

# *IMI-NFG's Mini Course on Chalcogenide Glasses*

## Lecture 11

# Photonics applications 5: photoresists

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## **Resources:**

1. IMI-NFG Video Lectures by Prof. Vlcek: Glasses for Lithography & Lithography for Glasses  
[http://www.lehigh.edu/imi/TutAdv\\_R.htm](http://www.lehigh.edu/imi/TutAdv_R.htm)
2. A. Kovalskiy et al. Chalcogenide glass e-beam and photoresists for ultrathin grayscale patterning, J. Micro/Nanolith. MEMS MOEMS 84, 043012 (2009)
3. H. Jain and M. Vlcek, Glasses for lithography, J. Non-Cryst. Solids 354 (2008) 1401–1406



1

## *Lithography:*

*Original:* From **Greek** *lithos*, 'stone' + *graphein*, 'to write', is a method for **printing** using stone

*Modern Technology:* Writing of a pattern or 3D relief images in film with the aim of transferring them subsequently to the substrate

*Microlithography* – patterning method which allows fabrication of features smaller than 10  $\mu\text{m}$

*Nanolithography* – patterning on a scale smaller than 100 nm

*Contact and/or proximity lithography* – patterning with photomask in direct contact with resist-coated substrate and/or small gap between them

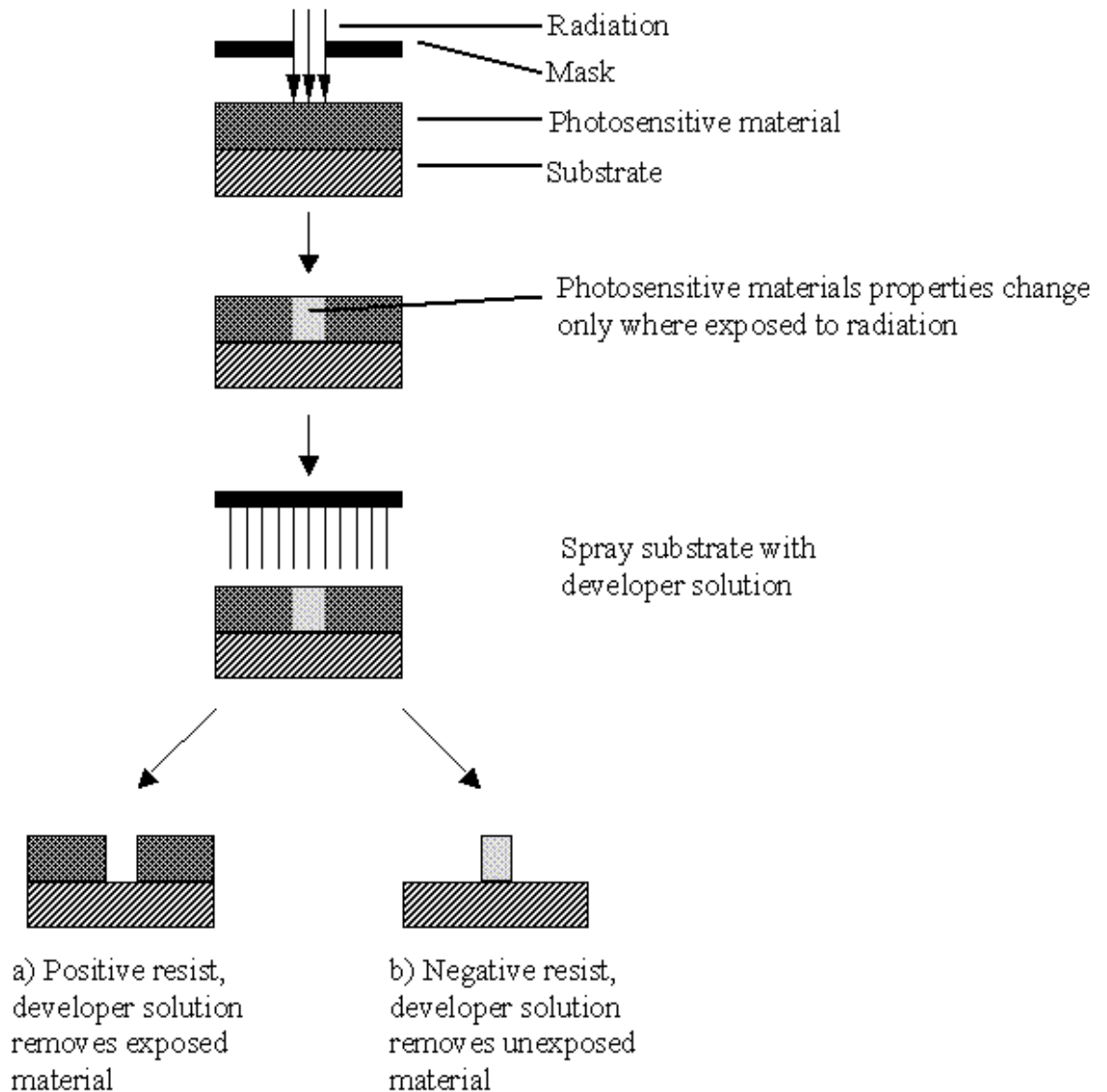
*Maskless lithography* - no mask is required to generate the final pattern e.g.:

- *electron beam lithography* – final patterns are created from digital design; computer controls the scan of an e-beam across a resist-coated substrate
- *interference lithography*



# Components of modern lithography

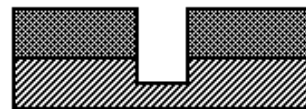
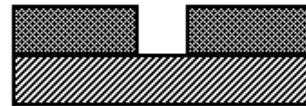
1. An exposure (irradiation) source
2. A mask and/or computer controlled scan of suitable beam across resist-coated substrate
3. A resist that stores the pattern
4. Know-how of a series of fabrication steps that would accomplish pattern transfer from the mask to resist and subsequently to substrate on which device is to be fabricated



<http://www.memsnet.org/mems/processes/lithography.html>

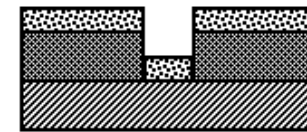
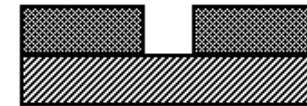
# Pattern transfer from photoresist to substrate

Subtractive Process



Pattern transfer  
by etching

Additive Process



Pattern transfer  
by lift off

Photolithography

Etch

Deposit

Strip Resist

- Pattern transfer from patterned photoresist to underlying layer by etching*
- Pattern transfer from patterned photoresist to overlying layer by lift-off*



# Most important properties of any resist

**Sensitivity to some radiation and proper technology of selective etching (simpler is better)**

**Resistant to agents used for substrate etching**

**High resolution – nano is better**

**Easy deposition – homogenous in properties and thickness**



# Radiation-induced processes in amorphous chalcogenides

## Structural changes:

- *changes of local atomic configuration*
- *polymerization – creating new bonds*
- *phase changes, including crystallization*

## Physico-chemical changes:

- *decomposition*
- *photo-vaporization*
- *photo-dissolution of certain metals*
- *thermoplastic changes*

All these processes can result in changes of optical and physico-chemical properties  $\Rightarrow$  Suitable for lithography resist



# Exposure with suitable radiation can change chemical resistance

What does it mean „suitable radiation“?

Band gap light ( $\approx 1 - 2.3$  eV)

UV or even visible light

e - beam

flux of ions

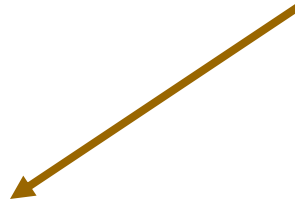
X -rays ...

Both dry and wet etching can be applied

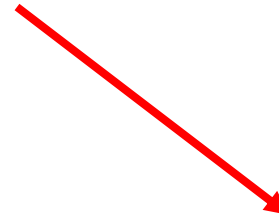
- *Wet etching – all photoinduced processes can be applied*
- *Dry etching – usually photo-dissolution of certain metals is applied*



# Amorphous chalcogenides



**insoluble  
in acid solutions**



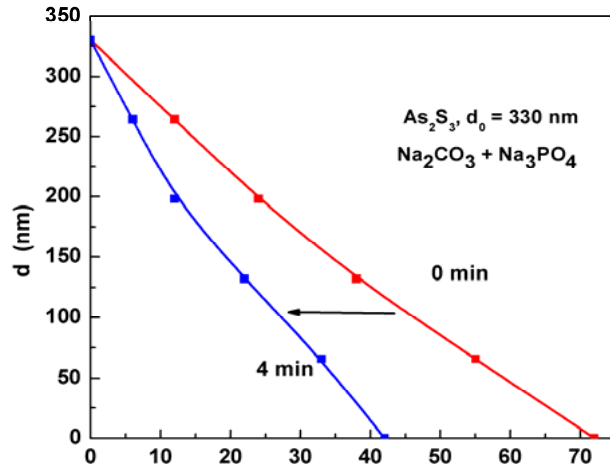
**relatively well soluble  
in alkaline solvents**

- **Dissolution rate in alkaline solvents can be influenced by exposure**
- **Both positive and negative etching can be achieved (even without Ag diffusion)**

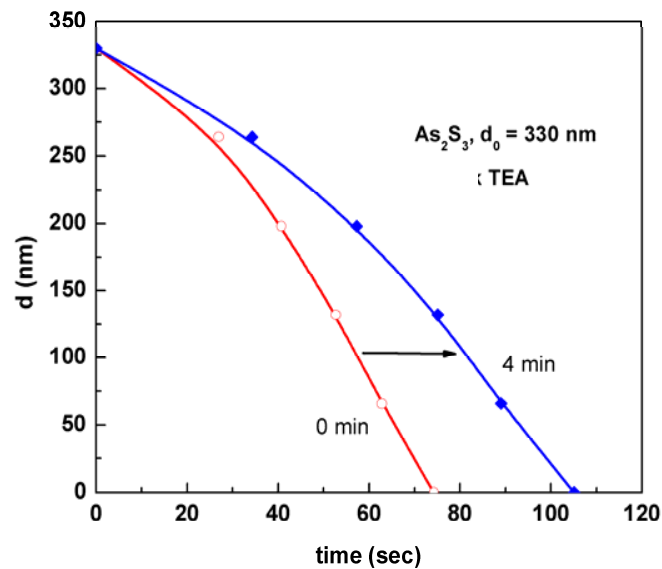




# Aqueous vs. organic solvent



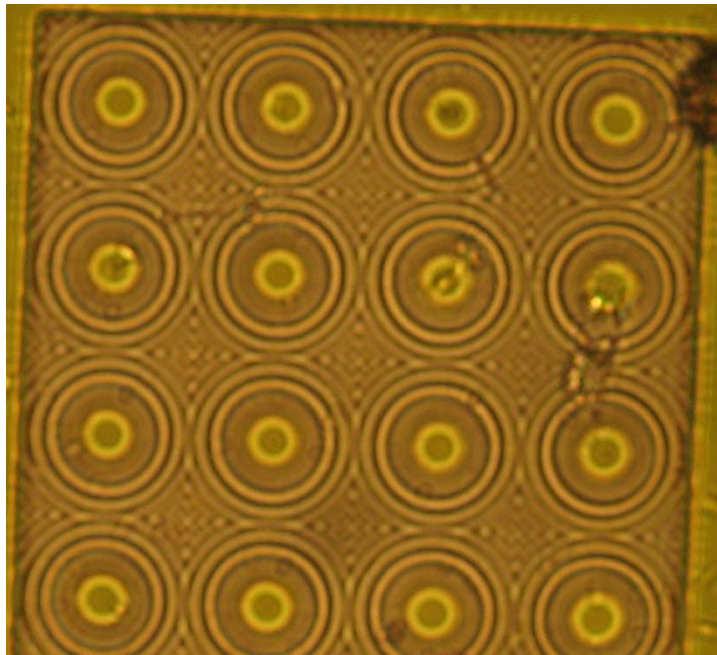
**Aqueous base**  
⇒  
**Positive etching**



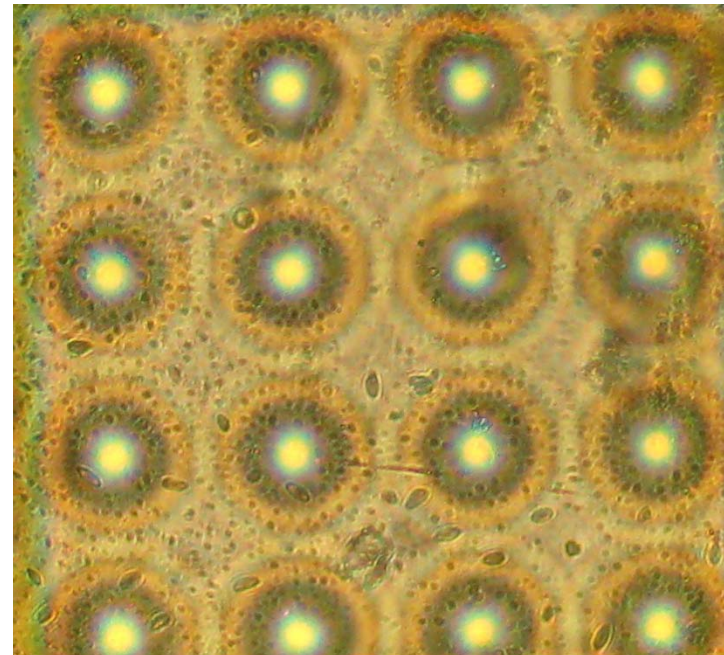
**Organic amine base**  
**e.g. triethylamine**  
⇒  
**Negative etching**

# Thinnest Fresnel Lens Array for IR Imaging

120  $\mu\text{m}$



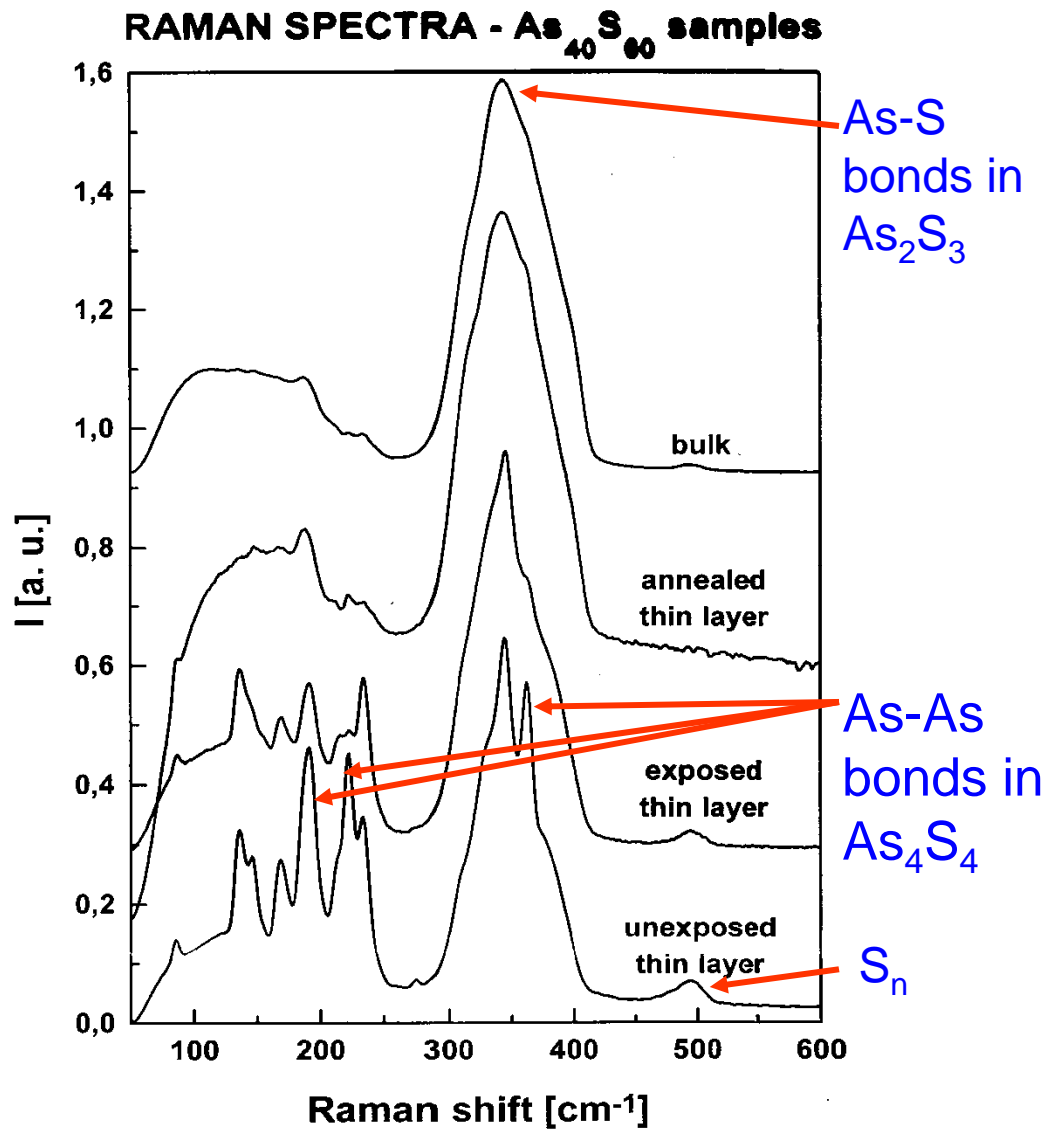
Focused on surface



At focal point  $\sim 300 \mu\text{m}$



# Raman spectra of ChG films and bulk



As deposited films contain  $\text{As}_4\text{S}_4$  and  $\text{S}_n$  fragments, which tend to disappear by light or heat.

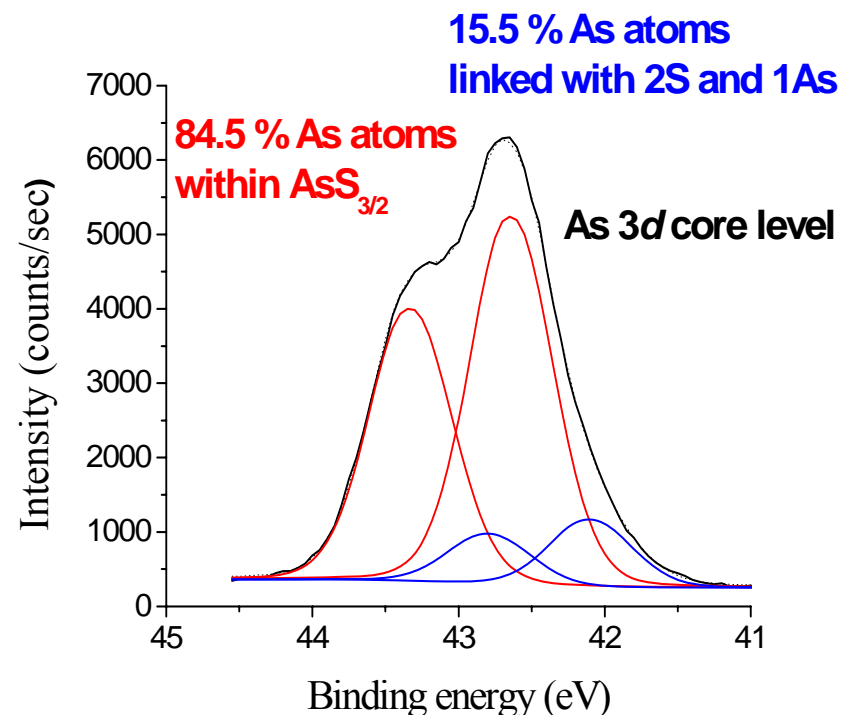
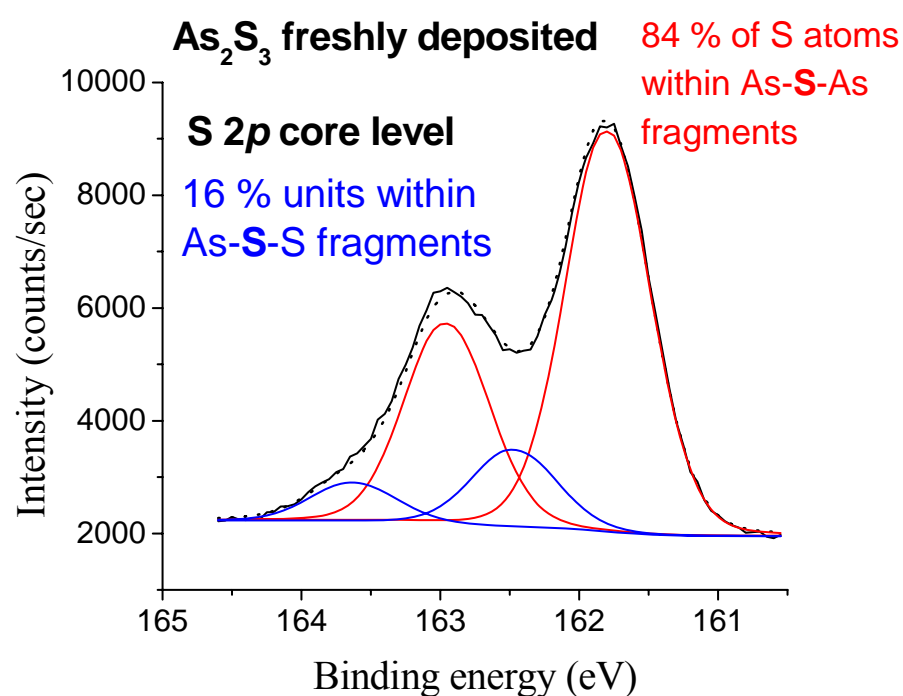
⇒

Increase in etching rate



# Structure of ChG thin films by high resolution XPS

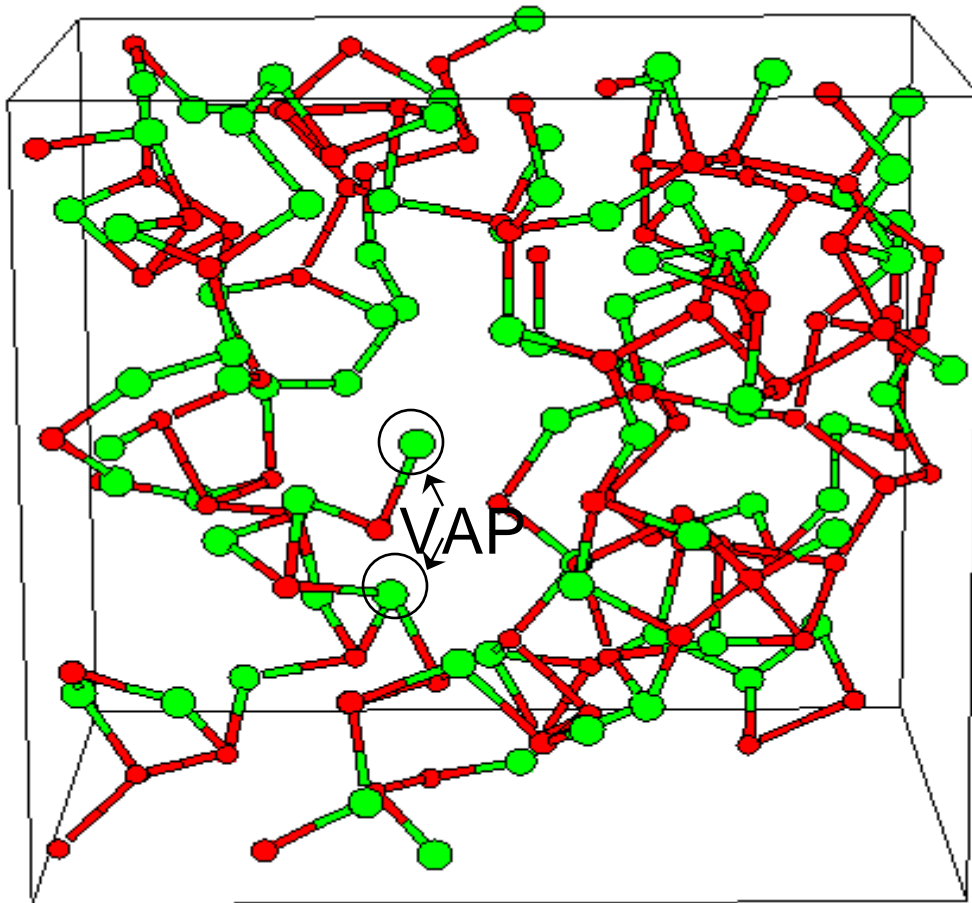
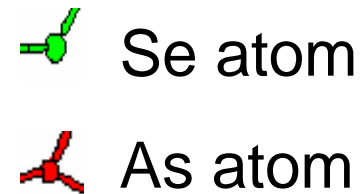
As<sub>2</sub>S<sub>3</sub> thin films deposited inside of XPS chamber without contact with air



Films have significant fraction (~16%) of S and As atoms in wrong bonds. Precision for such estimation depends on the parameters of spin orbit splitting



# First Principles MD simulation of a-As<sub>2</sub>Se<sub>3</sub>



Chemical disorder:

As-As and Se-Se

Coordination defects:

Se<sub>3</sub><sup>+</sup>, Se<sub>1</sub><sup>-</sup>, As<sub>4</sub><sup>+</sup>, As<sub>2</sub><sup>-</sup>

Valence alternation

pairs (VAP):



	<b>Chemical disorder</b>	Coordination defects
As	<b>High</b>	<b>Low</b>
Se	<b>High</b>	<b>High</b>

Li and Drabold (2001)



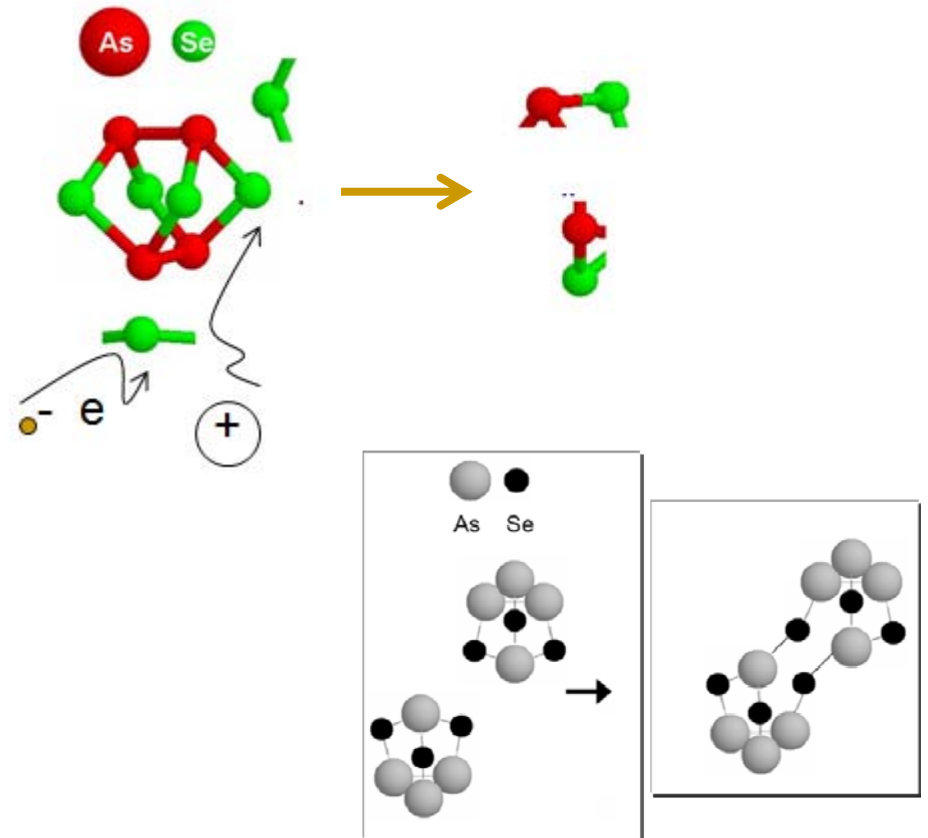
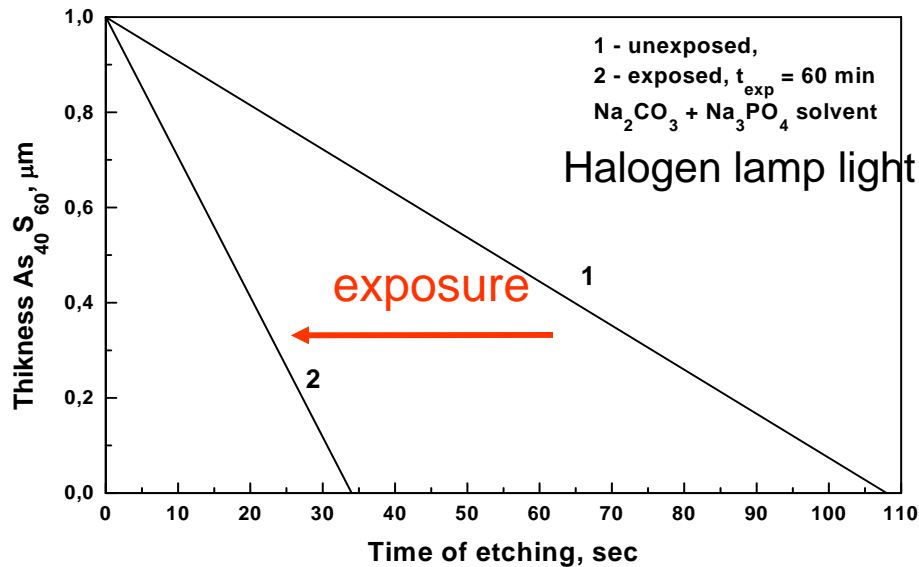
# Basic Mechanism of Positive Photo-etching

1.  $\text{As}_4\text{S}_4$  crystals dissolve much more slowly than  $\text{As}_2\text{S}_3$  crystals

+

light

2.  $\text{As}_4\text{S}_4 + \text{S}_n$  (in AP-  $\text{As}_2\text{S}_3$  film)  $\rightarrow$   $\text{As}_2\text{S}_3$  (in exposed film)



# Mechanism of **NEGATIVE** selective etching in non-aqueous amine based solvents

**Amines can promote the cleavage of sulfur rings (or chains)**



**Exposed parts – ammonolysis of heteropolar bonds (slow process)**



**Unexposed part – breaking of polymeric network through homopolar bonds (faster process)**



S.A. Zenkin, S.B. Mamedov, M.D. Mikailov, E. Yu. Turkina, I.Yu. Yusupov: Fizika i Khimiya Stekla 23 (5) (1997) 393





# Parameters influencing selectivity of wet etching

- **Sample composition, method and conditions of thin film preparation**
- **Prehistory of sample – virgin vs. annealed**
- **Exposure conditions ( $I$ ,  $\lambda$ ,  $T$ ,  $\tau$ , environment...)**
- **Etching conditions (composition of etching bath, pH, temperature..)**

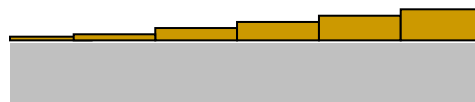
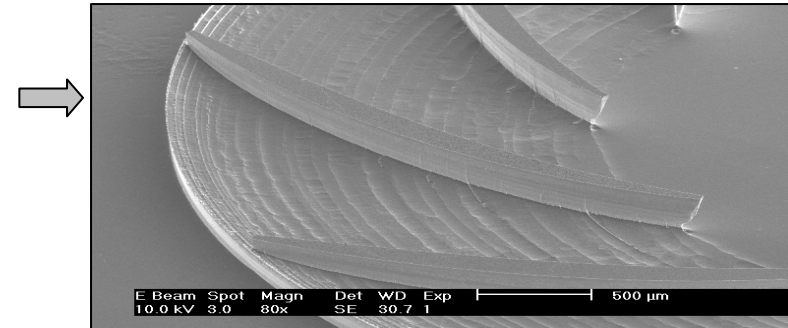




# Emerging needs: dry, grayscale lithography

## Motivation:

Gray scale profiles in photoresist film, for example development of microturbine compressor with variable height blades



Current

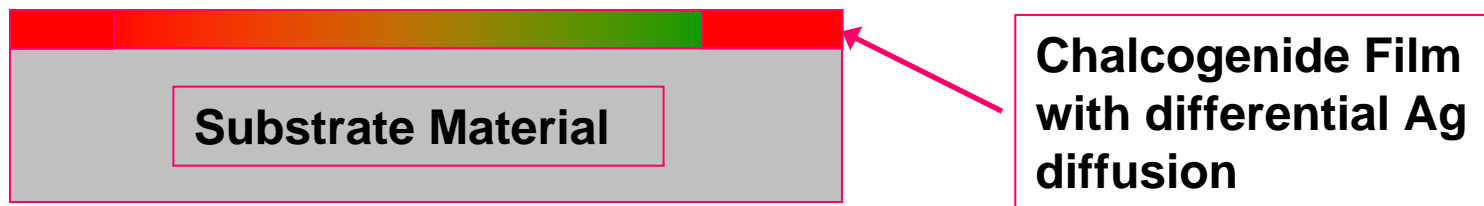


Desired

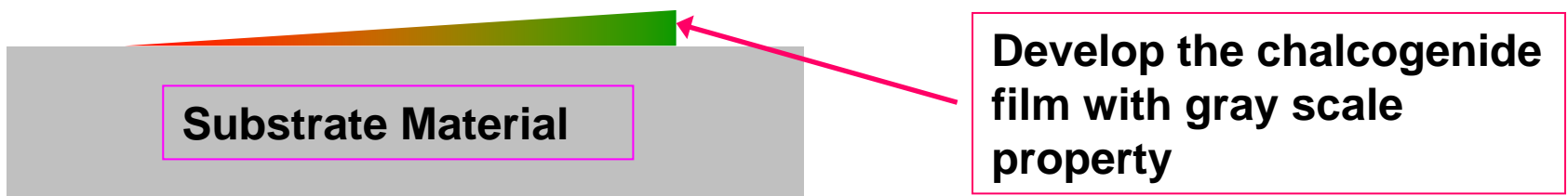


# Grayscale chalcogenide negative photoresist

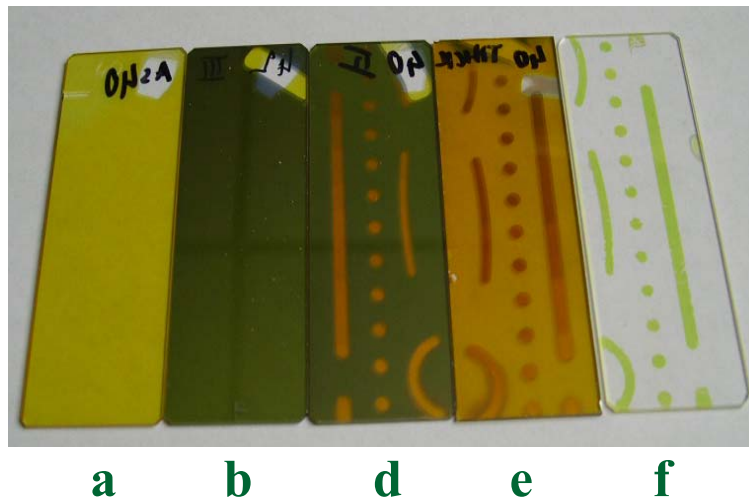
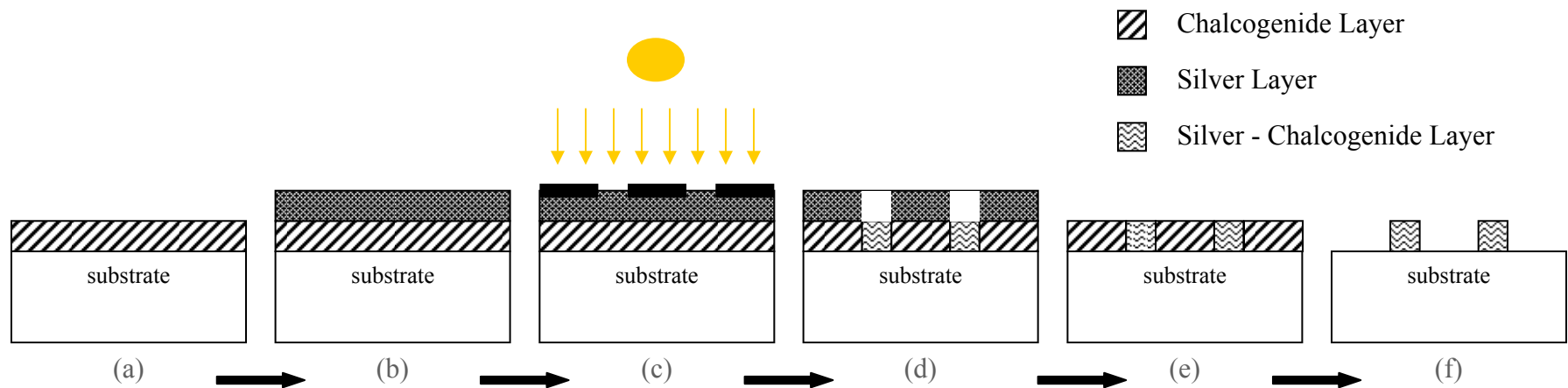
1. **Diffuse Ag into chalcogenide film through modulated halogen lamp light exposure**
2. **Remove excess of Ag for negative resist**



3. **Develop the chalcogenide film**
  - a. **Etch rates vary due to diffused Ag gradient**
  - b. **Reactive Ion Etching (RIE) – good dimensional control**



# Photodoping with Ag for Enhanced Selectivity



- (a) Deposition of chalcogenide layer
- (b) Deposition of silver layer
- (c) Exposure through mask
- (d) Silver diffusion
- (e) Removal of remaining silver
- (f) Removal of chalcogenide regions to create photoresist

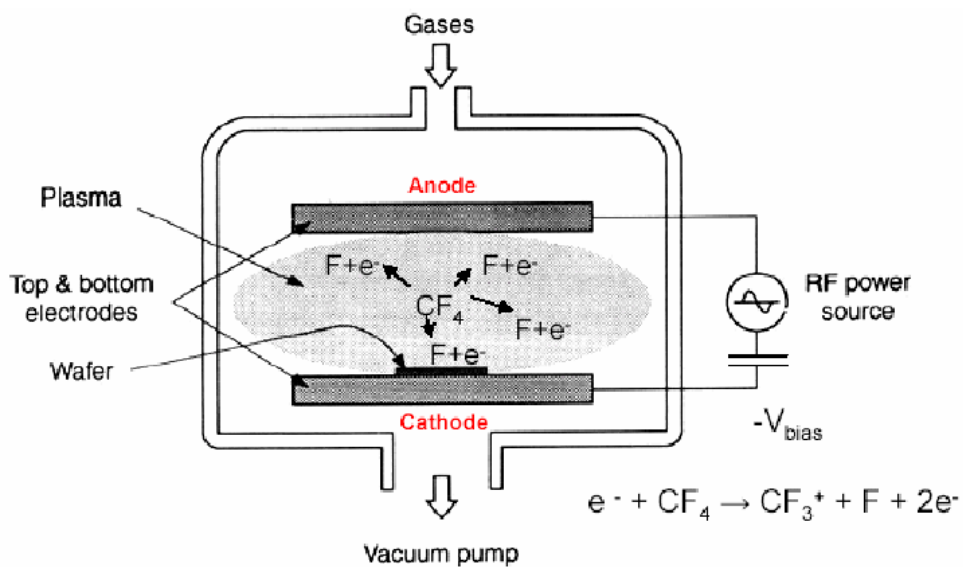


# DRY ETCHING

Plasma of ionized gases used to blast away atoms from the surface of the sample. (Also known as plasma etching)

Harsh conditions in plasma require hard photoresist ! including:

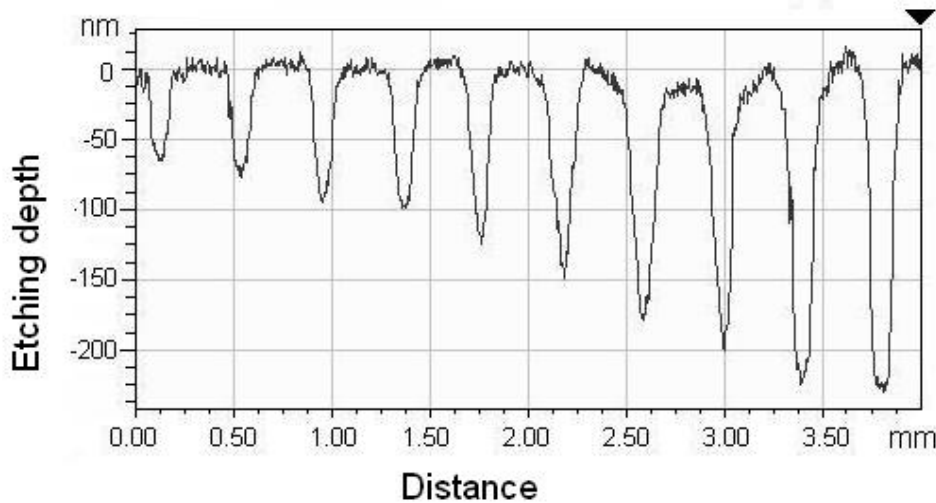
- high contrast of patterning
- resistance to aggressive, ionized gases



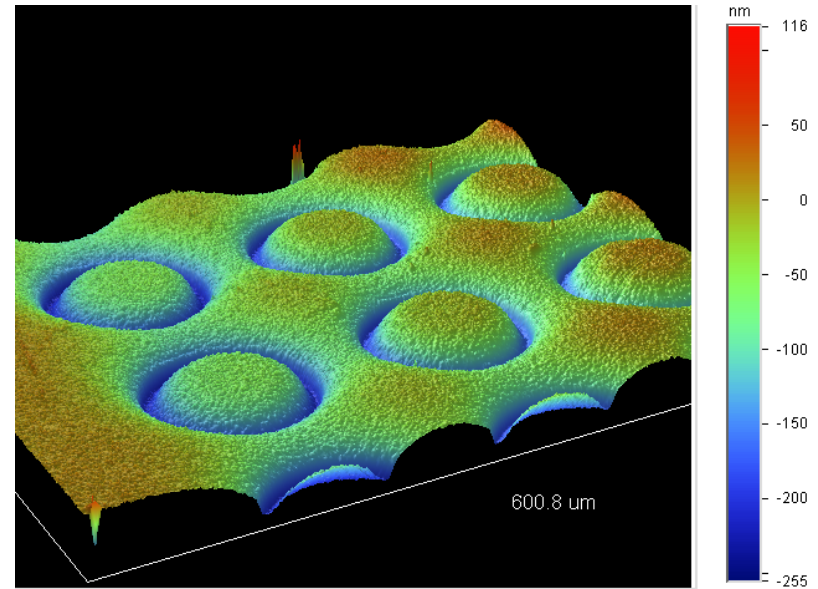
*Certain metals are usually added to ChG photoresist – Why?*  
**Combine photostructural and compositional changes from photodiffusion of metal (mainly Ag) in ChG.**



- Ag diffuses into As-S glass in step like fashion
- depth of diffusion - function of exposure dose

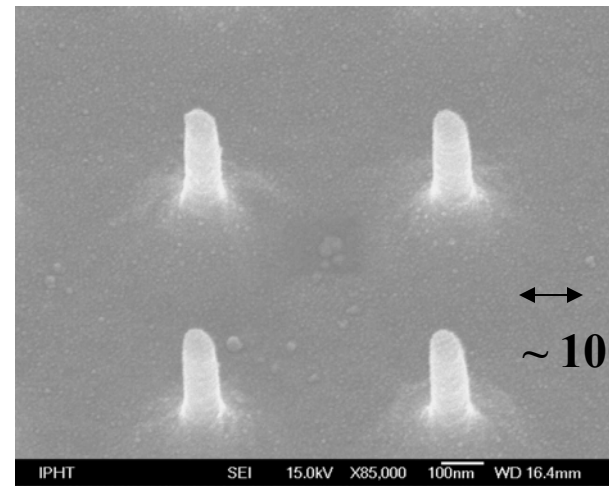
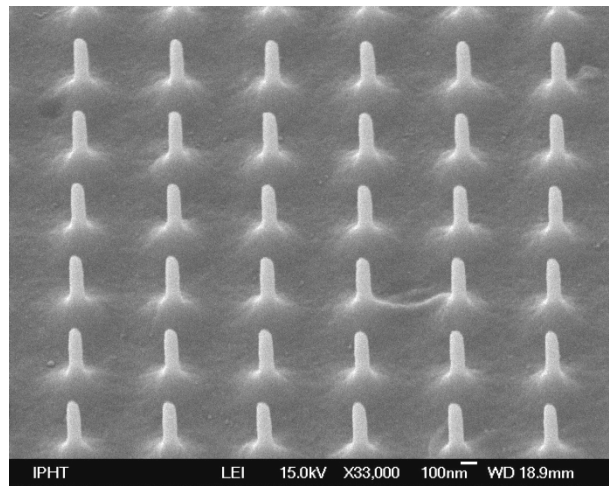
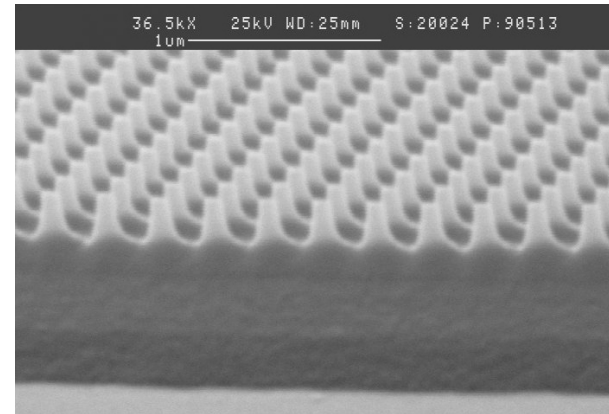
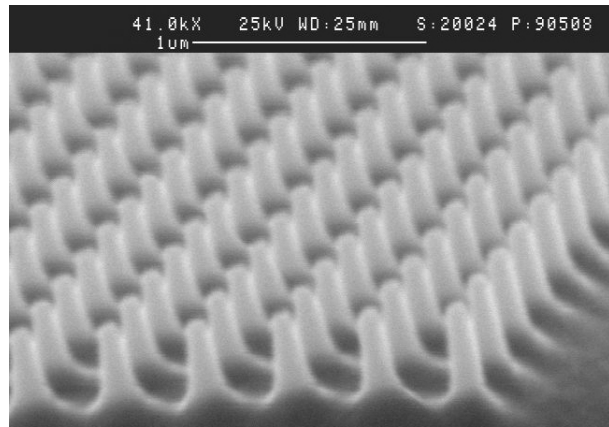


Profilogram demonstrating the change of etching depth with gradual variation of transparency of mask fragments.



Optical Profiler image demonstrating the possibility of smooth shaping with lens-like mask by photoinduced Ag diffusion into  $As_2S_3$  film with following dry etching (reverse image, depth of etching 200 nm).  $CF_4$  as the etchant gas, with pressure of 100 mTorr, an electrode power of 110 W,  $CF_4$  flow rate of 100 sccm and an etching time of 2 min

# Electron beam wet nanolithography

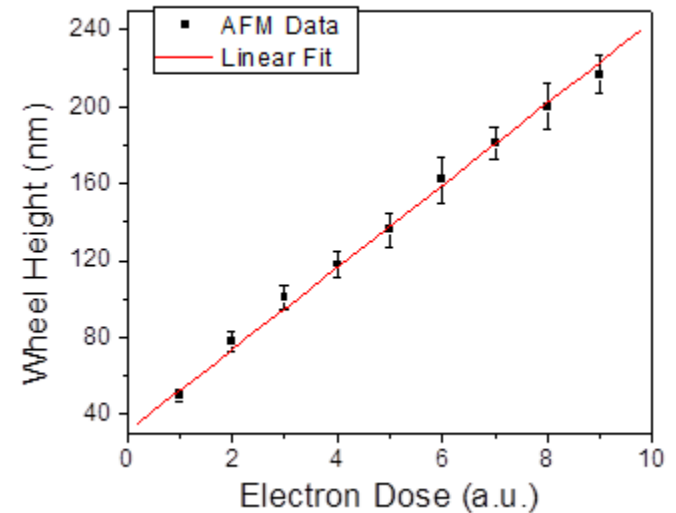
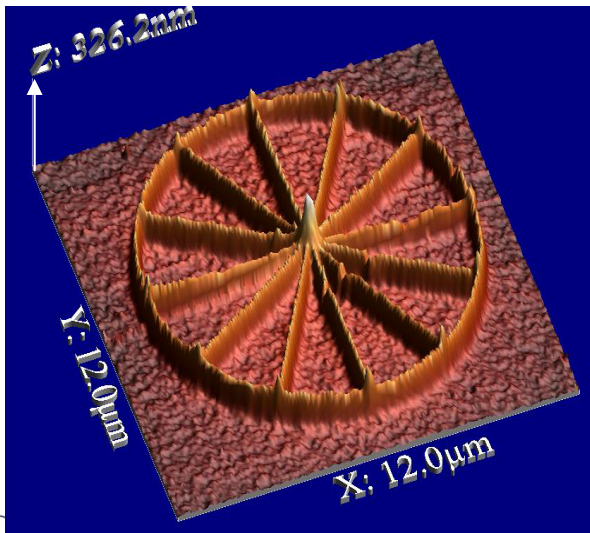
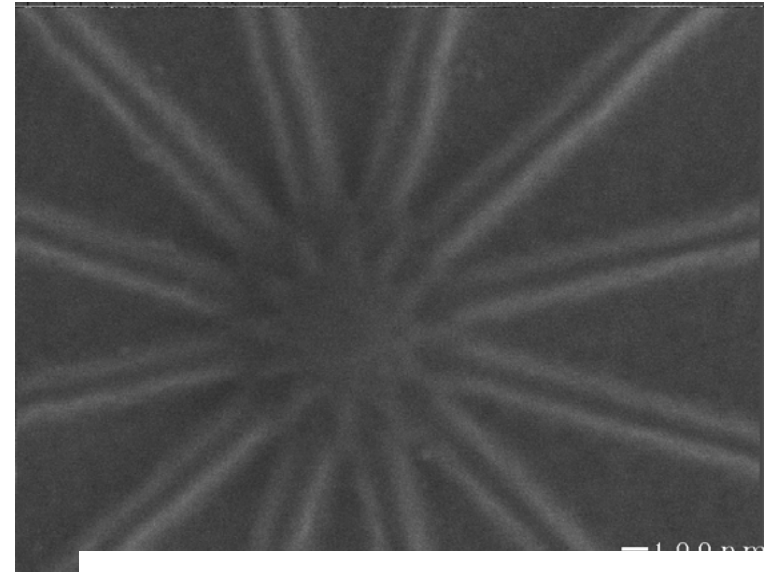
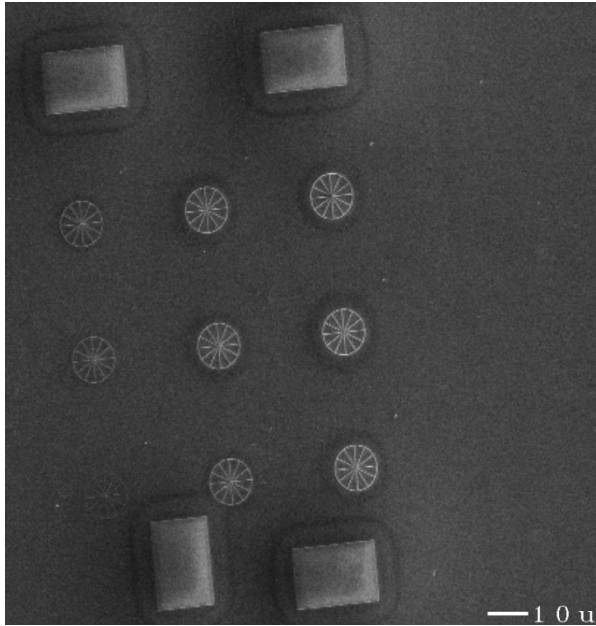


SEM pictures of pillar arrays in quadratic arrangement etched into  $As_{35}S_{65}$ . (a): diameter 122 nm, depth 410 nm, and period 400 nm (b): diameter 100 nm, depth 410 nm, and period 300 nm (c,d): diameter less than 100 nm, depth 300 nm, and period 350 nm, displayed at different magnifications



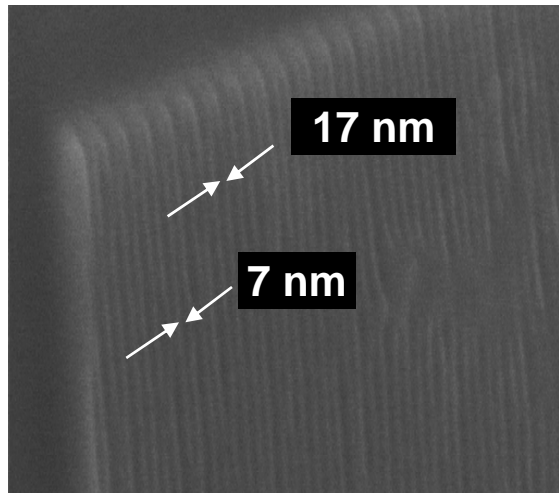


# High resolution capability



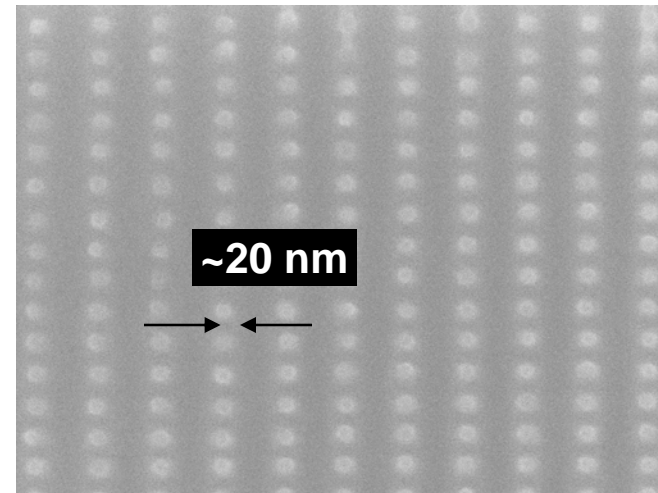
# E-beam Patterning of Chalcogenide Glasses

What is the resolution limit for ChG?



J. Neilson, A. Kovalskiy, et al. J. Non-Cryst. Solids 353 (2007) 1427

**Finest structural features on glasses**



unpublished

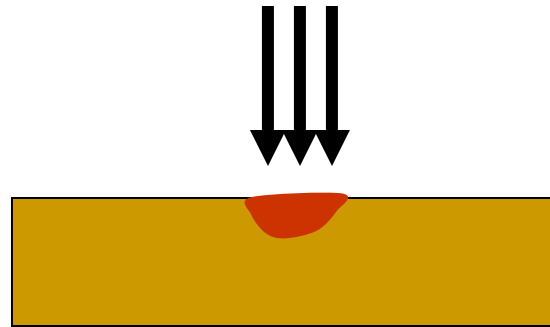
**Direct observation of separate e-beam spots**

**Wet Etching in Amine Solution**

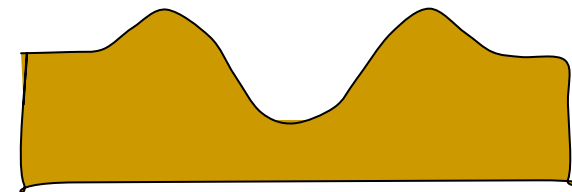




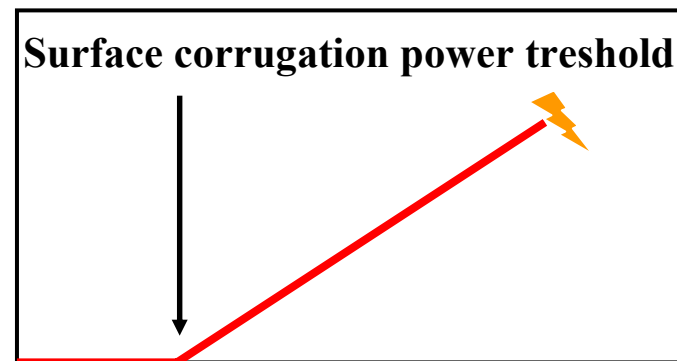
# Photoinduced local corrugation by high energy high intensity beam



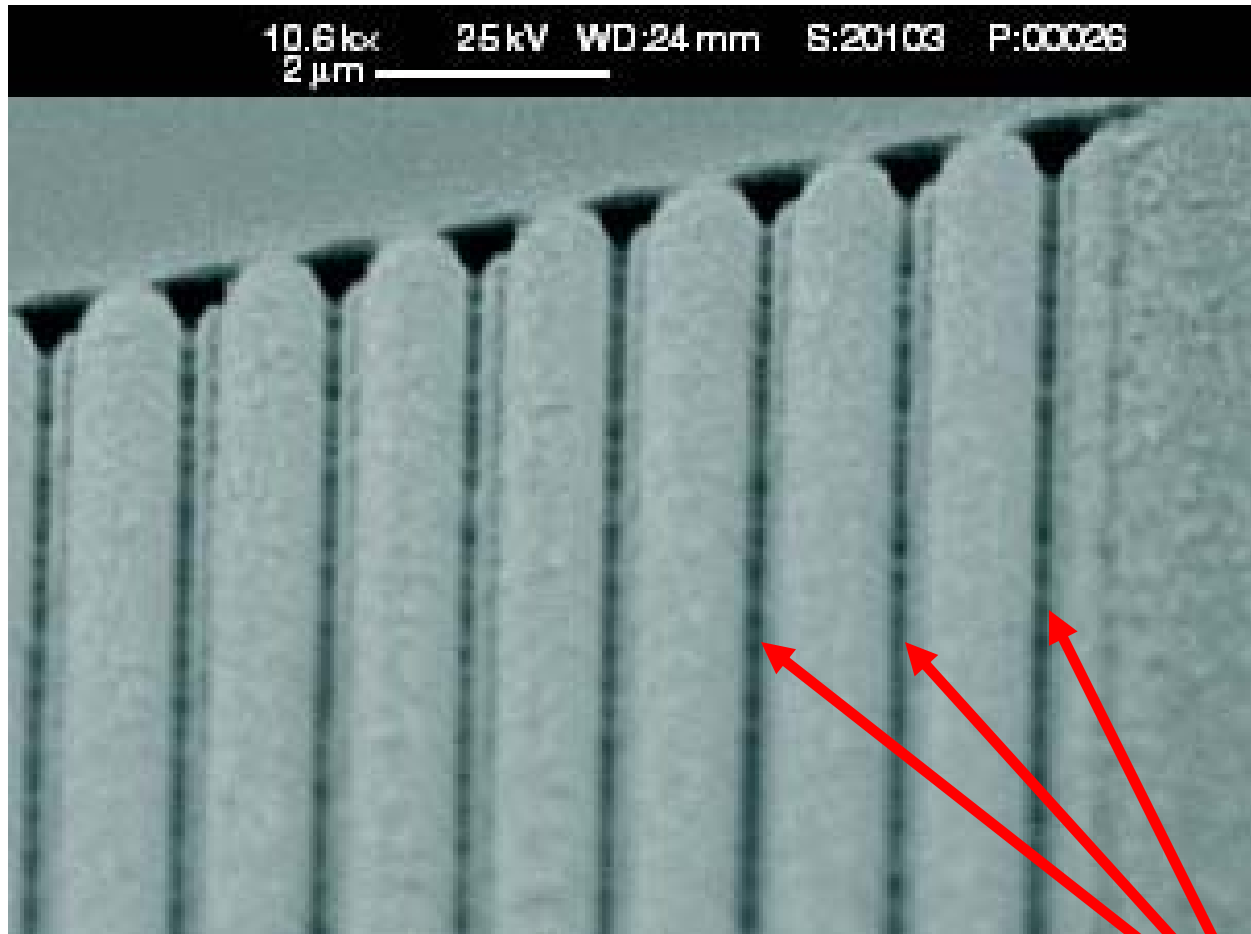
**Local heating + light field  
driven mass transport**



**Corrugated result**



# Laser writer DWL 66-UV, 244 nm – doubled Ar laser



Grating in  $\text{As}_{35}\text{S}_{65}$  layer with period of  $1.28 \mu\text{m}$ , and grooves of  $160 \text{ nm}$  bottom width and  $640 \text{ nm}$  depth, written with beam power of  $400 \text{ mW}$  at a scanning speed of  $30 \text{ mm/s}$

exposed



# Laser writer DWL 66-UV, 244 nm – doubled Ar laser

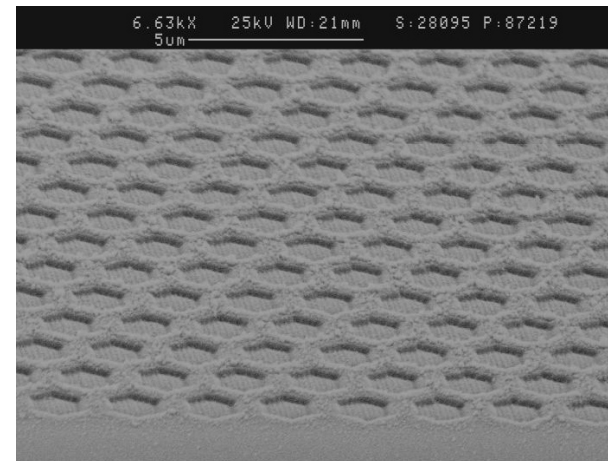
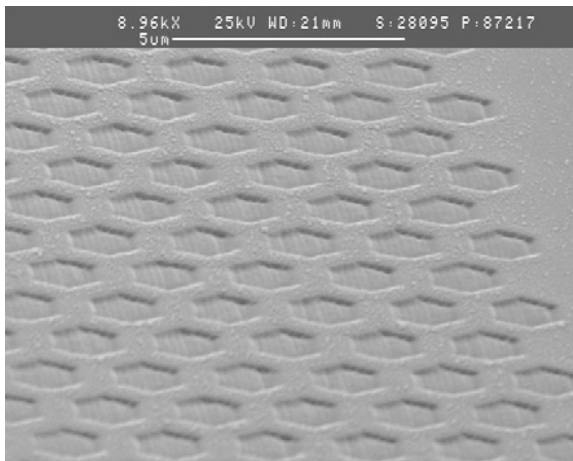
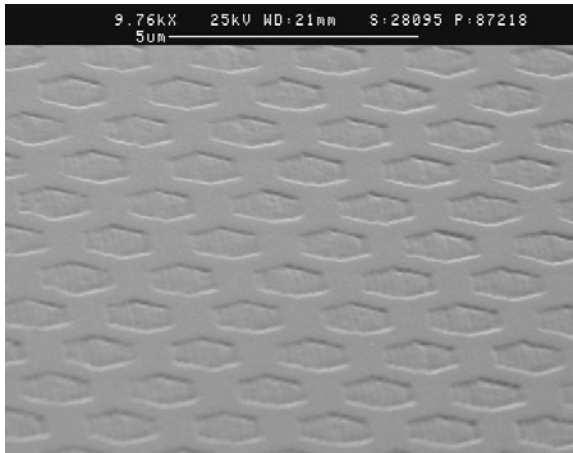


Table 1  
Surface corrugation threshold powers

Composition	Threshold power $P_{th}$ (mW)	$T_g$ (°C)
$As_{35}S_{65}$	0.35	130
$As_{40}S_{60}$	0.65	184 [12]
$As_{50}Se_{50}$	0.70	164 [12]
$Ge_{10}As_{40}Se_{50}$	0.75	200
$Ge_{30}In_{10}S_{60}$	0.90	300
$Ge_{30}Ga_{10}S_{60}$	1.25	330
$Ge_{40}S_{60}$	1.30	330

SEM pictures of 2D gratings fabricated by direct DUV laser writing technique and consisting of a trigonal air hole pattern written with a period of  $2.2 \mu\text{m}$  designed to exhibit hexagonal holes of  $1.6 \mu\text{m}$  width across flats in a  $700 \text{ nm}$  thick layer of  $As_{35}S_{65}$  written at  $0.4 \text{ mW}$  (up),  $0.5 \text{ mW}$  (left) and  $0.8 \text{ mW}$  (right) imaged at  $75^\circ$ .

For  $0.5 \text{ mW}$  the exposed power intensity and dose are  $0.7 \text{ MW/cm}^2$  and  $2.6 \text{ J/cm}^2$ .



# Summary

## Why Chalcogenide Glasses for Lithography?

- Sensitive to different radiations  $\Rightarrow$  Greater versatility
- Smaller molecular units  $\Rightarrow$  Higher resolution
- Much harder than polymers  $\Rightarrow$  Maintains shape
- Easier deposition of thin, uniform films  $\Rightarrow$  More accurate transfer of patterns (eliminates interference effects)
- Optical functionality  $\Rightarrow$  Etchless resist
- Metal photodissolution  $\Rightarrow$  Dry grayscale lithography possible
- Much simpler etching, without any other treatments
- Resistant to acids  $\Rightarrow$  Easier transfer of patterns in substrates
- Both positive and negative resists are possible

## Conclusion:

**ChG offer a new more powerful class of photoresists for more versatile lithography than currently available polymer resists**

