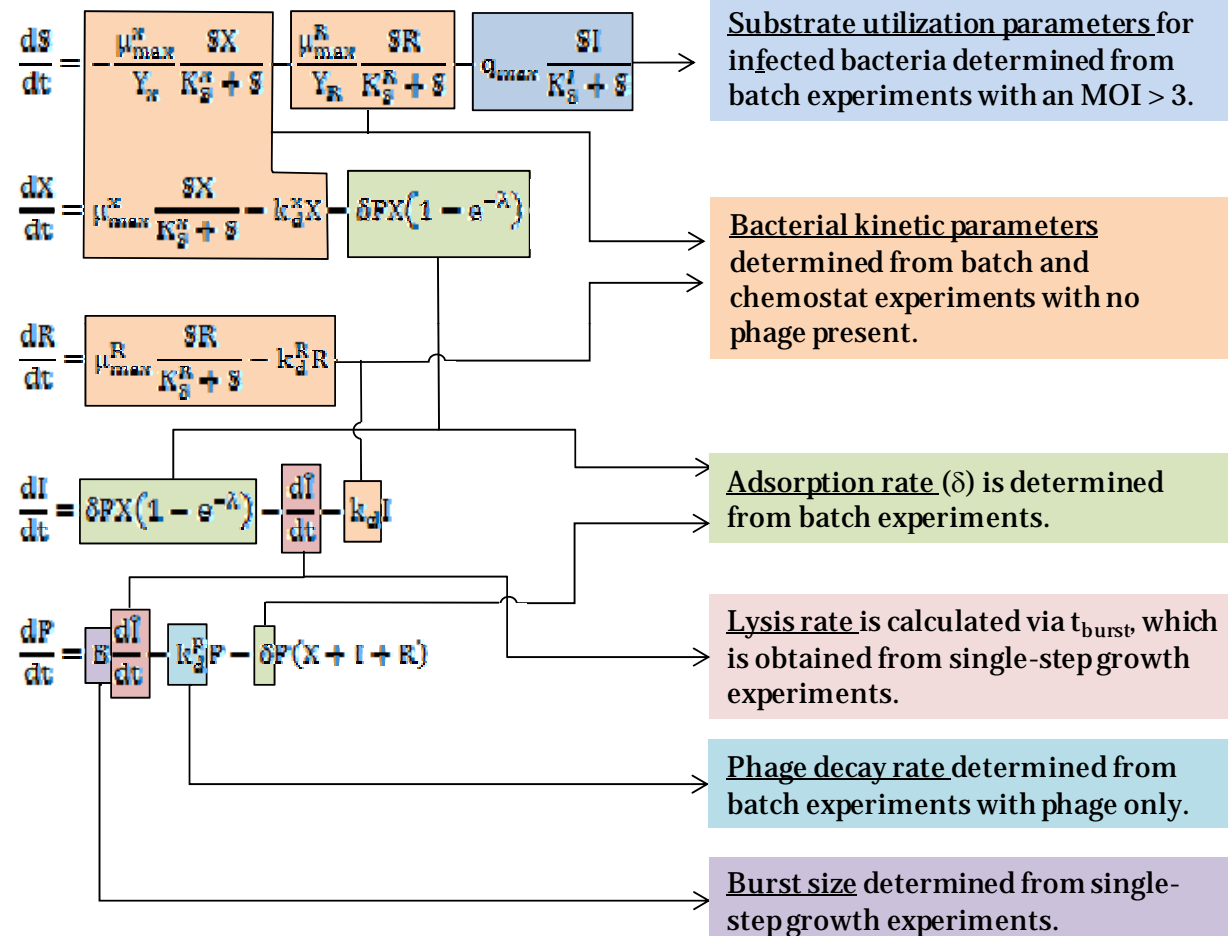
Infected Bacteria:
$$\frac{dI}{dt} = \delta P X (1 - e^{-\lambda}) - \frac{d\hat{I}}{dt} - k_d I$$

Resistant Bacteria:
$$\frac{dR}{dt} = \mu_{\max}^R \frac{SR}{K_S^R + S} - k_d^R R$$

Bacteriophages:
$$\frac{dP}{dt} = B \frac{d\hat{I}}{dt} - k_d^P P - \delta P (X + I + R)$$

Growth Substrate:
$$\frac{dS}{dt} = -\frac{\mu_{\max}^X}{Y^X} \frac{SX}{K_S^X + S} - \frac{\mu_{\max}^R}{Y^R} \frac{SR}{K_S^R + S} - q_{\max}^I \frac{SI}{K_S^I + S}$$

All the model parameters can be determined from independent experiments



Undergraduates are involved in all this research

- **Metabolic activity** – Long-term survival of attached and planktonic bacteria
- **Metabolic activity** – Development of method for determination of surface charge properties of macroscopic materials



- **Bacteriophage kinetics** – Effects of bacterial growth rate on phage burst size and time to burst.
- **Bacteriophage kinetics** – Factors that affect bacteriophage adhesion rate.

Theme...

You can learn a lot from
simple experiments

Environmental Biotechnology

