

Novel functionalities of chalcogenide glasses

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Lehigh University

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International Materials Institute for New Functionality in Glass

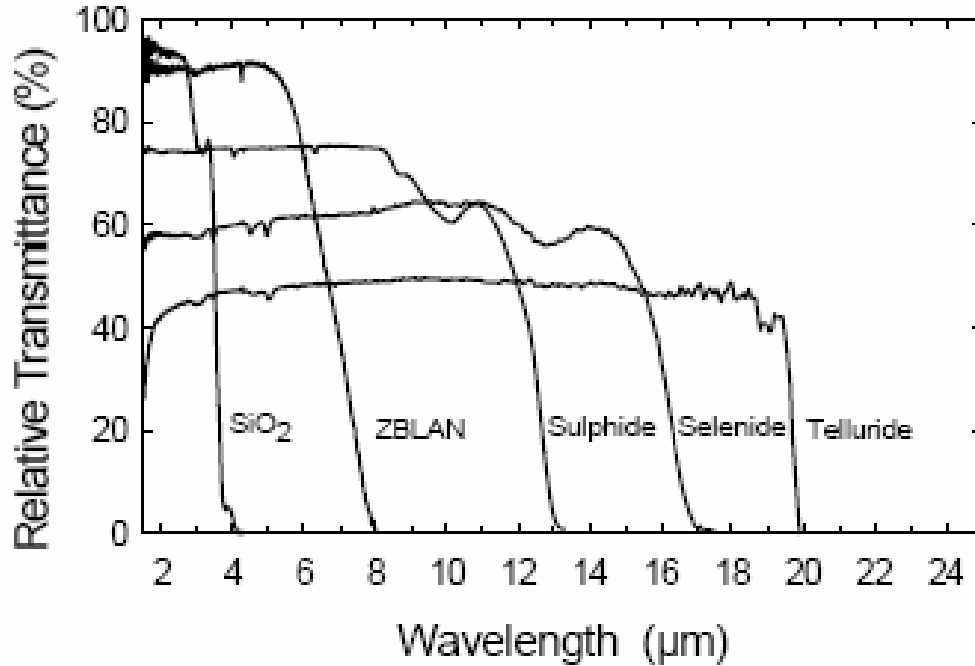


Outline

- ✓ Introduction
- ✓ Photosensitivity
- ✓ Structure of ChG
- ✓ Speed of photosensitivity
- ✓ Examples of New Functionalities

What are Chalcogenide Glasses?

After I.D. Aggarwal, J.S. Sanghera, "Development and Applications of Chalcogenide Glass Fibers at NRL"



- transmission in the infrared
- high refractive index (~ 1.75)
- sensitive to bandgap light (1 – 3 eV), e-beam, x-rays
- **photosensitive** (photodarkening & photoexpansion)

	T_g (°C)
Silicates	~ 700
As-Se	~ 150

Compounds of S, Se and Te e.g. elemental Se, Ge-Se, As-Se, As-S, Sb-Te,...

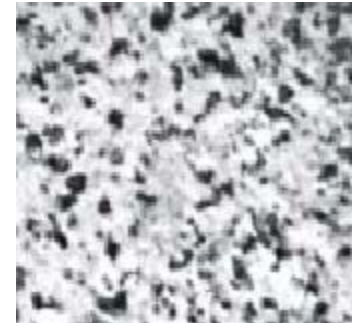


Recent new functionalities: CD-RW and DVD-RW: Phase-change memory

Laser power controls the switching between amorphous and crystalline states.



High power
→ amorphous

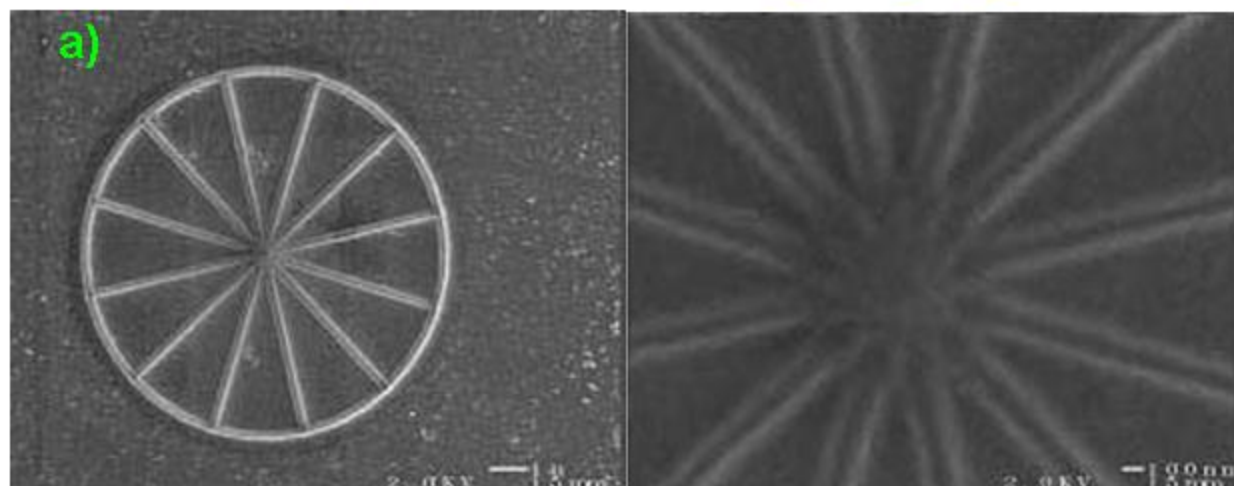


Medium power
→ crystalline

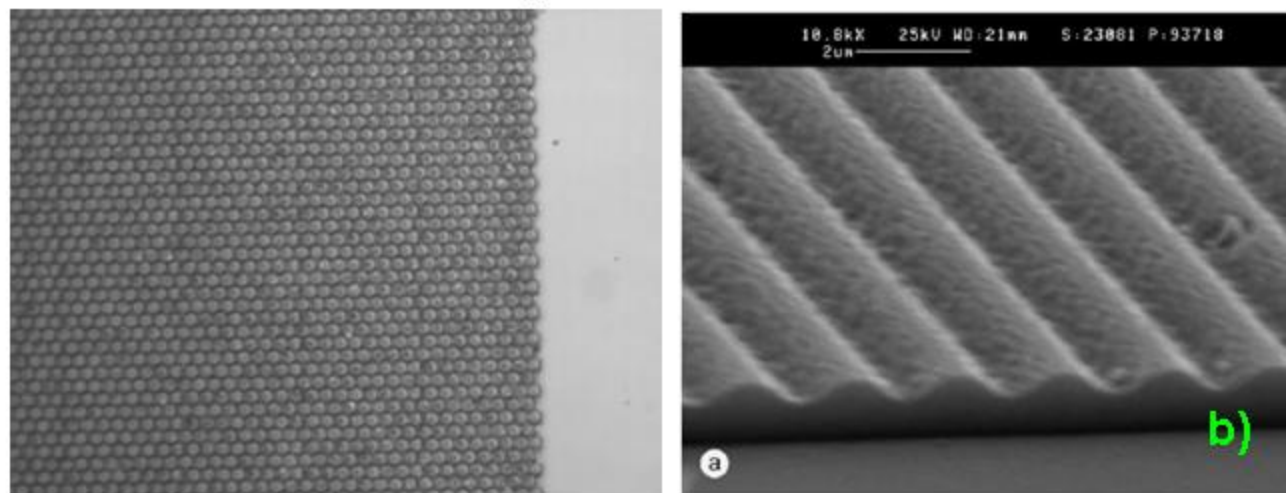
Recent new functionalities: FIR Night vision system on BWM 7 series



Micro/Nano Lithography



J.R. Neilson, A. Kovalskiy, M. Vlček, H. Jain, F.C. Miller, JNCS 353 (2007) 1427–1430.



Etchless Lithography

Optically written honeycomb structure with $\sim 1 \mu\text{m}$ radius

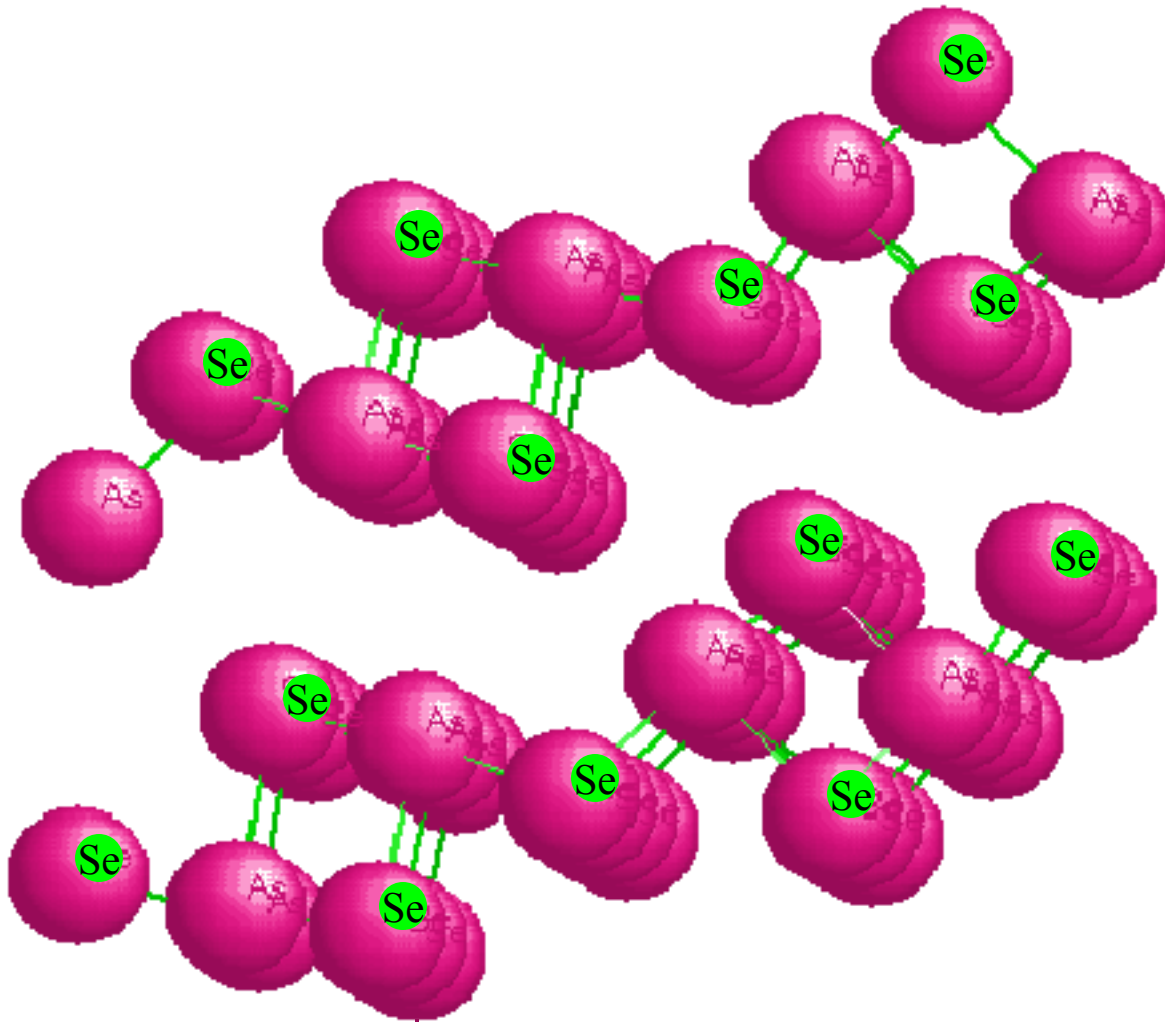
M. Vlček, S. Schroeter, J. Čech, T. Wágner, T. Glaser, J. Non-Cryst. Solids, 326&327 (2003) 515



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Crystal structure of As_2Se_3



Two-dimensional
layer structure

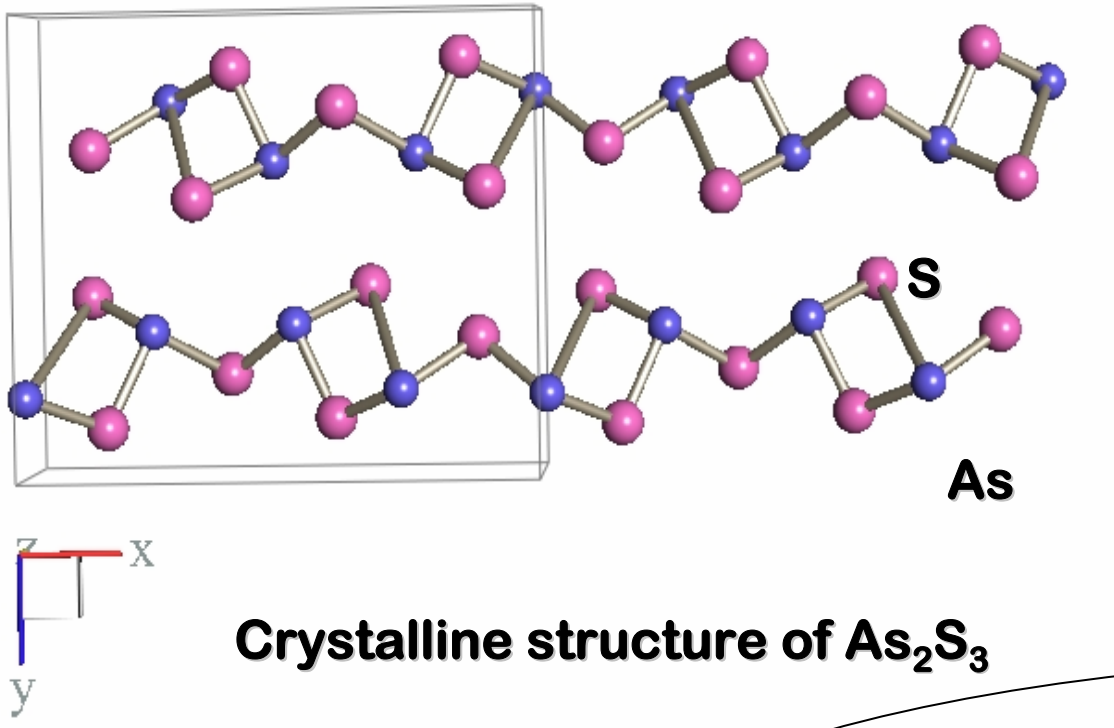
Covalent bonding:
8-N rule

Se: 2-fold

As: 3-fold

Each As bonds to Se

Each Se bonds to As



Layered structure

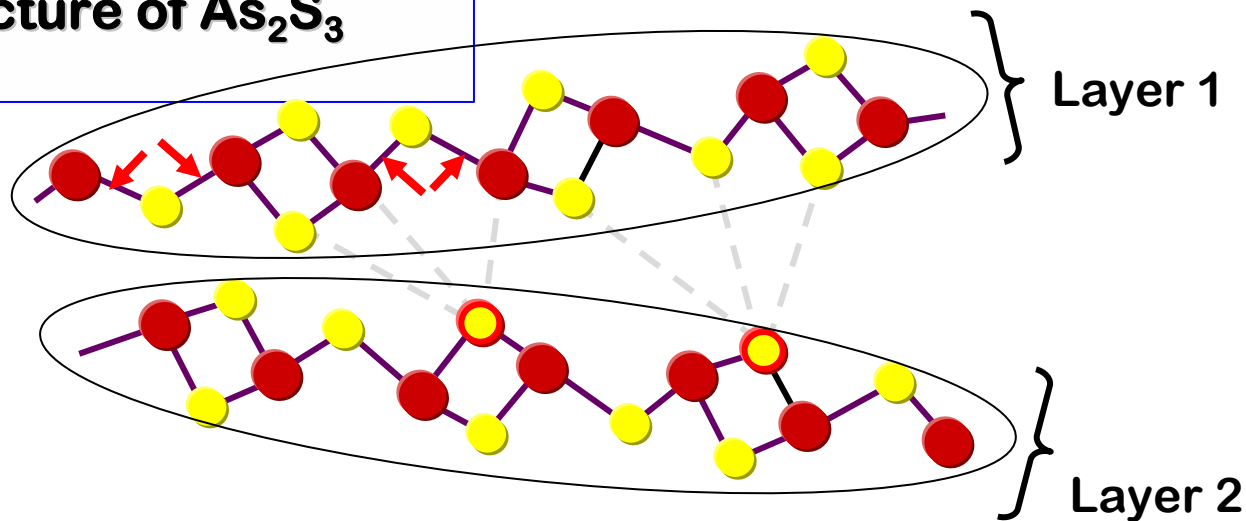
Covalent bonding:

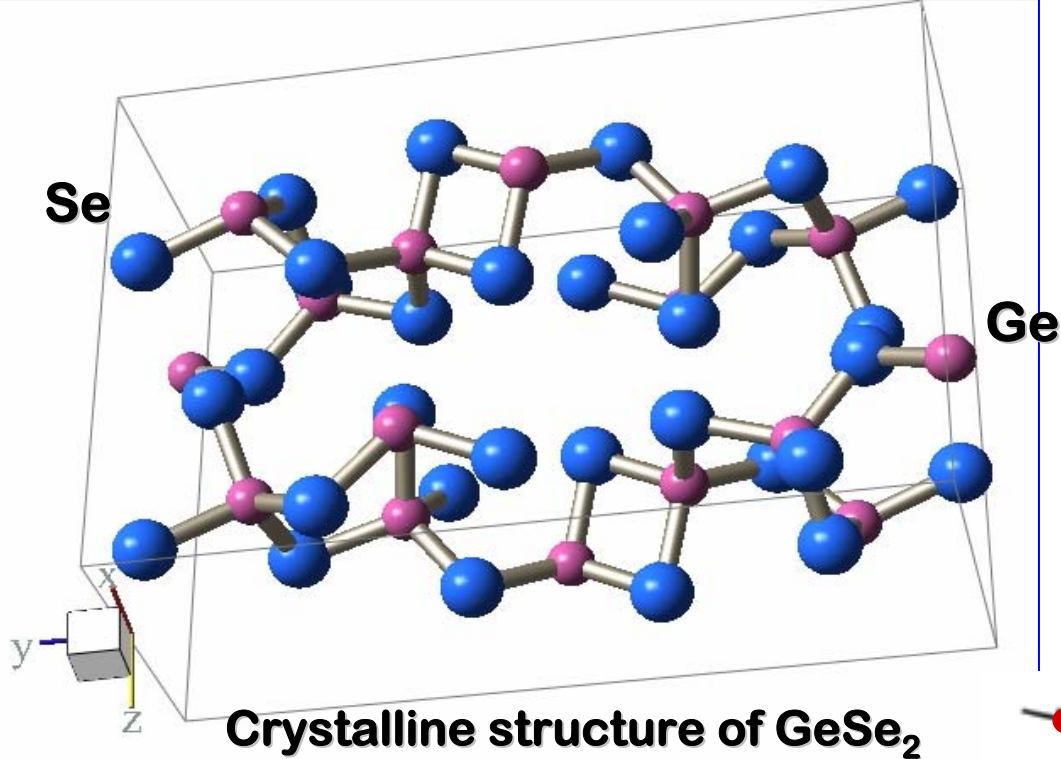
S: 2-fold

As: 3-fold

Each As bonds to S

Each S bonds to As





3D Four fold coordinated

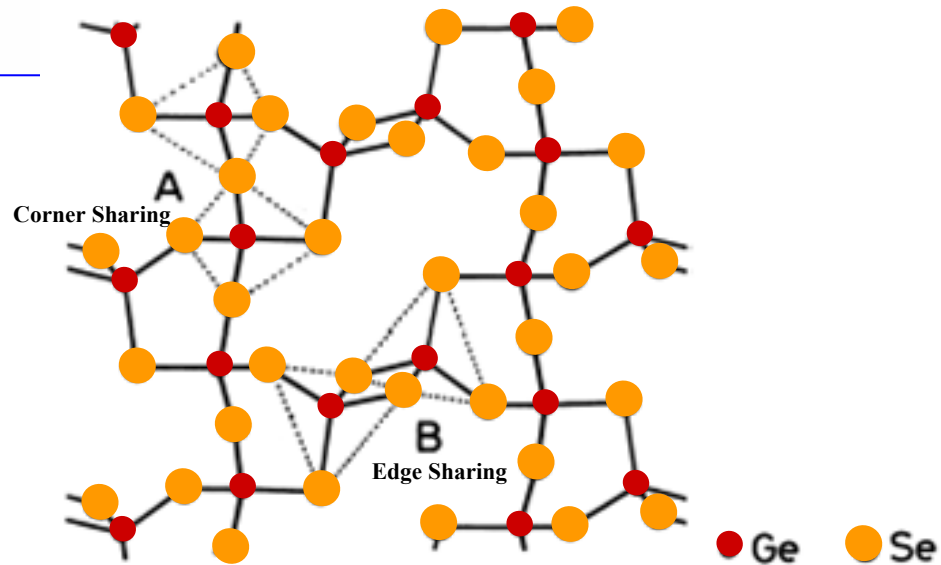
Covalent bonding:

Se: 2-fold

Ge: 4-fold

Each Ge bonds to Se

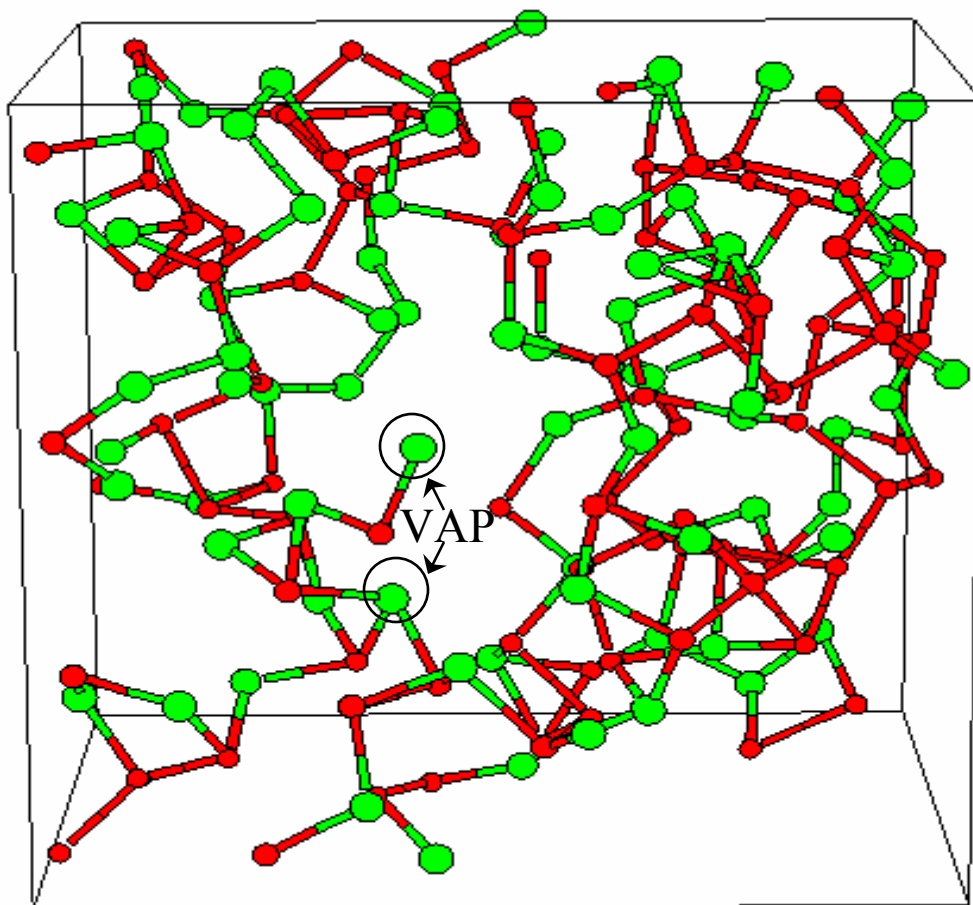
Each Se bonds to Ge



First Principles MD simulation of a-As₂Se₃ structure

 Se atom

 As atom



Chemical disorder:

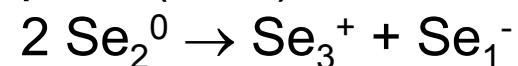
As-As and Se-Se

Coordination defects:

Se₃⁺, Se₁⁻, As₄⁺, As₂⁻

Valence alternation

pairs (VAP):

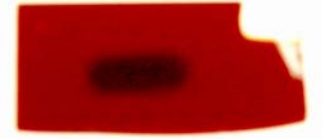


Li and Drabold (2001)

	Chemical disorder	Coordination defects
As	High	Low
Se	High	High

Wide composition range

As based films expand and photodarken



Ge- based films contract and photo bleach



Potential new applications:

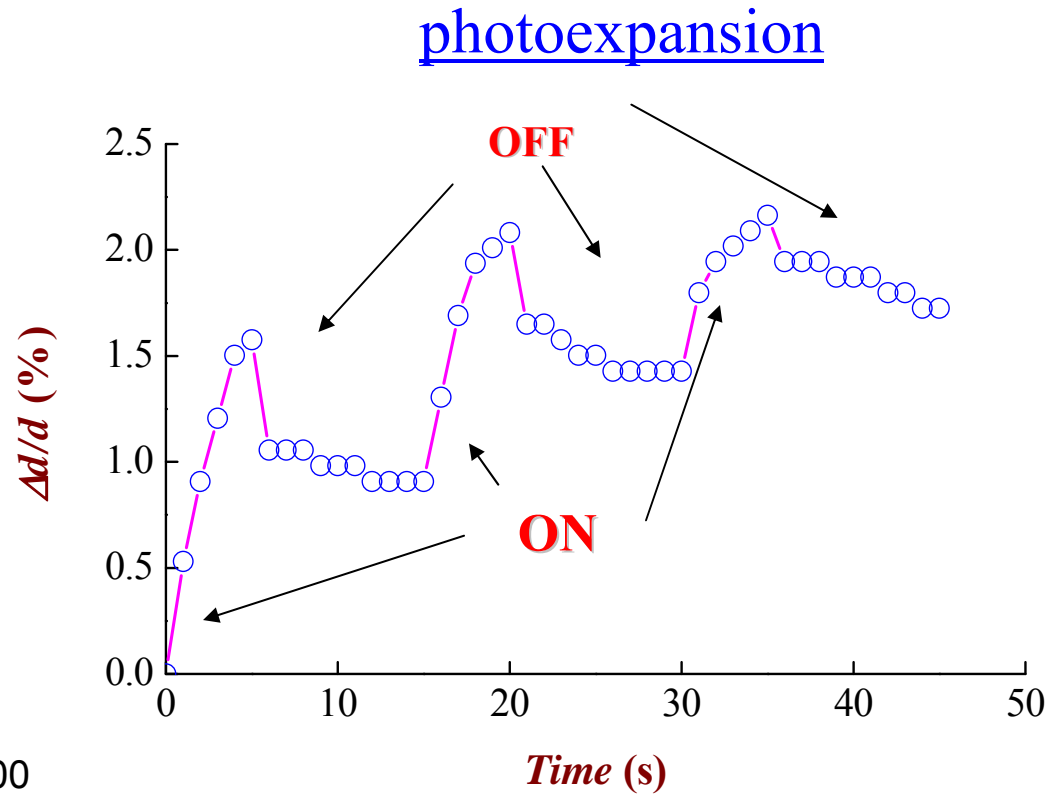
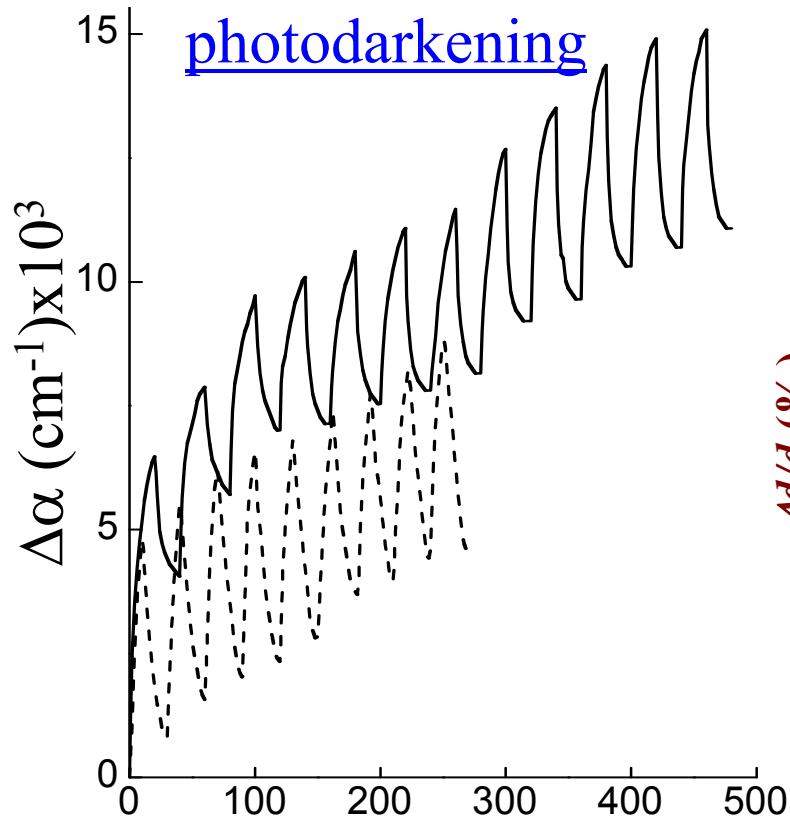
Creating micro and nano-sized optical components (lenses, gratings etc.)

Convex and concave structures can be developed by light on changing the composition

What happens at the atomic scale? Can we see similar features at an atomic scale??

* Kuzukawa, Ganjoo and Shimakawa; J. Non-Cryst. Solids (1998)

Temporary reversible effects



Change in absorptivity with time for $a\text{-As}_2\text{Se}_3$ films, $\Delta\alpha$, after illumination at 50 K (solid line) and 300 K (dashed line). Ar laser ON and OFF for 20 s each at 50 K; and 10/20 s at 300 K. (Ganjoo et al.)



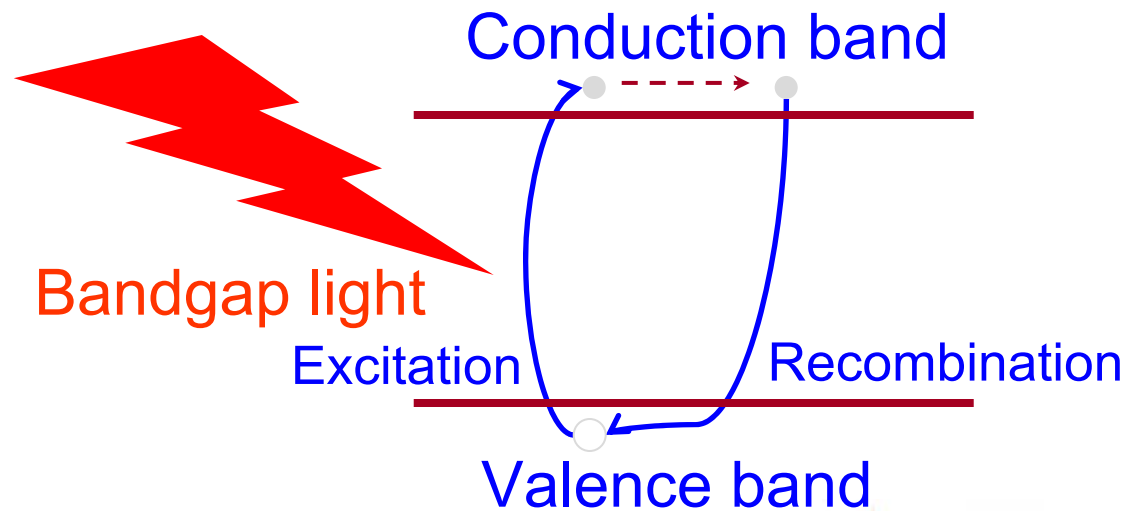
Photostructuring of ChG: Why ChG?

Based on group VI elements (S, Se, Te) as one of the major components. e.g. Se, sulfides or selenides of Ge, Sb or As, etc.

Materials that may show photosensitivity have:

- Low average coordination number
 - Low steric hindrance or large internal volume
 - Strong localization of light generated e-h pair: tight binding, lack of periodicity / disorder → Concentration of recombination energy in a small volume and change in valence of atoms before recombination.
- Also favor glass formation

Chalcogenide glasses are best suited for producing photosensitive phenomena by the near and above-bandgap-light illumination.



Consequence of photostructuring

What does it do?

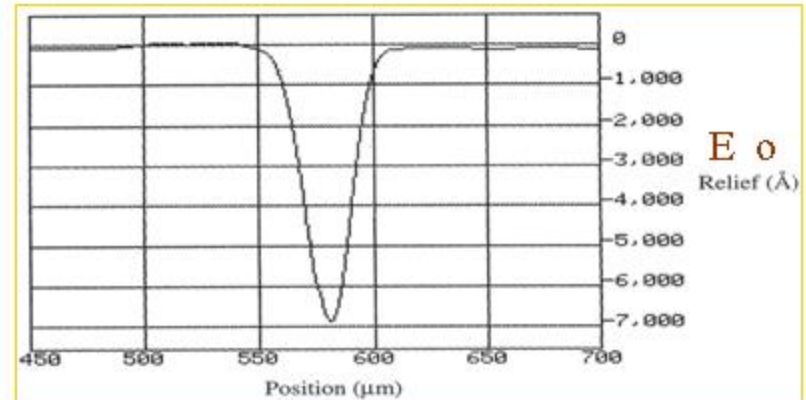
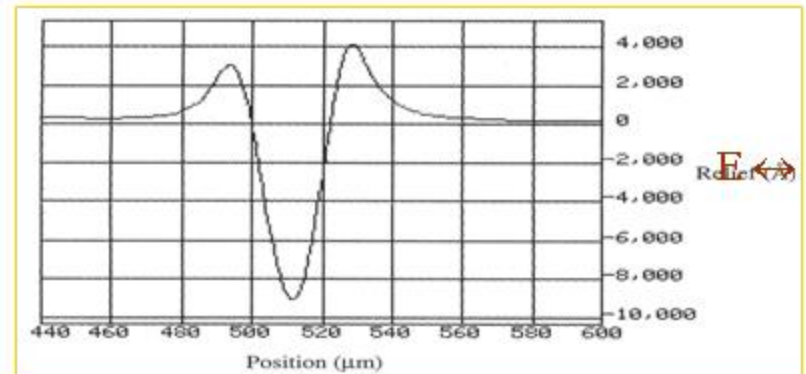
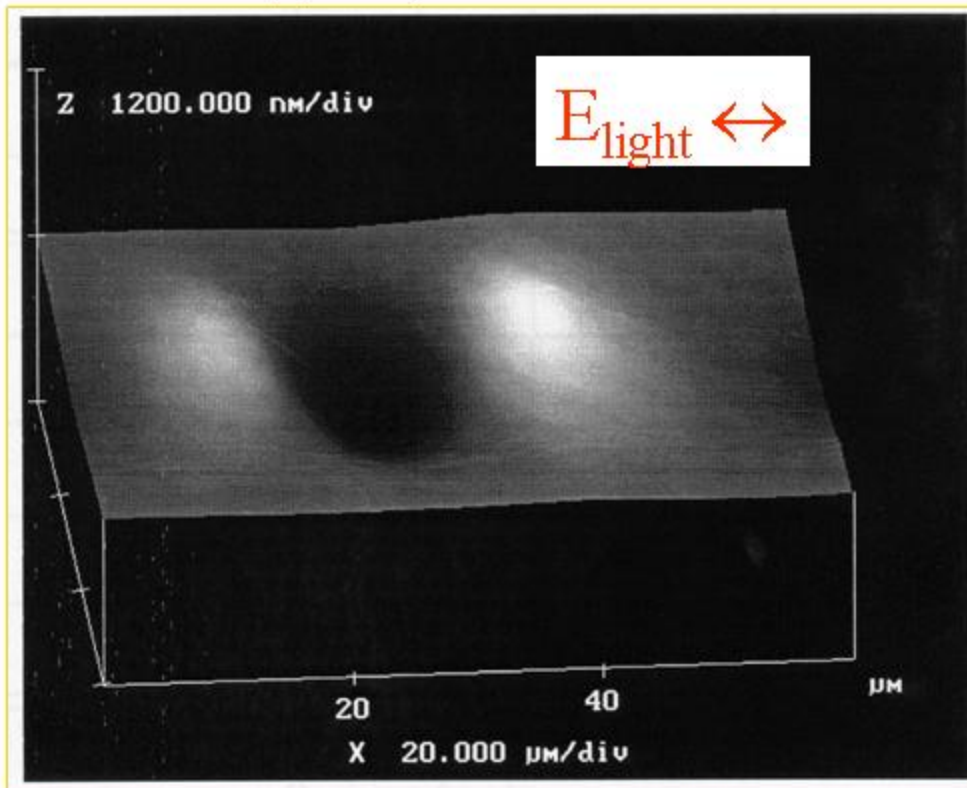
Miracles...

- **volume: integrated optics devices**
- **amorphization/ devitrification: CD-RW, DVD-RW**
- **mechanical properties - plasticity**
- **viscosity - athermal melting**
- **optical properties - darkening, birefringence**
- **electrical properties - conductivity, dielectric constant**
- **chemical properties - etching, dissolution**



Optical field-induced mass transport

Saliminia et al., (2000)



- A gaussian polarized Ar laser (514.5 nm) beam of circular x-section created an anisotropic crater on the surface of an a-As₂S₃ film.
- Circularly polarized light makes a dip with circular pile up.



Classification

Photoinduced Changes

Ability to recover

Permanent

Reversible

Metastable

Temporary

Effect of light polarization

Scalar

Vector

Permanent: can't be recovered w/o remaking the glass

Metastable: recovered on heating to $\sim T_g$

Temporary: recovered on removing the light

Scalar: don't depend upon the polarization of the light

Vector: depend upon the polarization of the light

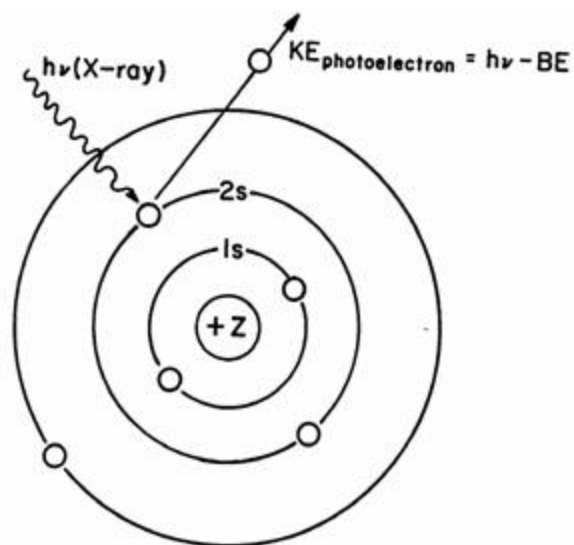
Temporary + Vector = Smart



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XPS with *in situ* laser irradiation



Monochromatic
x-ray beam

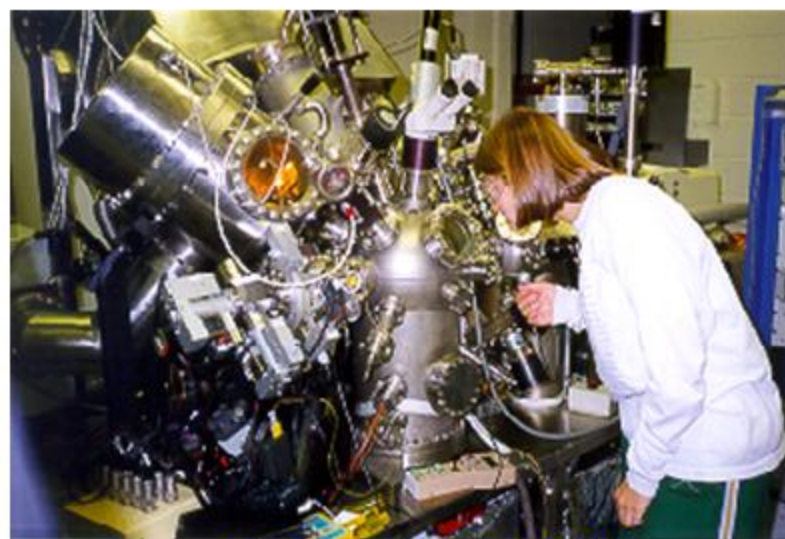
Electron
detector

He-Ne
Laser

Electron
flood gun

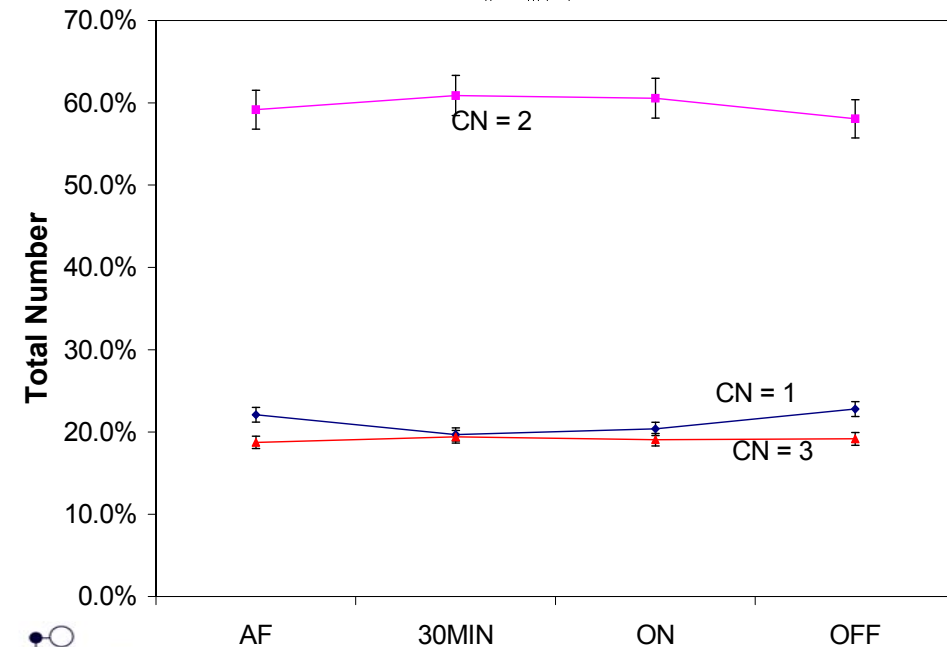
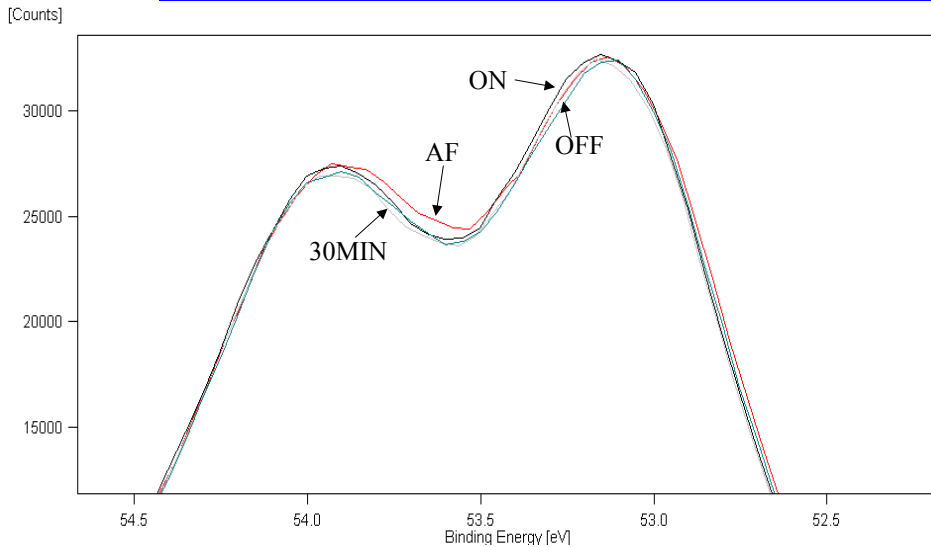
specimen

- Monochromatic x-rays => photoelectrons
- Electrons emitted with kinetic energies related to their binding energies
- Density of states
- Shift in peaks shows the change in the bonding character of the atoms



Scienta ESCA 300

Distribution of coordination configurations for Se in a-As₂Se₃

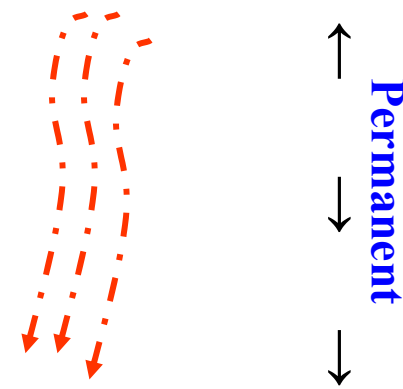


Laser irradiation

Se-[2(Se/As)]

Se-[3(Se/As)]

Se-(Se)



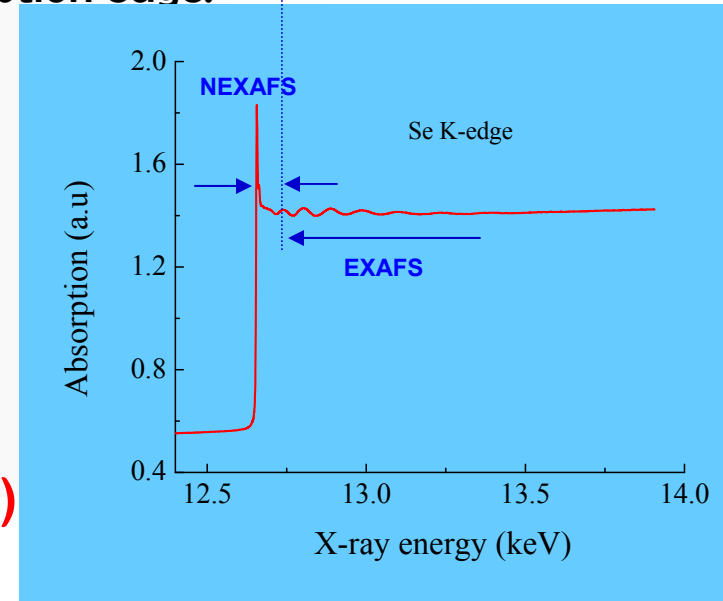
Laser reduces coordination defects around Se permanently

⇒ Optical annealing



X-ray absorption fine structure (XAFS)

X-rays of varying photon energies excite the electrons in a central atom (absorbed=> absorption edge)
Resulting photoelectrons have a low kinetic energy and are backscattered by the atoms surrounding the emitting atom. Probability of backscattering depends on the energy of the photoelectrons.
The net result is a series of oscillations on the high photon energy side of the absorption edge.



XAFS spectrum of Se K-edge

What can we obtain from EXAFS?

- ✓ Local structure around a specific element.
- ✓ Average inter atomic distance (R)
- ✓ Mean square relative displacement (MSRD)
- ✓ average coordination number (CN)



Experimental details

a-As₂S₃ films; a-GeSe₂ films

In-situ EXAFS at NSLS, BNL

For a-As₂S₃ films: beamline X19A; As (11.867 keV) and S (2.472 keV) K-edges
(Different spots, different scans)

For a-GeSe₂ films: beamline X18B; Ge (11.103 keV) and Se(12.658 keV) K-edges
(Same spot; one scan)

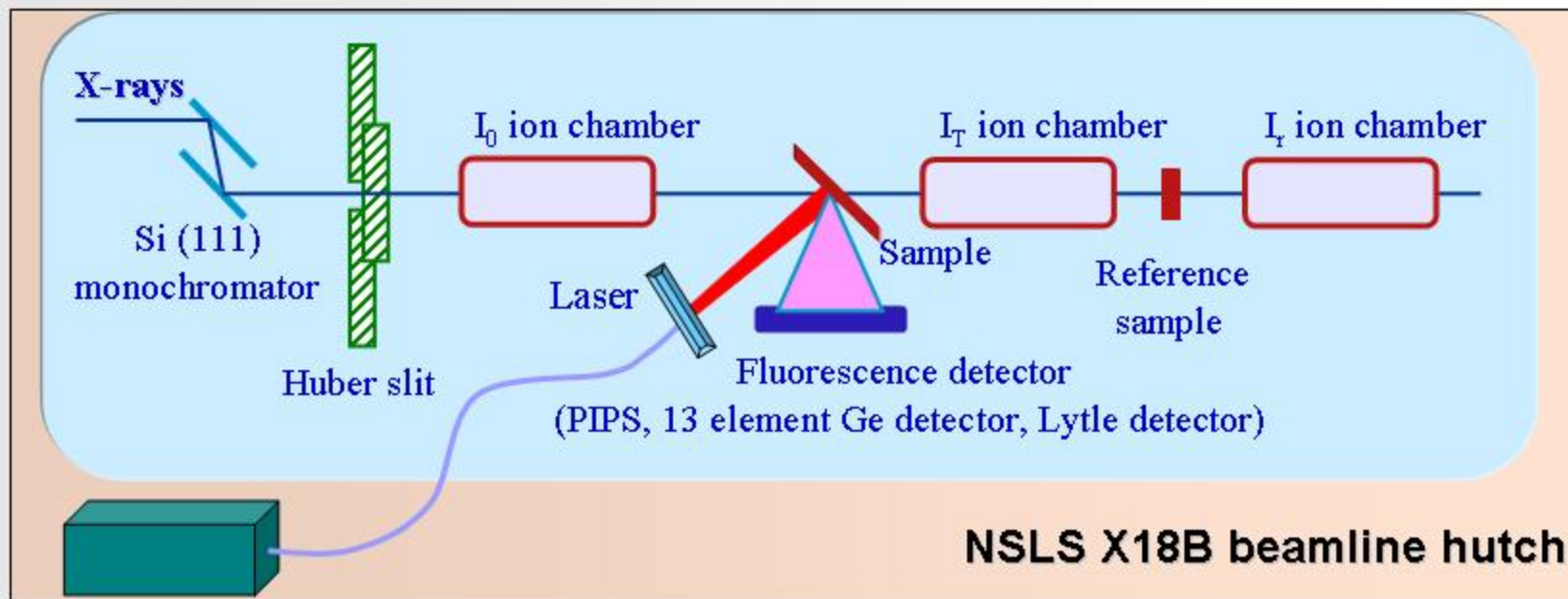
Data collected in fluorescence mode before (As prepared: AP), during (ON) and after laser illumination (OFF) states of the sample

Illumination sources

For a-As₂S₃ films: Ar⁺ laser (488 nm; 50 mW/cm²)

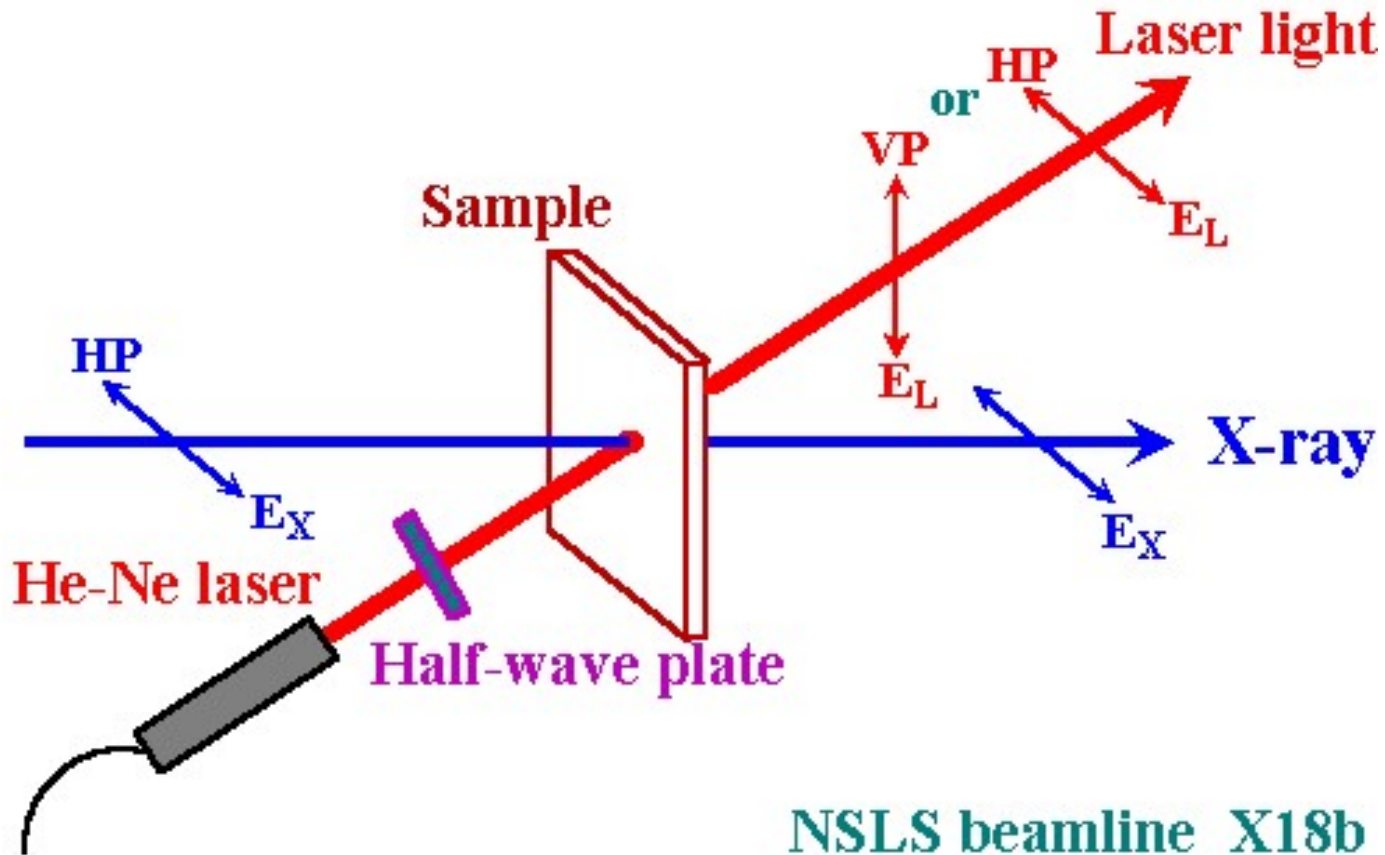
For a-GeSe₂ films: Semiconductor laser (633 nm; 50 mW/cm²)

In-situ experimental setup at X19A beamline



In situ EXAFS

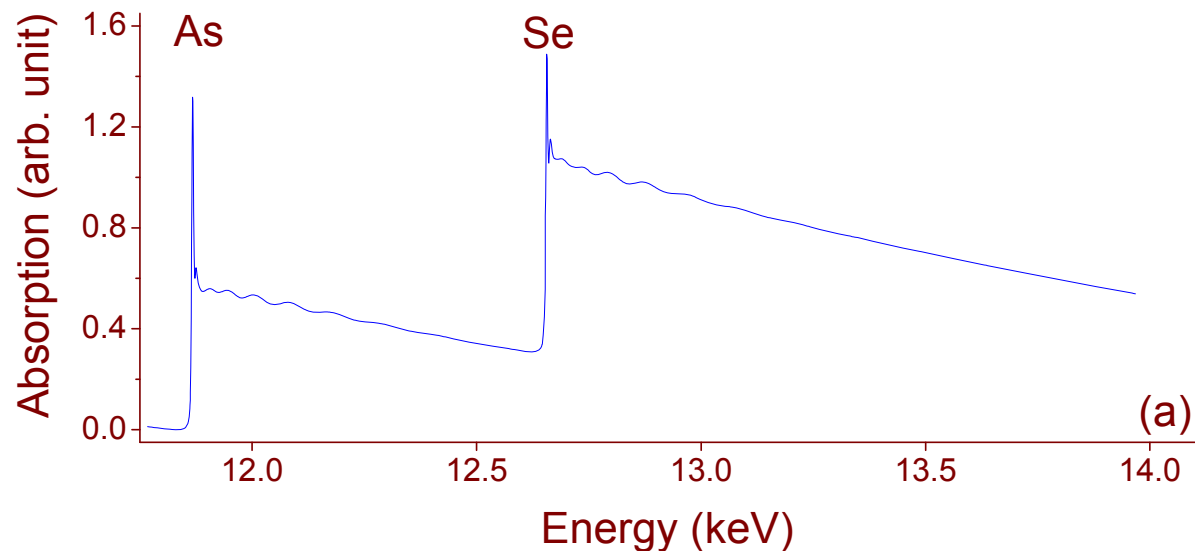
Synchrotron x-rays: linearly polarized



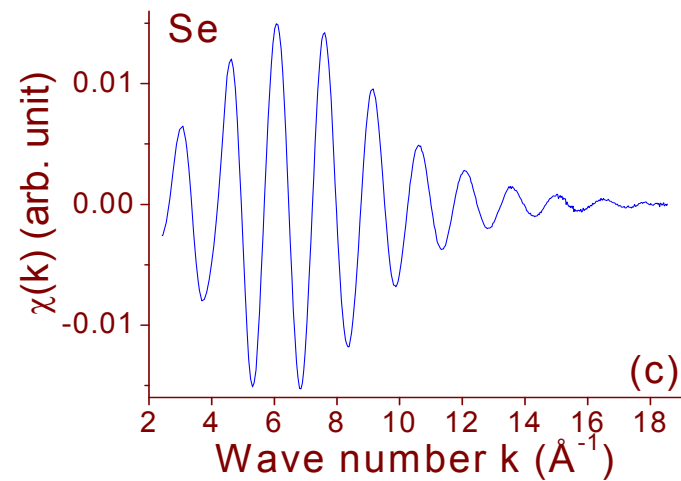
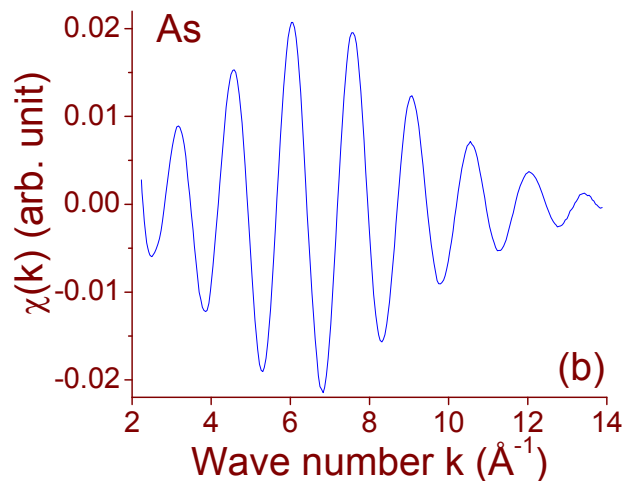
Looking for laser-induced polarization-dependent changes.

EXAFS Spectra

(a): X-ray absorption spectrum of an a-As₄₀Se₆₀ film beyond As and Se K-edges.

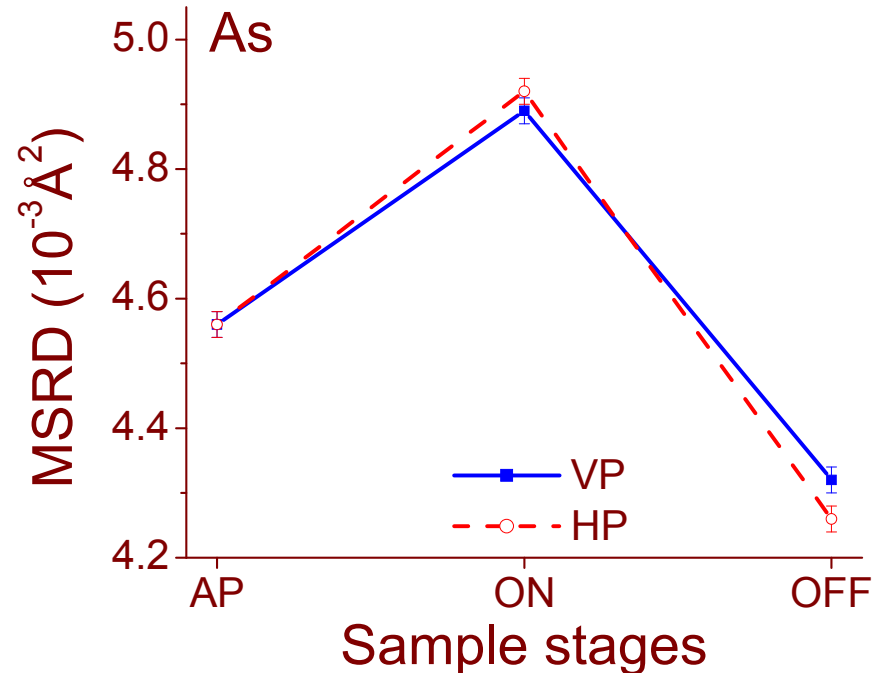
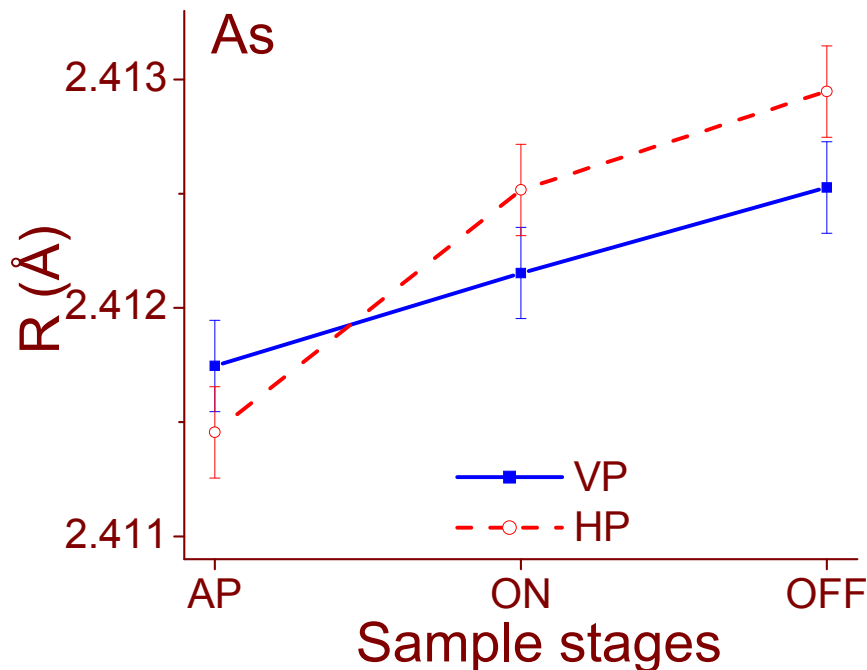


(b) and (c): The EXAFS oscillations derived from (a).



Structural changes around As atoms

Sample: as-prepared $\text{As}_{40}\text{Se}_{60}$ film



AP: as-prepared

ON: laser is on

OFF: laser is off

VP: laser has vertical polarization.

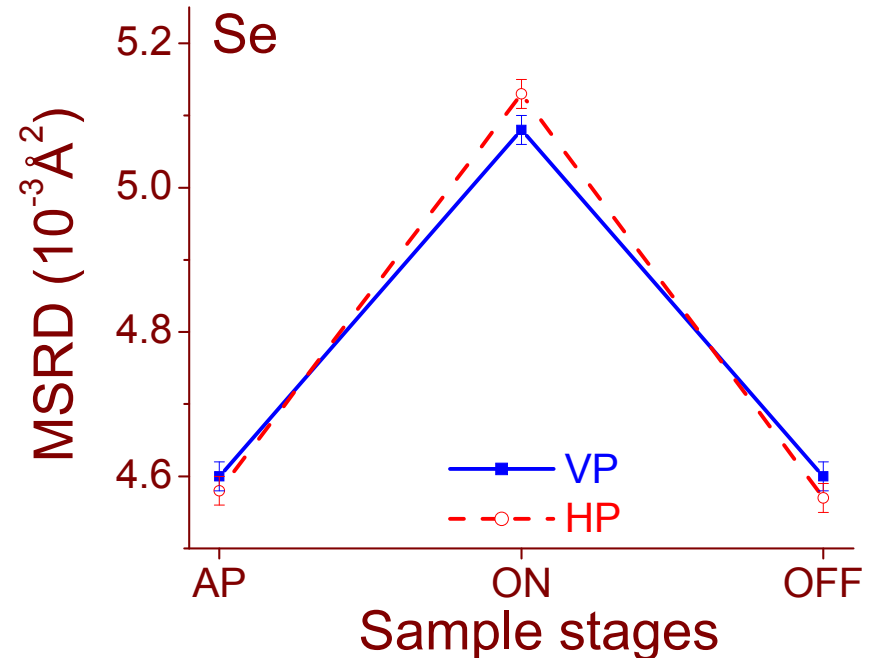
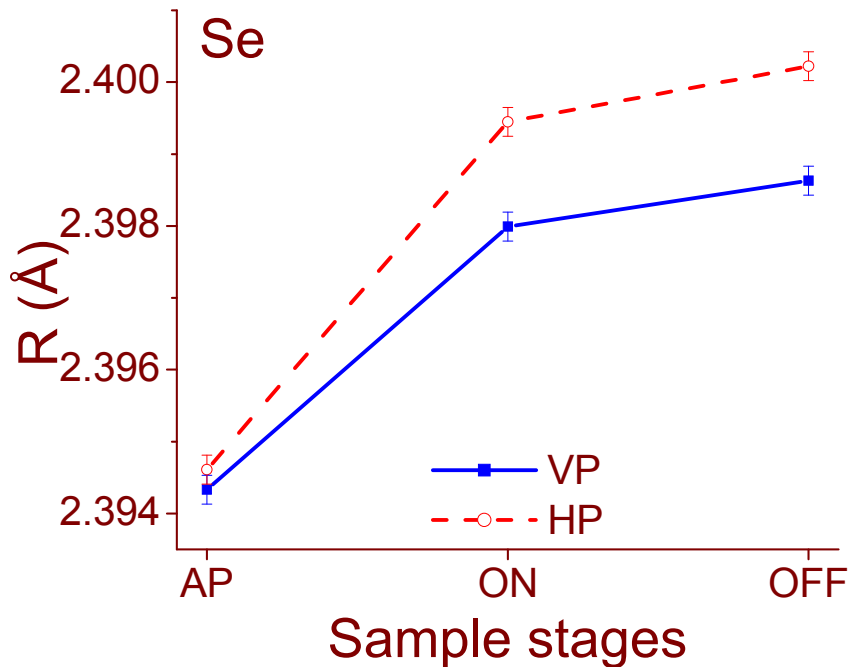
HP: laser has horizontal polarization

X-rays' polarization is horizontal.



Structural changes around Se atoms

Sample: as-prepared $\text{As}_{40}\text{Se}_{60}$ film



AP: as-prepared

ON: laser is on

OFF: laser is off

VP: laser has vertical polarization.

HP: laser has horizontal polarization

X-rays' polarization is horizontal.



Mechanisms of Scalar Changes

$R_{\text{As-NN}}$:

small permanent \uparrow expansion

$R_{\text{Se-NN}}$:

large permanent \uparrow expansion

1. Photo-chemical reaction

Microscopic heterogeneity in AP films

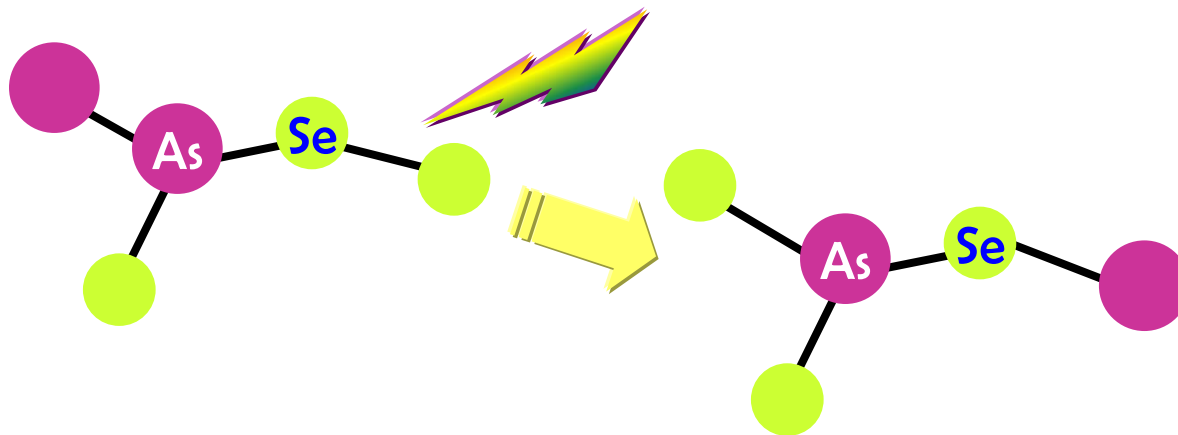


$R_{\text{As-As}} > R_{\text{As-Se}} > R_{\text{Se-Se}} \Rightarrow \uparrow R_{\text{Se-NN}}$ and $\downarrow R_{\text{As-NN}}$

However, experiments: \uparrow in both $R_{\text{Se-NN}}$ and $R_{\text{As-NN}}$

Covalent Radii:

As: 1.21 Å. Se: 1.17 Å



2. Strain relief

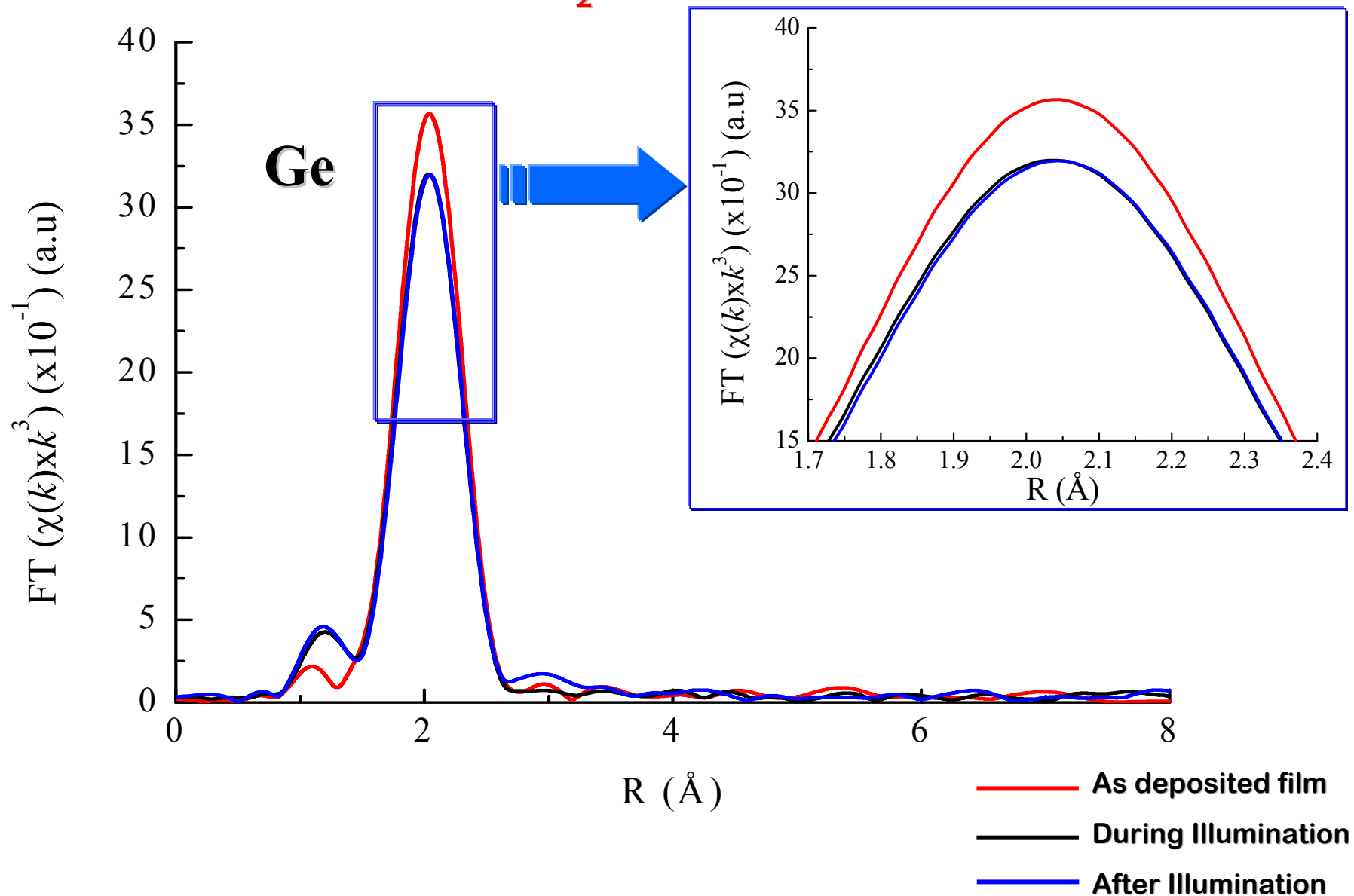
Intramolecular bonds in As-rich molecules are highly strained

\Rightarrow breaking of such molecules by light will $\uparrow R_{\text{As-Se}}$

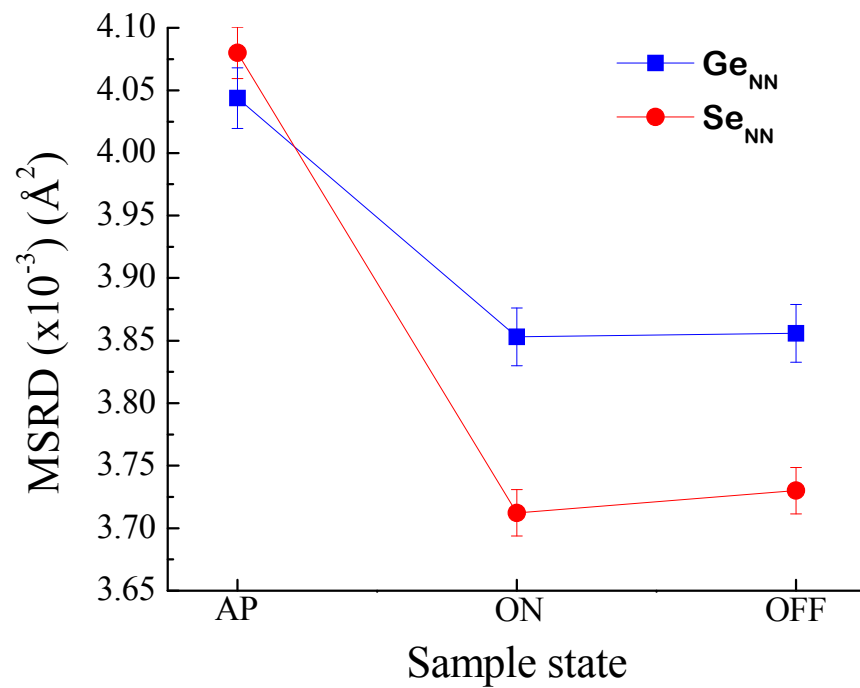
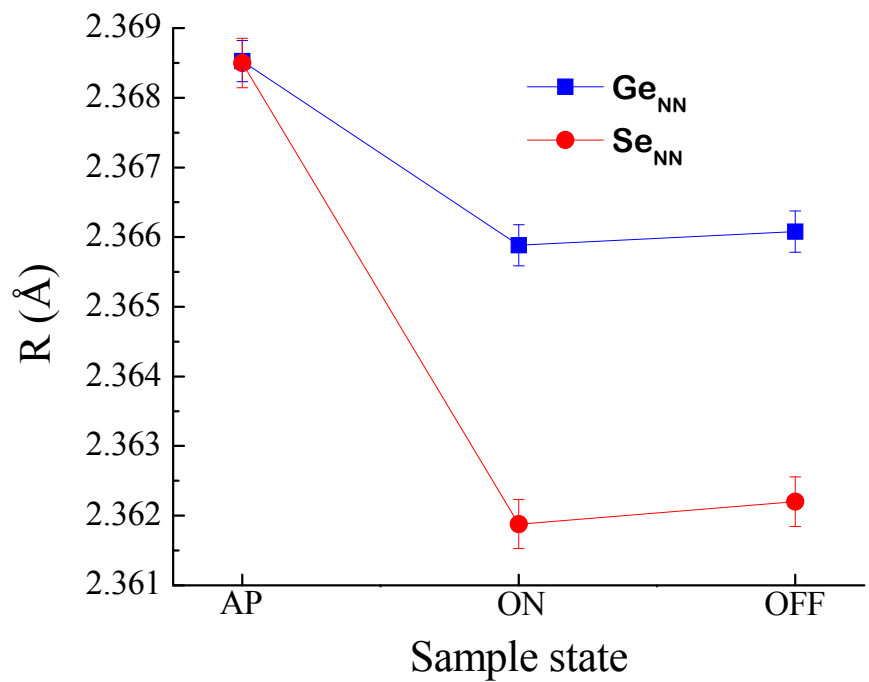
$\Rightarrow \uparrow R_{\text{Se-NN}}$ and $\uparrow R_{\text{As-NN}}$

1 + 2 \Rightarrow large \uparrow in $R_{\text{Se-NN}}$ & small \uparrow in $R_{\text{As-NN}}$

PRDF around Ge for α -GeSe₂ films



GeSe₂ EXAFS



a-GeSe₂ films:

Decrease in Ge_{NN} and Se_{NN} distances with illumination ⇒ **CONTRACTION IN VOLUME**

Mechanism of photoinduced changes

AP films: Chemical disorder: Ge-Se, Ge-Ge and Se-Se bonds

1. Photochemical reaction

Ge - Ge + Se - Se ⇒ 2 Ge - Se (similar to effect of annealing)
Ge-Se bonds energetically favored

Bond lengths from covalent radii:

Ge-Ge (2.44 Å) > Ge-Se (2.36 Å) > Se-Se (2.32 Å)

⇒ R_{Ge-NN} should decrease and R_{Se-NN} should increase;
but R_{Se-NN} is also decreasing

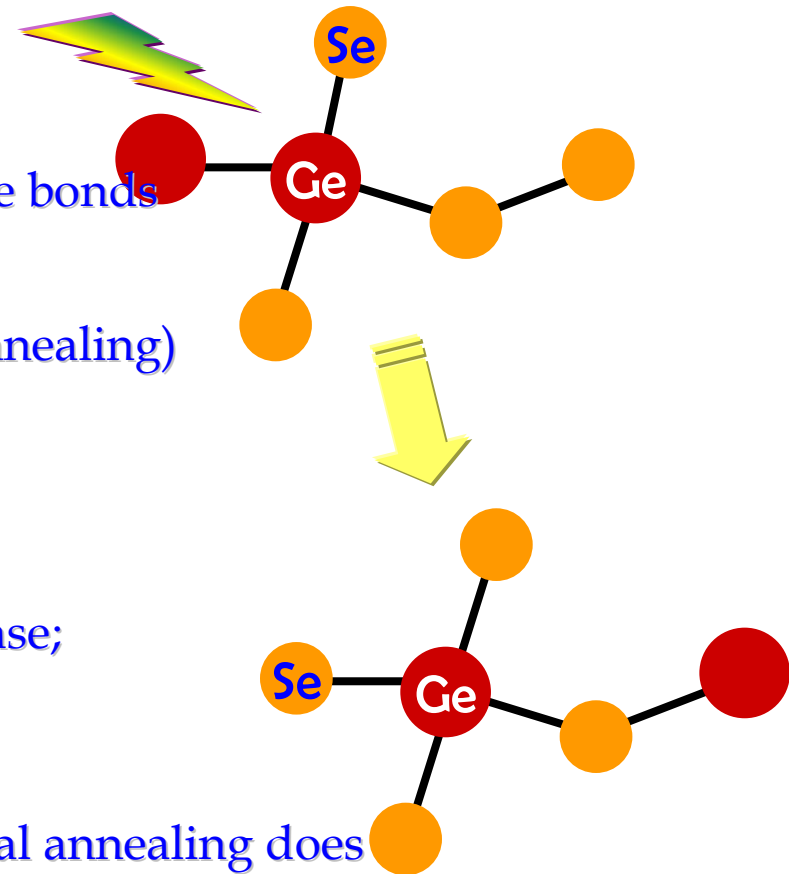
2. Strain relief

Light has similar effect on the NN distance as thermal annealing does

Light relieves highly strained atoms

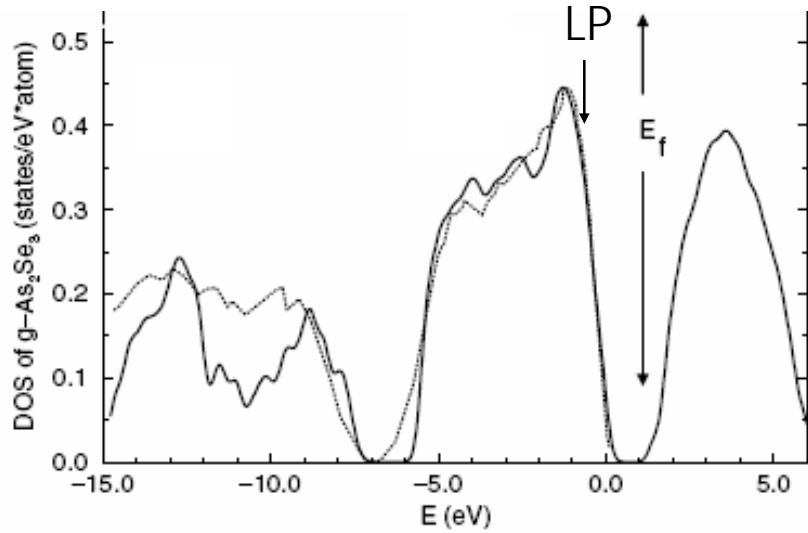
(mostly strained 2 fold Se atoms bonded both to Ge and Se)

Decrease in Se NN distances - **Experimentally observed by *in-situ* EXAFS**

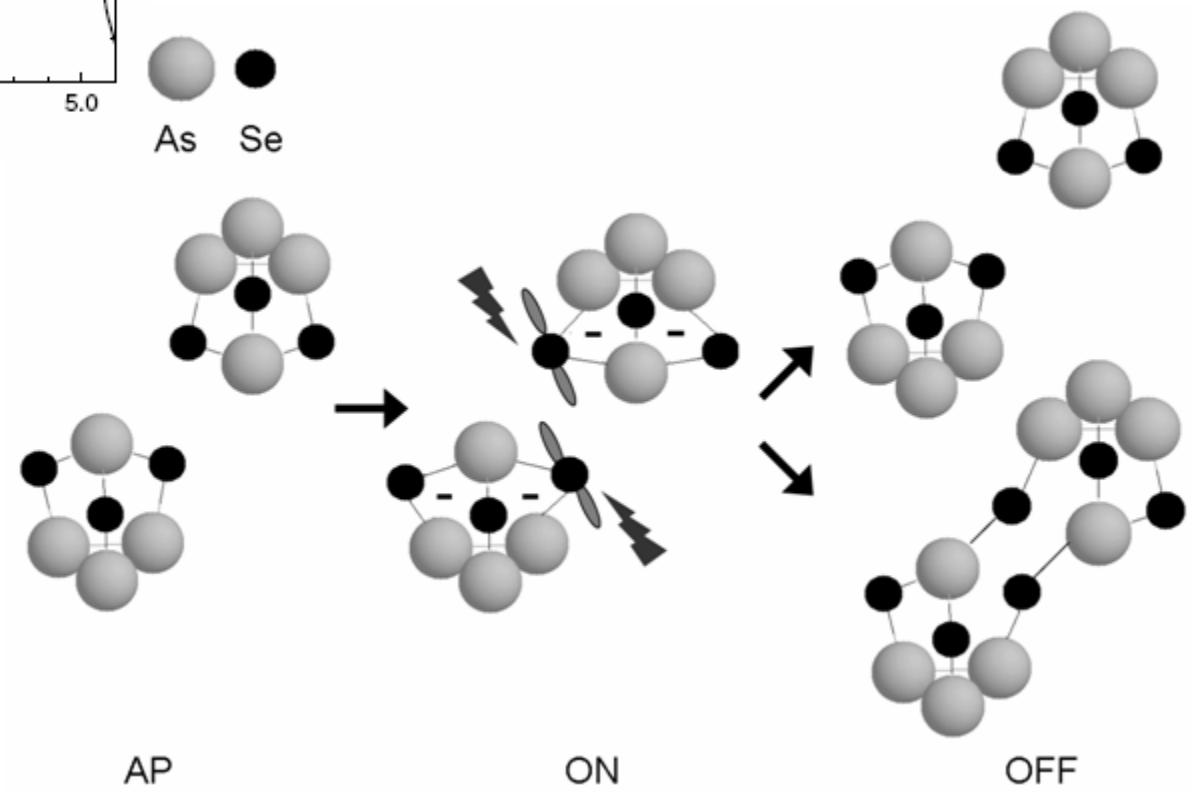


Mechanism of photo-structural change

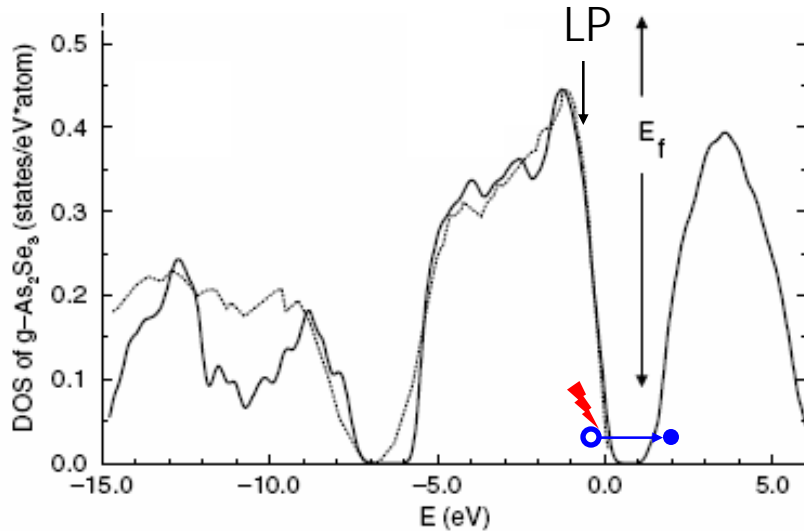
MD simulation shows Se 4p lone pair (LP) electrons occupy top of the valence band



Li, Drabold, et al. (2003)

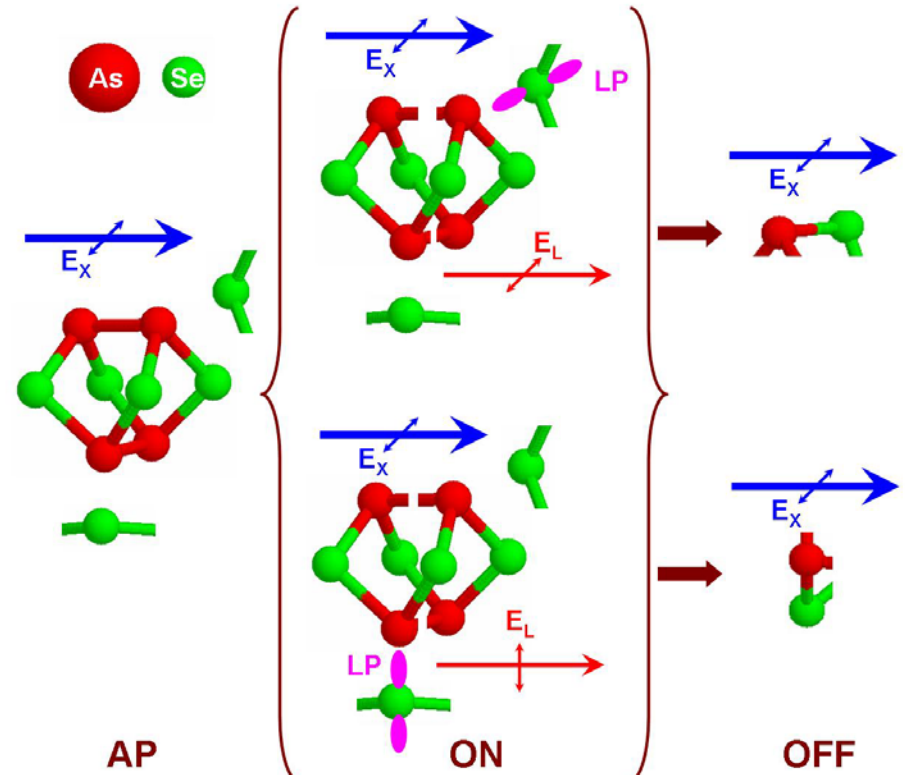


Mechanism of Vectoral Changes



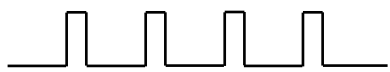
Li, Drabold, et al. (2003)

MD simulation shows Se 4p lone pair (LP) electrons occupy top of the valence band



- AP:** As-rich molecules (As_4Se_4) and Se-rich phase co-exist in AP $\text{a-As}_2\text{Se}_3$ film.
- ON:** As dangling bonds (from As-As bonds in As_4Se_4 molecules) react with preferentially excited Se 4p LP's (orbital $\parallel E_{\text{laser}}$), form anisotropic As-Se.
- OFF:** Anisotropic As-Se bonds can be detected by polarized X-rays.





Amorphous Semiconductors



Excited electronic carriers



Relaxation

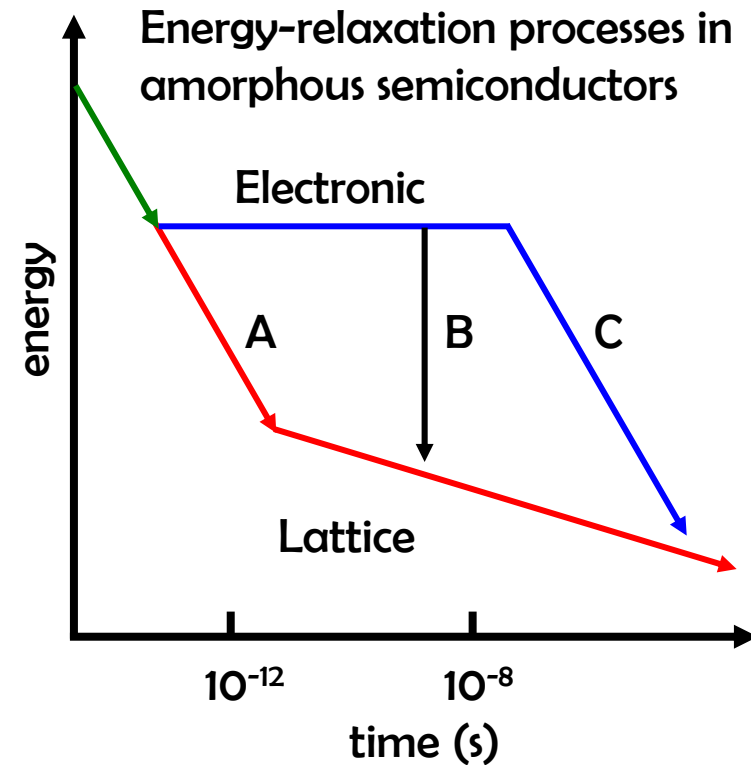
Within ps region, electrons relax to the bottom of conduction band (sometimes forming polarons)

**System → Equilibrium mainly through:
Electronic relaxations and lattice relaxations**

Carriers recombine radiatively or non-radiatively, and the electronic relaxation terminates

In capture process (trapping and detrapping) lattice distortion may be enhanced

Lattice relaxations may occur in time domain extending from ~ps to infinite times



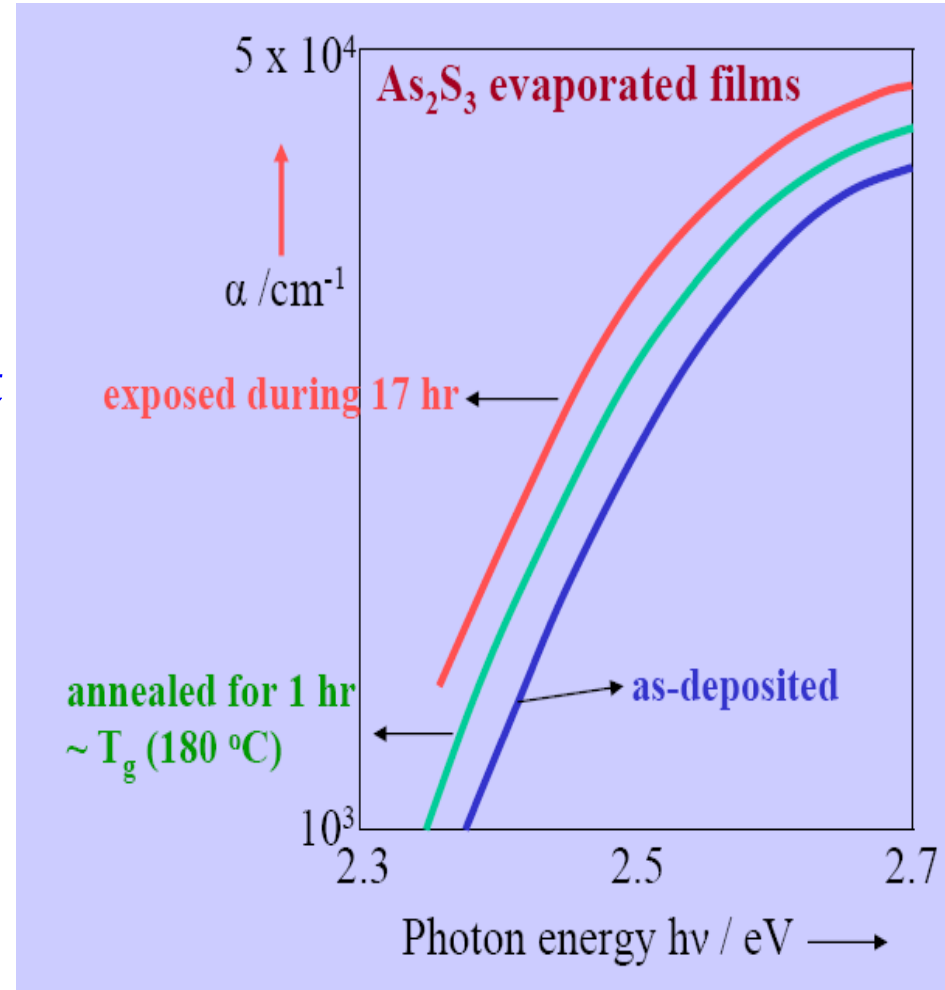
A: Non-radiative recombination
B: Radiative recombination
C: Capturing process

Photoinduced changes in absorption coefficient

Absorption edge believed to shift in parallel by annealing and illumination (Tanaka et. al., 1981)

Measurements after illumination (Metastable state only)

In-situ measurements at a single wavelength should represent changes at other wavelengths as well?

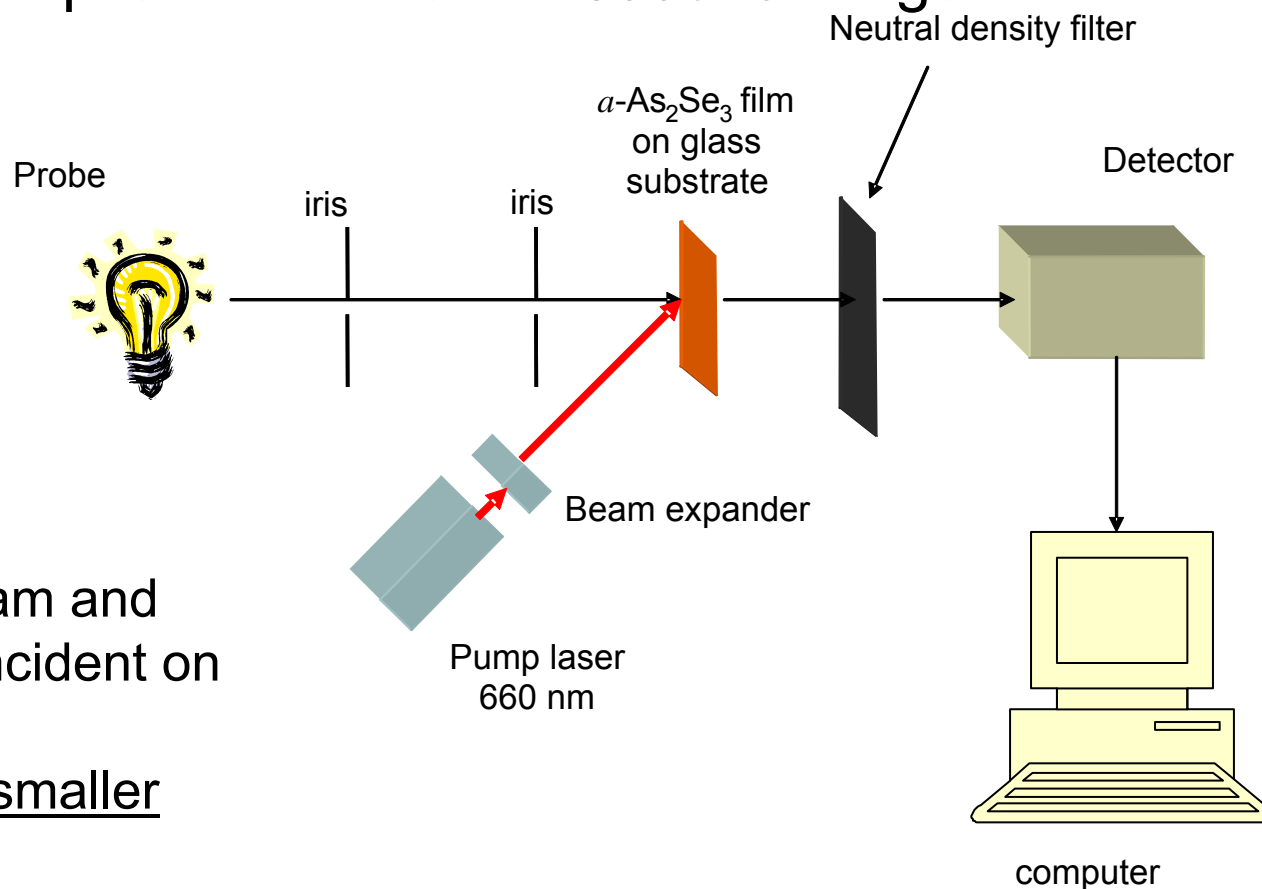


Tanaka et. al., JNCS (1981)

Speed of reversibility is crucial:

In situ vis-NIR spectroscopy

Use of an optical spectrometer (450 – 1000 nm) that allows real time data acquisition in the *millisecond* range.



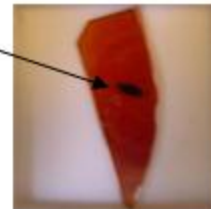
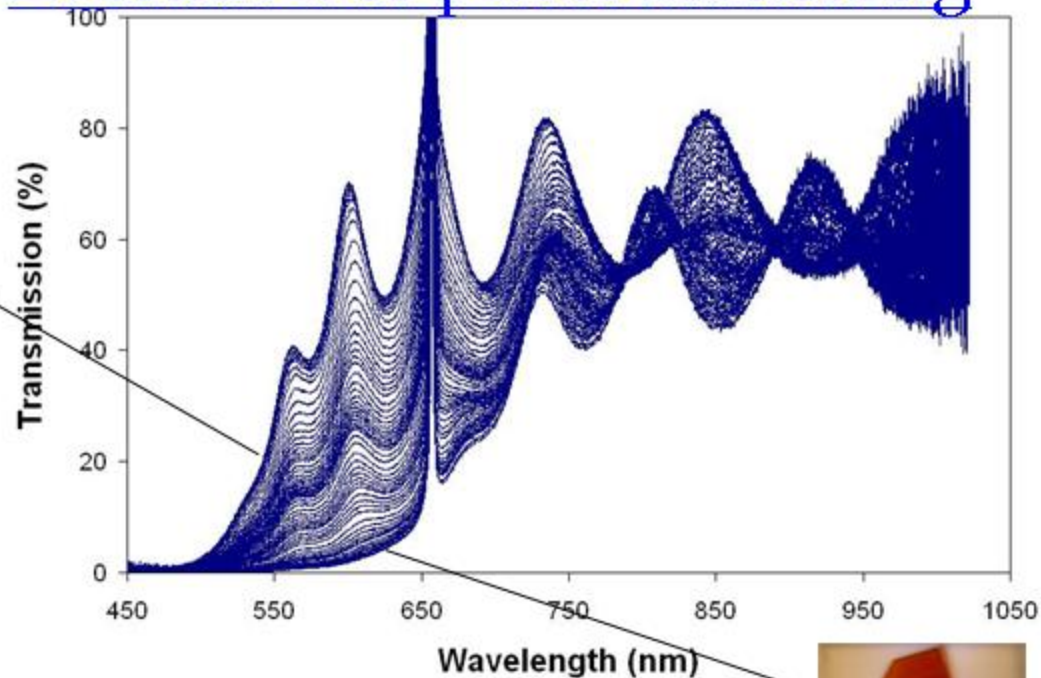
Light from probe beam and pump beam are coincident on the film.

Probe beam size is smaller than pump beam

Evolution of photodarkening



As prepared

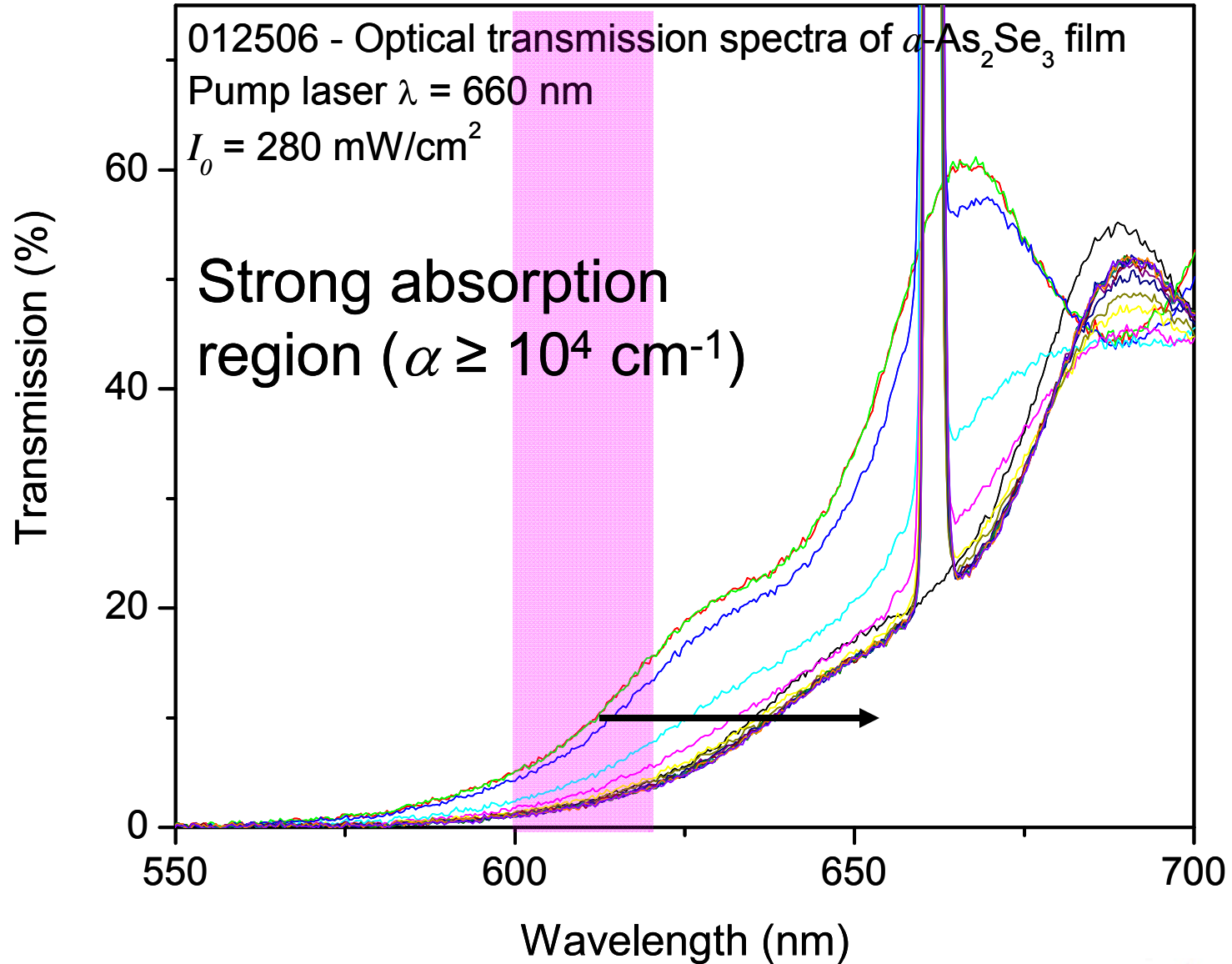


Fully photodarkened spot

The strength of fringes $\Delta I = (I_{\max} - I_{\min})$ i.e. the optical path first decreases and then increases in time \Rightarrow photodarkening and photoexpansion have different kinetics. Be careful when using the classical Swanepoel's method for data analysis of transmission spectra of thin films.

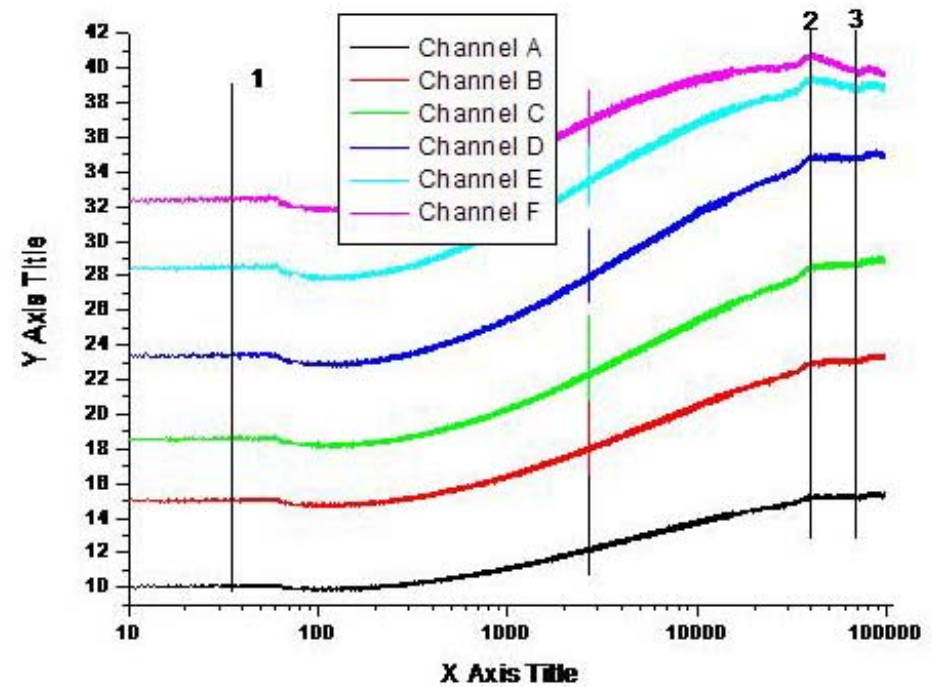
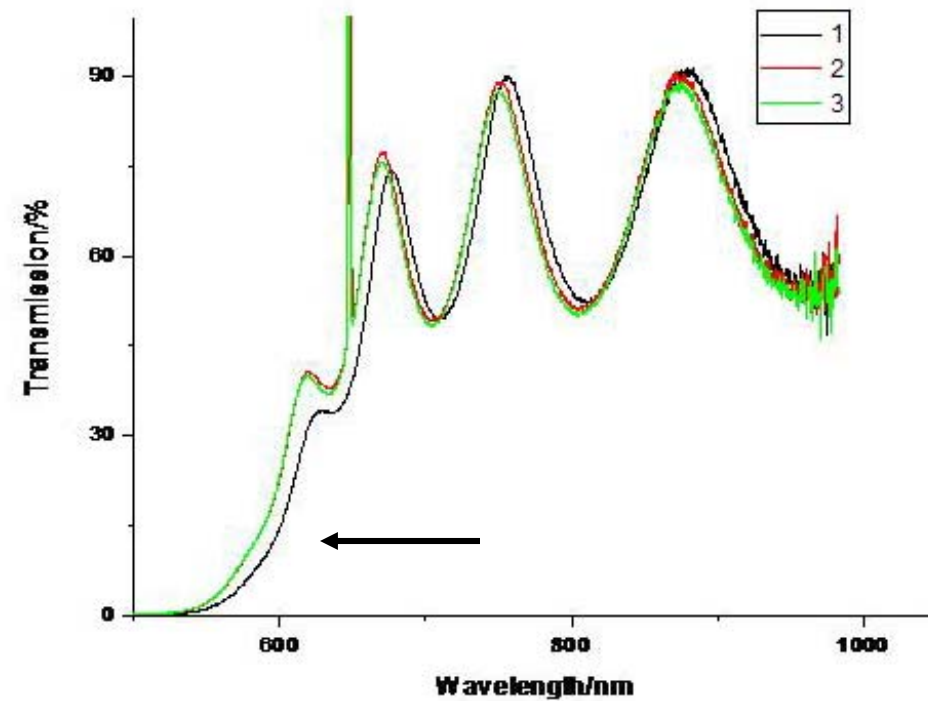
Tanaka: a- As_2S_3 indicate that the rate of photovolume expansion (a photostructural change) is greater than that of photodarkening for bandgap illumination.

PD kinetics in the strong absorption region ($\geq 10^4 \text{ cm}^{-1}$)

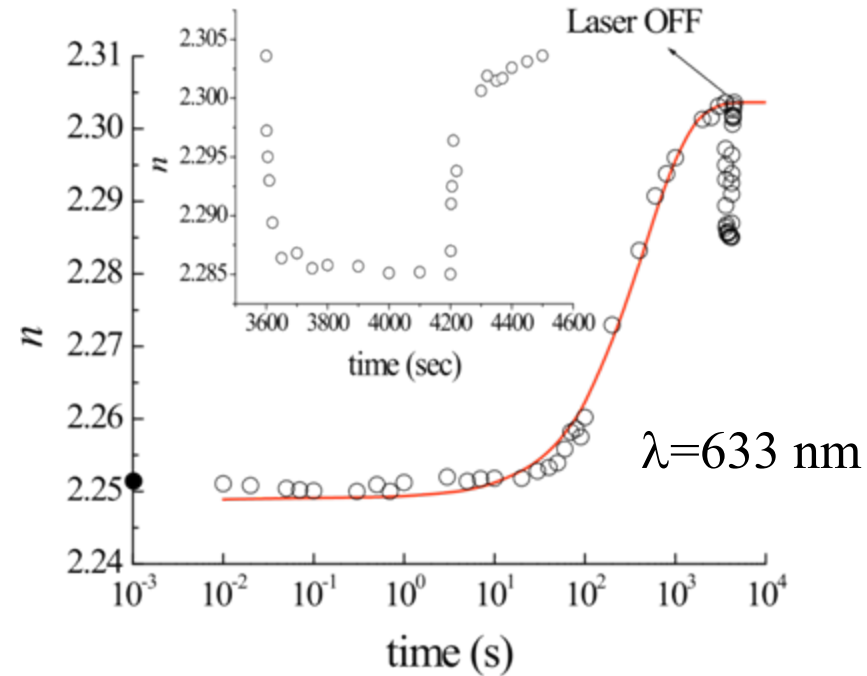
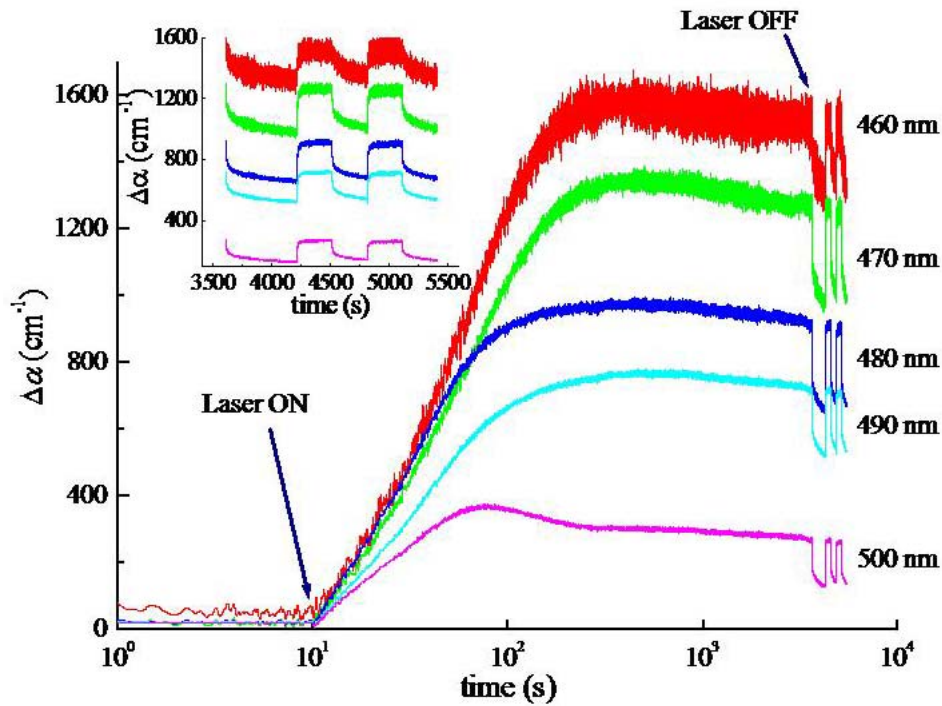


Photobleaching in Ge-Se glass

. Ge₂₂As₂₃Se₅₅, 660nm, 146mW/cm²



Evolution of photodarkening



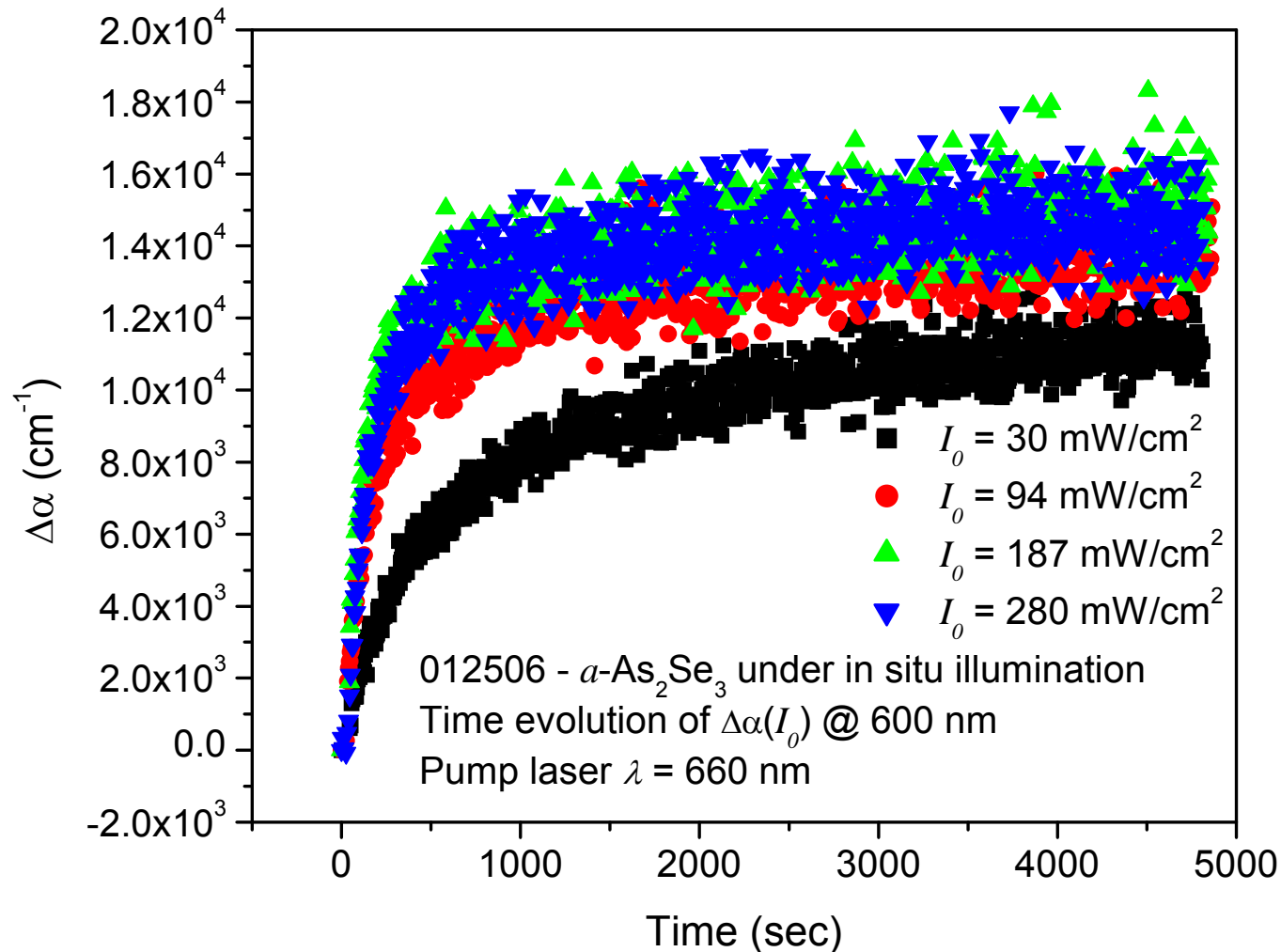
Initial photodarkening: $\underline{\text{As}}_2\underline{\text{S}}_3$: $\lambda_{\text{pump}}=488$ nm, $I_0=25$ mW/cm²

$$\Delta\alpha = [\alpha - \alpha(t=0)][1 - \exp(-t/\tau)]^\beta$$

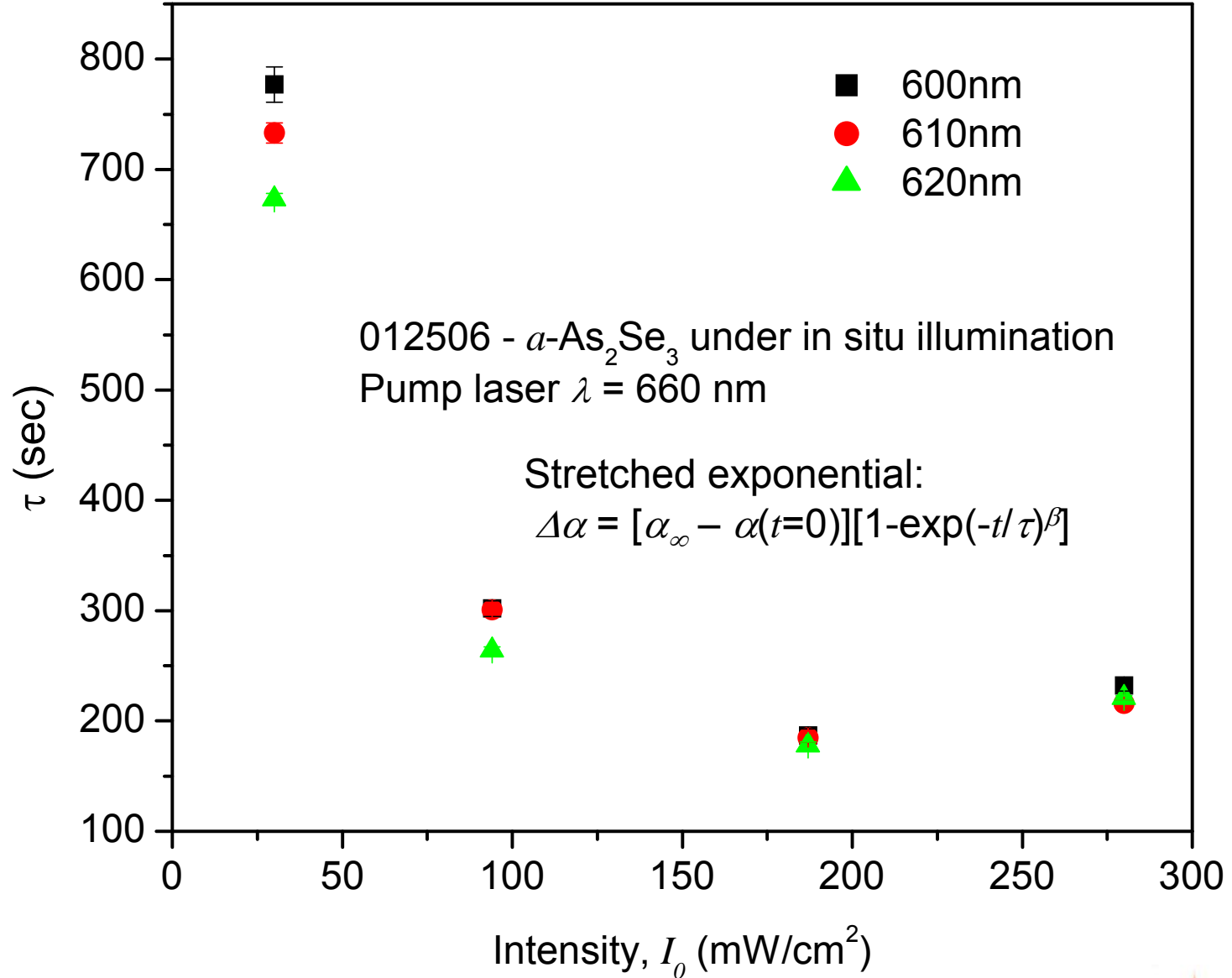
Ganjoo and Jain,
Phys. Rev. B **74**, 024201 (2006)

Photodarkening kinetics at various intensities

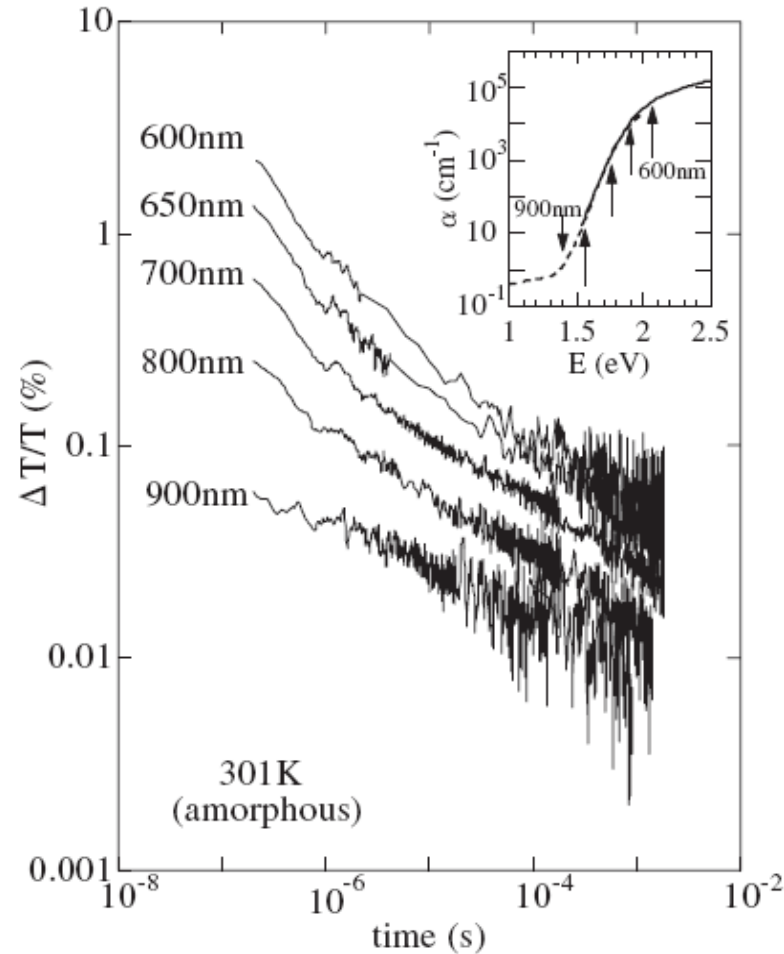
Plot of $\Delta\alpha(I_0)$ vs. t for $\lambda = 600$ nm. I_0 = laser intensity.



Plot of τ vs. I_0



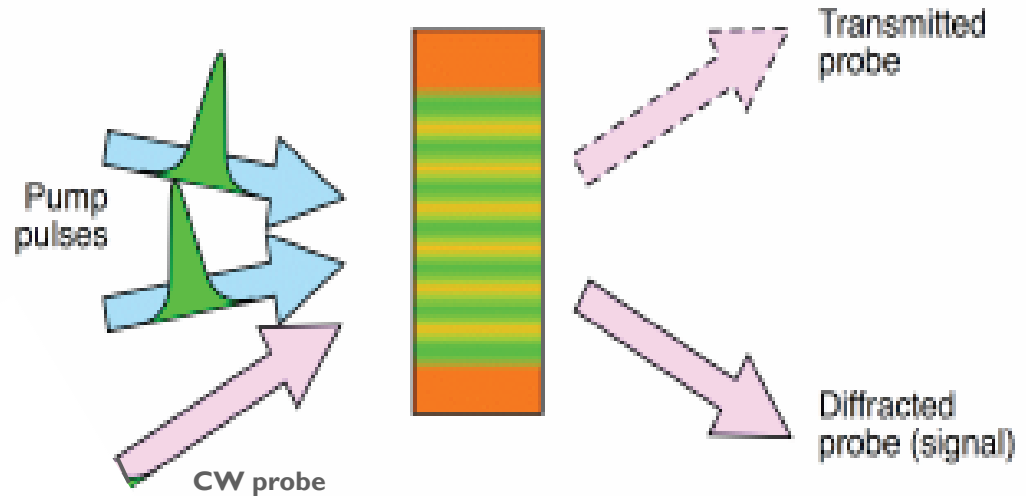
Fast optical changes



Decay of the transient part of photoinduced changes in transmission with time after pulsed laser illumination ($1.1 \text{ mJ}/\text{cm}^2$)

Sakaguchi and Tamura, Journal of Physics: Condensed Matter (2006)

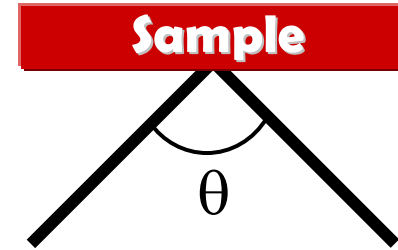
Fast photo-effects by transient grating method



- Two nearly equal intensity laser pulses made to cross within the sample at an angle
- Interference of two “writing pulses” within the sample writes a transient grating (by inducing a change in the refractive index)
 - ✓ The grating spacing varies with angle between the writing beams.
- The refractive index grating is read by diffracting a probe beam off the grating at the Bragg condition
- The diffracted probe light is collected by a high speed photomultiplier
- As grating disappears, the time dependence of the probe intensity reflects the decay of the change in refractive index and thus the carrier kinetics

Advantages of transient grating technique

$$\Lambda \propto \frac{1}{\theta}$$



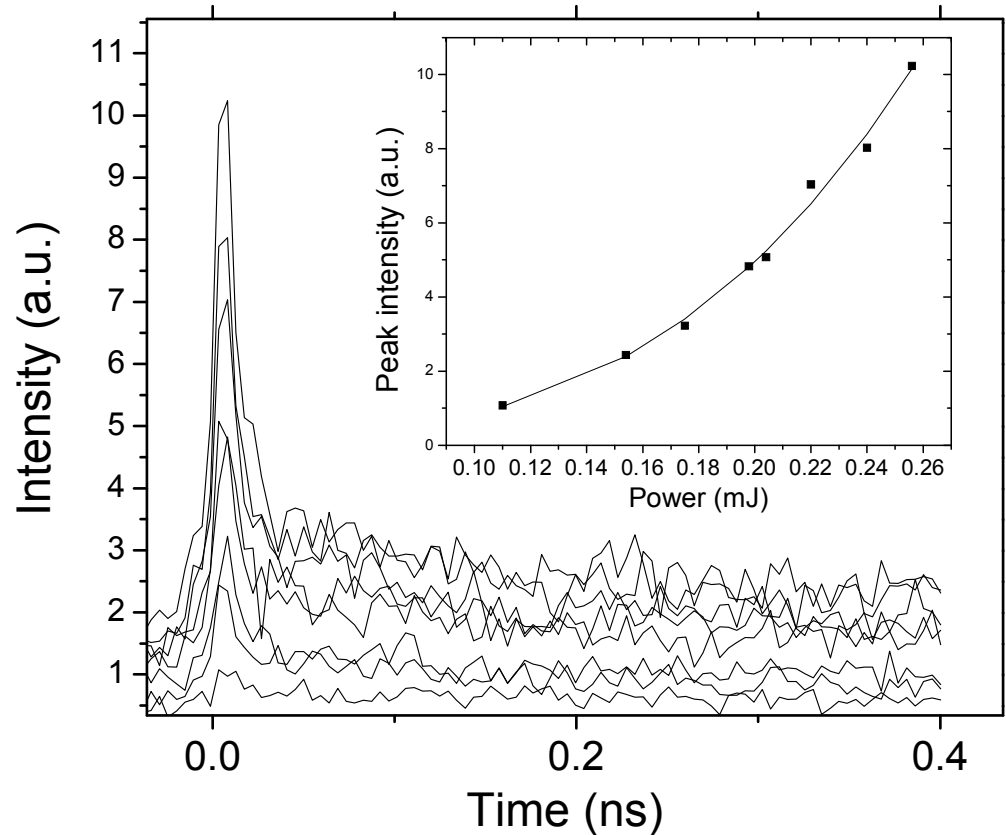
High signal/noise ratio

Can control the grating spacing (e.g. 0.675, 1.1 and 1.65 μm presently) by changing the angle between the two beams

Helpful in understanding the meaning of the time constants

Ultra fast photoinduced changes from 20 ps pulse

Transient Grating, Four-Wave Mixing method:
diffracted intensity



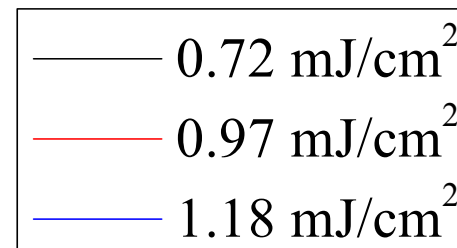
The photo-response to pulsed illumination is composed of a fast ~ 80 ps component followed by nanosecond component. The “ultra fast” component shows almost third-order power dependence indicating third-order nonlinear effect in As₅₀Se₅₀.



Grating spacing: $1.65 \mu\text{m}$



18.6°



$\Delta n \sim 1.21 \times 10^{-4}$

