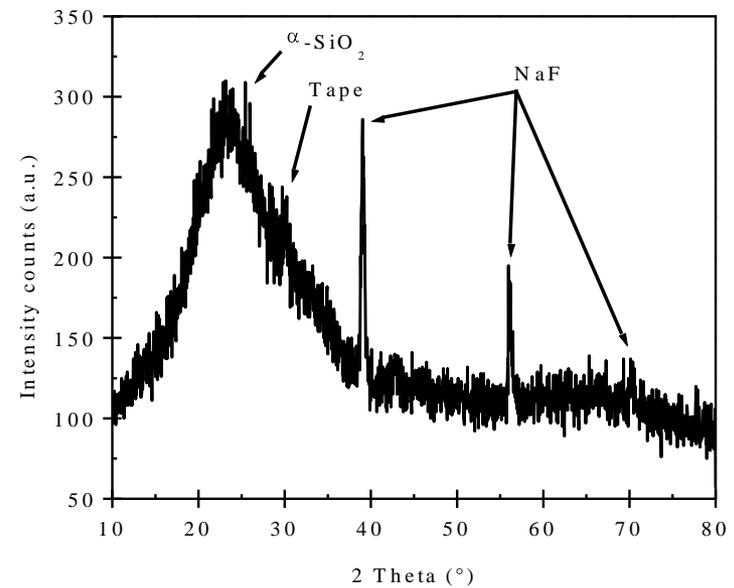
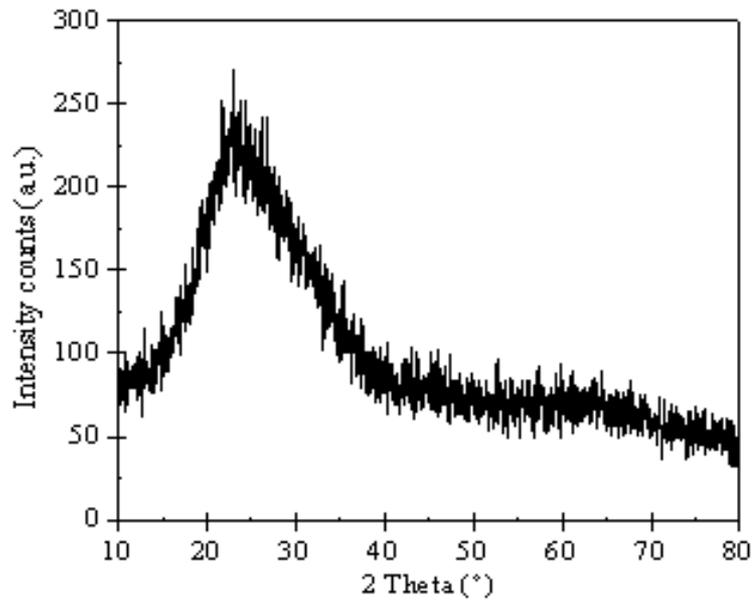


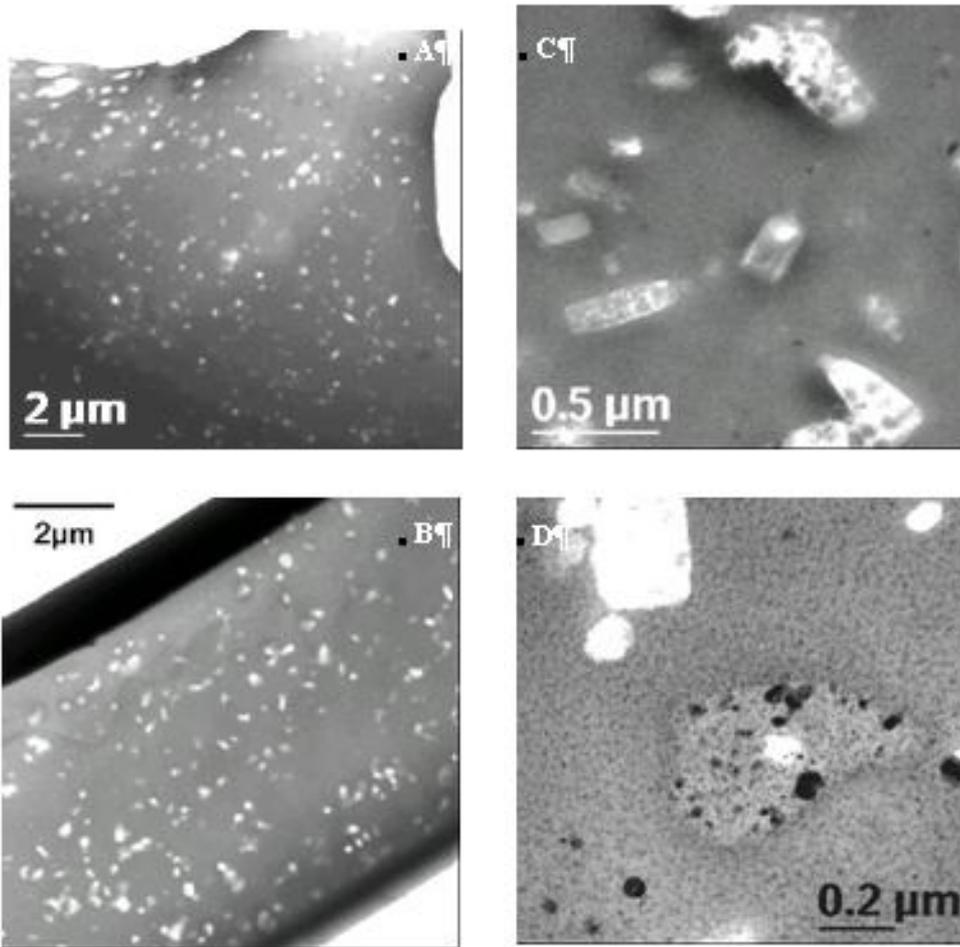
Crystallized phases: induced index change

Phase	PDF #	System	Group	a(Å)	b(Å)	c(Å)
NaF	36-1455	Cubic	Fm3m	4.633	4.633	4.633
NaBr	36-1456	Cubic	Fm3m	5.974	5.974	5.974
NaBr	27-0658	Cubic	N/A	12.133	12.133	12.133
Ag	04-0783	Cubic	Fm3m	4.086	4.086	4.086
Ag	41-1402	Hexagonal	P63/mmc	2.886	2.886	10.000
AgF	25-0762	Cubic	Pm3m	2.945	2.945	2.945
AgF	03-0890	Cubic	Fm3m	4.921	4.921	4.921
AgF	32-1004	Hexagonal	P63mc	3.246	3.246	6.226
AgF ₂	19-1134	Orthorhombic	N/A	5.813	5.529	5.073
AgF ₃	45-0159	Hexagonal	N/A	8.989	8.989	9.815
AgBr	06-0438	Cubic	Fm3m	5.774	5.774	5.774

XRD pattern of virgin and crystallized PTR



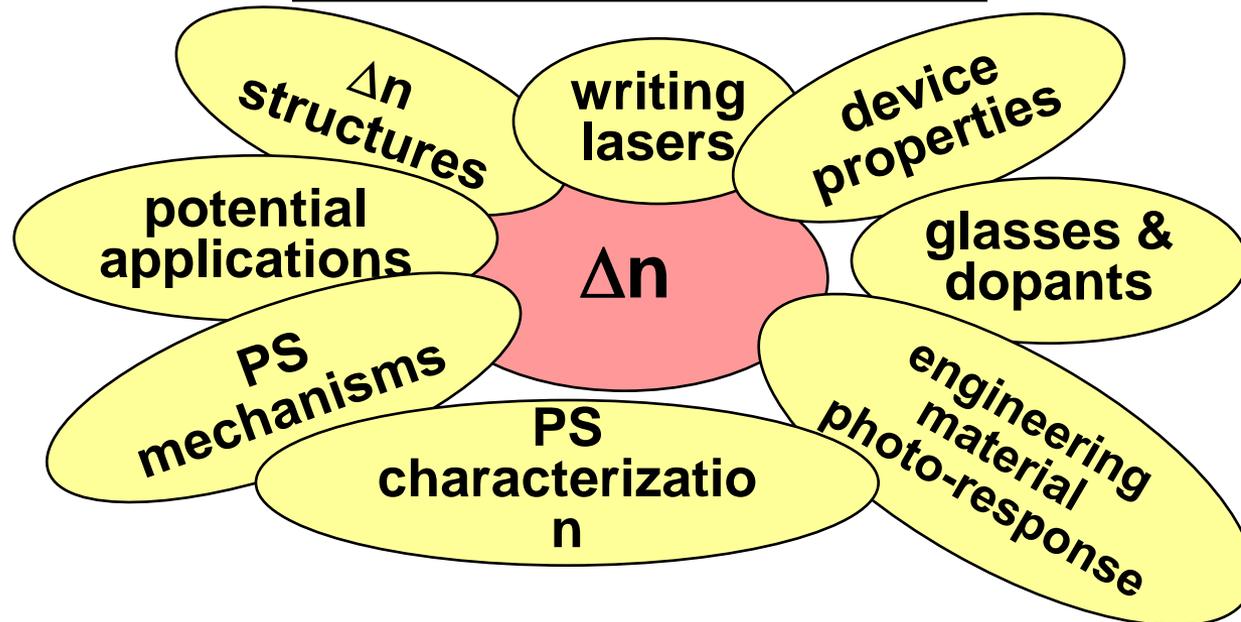
(A-B) Low- and (C-D) High-magnification TEM images of spontaneously crystallized PTR glass prepared by TP and FIB respectively.



Photosensitivity (PS)

permanent
refractive index change Δn
by laser exposure

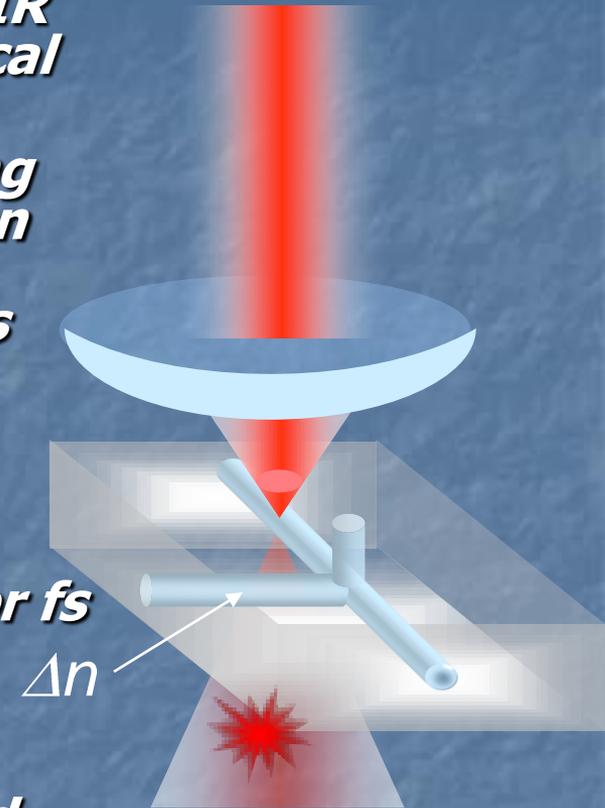
points of views



From "Photosensitivity, Fundamentals and Overview", H. Ebendorff-Heidepriem

Laser material modification: pulsed direct write or cw laser interaction, ablation

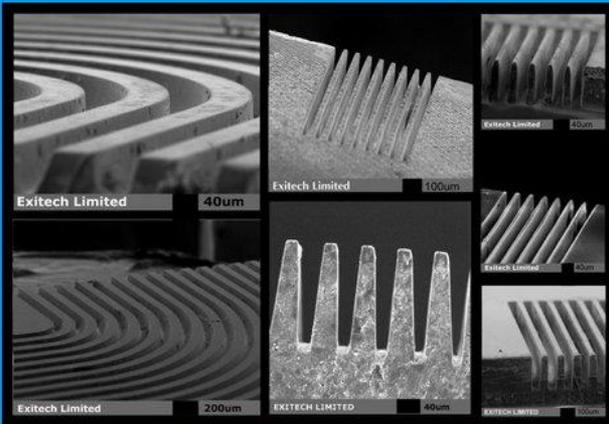
- *Focusing a cw source or a femtosecond near-IR beam in a transparent material produces a local change of the refractive index*
- *fs-regime writing allows volumetric processing and minimizes thermally induced defects often seen in ns experiments; lack of thermal "damage" to material results in clean features*
 - *Glass structure reorganization (bond bending and/or breaking)*
 - *Photoexpansion or densification*
 - *Refractive index modification (+ or -)*
- *Sub-micron precision $0.5 \mu\text{m}$ demonstrated for fs (Schaffer et al., Opt. Lett. 26, 2001)*
- *Real time serial fabrication, 3-D structuring possible, not amenable to high volume processing due to limitations of writing speed*



ns or other
"conventional"
exposure
faculty@university

fs exposure
with minimal debris
and thermal

Microchannels

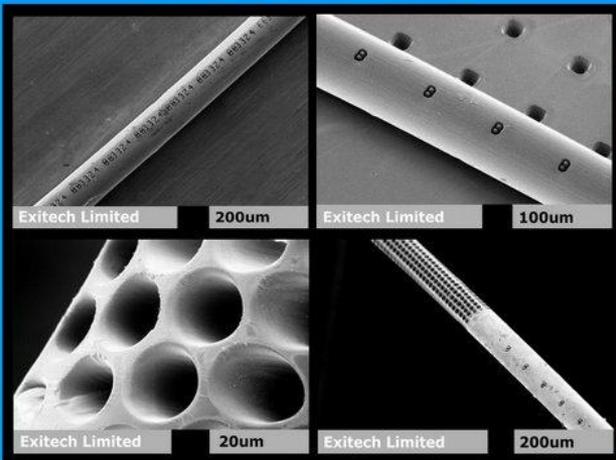


Examples of Processing of Glass and Ceramic Materials

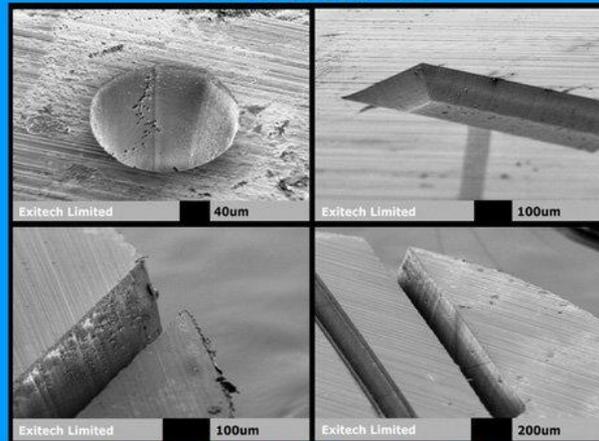
Excimer Laser Patterning



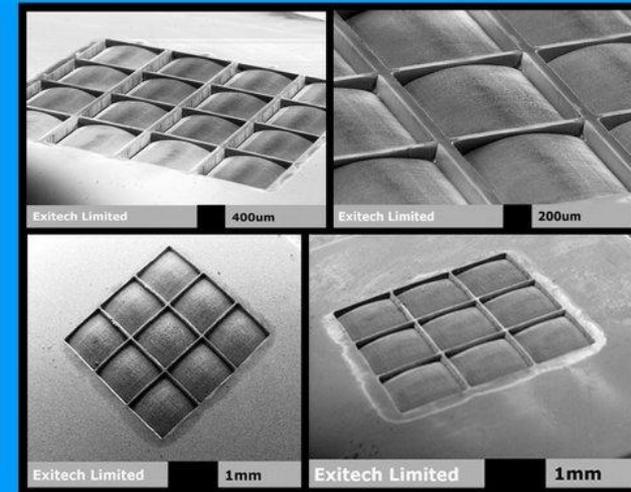
Optical Fibres



Femtosecond Laser Micromachining Ceramics



Microlenses



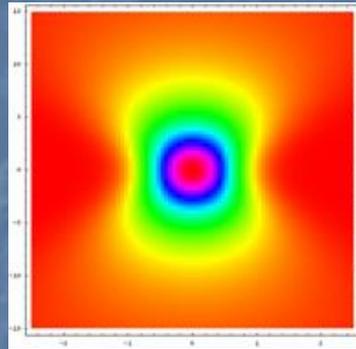
Courtesy Exitech Corp.

Spectra Physics "Hurricane" Laser

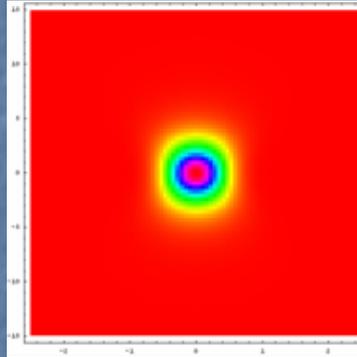
Really Need True 3D Patterns and a Cost Effective Processing Approach

Two regimes of *direct writing*

Dependence of axial shape of structural modification on writing approach



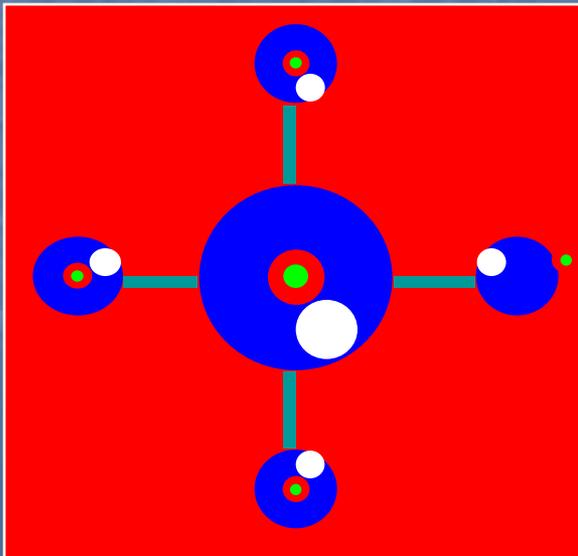
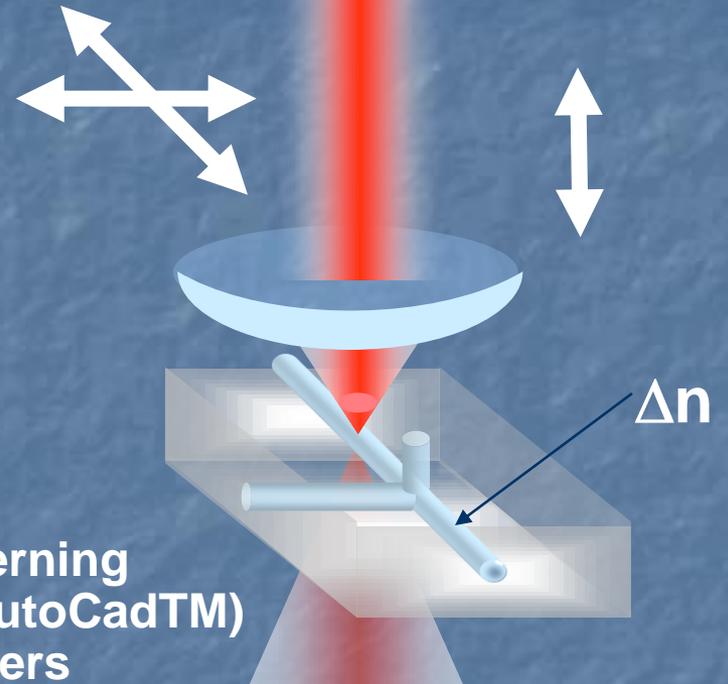
1-photon



2-photon

transverse writing

longitudinal writing



Direct-Write Patterning Using Various CAD (AutoCad™) Patterning Layers

- 355 nm
- 266 nm (high dose)
- 266 nm (low dose)
- 248 nm

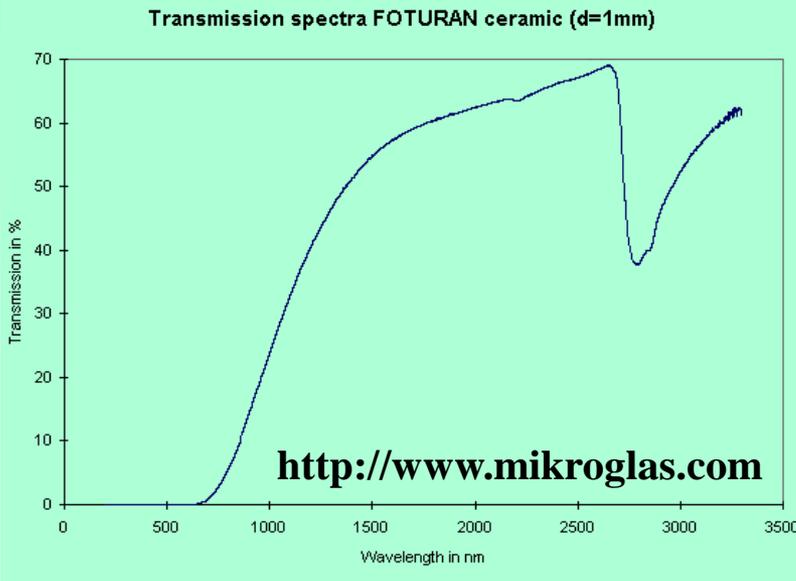
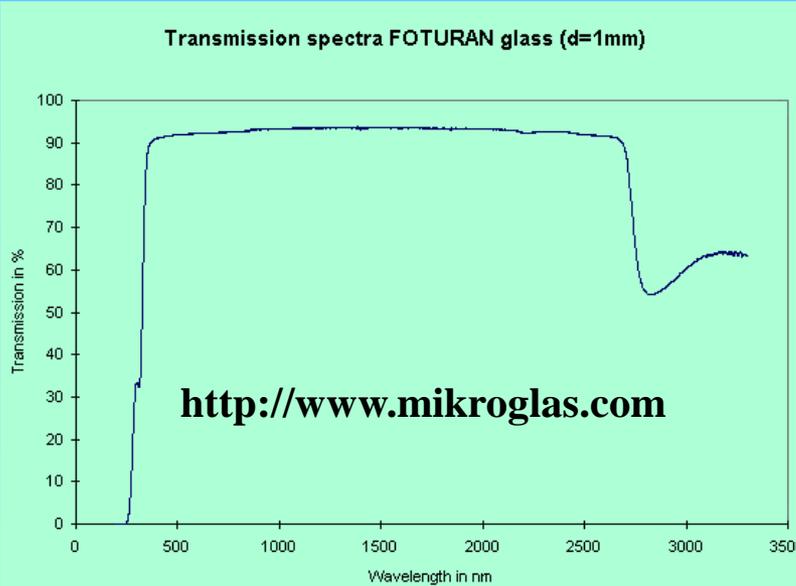
Additional "Layers" that can be added

Platinum metal deposition

Imbedded exposure

What is a Photostructurable Glass Ceramic Material or Photoceram

Example: Foturan™ (Schott Corp.)



Property	Foturan in the Vitreous State
Young's Modulus	$78 \times 10^3 \text{ N/mm}^2$
Poisson's Ratio	0.22
Knoop Hardness	4600 N/mm^2
Modulus of Rupture (MOR)	60 N/mm^2
Density	2.37 g/cm^3
Thermal Expansion	$8.6 \times 10^{-6}/\text{K}$
Thermal Conductivity	$1.35 \text{ W/mK @ } 20^\circ\text{C}$
Specific Heat	$0.88 \text{ J/gK @ } 25^\circ\text{C}$
Glass-ceramic Transformation Temperature	465°C
Electrical Conductivity	$8.1 \times 10^{12} \text{ Ohm-cm @ } 25^\circ\text{C}$ $1.3 \times 10^7 \text{ Ohm-cm @ } 200^\circ\text{C}$
Dielectric Constant	$6.5 @ 1\text{MHz, } 25^\circ\text{C}$

Schott/SGT April 2002

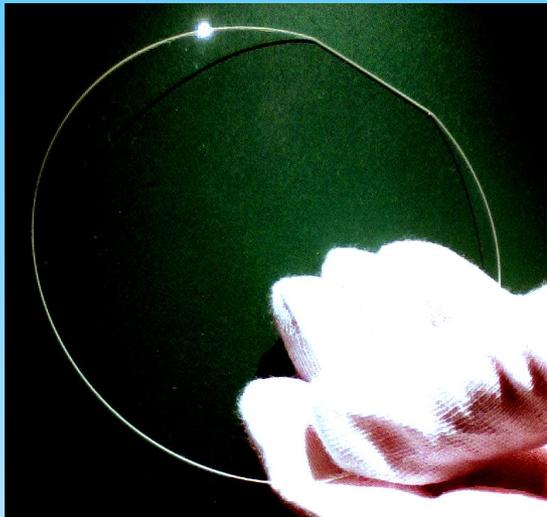
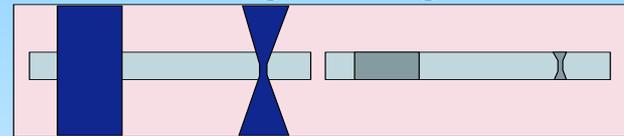
Copyright © 2002 by The Aerospace Corporation



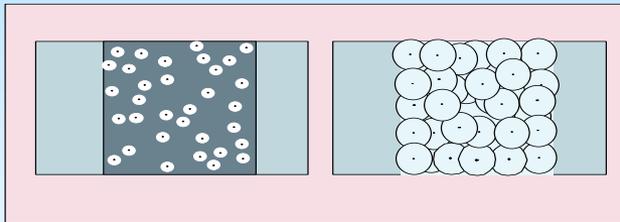
Processing Photoceramic Glasses

Typical Process Flow

Step 1: Illumination/Latent Image

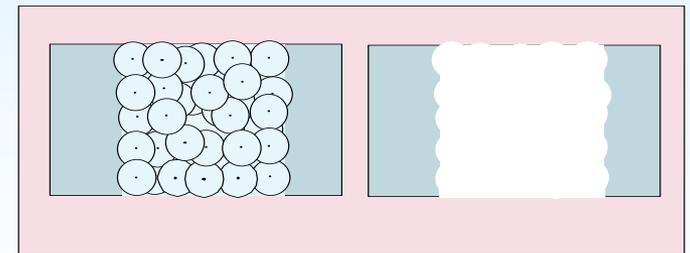


Step 2: Ceramization to a Meta-Silicate

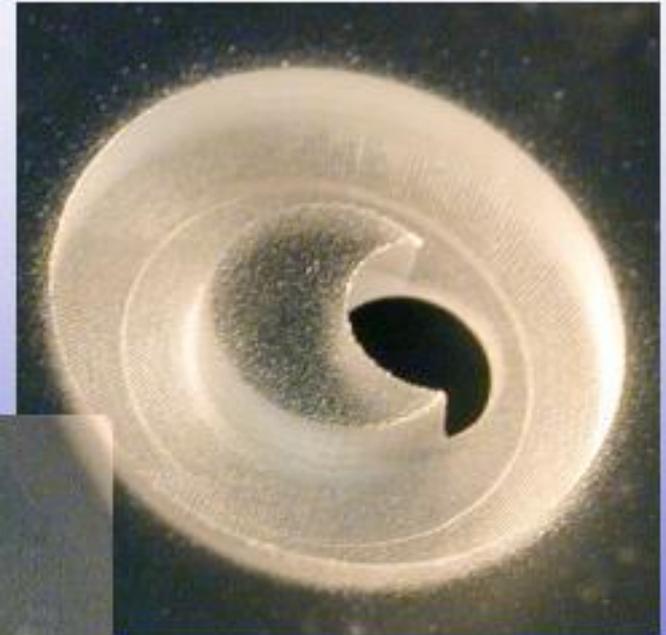
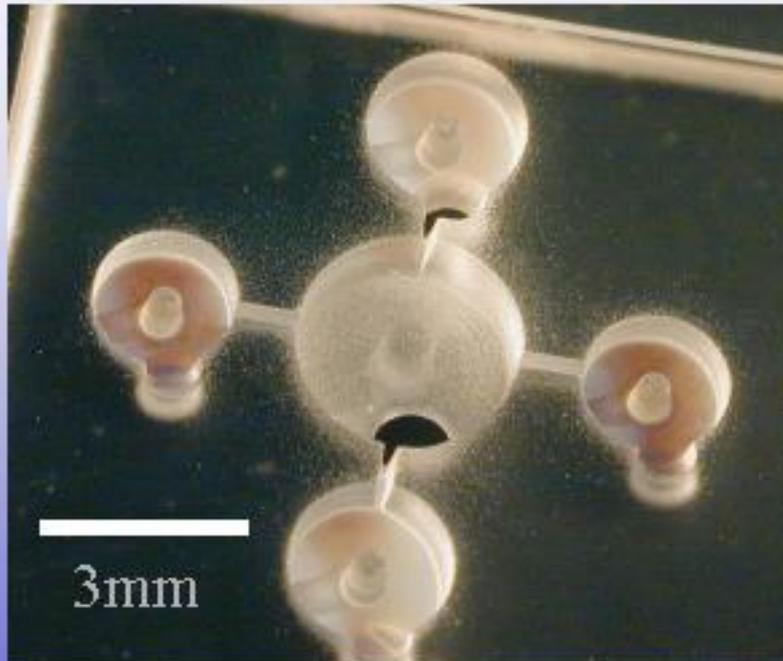


Step 3: Preferential Isotropic Etching

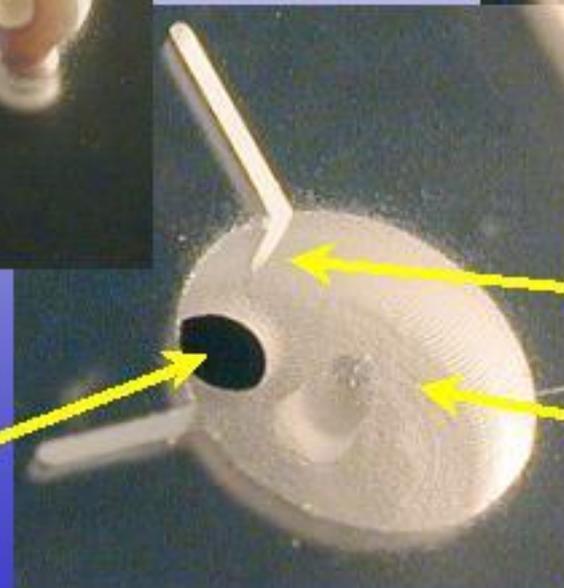
- Crystalline Li_2SiO_3 dissolves 20x faster than the amorphous glass in 5% hydrofluoric acid.
- $\text{Li}_2\text{SiO}_3 + 3\text{HF} \rightarrow 2\text{LiF} + \text{H}_2\text{SiF}_6 + 3 \text{H}_2\text{O}$



Exposure at Multiple Wavelengths Can Result in the Fabrication of Unique 3D Patterns



**Via hole through
1mm thick wafer
@355nm**



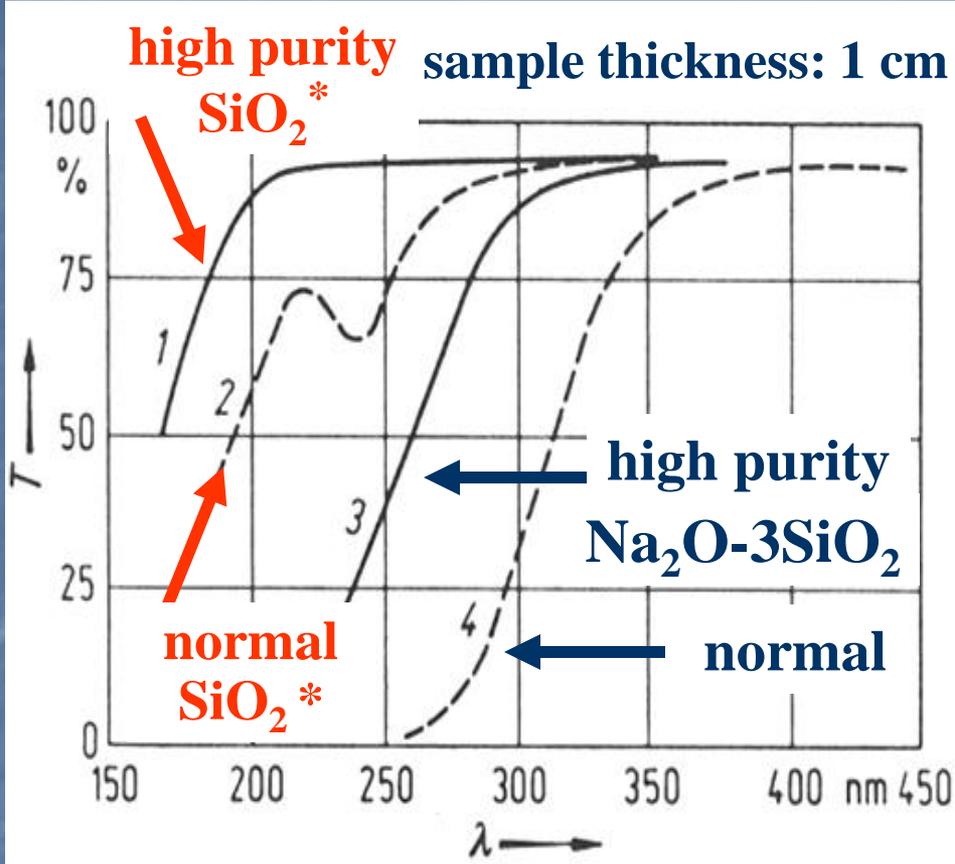
**500 microns deep
@248 nm**

**700 microns deep
@266nm**

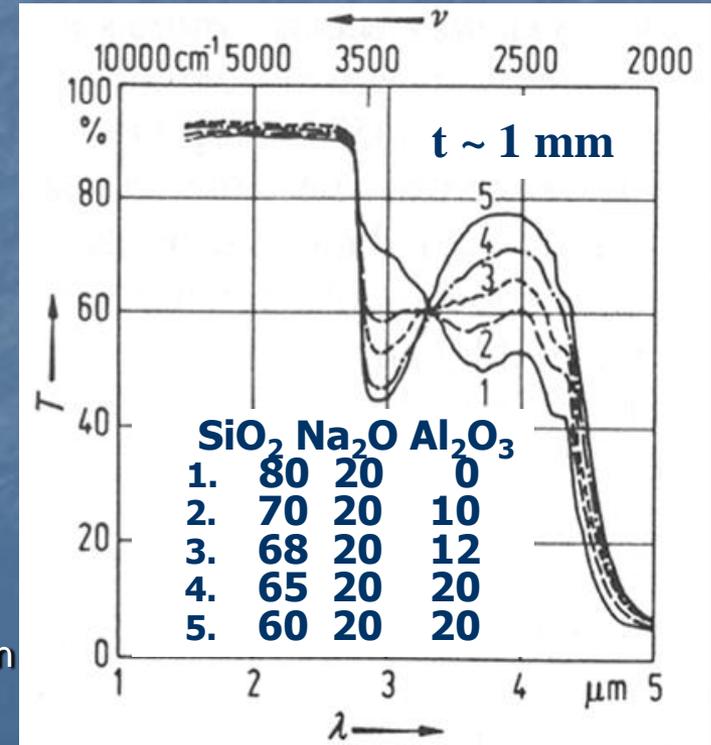
So...as we'd expect

- Chemistry dictates the structure of the material (purity matters)
- Structure dictates the properties
- Optical properties are dictated by chemistry and processing route (thermal history dictates $V, \rho \Rightarrow n$); impurities define intrinsic absorption properties (α, α_2)
- Thus...material's **photo-response** will be dependent on all of these attributes

What does this mean to absorption? network formers and modifiers



- Additions of Al_2O_3 and B_2O_3 improve the tetrahedral network structure, consuming NBO's and move the UV edge back up to higher frequencies.
- PbO which is present in moderate concentrations in may flint optical glasses, shifts (ν) the UV edge significantly.



• **Network:** SiO_2 **Modifier:** 25 mol% Na_2O

* BO: O-Si-O (covalent)

• Formation of NBO with alkali addition

NBO: Si-O⁻ Na⁺ (ionic-like)

• Lower BE electrons (red) shift UV edge

higher field strength ions shift less

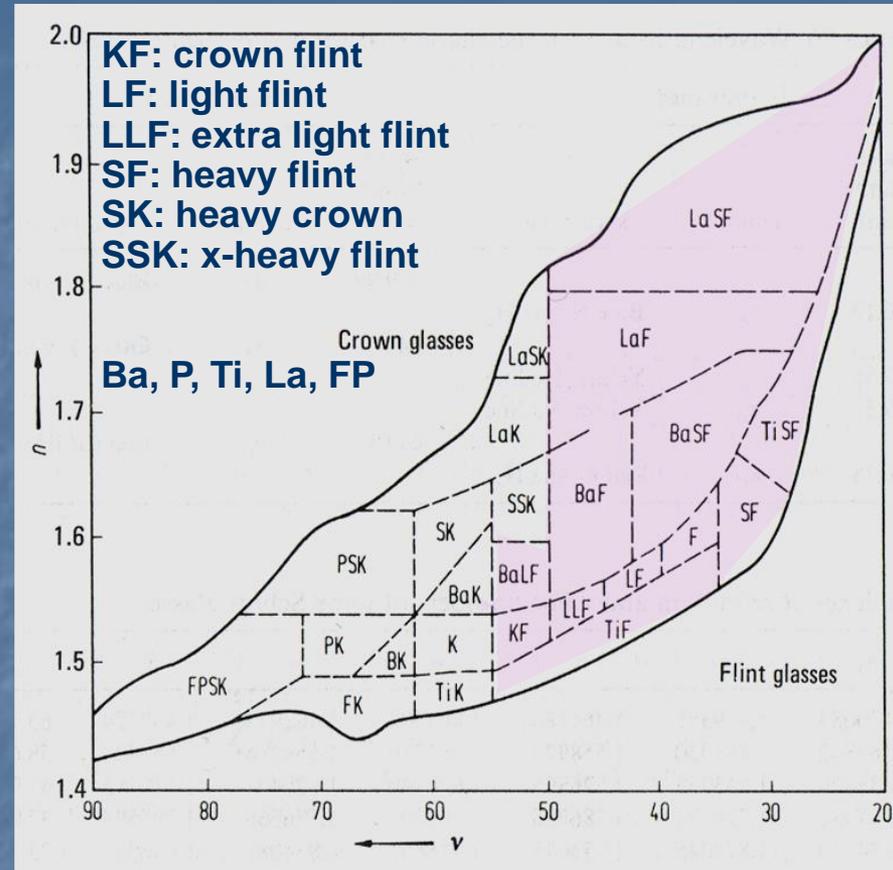
$\text{Li} < \text{Na} < \text{K}$, etc.

• Impurities: Fe, Mn, etc.

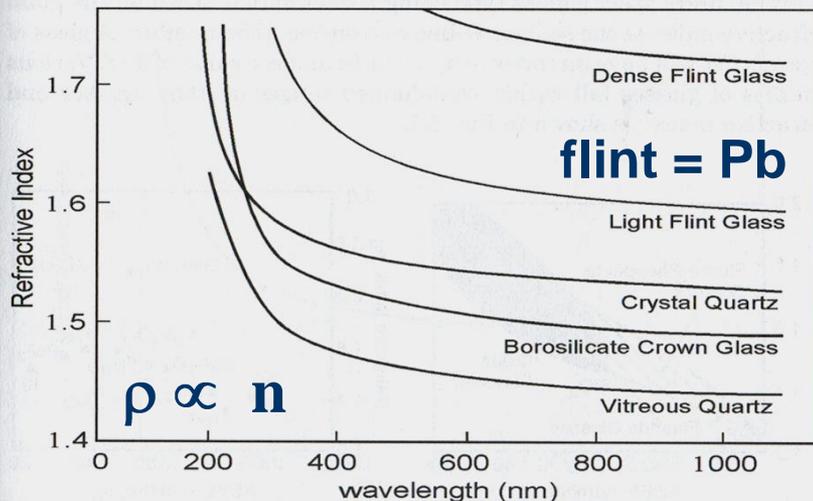
Absorption and Dispersion

Network UV edge position

- SiO₂ glass 155-160 nm (7.8 eV)
- GeO₂ glass 200 nm (6.2 eV)
- B₂O₃ glass 172 nm (7.2 eV)
- P₂O₅ glass 145 nm (8.6 eV)
 - P₂O₅ has a tetrahedral structure with a double-bonded oxygen
- Al₂O₃ sapphire (single crystal, annealed film)
 - sc 145 nm (8.55 eV)
 - film 182 nm (6.8 eV)



Dispersion of Typical Optical Glasses



- Li₂O-SiO₂ glass 188 nm (6.6 eV)
- Foturan: Ce, Al₂O₃, Ag, Zn, Sn
- Na₂O-SiO₂ glass 207 nm (6.0 eV)
- K₂O-SiO₂ glass 214 nm (5.8 eV)

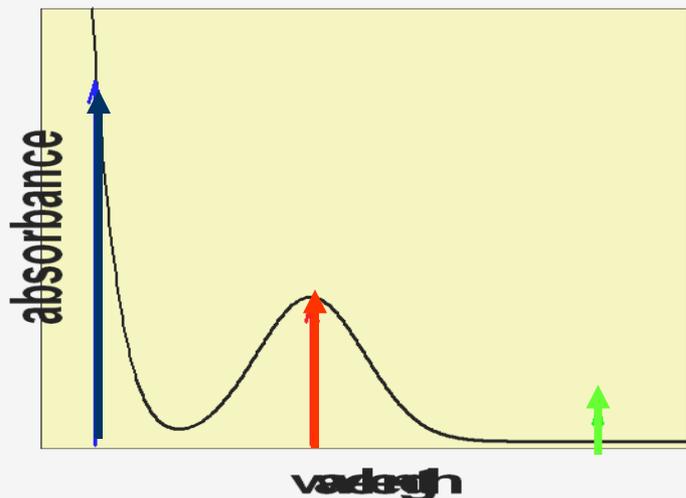
ss: Section being

Photo-induced property changes

- Exposure ($h\nu$) induced:
 - Structural reorganization (bond bending);
reversible **As_2S_3**
 - Structural reorganization (bond breaking)
permanent **As_2S_3 and other glasses**
 - Structural reorganization (melting and solidification: cooling rate causes $\Delta V, \Delta n$)
 - Crystallization - realized through exposure and heat treatment= \Rightarrow to yield new phase with:
 - Refractive index variation (Δn crystal $\neq \Delta n$ glass) **PTR**
 - Creation of a new phase with etch rate (contrast)
different than glass

Material absorption: response to laser light network structure, dopants

Material absorption spectra



spectral range	wave-length	laser type	Regime
VUV	157nm	F ₂	pulsed
UV	193nm	ArF excimer	pulsed
	244nm	Ar ⁺ 2.Harmonic	cw
	248nm	KrF excimer	pulsed
	266nm	Nd:YAG 3.Harm.	pulsed
	325nm	HeCd	cw
VIS	457 – 488nm	Ar ⁺ various lines	cw
NIR	800nm	Ti:sapphire	fs

UV-Edge

Excitation: $\lambda_{\text{edge}} \leq \lambda_{\text{Laser}}$
 $\lambda_{\text{glass}} \text{ band-gap}$

Ge-SiO ₂	157nm
PbO-SiO ₂	244 & 266nm
Zr-Ba-F	193 nm
Ga-La-S	244 nm
As-S	550 nm

Defect, Dopant absorption

$\lambda_{\text{defect,dopant}} \approx \lambda_{\text{Laser}}$

selective excitation

Ge-SiO ₂	244 & 248 nm
Eu ²⁺ , Ce ³⁺	244 & 248 nm
Eu ³⁺	466 nm
Ag ⁺	420 nm

Laser (writing) wavelength

$\lambda_{\text{laser}} \gg \lambda_{\text{glass}}$

Single vs multi-photon processes	
glass	800 nm
Ge-SiO ₂	488nm

Dopants/impurities and spectral regimes

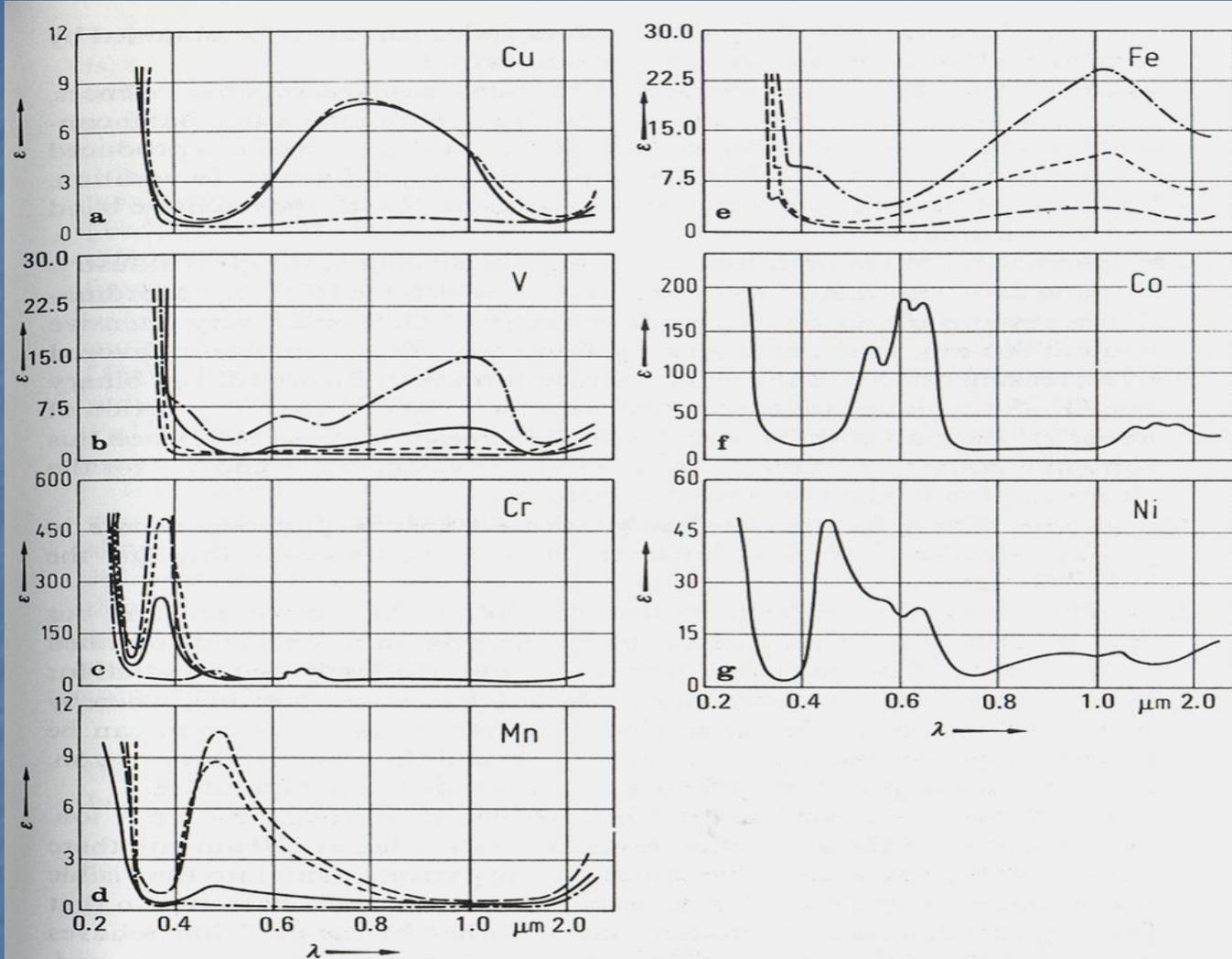


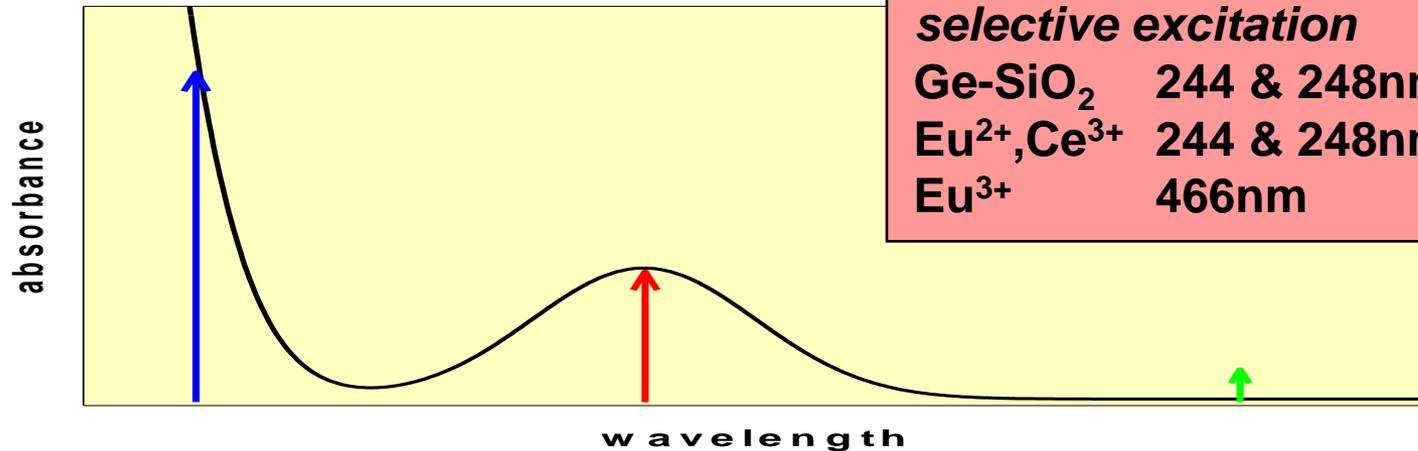
FIGURE 109a-g. Spectra of some transition elements in sodium silicate glasses, from Bamford [41, 42] (ϵ = molar normal extinction coefficient in $1/[\text{mole cm}]$).

— Na_2O content about 15 mole % } Melted in oxidizing atmosphere
 - - - Na_2O content about 40 mole % }
 — Na_2O content about 15 mole % } Melted in reducing atmosphere
 - - - Na_2O content about 40 mole % }

Now...what happens upon exposure to light?

- Absorption and other properties of material
- Form of the material (bulk, film, fiber)
- Desired modification we want
- Exposure conditions
 - Permanent, reversible, ablative

Glass \leftrightarrow Laser



Defect, Dopant

$$\lambda_{\text{defect,dopant}} \approx \lambda_{\text{Laser}}$$

selective excitation

Ge-SiO₂ 244 & 248nm

Eu²⁺, Ce³⁺ 244 & 248nm

Eu³⁺ 466nm

Transmission-Edge

$$\lambda_{\text{edge}} \geq \lambda_{\text{Laser}}$$

band-gap, strong absorption

Ge-SiO₂ 157nm

PbO-SiO₂ 244 & 266nm

Zr-Ba-F 193nm

Ga-La-S 244nm

High Transmission

$$\lambda_{\text{glass}} \gg \lambda_{\text{Laser}}$$

nonlinear absorption

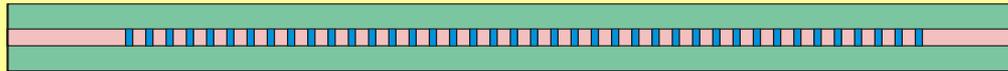
glass 800nm fs

Ge-SiO₂ 488nm

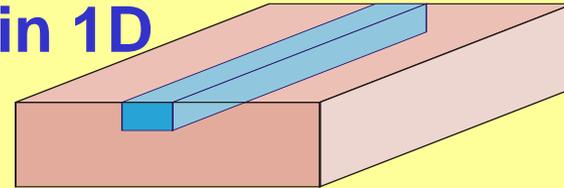
Δn structures

➤ grating: Δn modulation

- Bragg gratings: $\Lambda \geq 1 \mu\text{m}$
- Long period gratings: $\Lambda = \text{several } 100 \mu\text{m}$



➤ waveguides: Δn constant in 1D



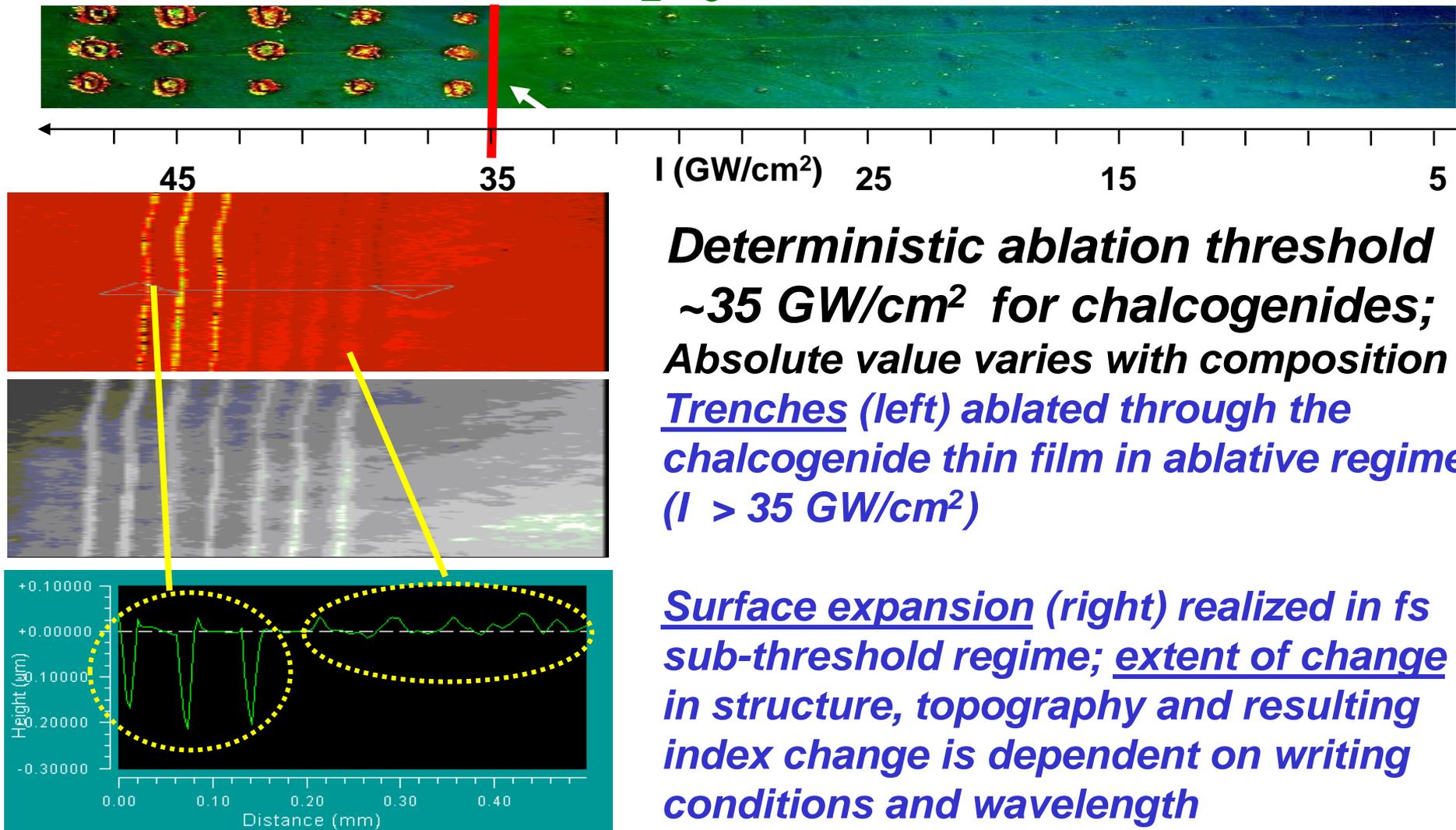
➤ combination

1. waveguide
2. Bragg grating

➤ uniform exposure: Δn constant

→ PS characterisation

Two distinct processing regimes of fs exposure: As₂S₃ films

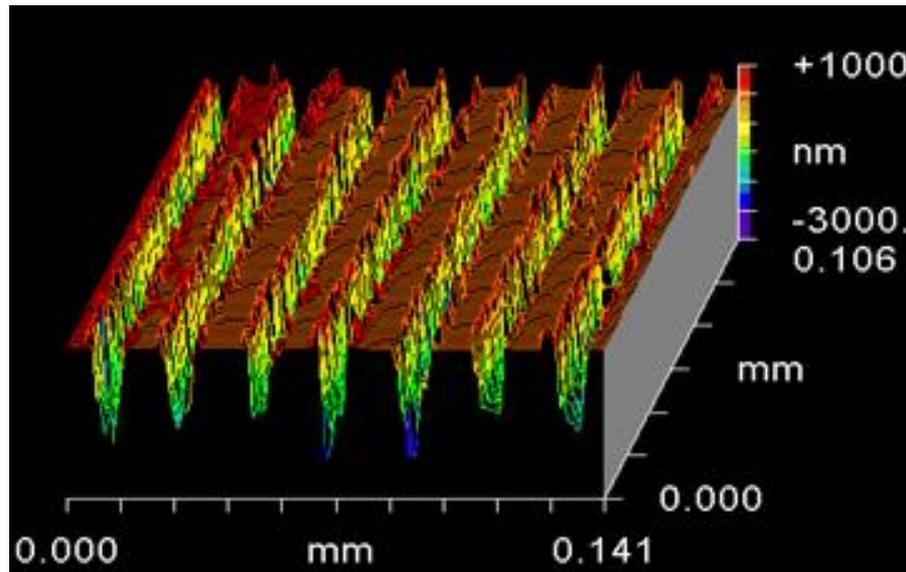


Deterministic ablation threshold
~35 GW/cm² for chalcogenides;
Absolute value varies with composition
Trenches (left) ablated through the
 chalcogenide thin film in ablative regime
 ($I > 35 \text{ GW/cm}^2$)

Surface expansion (right) realized in fs
 sub-threshold regime; extent of change
 in structure, topography and resulting
 index change is dependent on writing
 conditions and wavelength

Direct write fs laser micro-fabrication in As_2S_3

Micro-ablation of relief features (grating)

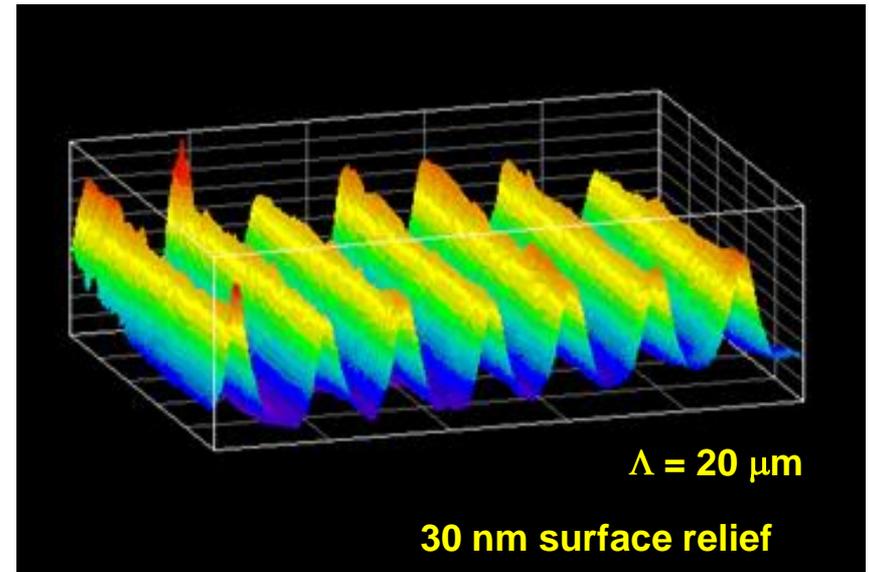


- Surface profile (Zygo New View white light interferometer microscope)
- Typical width of exposure features $\sim 10 \mu\text{m}$ (FWHM)

“Microfabrication of waveguides and gratings in chalcogenide thin films,” A. Zoubir et al., Technical Digest. CLEO pp 125-126 (2002)

“Direct femtosecond laser writing of optical waveguides in As_2S_3 thin films,” A. Zoubir, M. Richardson, C. Rivero, A. Schulte, C. Lopez, K. Richardson, Optics Letters 29 7 (2004)

Micro-restructuring of material Photo-induced expansion (phase grating)

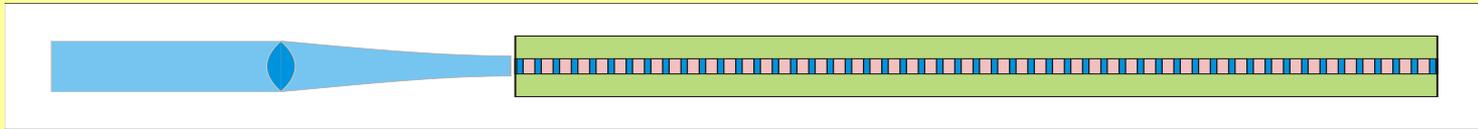


Design and Dimensions

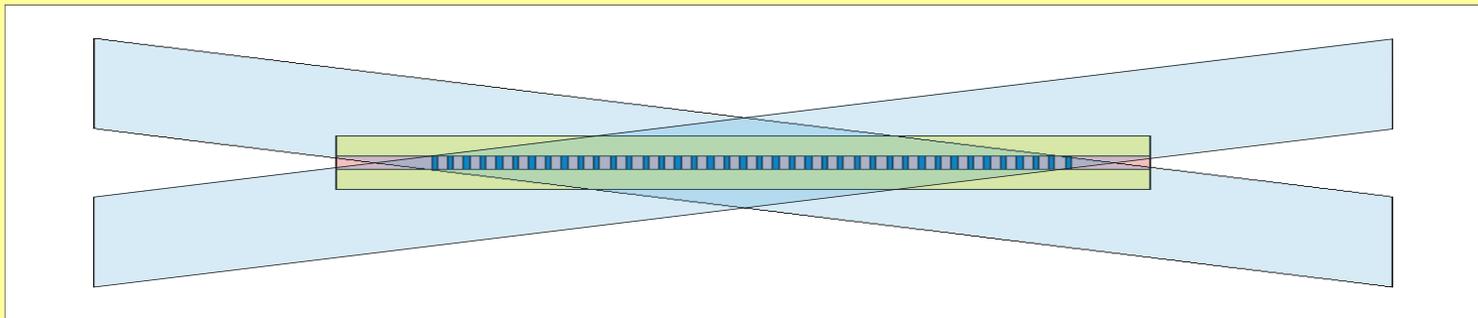
Δn structure		starting device	
<i>Bragg gratings</i>			
1D	Fiber Bragg gratings	1D	single-mode fibre
			channel waveguide in planar device
2D	planar gratings	2D	thin film on substrate
	grating limited to exposed surface	3D	bulk
3D	volume gratings, holograms	3D	bulk: $d = 2 - 7 \text{ mm}$ $d = 100\text{-}200\mu\text{m}$
<i>Long period gratings</i>			
1D	LPG in fibre	1D	single-mode fibre
<i>Waveguides</i>			
1D	channel	2D	thin film on substrate
		3D	bulk: $E_{\text{laser}} > E_{\text{band-gap}}$
>1D	multi-mode	3D	bulk: $E_{\text{laser}} < E_{\text{band-gap}}$

Fabrications of Gratings

internal (longitudinal)
self-written due to standing wave interference



external
interferometric, phase-mask, point-by-point



Glasses for Gratings

silica-based $\Delta n=10^{-5...3}$

- fibers, thin films → 1D or 2D gratings
- Ge-SiO₂: GODC @ 240nm
codopants: B, Sn 😊 / P ☹️
- Al-SiO₂: RE(Ce) doped
- P-SiO₂: Sn dopant
- H₂ treatment → PS increase

heavy metal fluoride $\Delta n=10^{-5...4}$

- undoped, $\lambda_w=193\text{nm}$ → 2D abs.-limited
- Ce³⁺, Eu²⁺ → fibres, thin film

oxide(Si,B,P,Ge):Eu³⁺ $\Delta n=10^{-6...5}$

- volume gratings by 466nm-laser

heavy metal oxide $\Delta n=10^{-2}$

- PbO-SiO₂, $\lambda_w=\text{UV}$, bulk → 2D abs.-limited
- PbO-GeO₂, $\lambda_w=\text{UV}$, thin film → 2D

Na-silicate/phosphate $\Delta n=10^{-?}$

- ion-exchange → wg → 1D or 2D grating

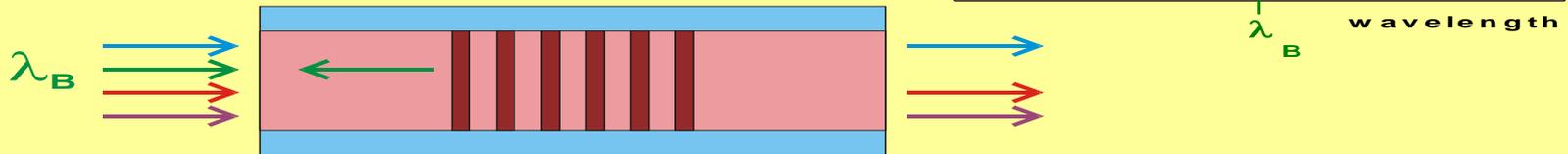
Fiber Bragg Gratings

uniform

const. grating period Λ

const. Δn_{mod} amplitude

phase fronts perpendicular



selectively reflected $\lambda_B = 2 n_{\text{eff}} \Lambda$

reflectivity $R = \tanh^2 (k \times \Delta n \times L)$

$R < 100\% \rightarrow \Delta\lambda \sim 0.1\text{nm} / R > 100\% \rightarrow \Delta\lambda$ broader

\Rightarrow diverse applications based on
selective separation of closely spaced λ 's

other types

variation of Λ , Δn_{mod} , phase front direction

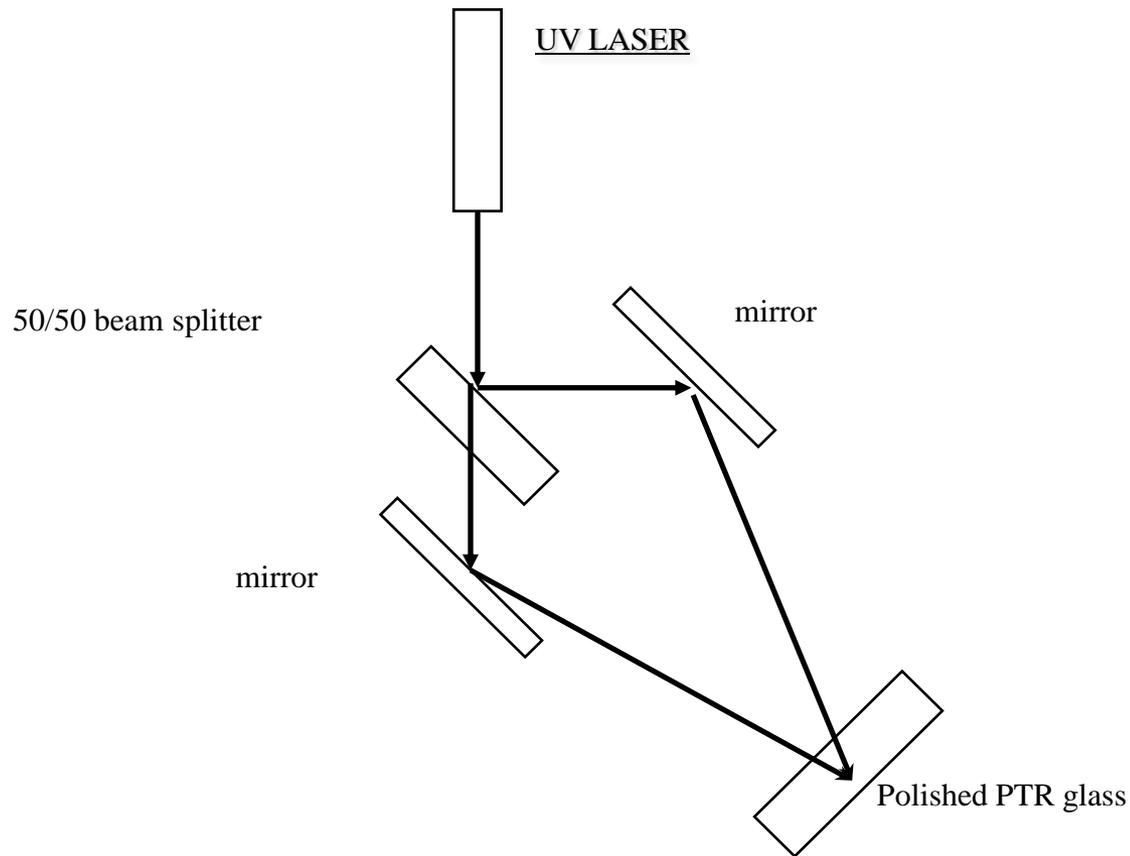
multiple gratings, phase shift gratings

temperature and strain dependence

n and $\Lambda = f(T, \text{strain}) \rightarrow \lambda_B = f(T, \text{strain})$

😊 sensors \leftrightarrow ☹️ DWDM

Schematic experimental set-up for hologram and grating writing



1D Gratings in planar devices

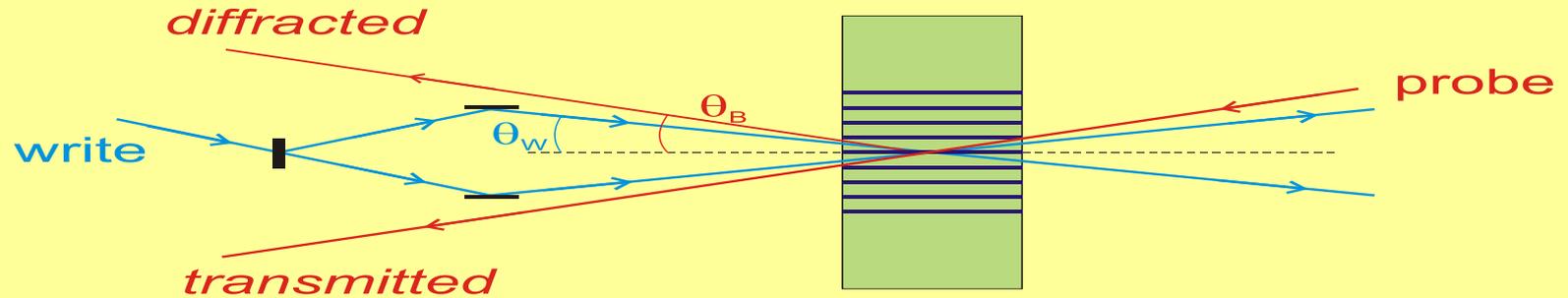
short device length
↓
short grating length needed
↓
high Δn required

e.g. $R=95\% \rightarrow \Delta n \cdot L = 1 \cdot 10^{-3} \text{ mm}$

	<u>fiber</u>	<u>planar</u>
Δn	$5 \cdot 10^{-5}$	$1 \cdot 10^{-3}$
L	2cm	1mm

Planar and volume Gratings

diffraction efficiency $\eta = I_R / I_0$



➤ planar wave approximation:

$$\eta = \sin^2 (\pi \cdot \Delta n \cdot L / \lambda_p / \cos \theta_B)$$

L = grating length → sample/film thickness

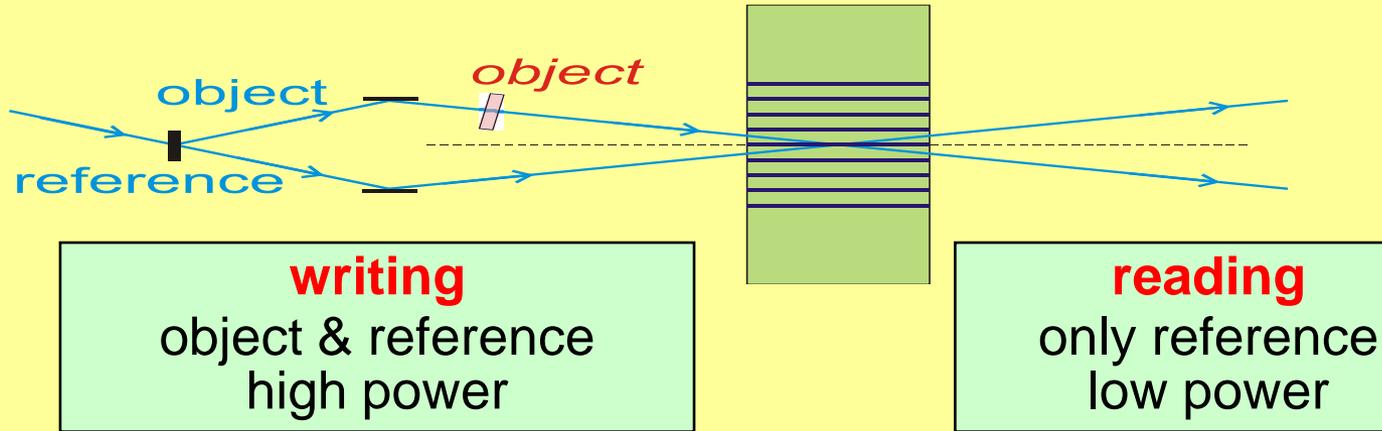
➤ absorption in the sample

→ effective thickness = $1 / \alpha_{\text{laser}}$

→ $\Delta n(z) = n_0 \cdot \exp(-\alpha_{\text{laser}} \cdot z)$

Planar and volume Gratings

holographic information storage



advantages of glass:

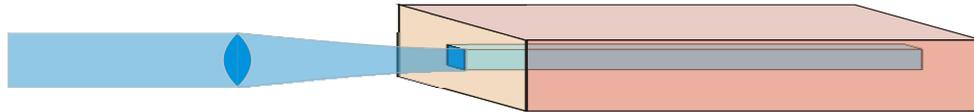
- long-time storage
 - room temperature operation
 - multiple readings without degradation
- if $\lambda_w \neq \lambda_{\text{peak}}$ of Eu^{3+}

demultiplexing

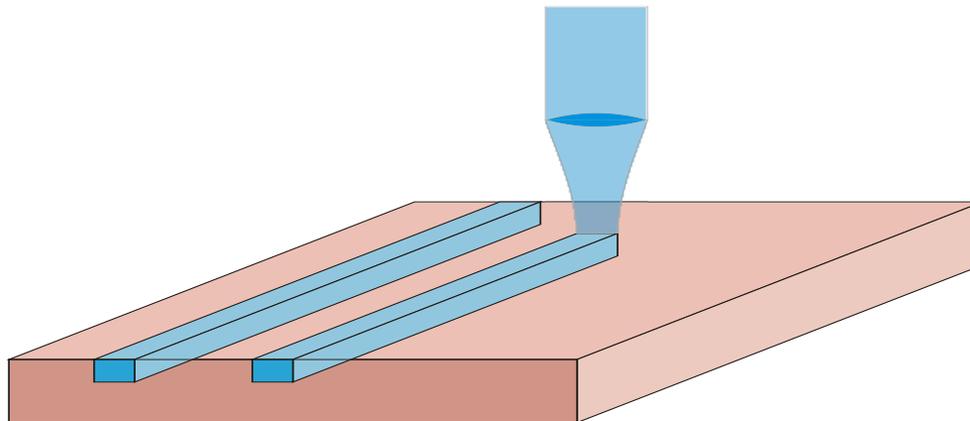
- frequency selective filters
 - tuning by sample rotation and tilting
- different $\theta_i \rightarrow$ different $\lambda_i = 2 \cdot \Lambda \cdot \sin \theta_i$

Fabrication of Waveguides

self-induced, self-written



Direct write by sample translation



Waveguides

applications

fabrication of channel waveguides
in integrated optical devices

- easy and fast process
- no sharp bends → low rad. losses

self-writing:

- buried waveguides in one step
- complex structures (Y-couplers, tapers)
by tailoring the writing beam shape

waveguide characterization

- waveguide image and mode-profile
- surface changes by AFM and profilometer
- Δn measurements:
from NA but modelling complex mode-profiles?
from beam output narrowing during self-writing

Glasses for Waveguides

GaLaS, FP:Ce,Eu, PbO-SiO₂, Ge-SiO₂

- 244nm cw, direct-writing

Na-borosilicate:Nd, Ge-SiO₂

- 455-488nm cw
- high transmission → self-writing

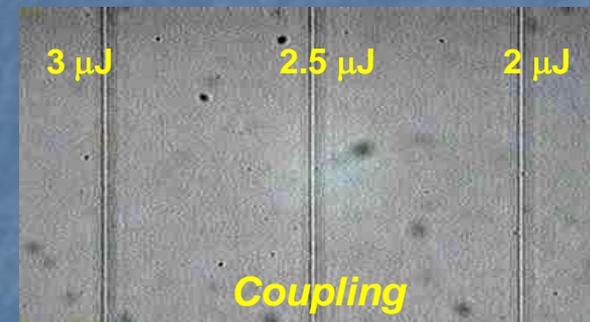
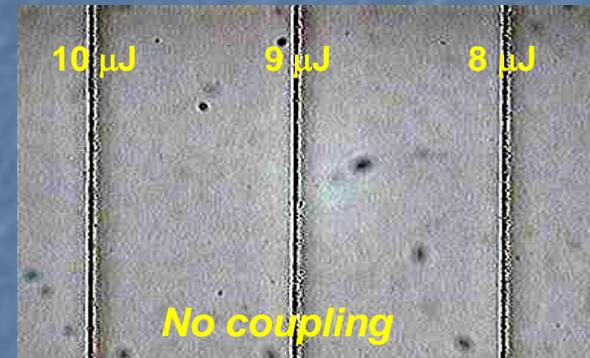
oxide, fluoride, sulfide

- 800nm fs, train of pulses

Material response: Direct-writing in fused silica

-tuning to absorption is only part of the issue

- Multi-photon exposure conditions
- 800 nm fs pulses; shown is dose
- Waveguide homogeneity highly dependent on irradiation parameters
- High pulse energy and/or slow translation speed induces too much inhomogeneity to support waveguiding
- Low pulse energy and/or fast translation speed results in not inducing a high enough Δn to support waveguiding

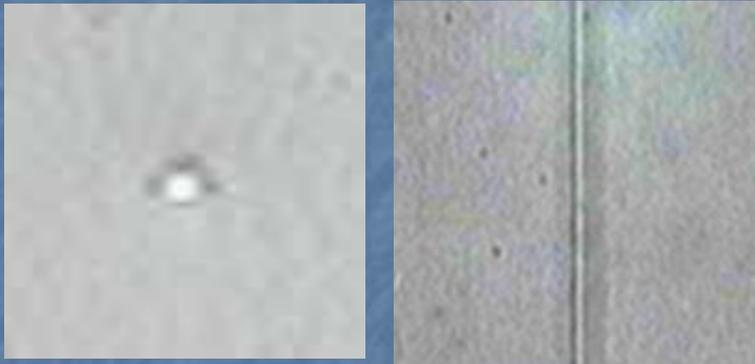


Direct-write in fused silica

- *The resulting refractive index change is estimated from the waveguide NA*

$$NA = \sqrt{2n\Delta n}$$

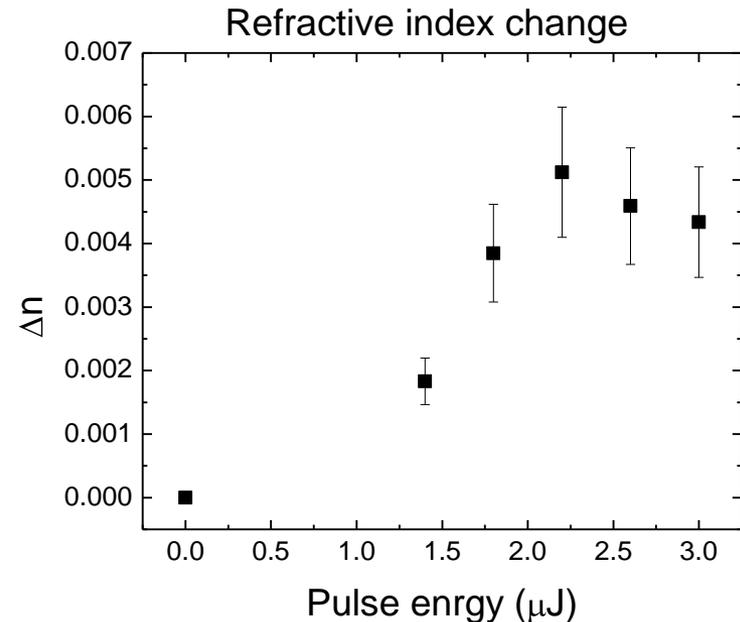
$$\Delta n \cong \frac{n_1 - n_0}{n_1}$$



Cross-sectional
view

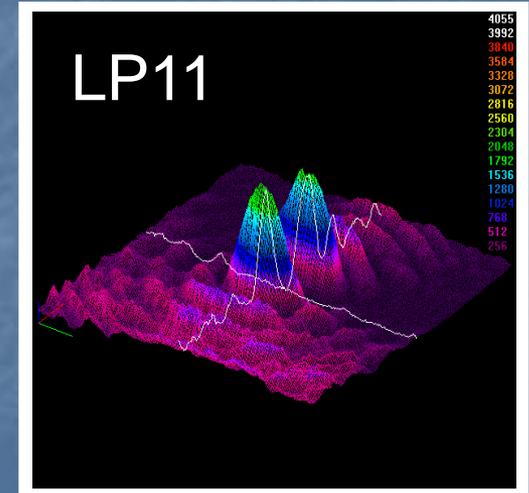
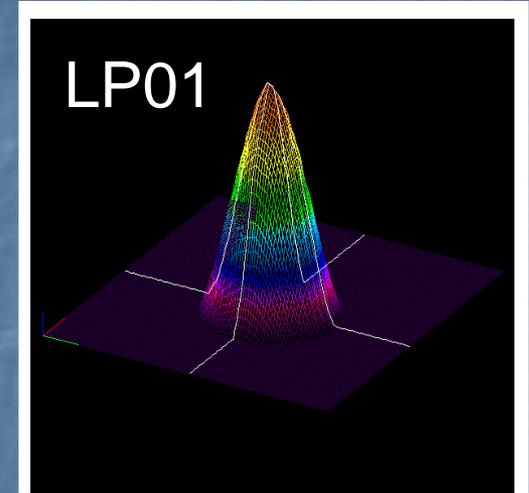
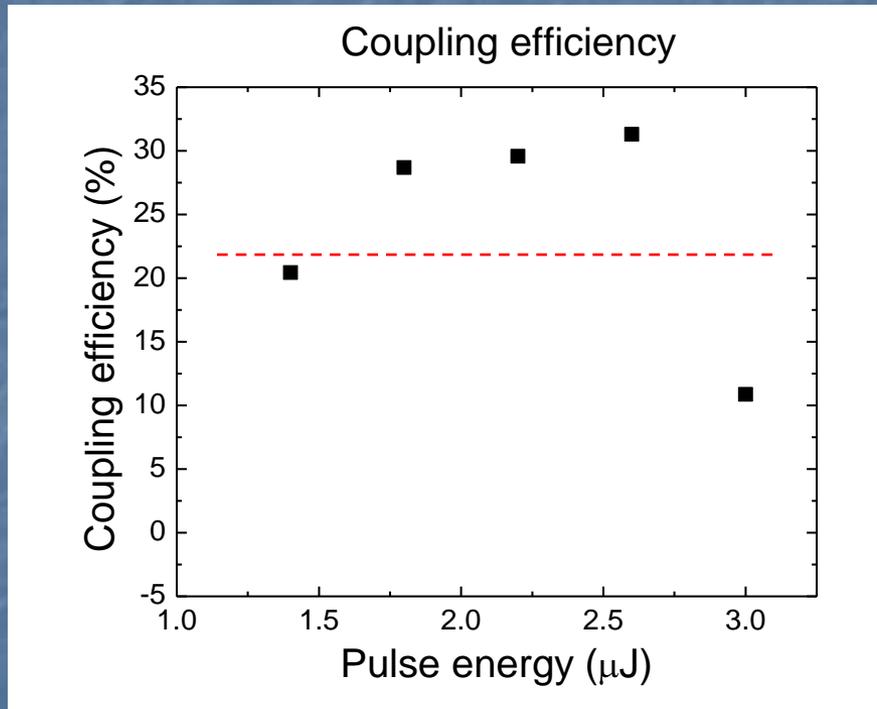
Transverse
view

Typical $\Delta n \sim 0.004$
Pulse energies \sim few μJ



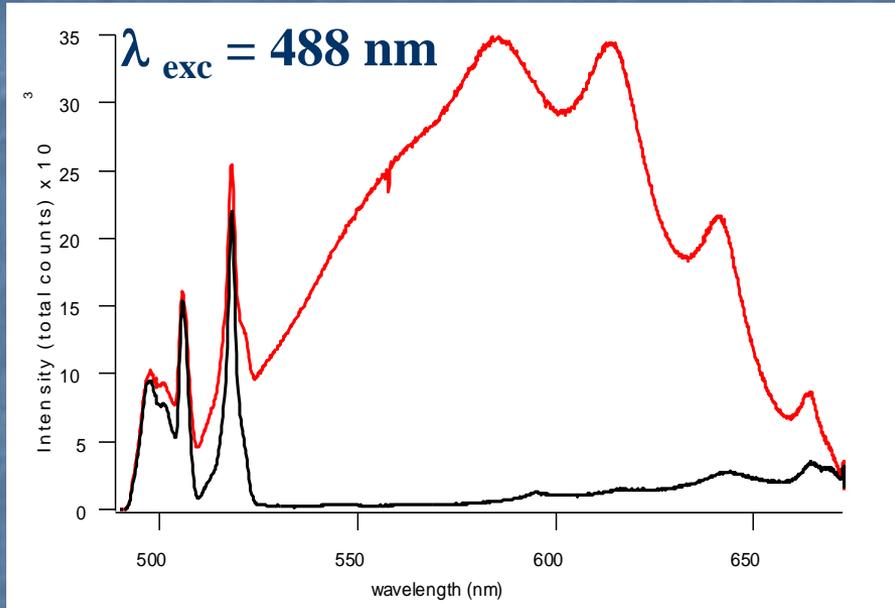
Direct-write in fused silica

- *Different modes are supported depending on the Δn created*

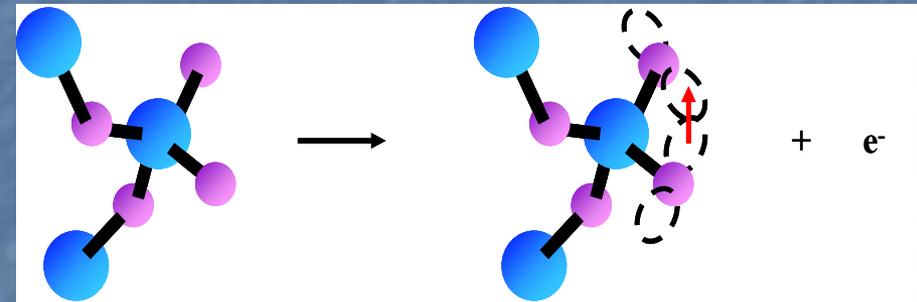


Coupling efficiency $\sim 30\%$

Direct-write in IOG-1 (phosphate) glass - fs (130 fs)-written (800 nm), 0.3 $\mu\text{J}/\text{pulse}$ (D. Krol, UC-Davis/LLNL)



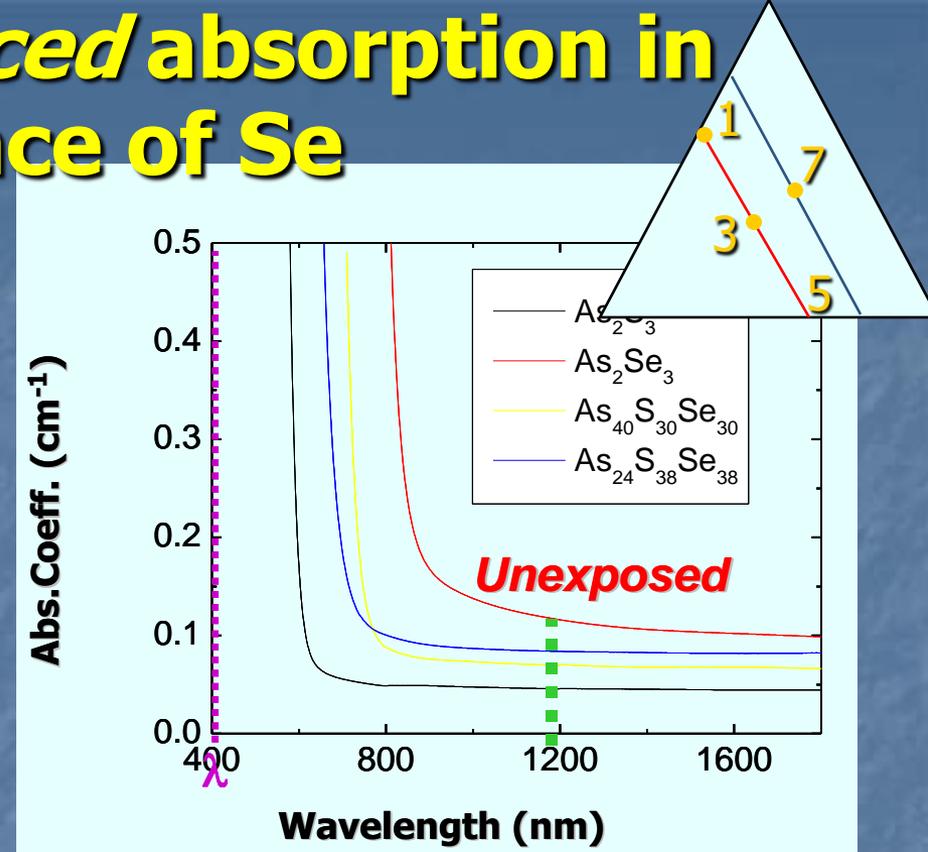
Fluorescence spectra (left) for modified (red) and unmodified (black) IOG-1 phosphate glass. $\lambda_{\text{exc}} = 488 \text{ nm}$. Increase emission is attributable to formation of POHC defect upon illumination via proposed mechanism below.



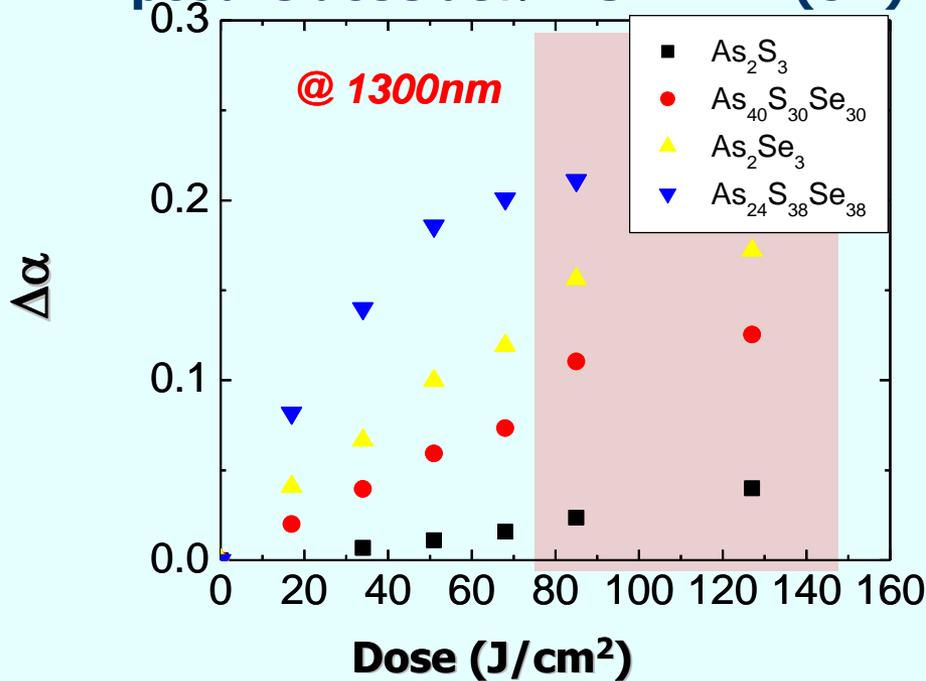
Proposed mechanism for the production of phosphorus-oxygen hole center (POHC) defects. The precursor consists of two non-bridging oxygen (NBO) atoms (pink) connected to a phosphorus (blue) atom, a defect that is common in phosphate glasses. A hole gets trapped on two orbitals of the two oxygen atoms to form the POHC. Resulting index change in exposed region of the glass is (-).

Absorption and *induced* absorption in ChG: influence of Se

- Substitution of As atoms by chalcogens has little effect on linear absorption (in 3->7 series)
- Nonlinear absorption changes
- Se content primary α driver; not only participant for $\Delta\alpha$

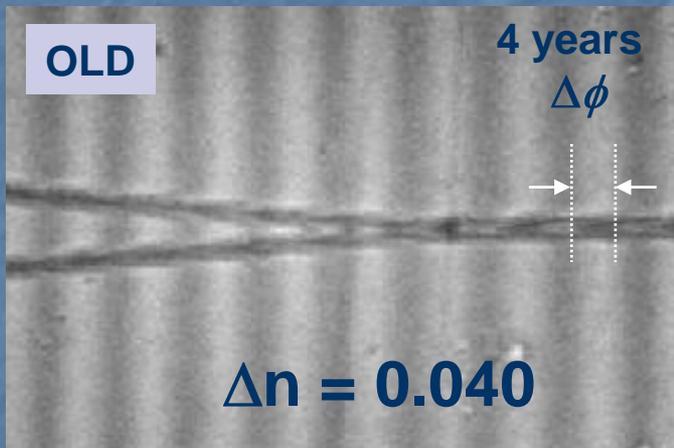
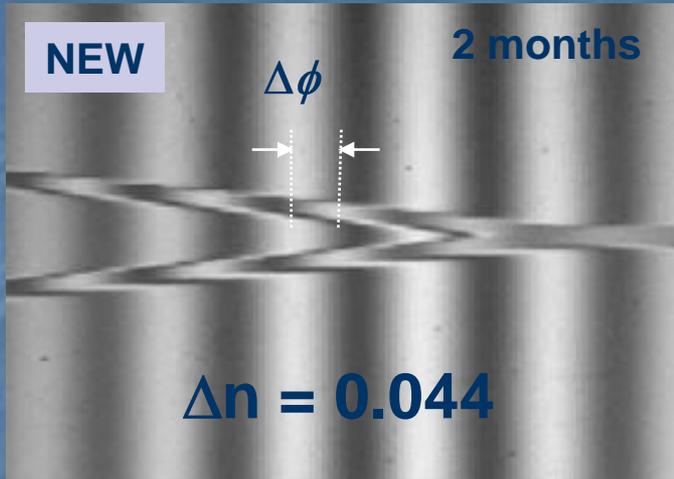


Exposure dose at $\lambda = 514 \text{ nm}$ (cw)



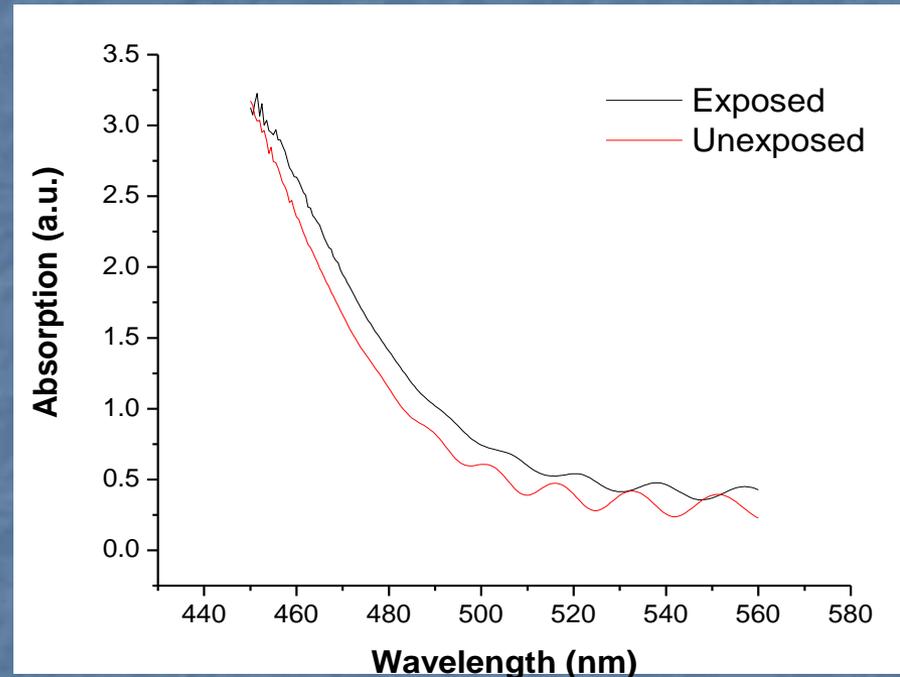
- Extent of modification ($\Delta\alpha$) increases with dose; effect is directly related to chalcogen (primarily Se) content
 - Not solely linear with chalcogen content; nonlinear absorption (two photon?)
- Influence of lone pair(s) of species affects n_2

Photo-induced index change: Δn measurement in As_2S_3



The Δn value relates to the measured phase shift $\Delta\phi$ by:

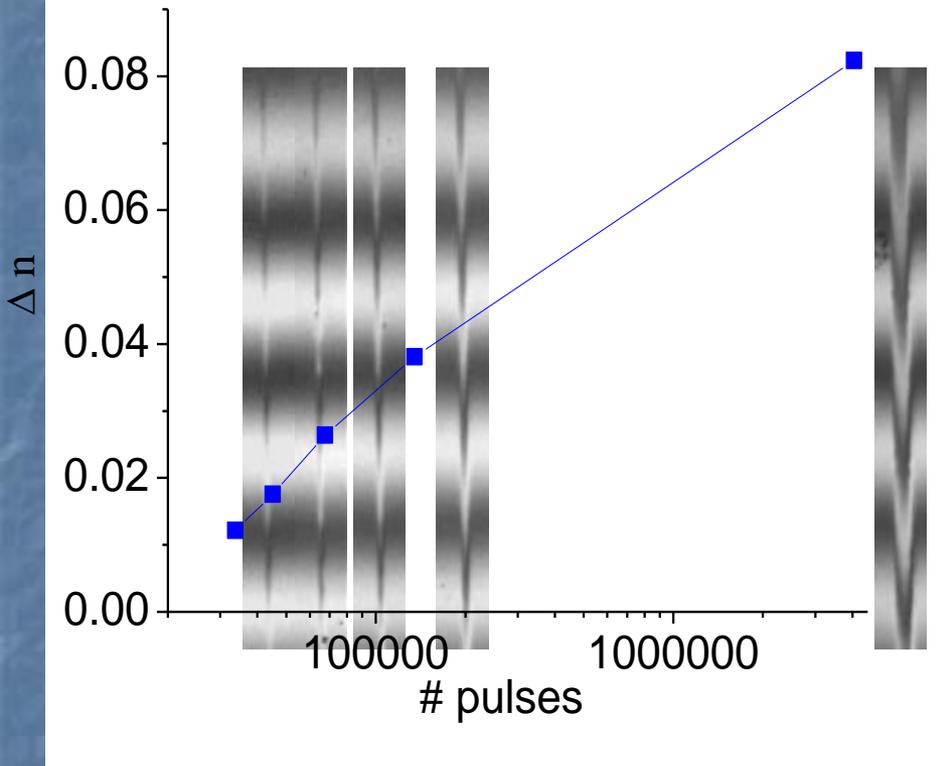
$$\frac{\Delta\phi}{2\pi} = \frac{2d}{\lambda} \Delta n$$



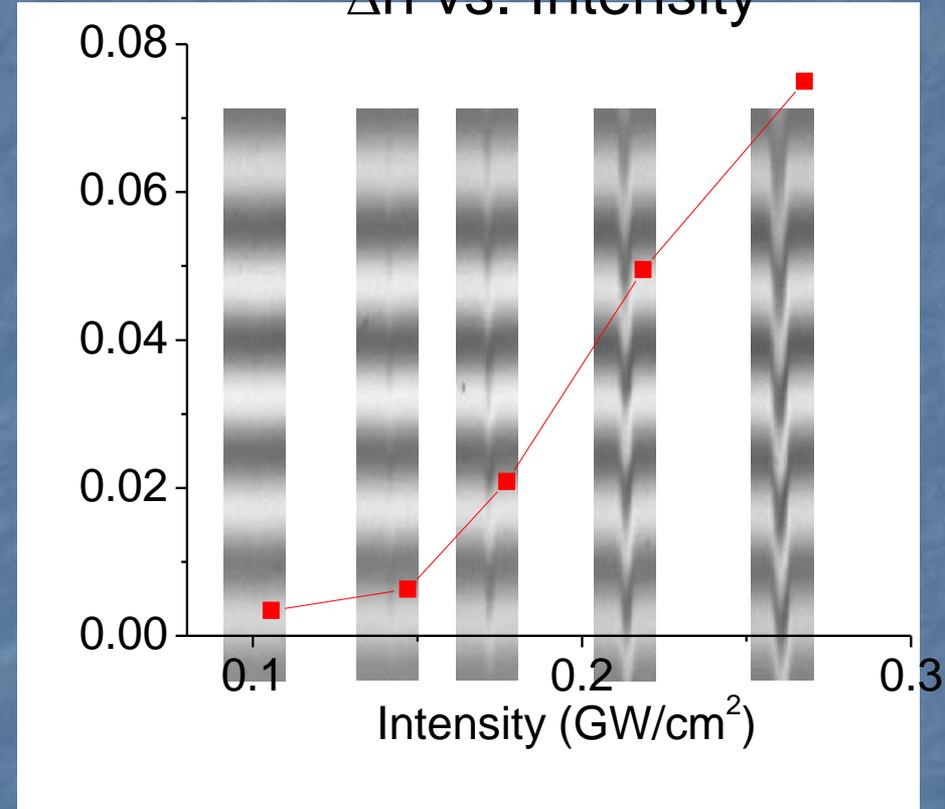
- Aging to remove deposition-induced stress to yield stable, relaxed network structure does not adversely affect resulting film photosensitivity.
- Measured Δn is similar to those from early studies using short wavelength light

Dose/Intensity-dependence on induced Δn ($\lambda=800$ nm, 100 fs pulses, 24 MHz rep rate): As_2S_3 films

Δn vs. Number of pulses

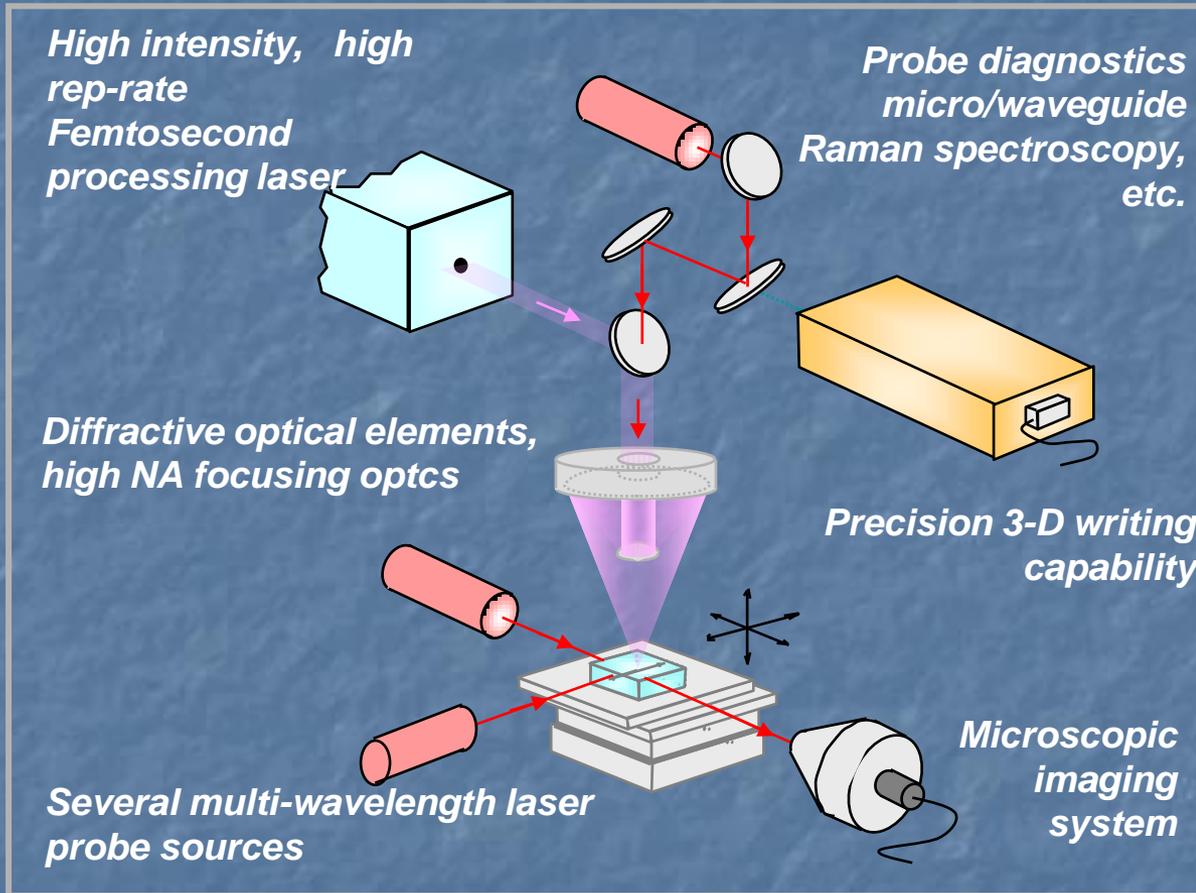


Δn vs. Intensity



The Δn values measured for As_2S_3 (> 0.08) are much larger than for oxide glasses: saturation?

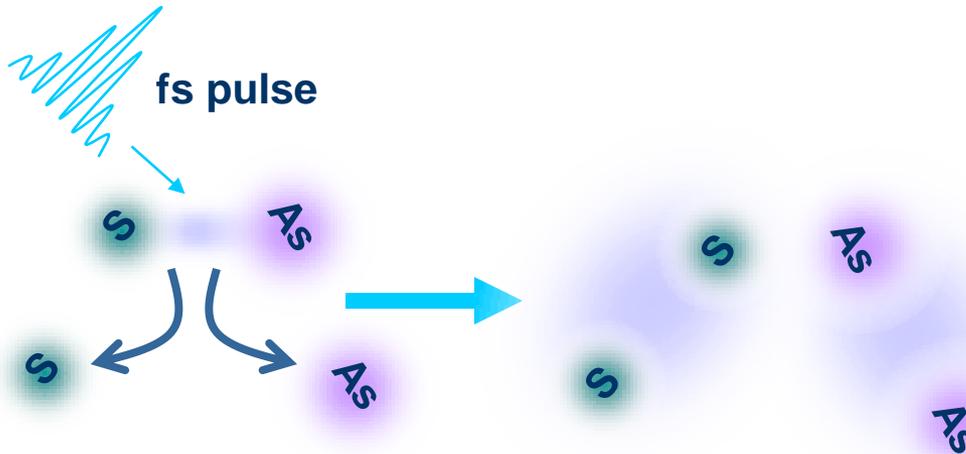
MHz fs laser machining system with in-situ microRaman spectroscopy



- **Goal: probe *dynamic* material response during laser writing to ascertain detailed knowledge of material modification mechanisms and kinetics**

Supported through NSF-MRI grant # DMR 0321110, "Development of a Femtosecond Laser-Materials Irradiation and In-Situ Probing Facility for Nano- and Micro-processing Applications and Student Training"

Free electron model in As_2S_3



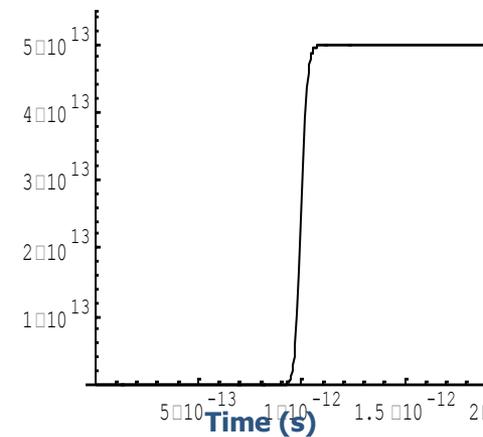
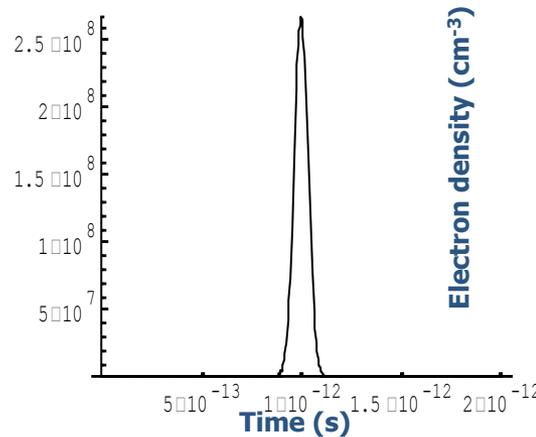
- ▶ **Photo-chemical: bond modification**
- ▶ **Photo-expansion: ΔV**
- ▶ **Photo-refraction: Δn**
- ▶ **Photo-darkening: $\Delta \alpha$**
- ▶ **Increase in thermal conductivity (via μTA): $\Delta \kappa$**

Avalanche ionization

$$\frac{\partial n_e}{\partial t} = \alpha I(t) n_e + \sigma_k I(t)^k$$

Multiphoton ionization

Intensity (W/cm^2)



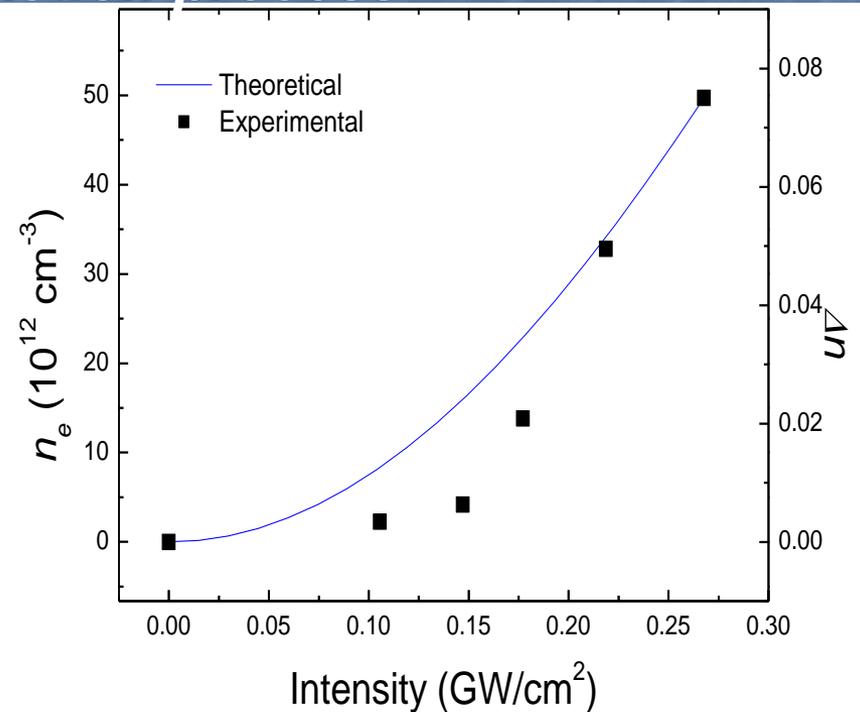
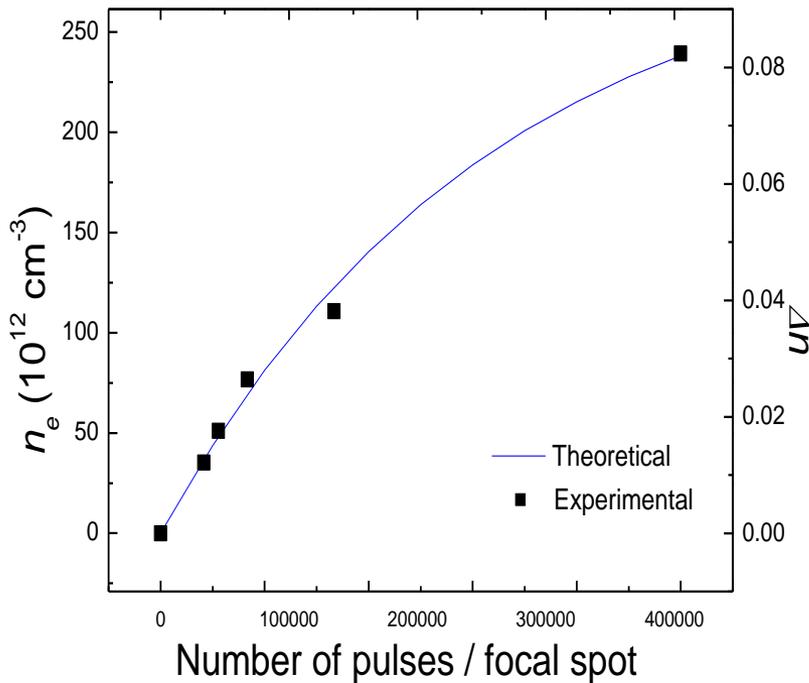
From Stuart, B.C. et al., *JOSA B*, Vol. 13, (1996) 459-468
 Structure of Glass: Section being lectured

Free electron model

Free electron density

Depletion parameter: maximum number of bonds available to participate in the photo-chemical reaction process

$$\frac{\partial n_e}{\partial t} = \sigma_k I(t)^k \cdot \frac{N_0 - n_e}{N_0}$$



Structure of Glass: Section being

Direct write in polymers: absorption

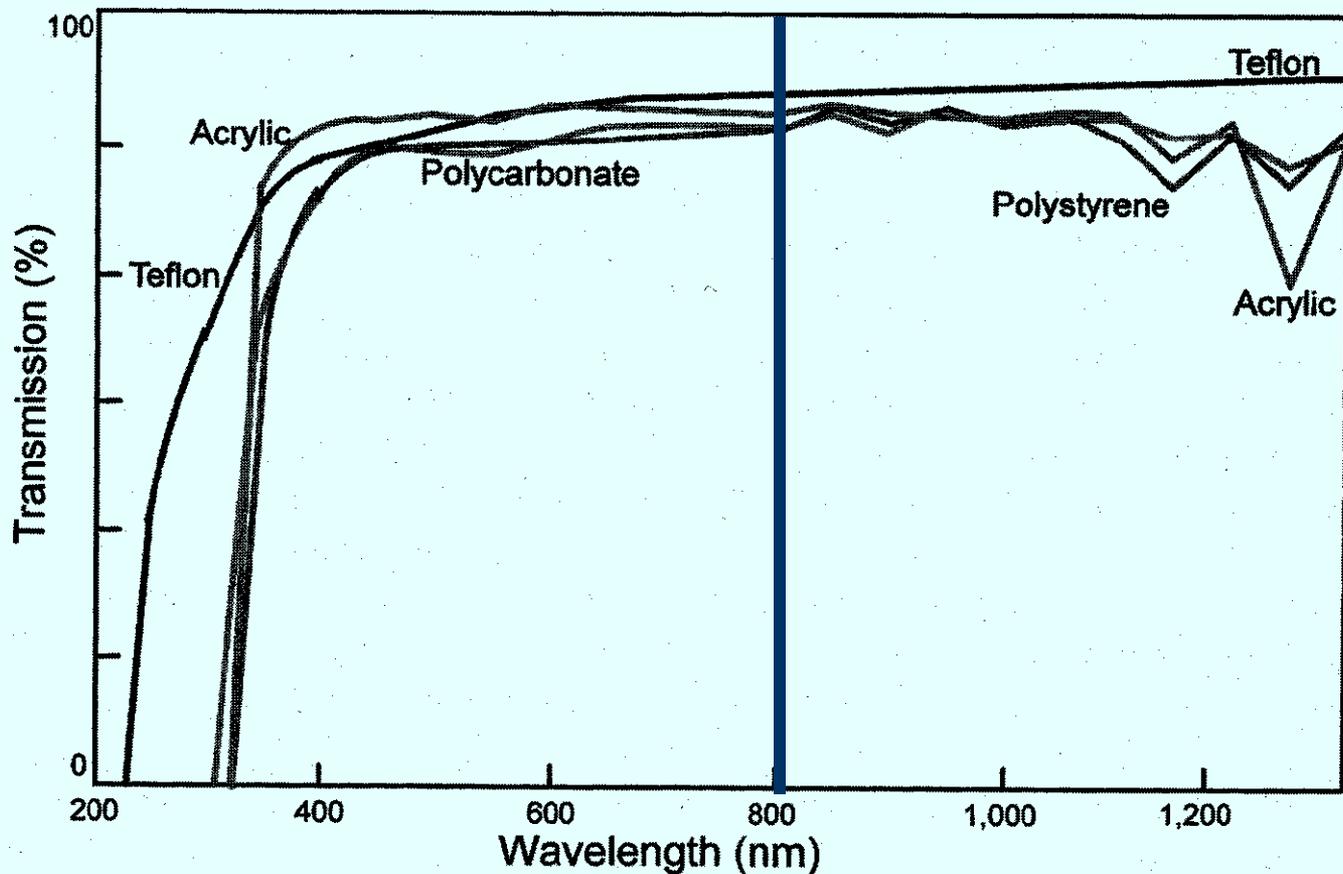


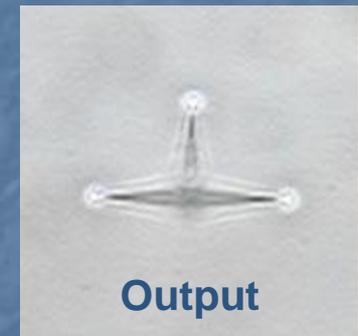
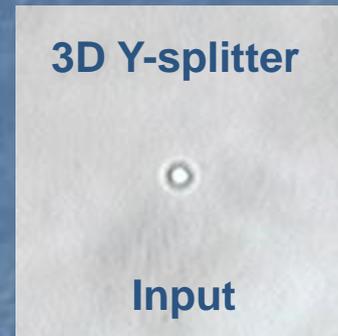
Figure 4.10: Transmission curves for a representative group of polymers. Loss in the IR comes from multiphonon absorption of light carbon-bonded species (C-H) (after Keyes 1995).

3-D writing in PMMA

fs writing (800 nm) in 25 MHz irradiation regime



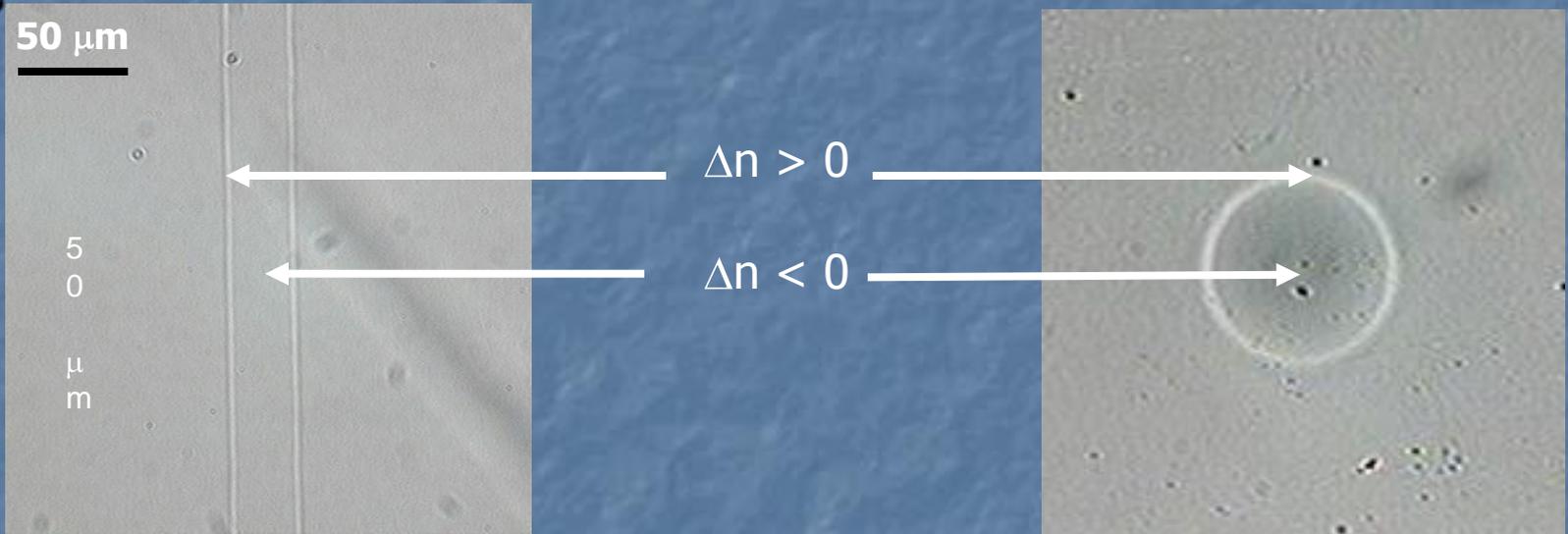
1-channel waveguides



*"Femtosecond laser fabrication of tubular waveguides in PMMA,"
A. Zoubir, C. Lopez, M. Richardson, K. Richardson, Optics Letters, in press, (2004)*

Direct-write in PMMA

- Low cost of production and ease of processing and fabrication
- Can be easily tailored to obtain the desired optical parameters (nonlinear coefficient, electro-optic coefficient, photosensitivity)
- Can be doped with conjugated chromophores or rare-earth ions

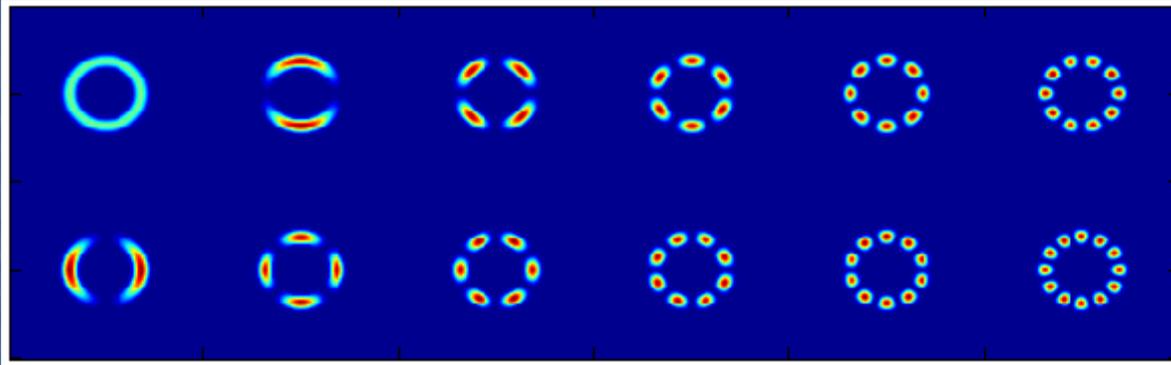


- Annular refractive index distribution caused by thermal expansion in the focus - resolved by a DIC microscope
- (-) induced index is similar for other chain-structured materials such as in glass materials such as phosphate glass (Schott IOG-1)

see "Chan et al., "Fluorescence Spectroscopy of Color Centers Generated in Phosphate Glasses after Exposure to Femtosecond Laser Pulses," 85 5 1037 (2002)"

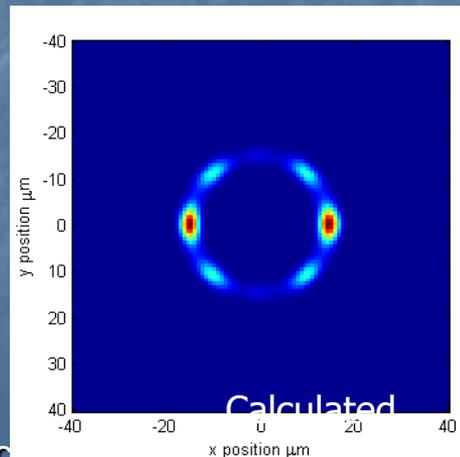
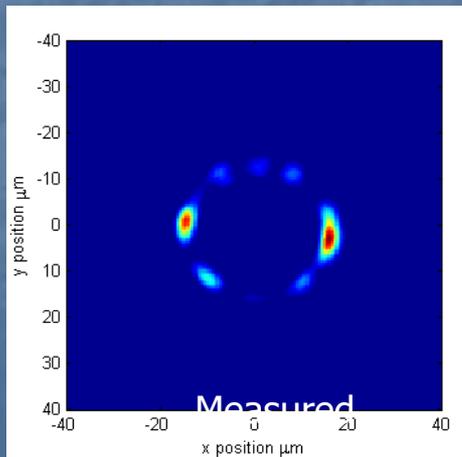
Direct-write in PMMA

Structures are highly multi-mode (large dimensions)



**Unusual mode
are allowed to
propagated in
such structures**

- *Near-field intensity distribution measured and calculated by the finite-difference method*



**Refractive
index change
estimated from
simulation:
 $\Delta n \sim 0.002$**